

Phys 327 HW6

Trevor McCreary
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① Griffiths 5.9, modified

In example 5.1 & Problem 5.5b we ignored spin.

Do it now for particles of spin $1/2$. Construct the four lowest-energy configurations, and specify their energies and degeneracies. Suggestion: use

notation $\psi_{n_1 n_2} |s m\rangle$, where

$$\psi_{n_1 n_2}(x_1, x_2) = \psi_{n_1}(x_1) \psi_{n_2}(x_2)$$

With spin, two particles occupy the state:

$$\psi_{n_1 n_2}(x_1, x_2) \chi(1, 2)$$

or $\psi_{n_1 n_2} |s m\rangle$

The state ψ_{11} is symmetric, so it must be paired with the antisymmetric singlet configuration:

1st State: $\boxed{\psi_{11} | 0 0 \rangle}$

It has energy $(1^2 + 1^2) \hbar^2 = \boxed{\frac{\pi^2 \hbar^2}{ma^2}}$

The first excited state ψ_{12} is antisymmetric and so may be paired with any of the three symmetric triplet configurations:

Next 3 states: $\boxed{\begin{array}{l} \psi_{12} | 1 1 \rangle \\ \psi_{12} | 1 0 \rangle \\ \psi_{12} | 1 -1 \rangle \end{array}}$

Each state has energy $(1^2 + 2^2) \hbar^2 = \boxed{\frac{5\pi^2 \hbar^2}{2ma^2}}$

② Griffiths, 5.29

Suppose you have three particles in a potential such that the wave function is separable into products of one-particle states. Also suppose that there are only three distinct one-particle states $\psi_a(x_1)$, $\psi_b(x_2)$, and $\psi_c(x_3)$.

- How many different three-particle states can you construct if the particles are:

a) distinguishable, b) identical bosons, c) identical fermions?

The particles do not necessarily have to be in different states: if the particles are distinguishable, one possible state is $\psi_a(x_1)\psi_a(x_2)\psi_a(x_3)$

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a) For distinguishable particles, just count

the # of possible permutations of

$$\psi_a(x_1)\psi_b(x_2)\psi_c(x_3)$$

$$\rightarrow 3 \cdot 3 \cdot 3 = \boxed{27}$$

b) For identical bosons, the ordering of states doesn't matter.

$$\psi_a\psi_a\psi_a$$

$$\psi_c\psi_c\psi_c$$

$$\psi_b\psi_b\psi_b$$

$$\psi_c\psi_c\psi_b$$

$$\psi_c\psi_c\psi_a$$

$$\psi_a\psi_b\psi_c$$

$$\psi_a\psi_a\psi_b$$

$$\psi_a\psi_a\psi_c$$

$$\psi_b\psi_b\psi_a$$

$$\psi_b\psi_b\psi_c$$

$$\Rightarrow \boxed{10}$$

c) No two identical fermions can occupy the same state so the only possible state is $\uparrow\downarrow\uparrow\downarrow\uparrow\downarrow \rightarrow \boxed{1}$

③ Griffiths 5.21

The density of copper is 8.96 g/cm^3 and its atomic weight is 63.5 g/mol .

a) Calculate the Fermi energy for copper. Assume $d=1$ and give answer in eV.

$$E_F = \frac{\hbar^2}{2m} (3\rho\pi^2)^{2/3}$$

$$\rho \equiv \frac{Nd}{V} = \frac{M_{\text{total}}}{M V} = \frac{\rho_m}{\mu} = \frac{8.96 \text{ g/cm}^3}{63.5 \text{ g/mol}} = 0.141 \text{ mol/cm}^3$$

$$= 1.41 \times 10^5 \text{ mol/m}^3$$

$$= 8.5 \times 10^{28} \text{ atoms/m}^3$$

$$\rightarrow E_F = \frac{(1.05 \times 10^{-34} \text{ Js})^2}{2(9.11 \times 10^{-31} \text{ kg})} \left[3(8.5 \times 10^{28} \text{ atoms/m}^3) \pi^2 \right]^{2/3}$$

$$E_F = 1.1 \times 10^{-18} \text{ J} \times \frac{1 \text{ eV}}{1.6 \times 10^{-19} \text{ J}} = \boxed{7 \text{ eV}}$$

b) What is the correspond electron velocity?

$$E_F = \frac{1}{2} m v^2 \rightarrow v = \sqrt{2E_F/m} = \sqrt{2(1.1 \times 10^{-18} \text{ J}) / (9.11 \times 10^{-31} \text{ kg})}$$

$$\rightarrow \boxed{v \approx 1561 \text{ km/s}}$$

$$\frac{v}{c} = \frac{1561}{30000} = 0.005 \rightarrow \text{electrons non-relativistic} \checkmark$$

c) At what temperature does the Fermi energy equal its characteristic thermal energy?

$$E_F = k_B T$$

$$\rightarrow T = \frac{1.1 \times 10^{-18} \text{ J}}{1.38 \times 10^{-23} \text{ J/K}} = \boxed{79710 \text{ K}}$$

d) Calculate the degeneracy pressure of copper in the free electron gas model.

$$P = \frac{(3\pi^2)^{2/3} \hbar^2}{5m} n^{5/3}$$

$$= \frac{(3\pi^2)^{2/3} (1.05 \times 10^{-34} \text{ Js})^2}{5(9.11 \times 10^{-31} \text{ kg})} (8.5 \times 10^{28} \text{ atoms/m}^3)^{5/3}$$

$$\rightarrow \boxed{P = 3.8 \times 10^{10} \text{ N/m}^2}$$

④ Griffiths, Problem 5.30

Calculate the Fermi Energy for non-interacting electrons in a two-dimensional infinite square well. Let σ be the number of free electrons per unit area.

In two dimensions, Eq. 5.50 reduces to:

$$E_{\text{any}} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_x^2}{l_x^2} + \frac{n_y^2}{l_y^2} \right) = \frac{\hbar^2 k^2}{2m}, \quad \vec{k} = (k_x, k_y)$$

For a two-dimensional k_x - k_y space, with lines at $k_x = \pi/l_x, 2\pi/l_x$ and $k_y = \pi/l_y, 2\pi/l_y$, each intersection point represents a distinct one-particle stationary state. Each state occupies an area

$$\frac{\pi^2}{l_x l_y} = \frac{\pi^2}{A}$$

If we have N atoms, each has d electrons; no two electrons can occupy the same state. They will fill one quadrant of a circle, in k_x - k_y space, with radius k_F given by:

$$\frac{1}{A} (\pi k_F^2) = \frac{2Nd}{2} \left(\frac{\pi^2}{A} \right) \rightarrow k_F = \left(\frac{2Nd\pi}{A} \right)^{1/2}$$

$$\text{or } k_F = \sqrt{2\pi\sigma}$$

$$\Rightarrow E_F = \frac{\hbar^2 k_F^2}{2m} = \boxed{\frac{\hbar^2 \pi \sigma}{m}}$$