

1. From Text book • $\beta \equiv \frac{1}{V_m} \left(\frac{\partial V_m}{\partial T} \right)_P$, eq. 4.82, $\kappa \equiv -\frac{1}{V_m} \left(\frac{\partial V_m}{\partial P} \right)_T$, eq. 4.33

a) $\left(\frac{\partial P}{\partial T} \right)_{V_m} = \frac{\beta}{\kappa}$? Cycle Rule: $\left(\frac{\partial P}{\partial T} \right)_{V_m} \left(\frac{\partial V_m}{\partial P} \right)_T \left(\frac{\partial T}{\partial V_m} \right)_P = -1$

$\left(\frac{\partial P}{\partial T} \right)_{V_m} = - \frac{1}{\left(\frac{\partial V_m}{\partial P} \right)_T \left(\frac{\partial T}{\partial V_m} \right)_P} = - \frac{1}{\left(\frac{\partial V_m}{\partial P} \right)_T \left(\frac{\partial V_m}{\partial T} \right)_P}$

$\beta = \frac{1}{V_m} \left(\frac{\partial V_m}{\partial T} \right)_P \Rightarrow \left(\frac{\partial V_m}{\partial T} \right)_P = \beta (V_m)^*$

$\kappa = -\frac{1}{V_m} \left(\frac{\partial V_m}{\partial P} \right)_T \Rightarrow \left(\frac{\partial V_m}{\partial P} \right)_T = \kappa (-V_m)^*$

$\therefore \left(\frac{\partial P}{\partial T} \right)_{V_m} = - \frac{\beta (V_m)}{\kappa (-V_m)} = \frac{\beta}{\kappa}$ Matches!

b) $C_{P,m} - C_{V,m} = \frac{V_m T \beta^2}{\kappa}$? Use entropy (dS_m) expressions to relate $C_{P,m}$ & $C_{V,m}$; Both derived in class

(i) $dS_m = \frac{C_{V,m}}{T} dT + \left(\frac{\partial P}{\partial T} \right)_{V_m} dV_m$

(ii) $dS_m = \frac{C_{P,m}}{T} dT - \left(\frac{\partial V_m}{\partial T} \right)_P dP$

Equate dS_m expressions

$\therefore \frac{C_{V,m}}{T} dT + \left(\frac{\partial P}{\partial T} \right)_{V_m} dV_m = \frac{C_{P,m}}{T} dT - \left(\frac{\partial V_m}{\partial T} \right)_P dP$

$\left(\frac{\partial P}{\partial T} \right)_{V_m} dV_m + \left(\frac{\partial V_m}{\partial T} \right)_P dP = \frac{C_{P,m}}{T} dT - \frac{C_{V,m}}{T} dT$

$dT = \left[\left(\frac{\partial P}{\partial T} \right)_{V_m} dV_m + \left(\frac{\partial V_m}{\partial T} \right)_P dP \right] \cdot \frac{T}{C_{P,m} - C_{V,m}}$

$\left(\frac{\partial T}{\partial P} \right)_{V_m} = \frac{T}{C_{P,m} - C_{V,m}} \left(\frac{\partial V_m}{\partial T} \right)_P$ Isn't necessary here!

$\left(\frac{\partial T}{\partial V_m} \right)_P = \frac{T}{C_{P,m} - C_{V,m}} \left(\frac{\partial P}{\partial T} \right)_{V_m} = \frac{T}{C_{P,m} - C_{V,m}} \left(\frac{\beta}{\kappa} \right)$ Proved in part (a)

$\hookrightarrow \beta = \frac{1}{V_m} \left(\frac{\partial V_m}{\partial T} \right)_P \Rightarrow \left(\frac{\partial V_m}{\partial T} \right)_P = \beta (V_m)^* \Rightarrow \left(\frac{\partial T}{\partial V_m} \right)_P = \frac{1}{\beta V_m}$

$\therefore \frac{1}{\beta V_m} = \frac{T}{C_{P,m} - C_{V,m}} \left(\frac{\beta}{\kappa} \right) \Rightarrow C_{P,m} - C_{V,m} = \frac{T \beta^2 V_m}{\kappa}$ Matches!

'Divide' dT
by dP , const V_m
'Divide' dT
by dV_m , const P

Make
Substitution

2. $Z = \frac{P v_m}{RT} = 1 + B'P + C'P^2$ where B' & C' are constants, no T dependence!

(i) $\left(\frac{\partial h_m}{\partial P}\right)_T$, (ii) $\left(\frac{\partial h_m}{\partial P}\right)_{S_m}$, (iii) $\left(\frac{\partial h_m}{\partial T}\right)_P$, (iv) $\left(\frac{\partial h_m}{\partial T}\right)_{S_m}$ in terms of $B', C', R, P, T, C_{p,m}$

(i) $\left(\frac{\partial h_m}{\partial P}\right)_T$ FE: $dh_m = TdS_m + v_m dP \rightarrow$ 'Divide' by dP , entropy constant T

$\left(\frac{\partial h_m}{\partial P}\right)_T = T \left(\frac{\partial S_m}{\partial P}\right)_T + v_m \xRightarrow[\text{Maxwell relation}]{} \left(\frac{\partial h_m}{\partial P}\right)_T = -T \left(\frac{\partial v_m}{\partial T}\right)_P + v_m$

$\left(\frac{\partial v_m}{\partial T}\right)_P \cdot v_m = \frac{RT}{P} (1 + B'P + C'P^2)$ $\left(\frac{\partial h_m}{\partial P}\right)_T = 0$

$\left(\frac{\partial v_m}{\partial T}\right)_P = \frac{\partial}{\partial T} \left[\frac{RT}{P} (1 + B'P + C'P^2) \right] = \frac{R}{P} (1 + B'P + C'P^2)$

$\left(\frac{\partial h_m}{\partial P}\right)_T = -T \left[\frac{R}{P} (1 + B'P + C'P^2) \right] + v_m = -v_m + v_m = 0$

(ii) $\left(\frac{\partial h_m}{\partial P}\right)_{S_m}$ FE: $dh_m = TdS_m + v_m dP \rightarrow$ 'Divide' by dP , entropy constant S_m

$\left(\frac{\partial h_m}{\partial P}\right)_{S_m} = v_m = \frac{RT}{P} (1 + B'P + C'P^2)$

(iii) $\left(\frac{\partial h_m}{\partial T}\right)_P$

$\left(\frac{\partial h_m}{\partial T}\right)_P = C_{p,m}$ This is the definition of $C_{p,m}$

(i) $\left(\frac{\partial h_m}{\partial T}\right)_{s_m}$ FE: $dh_m = T ds_m + v_m dp$ → 'D.U.' by dT , enforcing constant s_m

$$\left(\frac{\partial h_m}{\partial T}\right)_{s_m} = v_m \left(\frac{\partial p}{\partial T}\right)_{s_m}$$

No direct relation
reciprocal

Cyclic Rule: $\left(\frac{\partial p}{\partial T}\right)_{s_m} \left(\frac{\partial s_m}{\partial p}\right)_T \left(\frac{\partial T}{\partial s_m}\right)_p = -1$

$$\left(\frac{\partial p}{\partial T}\right)_{s_m} = \frac{-1}{\left(\frac{\partial s_m}{\partial p}\right)_T \left(\frac{\partial T}{\partial s_m}\right)_p} = - \frac{\left(\frac{\partial s_m}{\partial T}\right)_p}{\left(\frac{\partial s_m}{\partial p}\right)_T} = \frac{\left(\frac{\partial s_m}{\partial T}\right)_p}{\left(\frac{\partial v_m}{\partial T}\right)_p}$$

FE: $dh_m = T ds_m + v_m dp$ → 'D.U.' by dT , enforcing constant p

$$\left(\frac{\partial h_m}{\partial T}\right)_p = T \left(\frac{\partial s_m}{\partial T}\right)_p \xRightarrow{\text{Definition}} \left(\frac{\partial s_m}{\partial T}\right)_p = \frac{C_{p,m}}{T}$$

$$\therefore \left(\frac{\partial p}{\partial T}\right)_{s_m} = \frac{C_{p,m}}{T \left(\frac{\partial v_m}{\partial T}\right)_p} \xRightarrow{\text{Plug back}} \left(\frac{\partial h_m}{\partial T}\right)_{s_m} = v_m \left(\frac{C_{p,m}}{T \left(\frac{\partial v_m}{\partial T}\right)_p}\right) \xRightarrow{\text{reciprocal}} \frac{v_m \cdot C_{p,m}}{T} \left(\frac{\partial T}{\partial v_m}\right)_p$$

$$\left(\frac{\partial T}{\partial v_m}\right)_p \cdot T = \frac{P v_m}{\rho(1 + B'P + C'P^2)}$$

$$\left(\frac{\partial T}{\partial v_m}\right)_p = \frac{\partial}{\partial v_m} \left[\frac{P v_m}{\rho(1 + B'P + C'P^2)} \right] = \frac{P}{\rho(1 + B'P + C'P^2)}$$

$$\Rightarrow \left(\frac{\partial h_m}{\partial T}\right)_{s_m} = \frac{C_{p,m} \cdot v_m}{T} \left[\frac{P}{\rho(1 + B'P + C'P^2)} \right] = \frac{C_{p,m}}{T} \left[\frac{P v_m}{\rho(1 + B'P + C'P^2)} \right]$$

$$\therefore \left(\frac{\partial h_m}{\partial T}\right)_{s_m} \xRightarrow{\text{use EOS}} \frac{C_{p,m}}{T} \cdot T = C_{p,m} \quad \left(\frac{\partial h_m}{\partial T}\right)_{s_m} = C_{p,m}$$

3. EOS: $P(u_m - b) = RT$ where b and $C_{v,m}$ are assumed to be constants

a) Show $u_m = u_m(T)$ Go w/ $u_m(T, v_m)$ total diff? to get a $C_{v,m}$ term!

$$du_m = \left(\frac{\partial u_m}{\partial T} \right)_{v_m} dT + \left(\frac{\partial u_m}{\partial v_m} \right)_T dv_m$$

$$= C_{v,m} dT + \left(\frac{\partial u_m}{\partial v_m} \right)_T dv_m$$

Use FE: $du_m = T ds_m - P dv_m$ 'Divide' by dv_m , enforcing constant T

$$\left(\frac{\partial u_m}{\partial v_m} \right)_T = T \left(\frac{\partial s_m}{\partial v_m} \right)_T - P \xrightarrow[\text{Maxwell Relation}]{\text{}} \left(\frac{\partial u_m}{\partial v_m} \right)_T = T \left(\frac{\partial P}{\partial T} \right)_{v_m} - P$$

$$\left(\frac{\partial P}{\partial T} \right)_{v_m} : P = \frac{RT}{v_m - b} ; \left(\frac{\partial P}{\partial T} \right)_{v_m} = \frac{R}{v_m - b}$$

$$\therefore \left(\frac{\partial u_m}{\partial v_m} \right)_T = T \left(\frac{R}{v_m - b} \right) - P = P - P = 0$$

$$\Rightarrow du_m = C_{v,m} dT + 0 \quad \text{hence } u_m = u_m(T)$$

b) Show $\frac{C_{p,m}}{C_{v,m}} = \text{constant}$

Definition: $dh_m = du_m + P dv_m \rightarrow$ 'Divide' by dT , enforcing constant P

$$dh_m = C_{p,m} dT + \left[v_m - T \left(\frac{\partial v_m}{\partial T} \right)_P \right] dP$$

We are looking for $C_{p,m}$, so hold P constant!

Plug in Definition $\Rightarrow C_{p,m} dT = C_{v,m} dT + P dv_m$

$$dv_m : v_m = \frac{RT}{P} + b ; dv_m = \frac{R}{P} dT$$

$$\therefore C_{p,m} dT = C_{v,m} dT + P \left(\frac{R}{P} dT \right) = C_{v,m} dT + R dT$$

$$C_{p,m} dT - C_{v,m} dT - R dT = 0$$

$$(C_{p,m} - C_{v,m} - R) dT = 0 \Rightarrow C_{p,m} - C_{v,m} - R = 0$$

$$\therefore C_{p,m} = C_{v,m} + R \Rightarrow \frac{C_{p,m}}{C_{v,m}} = 1 + \frac{R}{C_{v,m}}$$

constant as R is constant and $C_{v,m}$ is assumed constant

Derived in
class

4. Expr. for dS_m in terms of $P, T, V_m, C_{P,m}, C_{V,m}$ Start with $S_m(P, V_m)$
 $S_m = S_m(P, V_m) \rightarrow dS_m = dS_m(P, V_m)$

$$dS_m = \left(\frac{\partial S_m}{\partial P} \right)_{V_m} dP + \left(\frac{\partial S_m}{\partial V_m} \right)_P dV_m$$

$\left(\frac{\partial S_m}{\partial P} \right)_{V_m}$ • Use: $dU_m = T dS_m - P dV_m$ 'Divide' by dP , const. V_m
 $\left(\frac{\partial U_m}{\partial P} \right)_{V_m} = T \left(\frac{\partial S_m}{\partial P} \right)_{V_m} \Rightarrow \left(\frac{\partial S_m}{\partial P} \right)_{V_m} = \frac{1}{T} \left(\frac{\partial U_m}{\partial P} \right)_{V_m}$

$\left(\frac{\partial S_m}{\partial V_m} \right)_P$ • Use: $dH_m = T dS_m - V_m dP$ 'Divide' by dV_m , const. P
 $\left(\frac{\partial H_m}{\partial V_m} \right)_P = T \left(\frac{\partial S_m}{\partial V_m} \right)_P \Rightarrow \left(\frac{\partial S_m}{\partial V_m} \right)_P = \frac{1}{T} \left(\frac{\partial H_m}{\partial V_m} \right)_P$

$$\Rightarrow dS_m = \frac{1}{T} \left(\frac{\partial U_m}{\partial P} \right)_{V_m} dP + \frac{1}{T} \left(\frac{\partial H_m}{\partial V_m} \right)_P dV_m$$

$\left(\frac{\partial U_m}{\partial P} \right)_{V_m}$ • Use: $dU_m = C_{V,m} dT + \left[T \left(\frac{\partial P}{\partial T} \right)_{V_m} - P \right] dV_m$ (Deriv in class)

$$\left(\frac{\partial U_m}{\partial P} \right)_{V_m} = \frac{\partial}{\partial P} \left[C_{V,m} dT + \left[T \left(\frac{\partial P}{\partial T} \right)_{V_m} - P \right] dV_m \right]_{V_m} = C_{V,m} \left(\frac{\partial T}{\partial P} \right)_{V_m}$$

$\left(\frac{\partial H_m}{\partial V_m} \right)_P$ • Use: $dH_m = C_{P,m} dT + \left[V_m - T \left(\frac{\partial V_m}{\partial T} \right)_P \right] dP$ (Deriv in class)

$$\left(\frac{\partial H_m}{\partial V_m} \right)_P = \frac{\partial}{\partial V_m} \left[C_{P,m} dT + \left[V_m - T \left(\frac{\partial V_m}{\partial T} \right)_P \right] dP \right]_P = C_{P,m} \left(\frac{\partial T}{\partial V_m} \right)_P$$

$$\Rightarrow dS_m = \frac{C_{V,m}}{T} \left(\frac{\partial T}{\partial P} \right)_{V_m} dP + \frac{C_{P,m}}{T} \left(\frac{\partial T}{\partial V_m} \right)_P dV_m$$

Can be derived from a given EOS!

5. We are interested in the thermodynamic properties of a strip of rubber as it is stretched. Consider n moles of pure ethylene propylene rubber (EPR) that has an unstretched length z_0 . If it is stretched by applying a force F , it will obtain an equilibrium length z given by $F = kT(z - z_0)$, where k is a positive constant.

Rubber Elastic Expansion

$$C_{z,m} = \left(\frac{\partial U_m}{\partial T} \right)_z = a + bT \quad \text{for constant } z$$

a) $dU = dQ_{rev} + dW_{rev}$

second law: $dQ_{rev} = TdS$ } $dU = TdS + Fdz$

work from mechanics: $dW_{rev} = Fdz$ } plug in $F_{exp} \Rightarrow dU = TdS + kT(z - z_0)dz$

de Helmholtz free energy: $A = U - TS$

Total differential: $dA = dU - TdS - SdT$

Result from dU: $dA = (TdS + kT(z - z_0)dz) - TdS - SdT$
 $\Rightarrow dA = kT(z - z_0)dz - SdT$

b) $S = S(T, z) \rightarrow dS = dS(T, z)$

total differential: $dS = \left(\frac{\partial S}{\partial T} \right)_z dT + \left(\frac{\partial S}{\partial z} \right)_T dz$

$\left(\frac{\partial S}{\partial T} \right)_z$ use dU! \cdot Use: $dU = TdS + kT(z - z_0)dz$ 'Divide' by dT on both constant z
 $\Rightarrow \left(\frac{\partial U}{\partial T} \right)_z = T \left(\frac{\partial S}{\partial T} \right)_z \Rightarrow \left(\frac{\partial S}{\partial T} \right)_z = \frac{1}{T} \left(\frac{\partial U}{\partial T} \right)_z = \frac{n}{T} \left(\frac{\partial U_m}{\partial T} \right)_z$

$\therefore \left(\frac{\partial S}{\partial T} \right)_z = \frac{n}{T} (a + bT) = \frac{nC_{z,m}}{T}$

$\left(\frac{\partial S}{\partial z} \right)_T$ use dA! \cdot Use: $dA = kT(z - z_0)dz - SdT$ 'Divide' by dT on both constant z : $\left(\frac{\partial A}{\partial T} \right)_z = -S$

'Divide' by dz on both constant T : $\left(\frac{\partial A}{\partial z} \right)_T = kT(z - z_0)$

order of differentials does not matter!
 $\left(\frac{\partial S}{\partial z} \right)_T \cdot -S = \left(\frac{\partial A}{\partial T} \right)_z \Rightarrow -\left(\frac{\partial S}{\partial z} \right)_T = \frac{\partial}{\partial z} \left[\left(\frac{\partial A}{\partial T} \right)_z \right]_T = \frac{\partial}{\partial T} \left[\left(\frac{\partial A}{\partial z} \right)_T \right]_z$
 $\therefore -\left(\frac{\partial S}{\partial z} \right)_T = \frac{\partial}{\partial T} [kT(z - z_0)]_z \Rightarrow \left(\frac{\partial S}{\partial z} \right)_T = -k(z - z_0)$

Combine!
 $dS = \frac{nC_{z,m}}{T} dT - k(z - z_0)dz$

c) $U = U(T, z) \rightarrow dU = dU(T, z)$ (defining, also shown in b)

total differential: $dU = \left(\frac{\partial U}{\partial T}\right)_z dT + \left(\frac{\partial U}{\partial z}\right)_T dz = nC_{v,m} dT + \left(\frac{\partial U}{\partial z}\right)_T dz$

$\left(\frac{\partial U}{\partial z}\right)_T$ We: $dU = TdS + kT(z-z_0)dz$ 'Divide' by dz on RHS, constant T

$$\left(\frac{\partial U}{\partial z}\right)_T = T \left(\frac{\partial S}{\partial z}\right)_T + kT(z-z_0)$$

$$= T \left(\underset{\text{(from b)}}{-k(z-z_0)}\right) + kT(z-z_0) = 0$$

$\therefore dU = nC_{v,m} dT$

d) $F_u = \left(\frac{\partial U}{\partial z}\right)_T = 0$ This was found in part c!

$$F_s = -T \left(\frac{\partial S}{\partial z}\right)_T = -T \left(\underset{\text{(from b)}}{-k(z-z_0)}\right) = kT(z-z_0)$$

$$F_s = kT(z-z_0)$$

e) Adiabatically $\rightarrow dQ_{rev} = 0 \therefore dU = kT(z-z_0)dz$

$z_i \rightarrow z_f$ as we're stretching, T is a positive const (initial T) by const., k is a const (positive), so $dU > 0$ must be true! Thus for, the temperature will increase as the BPR's internal energy is increasing.

6. Answer the following reflection questions (5 points):

a. What about the way this class is taught is helping your learning?

b. What about the way this class is taught is inhibiting your learning?

A) The use of class time with examples and discussion helps my learning the most. The examples are challenging and serve as good parallels to the homework assignments. Being able to reference the notes, as I did through a lot of this assignment, makes the problems much more manageable. The enthusiasm Dr. Warburton has for the class's learning also encourages my learning, as I am more engaged and thus take in more of the information. The content is ever increasing in difficulty, but the way that the examples scale with this difficulty greatly helps my learning.

B) There is nothing currently inhibiting my learning. The class is not moving too fast, nor is it moving too slow. The examples are challenging but in a good way, discussion questions are helpful in engaging group discussion and invoking problem solving skills, and the homework enforces key topics from lecture. The way this class is taught currently does nothing but help my learning.