



EMAC 276 Final Exam Review



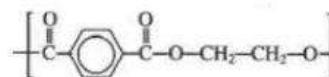
Exam Overview

9 – 11 AM 5/2

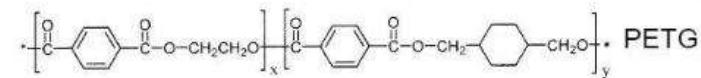
- 120 minutes
- Open-Book/Open-Note
- Only material after Midterm
- True/False Questions
- Multiple Choice
- Two short response

Polyesters - Poly(ethylene terephthalate) (2-6)

- T_g: 72-100 °C
- T_m: 255-265 °C
- Bisphenol-A free PETG to replace PC in teeth aligners, water/milk bottles, medical containers
- PETG contains glycol comonomer and is less brittle than PET
- Applications:
 - Injection molding and bottles
 - Fibers for textiles and tire
 - Films and sheets
 - Multilayer film coextrusion
 - 1D photonic crystals
 - 3M Enhanced Specular Reflection (ESR) films



Homopolymers and copolymers (e.g., PETG)
"G" means other "glycols"

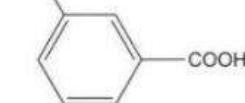


Diacid	+	Diol	→	Polyester
TPA	+	EG	→	PET
TPA	+	EG + CHDM	→	PETG (copolyester where EG > CHDM)
TPA	+	CHDM + EG	→	PCTG (copolyester where CHDM > EG)
TPA + IPA	+	CHDM	→	PCTA (copolyester)
TPA	+	CHDM	→	PCT

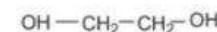
EASTMAN



Terephthalic acid (TPA)



Isophthalic acid (IPA)



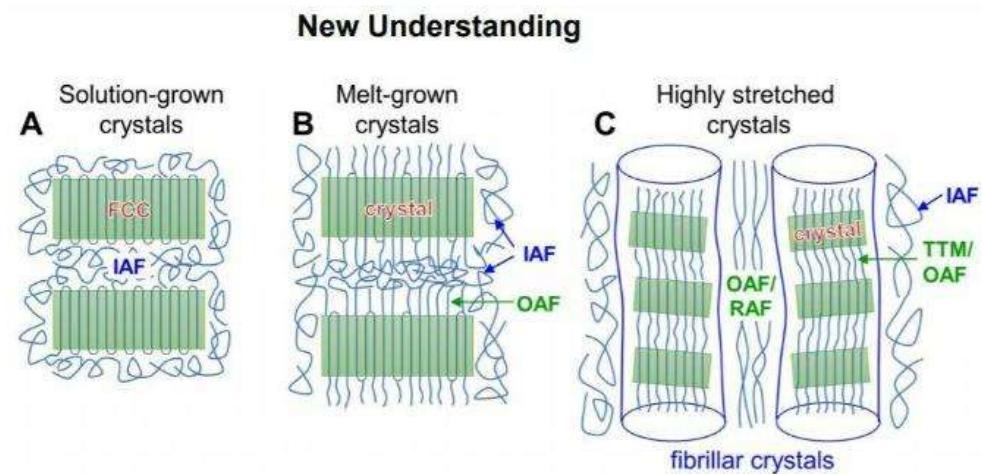
Ethylene Glycol (EG)



1,4 Cyclohexanedimethanol (CHDM)

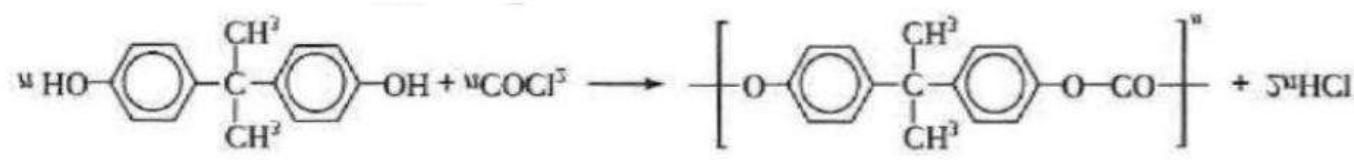
Polyesters - Poly(ethylene terephthalate) (2-6)

- PET has moderate crystallinity (0-40%)
- Isotropic amorphous fraction (IAF)
- Taut-tie molecules (TTM)
- Rigid amorphous fraction (RAF) enhances
 - Mechanical properties
 - Gas barrier properties
 - Electrical properties



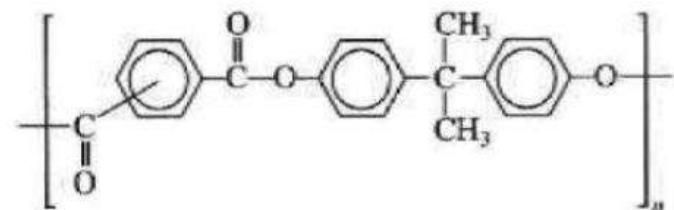
Polyesters - Polycarbonate (18-21)

- Synthesis:
 - Ester exchange
 - Phosgenation
- High toughness
- Amorphous
- Tg: 145 °C
- High transparency & Refractive Index
- Applications
 - Construction materials
 - Data storage
 - Automotive, aircraft and security
 - Medical applications

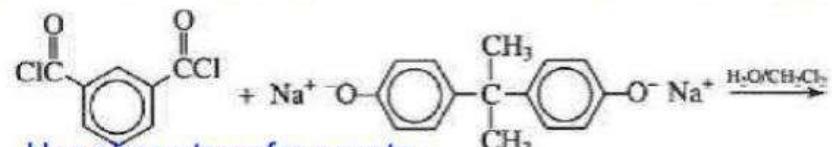


Polyesters - Aromatic Polyesters (22)

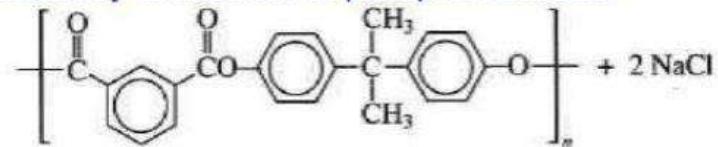
- Also called polyarylates
- Amorphous
- Tg: 190 °C
- Tougher and better heat resistance than PC
- Applications
 - Automobiles
 - Electronics
 - 5G antennas
 - Fire helmets/shields



1) Interfacial polycondensation (low temp.)

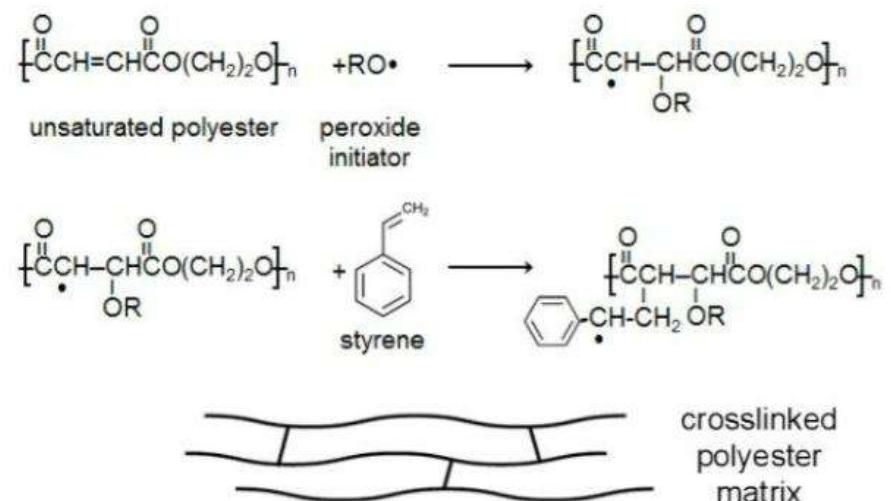


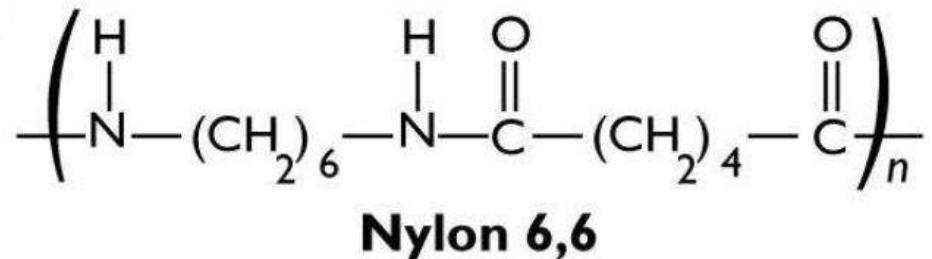
Use phase transfer agents:
quaternary ammonium or phosphonium salts



Polyesters - Unsaturated Polyester Resins (27-32)

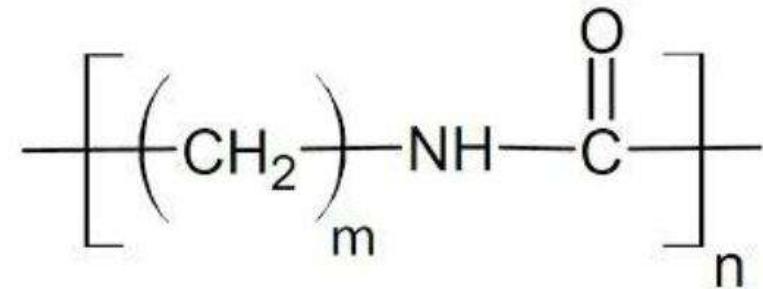
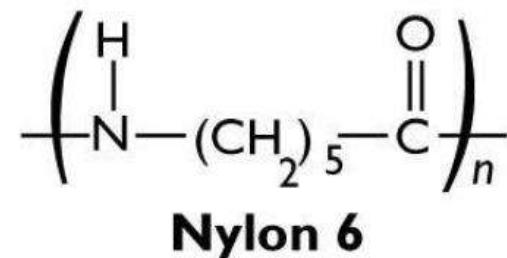
- Made by curing linear low MW unsaturated polyester, dissolved in vinyl monomer (styrene) with inhibitor
- The unsaturated component in UPRs is maleic/fumaric acid or maleic anhydride
- Rigidity depends on structure of saturated modifying acid
- Thermoset, insoluble and infusible after curing
- Applied hand lay-up or spray-up
- Applications
 - Transportation
 - Construction





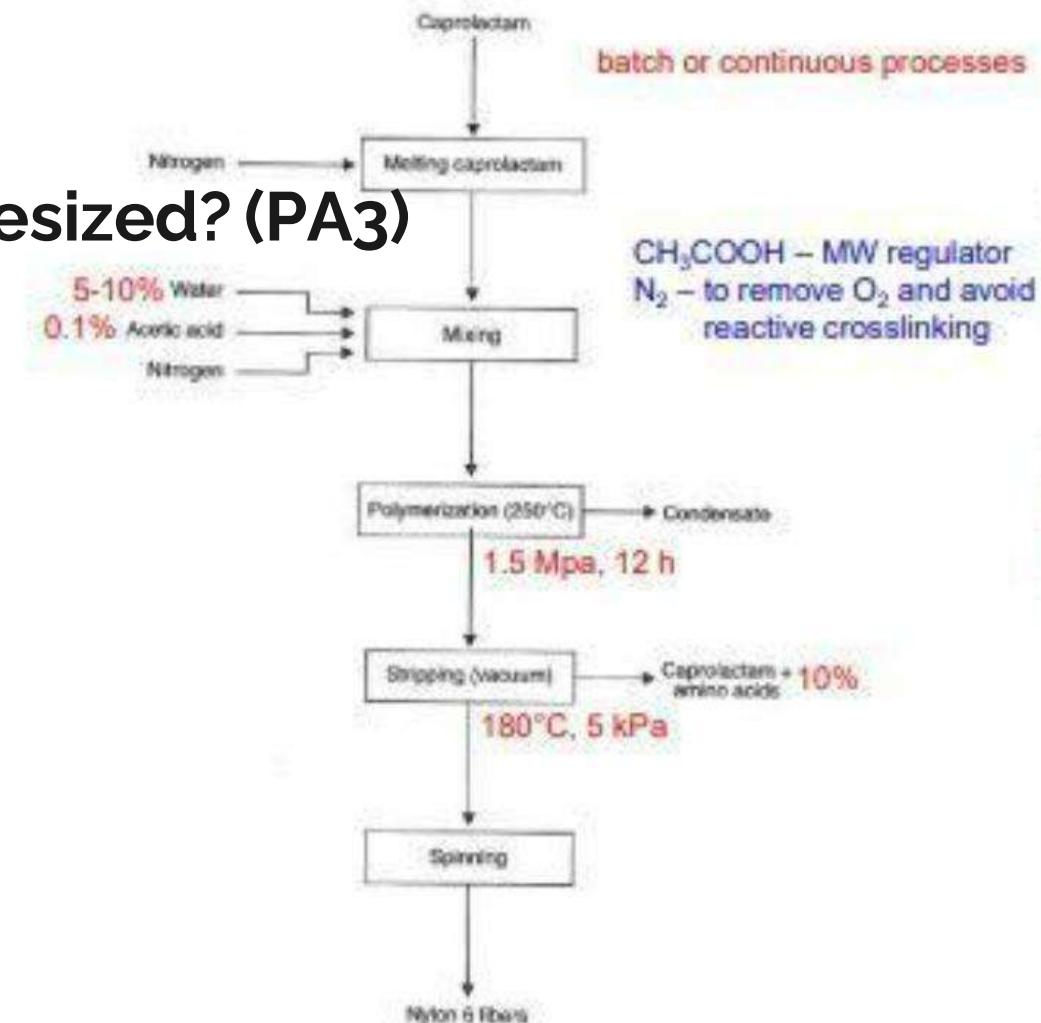
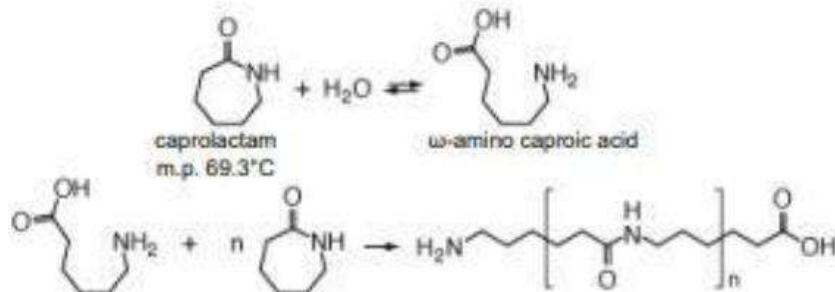
What are Polyamides (PA2)?

- Chemically (-NH-CO-)
 - n-Nylons
 - Nylon m,n
 - Copolymers
- Common PAs
 - Nylon 6
 - Nylon 11
 - Nylon 12
 - Nylon 6,6,



How are Polyamides Synthesized? (PA3)

- Polycondensation Rxns are common
 - Nylon 6,6 from adipic acid and hexamethylenediamine
- Ring Opening Polymerizations
 - Nylon 6 with caprolactam
- How does this scale up?
 - Hydrolytic Ring Opening Polymerization (Industry) with lactams
 - E.g. Nylon 12 with lauryl lactam





Properties of Polyamides (PA6)

Advantages

- High T_m
- High tensile strength
- High abrasion resistance
- Low linear CTE
- Good solvent resistance
- Good O₂ barrier
- Tailored properties with copolymers

Disadvantages

- Absorb moisture
- Low melt viscosity
- Poor UV resistance (need stabilizers)
- Easily oxidized
- Volume shrinkage after crystallization
- Electrical and mechanical properties affected by moisture

Why do we get really nice properties? \Rightarrow Hydrogen Bonding!



Can we modify Polyamide? (PA8)

YES!

- Nucleating Agents (e.g. silica)
- Heat stabilizers (e.g. copper salts)
- Light Stabilizers (e.g. carbon black)
- Impact Modifiers (e.g. rubber-type materials)
- Other Fillers (e.g. plasticizers)
- Reinforcing Fillers (e.g. glass fibers)

Where can we find Polyamides? (PA9-10)

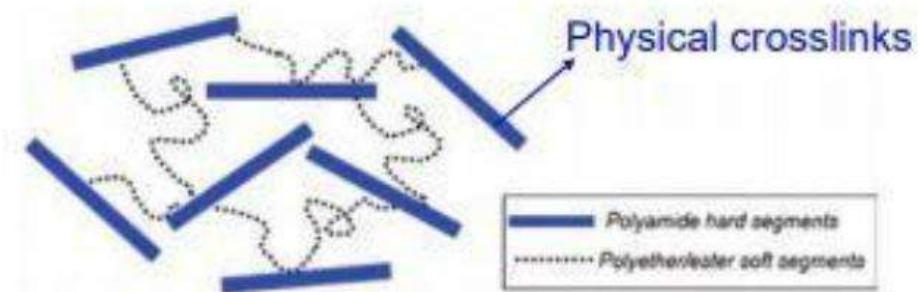
EVERYWHERE

- Fiber applications (e.g. monofilaments)
- Plastics (e.g. clothing or gears)
- Adhesives (e.g. flexible adhesives)
- Appliances (e.g. personal care)
- Automotive
- Consumer Equipment
- Electrical
- Hardware
- Packaging



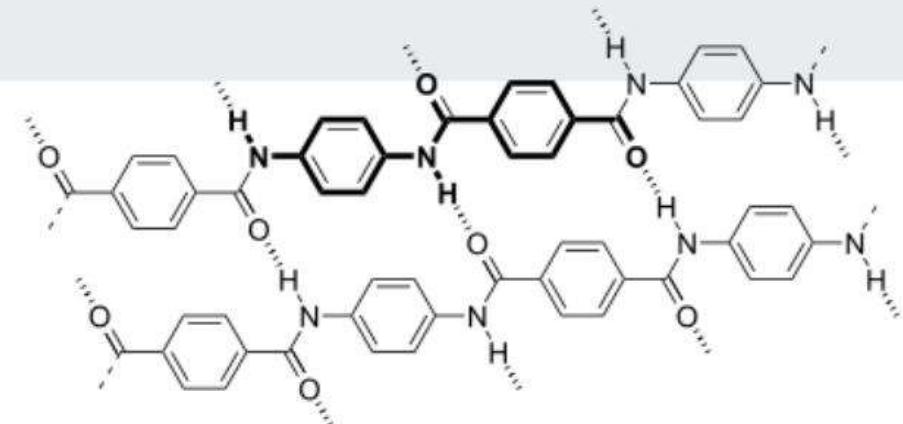
What are PEBA^X/TPEs? (PA11-13)

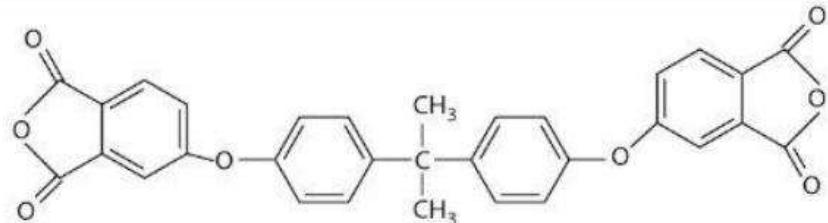
- Polyamide Thermoplastic Elastomers
- Block copolymers
 - Hard segments (e.g. PA 6, PA, 11, PA 12)
 - Soft segments (e.g. PTMO, PEOP, PPO)
- Synthesis = Melt Polycondensation
 - Polyamide + diacid
 - Polyol + polyamide
- High Performance Polymer with High Performance Properties (and \$\$\$)
 - Great Abrasion Resistance and Flexibility
 - Great Chemical Resistance and Mechanical Properties
 - UV Stability
- Multiple Applications



What are Aramids? (PA14-18)

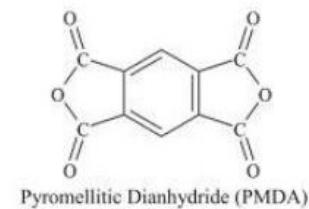
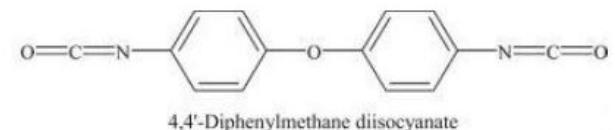
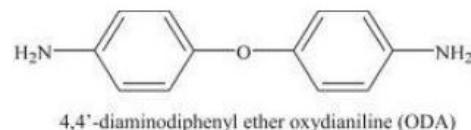
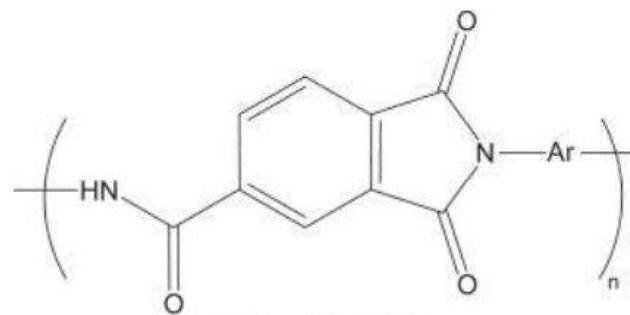
- Aramids = aromatic polyamides
- Can be partially aromatic (multiple copolymers available)
- Example: Kevlar
 - Poly(p-phenylene terephthalamide)
 - Solution Polymerization
 - Great properties (i.e. highest strength among commercial fibers)
 - Highly Crystalline
 - Forms LC in sulfuric acid (Kwolek + Fiber spinning)





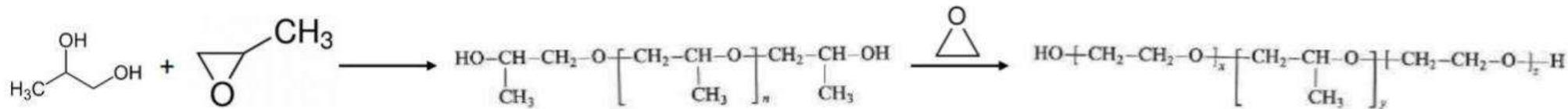
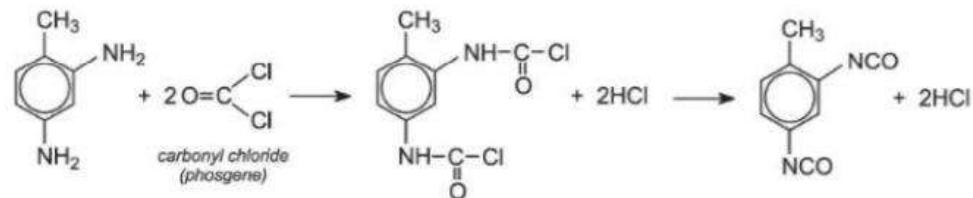
What are Polyimides? (PA19-21)

- Chemically (-CO-NR-CO-)
- One or Two step synthesis
- Properties
 - Heat resistance (high Tg)
 - Dielectric Strength
 - Oxidative Resistance
 - Flame Retardant
- Applications
 - Electric motors
 - Jet Engines
 - Flexible Electronics
- Modifications
 - PAIs = Poly(amide-imides)
 - PEIs = Poly(ether0imides)



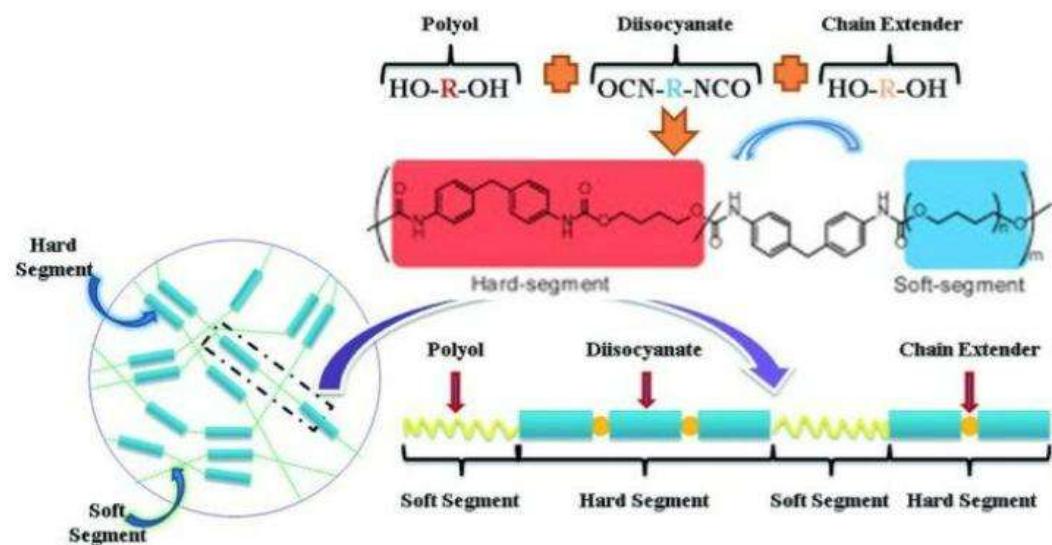
Polyurethanes - Formulation (3-11)

- Made up of urethane groups $-\text{NH}-(\text{C=O})-\text{O}-$
- Isocyanates
 - Made via phosgenation
 - Highly reactive from phosgenation
- Polyols
 - Polyethers polyols (PEPs)
 - Polyesters polyols (PESPs)
- Chain Extenders
 - Aromatic diamines and aliphatic or aromatic hydroxyl compounds
- Catalysts



Polyurethanes - Cast PU Elastomers (12-14)

- Diisocyanates + Polyols (soft segment) between Diisocyanates + Chain Extenders (hard segment)
- High abrasion and tear resistance
- Chemical resistance
- Maximum service temp: 120 °C
- Applications: rollers, pipes, molds
- Cannot remelt due to cast and crosslinking



Polyurethanes - Thermoplastic PU Elastomers (15)

- Linear block copolymers
- Can be melted like thermoplastics
- Mechanical strength is temperature sensitive
- Applications:
 - Medical grade catheter,
 - Wire jacketing
 - Adhesives



Microphase separated morphology

— Soft phase of TPU

█ Hard phase of TPU (Crystalline)

Table 7.6 Some properties of TPUs based on polycaprolactone polyester

Hardness	80 Shore A	55 Shore D
Tensile strength (MPa)	52	43
Elongation (%)	490	350
Modulus 100% (MPa)	6	20

Polyurethanes - PU fibers & Millable PU Rubbers (16-17)

- PU Fibers (Spandex)
 - Chemical crosslinks through biuret and allophanate structures at elevated temperatures during the curing of fibers.
 - Replaces natural rubber fibers in many applications
- Millable PU Rubbers
 - Crosslinked and processed using conventional rubber roll mills
 - Superior abrasion resistance and mechanical strength

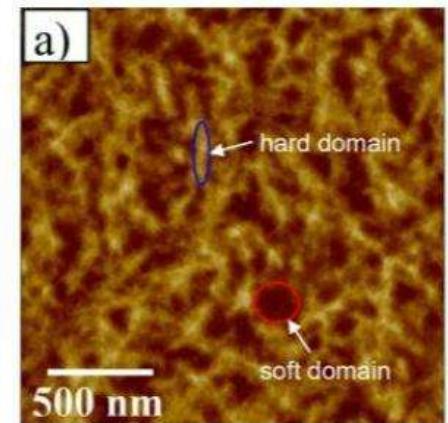
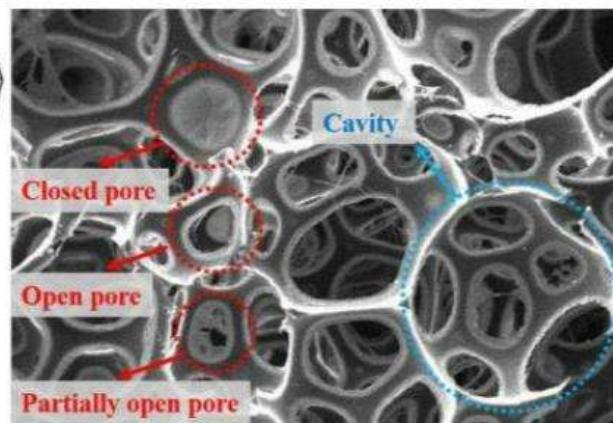
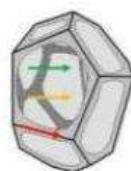
Uses

athletic, aerobic
exercise apparel
swimsuits
ski pants
disco jeans
hosiery
socks
underwear
surgical hose
support hose
motion capture suits
Home furnishings



Polyurethanes - PU Foams (18-22)

- Versatile foams, brittle to flexible
- Rigid foams contain more closed cell structure and adds resilience
 - Used for building thermal insulation
- Flexible foams contain more open cell structure
 - Used for cushioning and textiles
- Adding water acts as foaming agent and potential crosslinker



Polym. Adv. Technol. 2018, 29, 852-859; *J. Macromol. Sci. Part B: Phys.* 2003, 42, 6, 1125-1139.

Polyurethanes - PU by Reaction Injection Molding (23-24)

- High pressure intensive mixing
- Employs continuous circulation of reactants
- Quick injection into molds
- 1-2 post-curing time
- Can be reinforced with glass (RRIM) to provide a high flexural modulus and low thermal expansion coefficient

Table 7.12 Some properties of PU/urea RIM for automotive applications [68]

<i>Property</i>	<i>Unfilled</i>	<i>15.8% Glass filled</i>
Density (g/cm ³)	1.02	1.07
Hard segment content (%)	41.1	30.9
Flexural modulus (MPa)		
-29°C	849	618/392*
22°C	405	360/194*
70°C	304	260/137*
Tensile strength (MPa)	26.7	15.2
Elongation (%)	195	125/180*
Coefficient of thermal expansion ×10 ⁻⁶ /°C (from 66°C to 121°C)	170	51/145*

* parallel/perpendicular

Polyurethanes - Coatings, Adhesives, & Sealants (25-28)

Coatings

- Strongly adheres to substrate from its polar nature
- Tough and flexible, crack resistant

Adhesives

- Can use water dispersions
- Sticks on most surfaces, whether porous or not
- Not as strong as epoxy adhesives

Sealants

- Stronger adhesion than Si sealants

Typical Applications

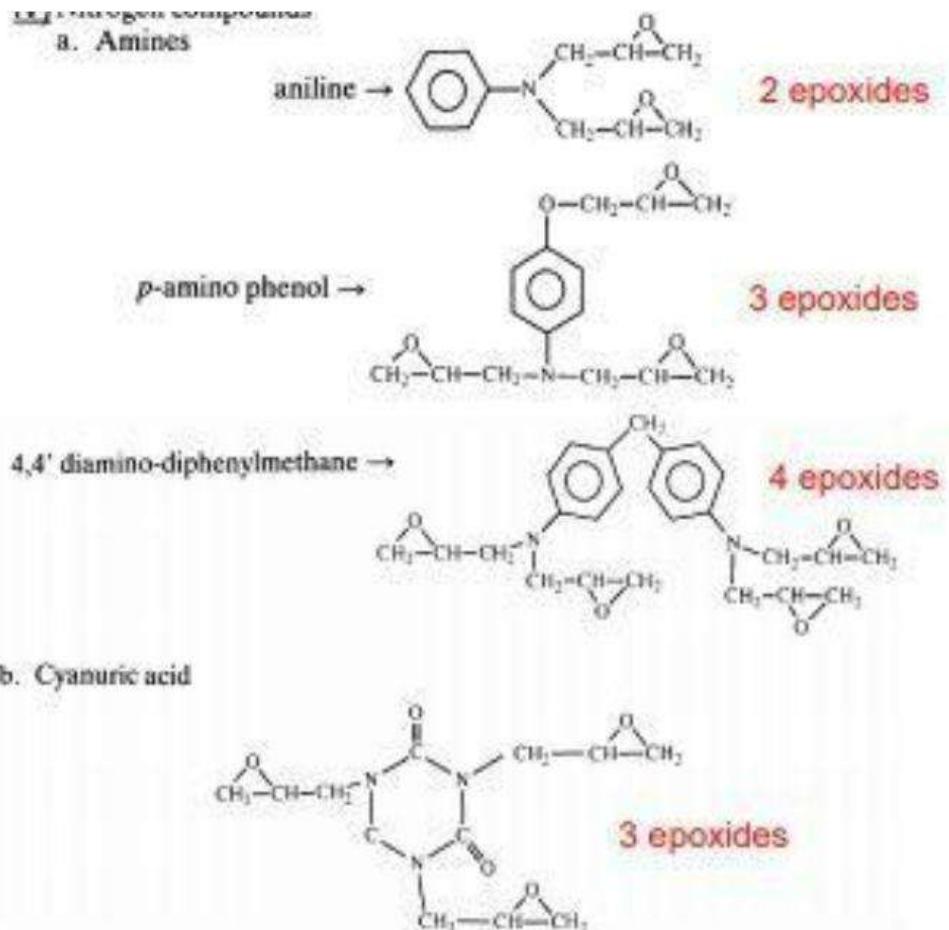


Difference between PU and silicone sealants

Sealants	PU	Silicone
UV resistance	Not good, need additives	Good
Lifespan	5 years	20 years
Cost	Less expensive	More expensive
Versatility	Less	More
Wood sealing	Good adhesion	Poor adhesion

What are Epoxies? (EP2-3)

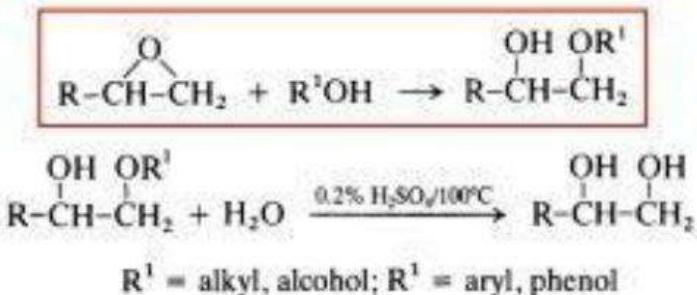
- Three-member ring
- Thermoset polymer resin
- General Properties
 - High strength and large cohesive force
 - Good adhesion
 - Excellent stability
 - Flexibility and diversity in design



How do we make Epoxies? (EP4-14)

Variety of ways!

- Phenols, Alcohols, Carboxylic or Fatty Acids, Nitrogen compounds
- Epoxy Prepolymers
 - Bisphenol A Epoxy (Taffy Method)
 - Fusion Method
- Curing
 - Amines (most versatile)
 - Diacids/Anhydrides
 - Various Hardeners have different effects

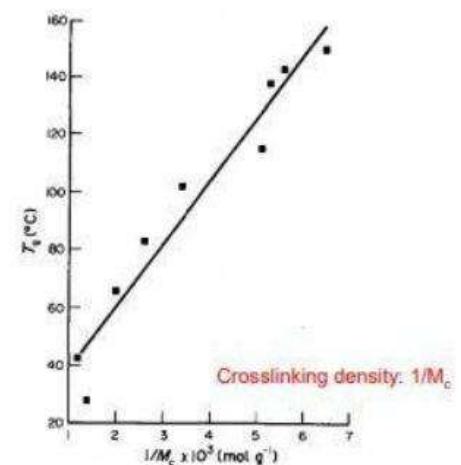


Hardeners	Advantages	Disadvantages
Aliphatic amines	Convenient, low cost, RT cure, low viscosity	Skin irritant and blushes
Aromatic amines	Moderate heat resistance, chemical resistance	Solids at RT, long and elevated cures
Polyamides	RT cure, flexibility, toughness, low toxicity	High cost, high viscosity, low HDT
Anhydrides	Heat and chemical resistance	Long, elevated cures
Polysulfides	Moisture insensitive, quick set	Odor, poor HDT
Catalysts	Long pot life, high HDT	Long, elevated cures, poor moisture resistance
Phenol/formaldehyde	HDT, chemical resistance, hardness	Solid, weatherability

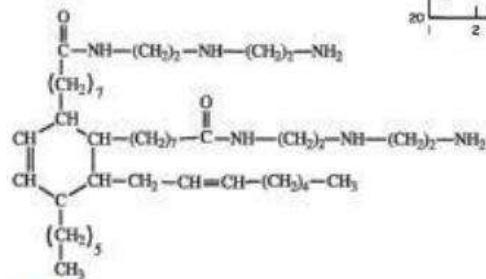
Can we modify Epoxies? (EP17-27)

OF COURSE WE CAN

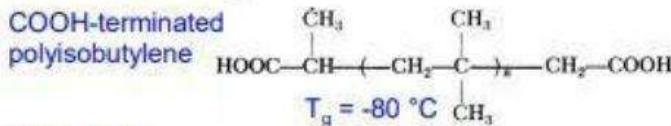
- Structure Property Relationship
 - T_g is proportional to $1/M_c$
- Curing agent affects mechanical properties
- Additives
 - Diluents can be either reactive or unreactive (reduce viscosity, increase wettability)
 - Flexibilizers are reactive (reduce brittleness, increase flexibility)
 - Inorganic fillers (reduce shrinkage and cost)
- Toughening (THINK: airplanes)
 - Resinous additives (i.e. aliphatic polyamides or nylon)
 - Elastomers (BAN, Polysiloxanes, fluoroelastomers)



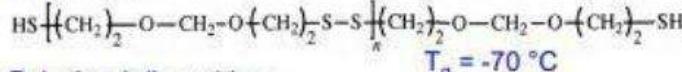
Amides



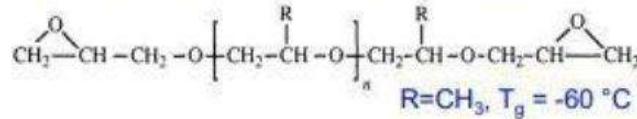
COOH-terminated
polyisobutylene



Polysulfides



Polyglycol diepoxides



$\text{R}=\text{CH}_3, T_g = -60^\circ\text{C}$



Applications of Epoxy Composites/Coatings (EP28-30,34)

- Electric Vehicles (Composites)
 - Carbon fiber composites
 - Printed Circuit Boards (PCBs) (Composites)
 - Epoxy laminates
 - Need low dielectric constant and low dielectric loss
 - Wind Turbine Blades (Composites)
 - Epoxy laminates
 - Industrial Flooring (Coating)
 - Waterproof Roofing (Coating)
 - Concrete Repair (Coating)
-
- More on Coatings
 - Surface Preparation
 - Solution Coating
 - Control film thickness
 - Dip Coating
 - Good control
 - Water Dispersions
 - Strong mechanical shearing
 - Difficult to achieve uniform dispersions
 - Powder Coatings
 - 100% solid
 - Electrical resistance
 - Corrosion resistance



General Tips for Studying and Last Minute Review

- 120 minutes - pace yourself
- ***Focus on structure-property relationships
- Don't focus on reaction mechanisms
- Compare types of polymers in each section and how they are processed and applied in industry