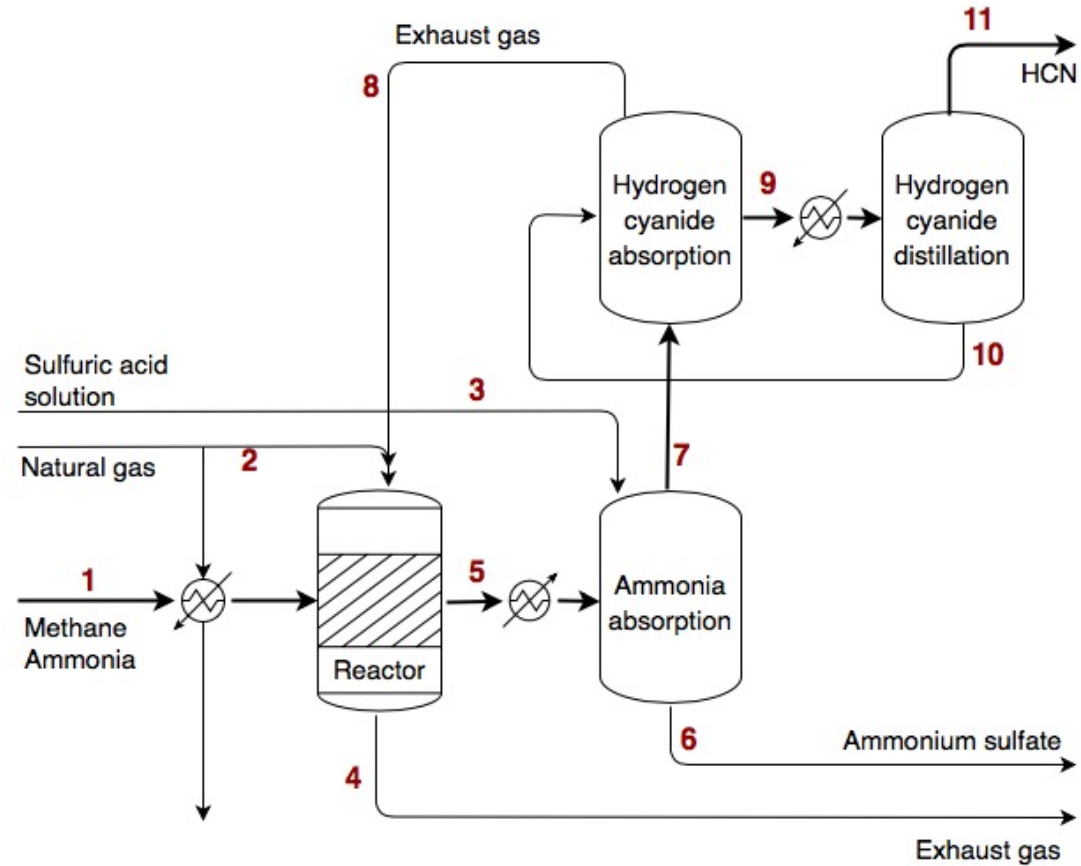


Case Studies in Process Design II

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Reactor flow sheet

General Modelling Approach

- Four cases considered:

	Case 1	Case 2	Case 3	Case 4
Gas model	ideal gas law	ideal gas law	Peng-Robinson-EOS	Peng-Robinson-EOS
Heating mode	cross-current	cross-current	cross-current	co-current
Absorbers	shortcut model	shortcut model	non-ideal Eq.-model	non-ideal Eq.-model
Distillation	Fenske-Underwood-Gilliland	Fenske-Underwood-Gilliland	NRTL	NRTL
ΔP considered	no	yes	yes	yes

- Overall optimization with a “black box” approach

ODE-system governing the reactor

$$\frac{dF_{N_2}}{dV} = r_2$$

$$\frac{dF_{CH_4}}{dV} = -r_1$$

$$\frac{dF_{NH_3}}{dV} = -r_1 - 2 \cdot r_2$$

$$\frac{dF_{H_2}}{dV} = 3 \cdot r_1 + 3 \cdot r_2$$

$$\frac{dF_{HCN}}{dV} = r_1$$

$$\frac{dT}{dV} = \frac{a \cdot U \cdot (T_a - T) - (r_1 \cdot \Delta H_{R_1} + r_2 \cdot \Delta H_{R_2})}{\sum_i F_i \cdot c_{p,i}}$$

$$\frac{dT_a}{dV} = \begin{cases} \frac{a \cdot U \cdot (T - T_a)}{\dot{m}_{Flu-Gas} \cdot c_{p,Flu-Gas}} & \text{for co-current heating} \\ 0 & \text{for cross-current heating} \end{cases}$$

$$\frac{dP}{dV} = \begin{cases} \frac{32 \cdot f_f \cdot \rho_{in} \cdot Q_{in} \cdot Q}{4 \cdot D^3} & \text{for considering a pressure drop} \\ 0 & \text{without pressure drop} \end{cases}$$

Assumptions for Ammonia Absorber

General

- Isothermal column
- Instantaneous & irreversible reaction
- Heat of absorption neglected
- Outlet temperature for liquid and gas alike
- No HCN absorption in water
- No pressure drop
- 20 % excess of sulfuric acid

Ideal case

- Heat capacities are T independent

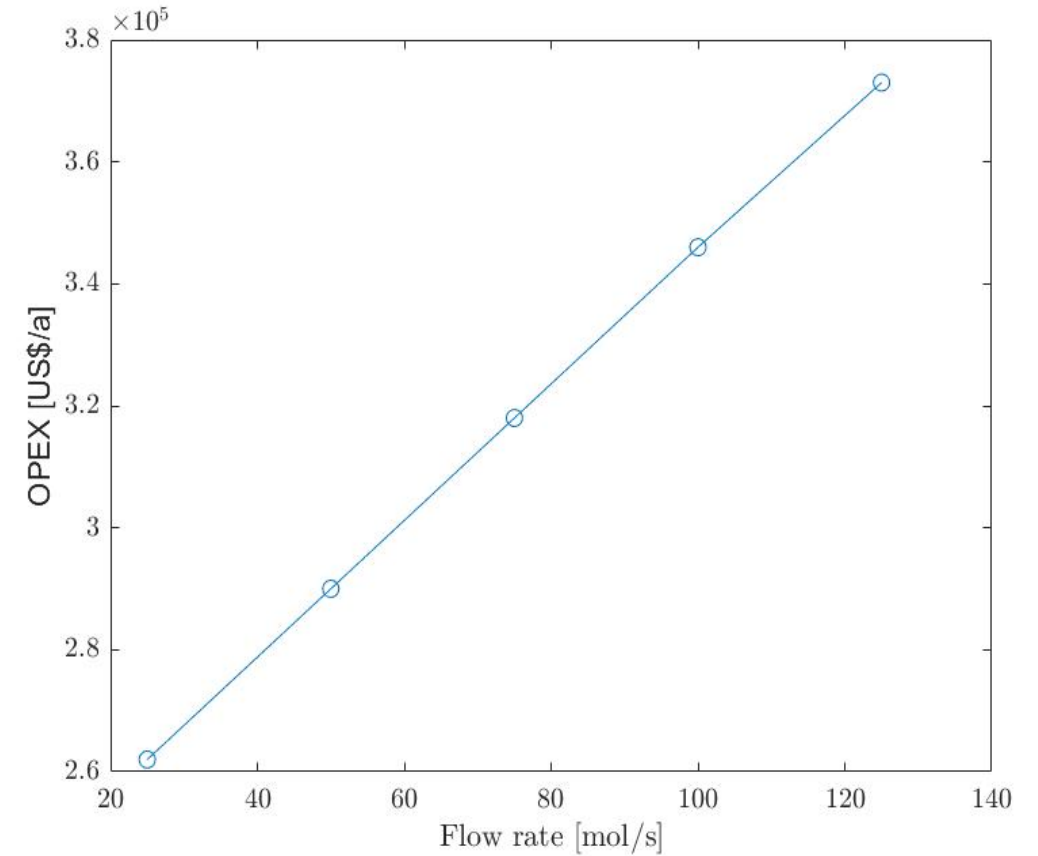
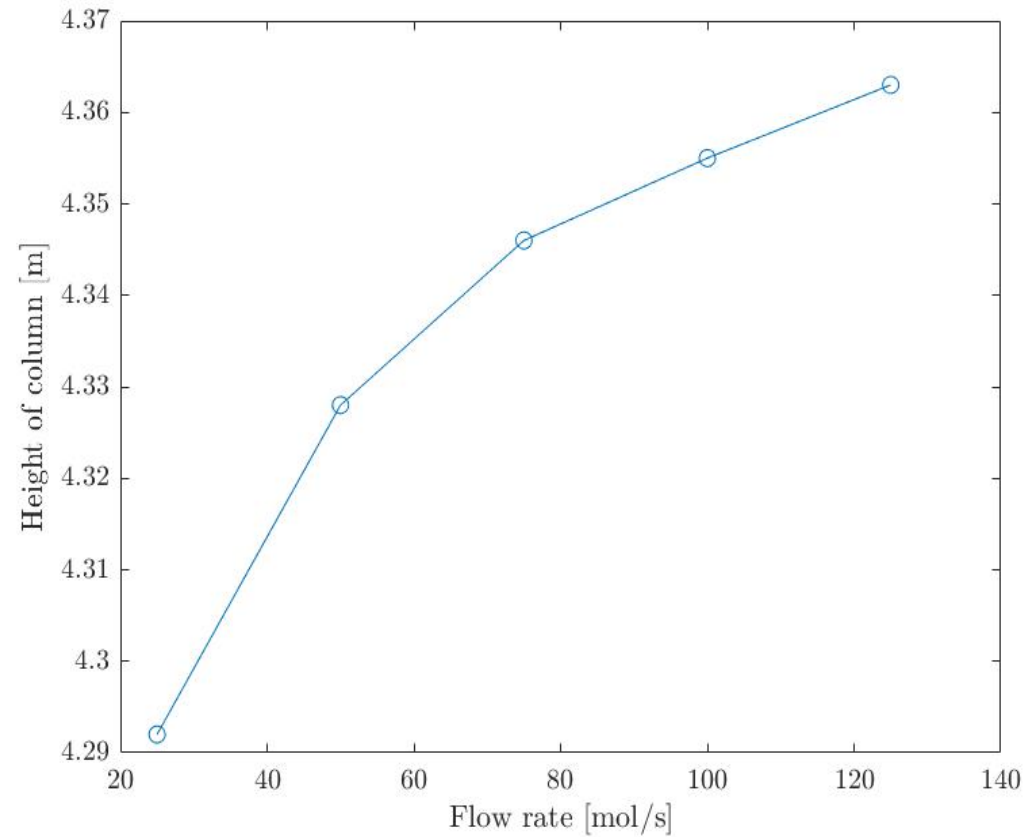
$$\Delta_f H(T) = \Delta_f H^0 + c_p \cdot (T - T_0),$$

Non ideal case

- Heat capacities are T dependent

$$\Delta_f H_i(t) = \Delta_f H_i^0 + \left(A_i \cdot t + \frac{B_i \cdot t^2}{2} + \frac{C_i \cdot t^3}{3} + \frac{D_i \cdot t^4}{4} - \frac{E_i}{t} + F_i - H_i \right)$$

Sensitivity Analysis of the Ammonia Absorber



Made assumptions for the HCN absorber

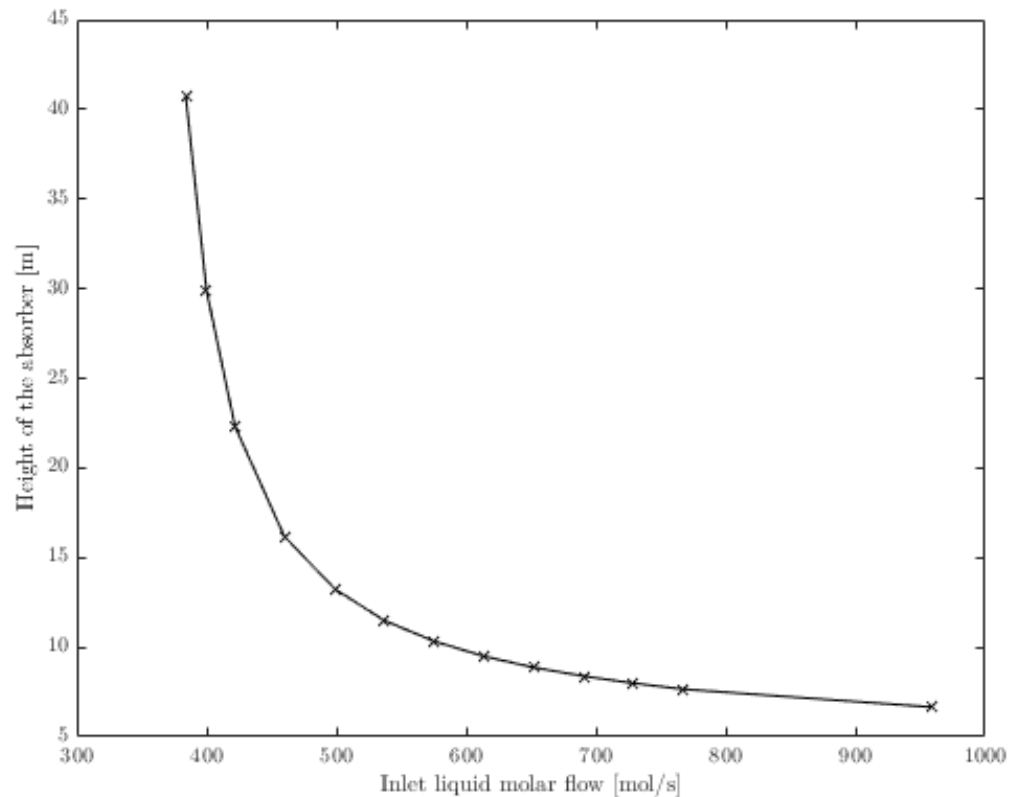
Ideal case

- Isothermal column
- Heat of absorption neglected
- Outlet temperature same for gas and liquid
- Only HCN is absorbed
- Heat capacities not temperature dependent
- No HCN in liquid inlet stream
- Resistivity in the liquid phase neglected

Non-ideal case

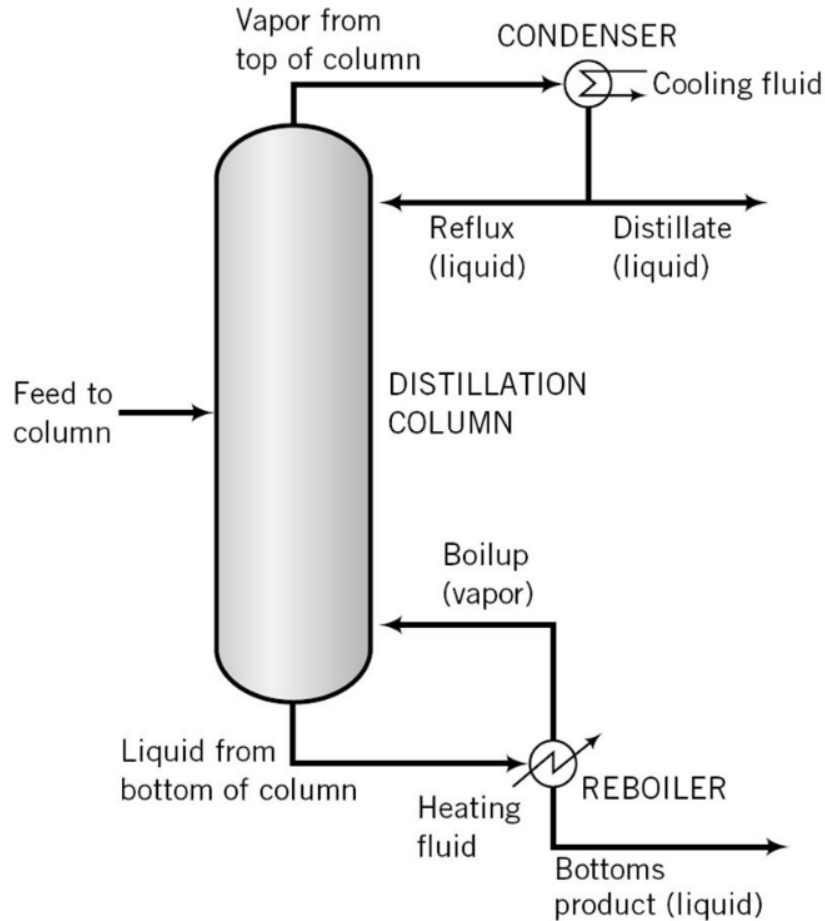
- Isothermal column
- Heat of absorption neglected
- Outlet temperature same for gas and liquid
- Only HCN is absorbed

Sensitivity analysis of the HCN Absorber



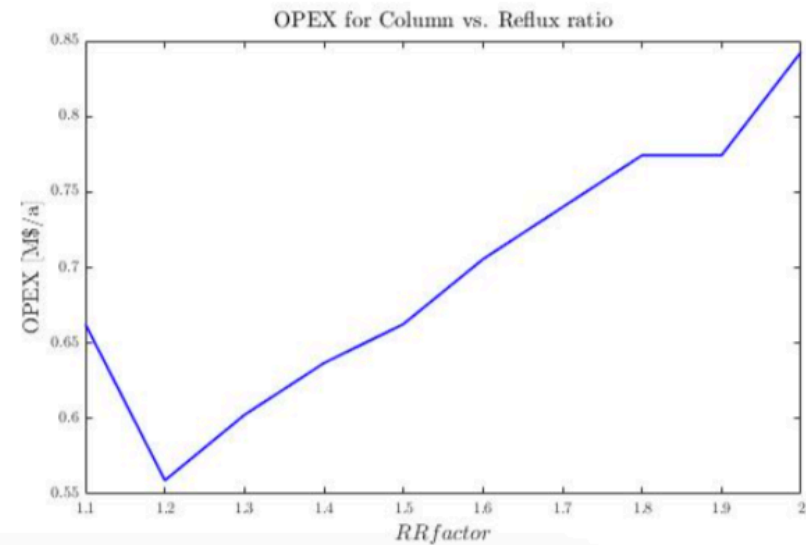
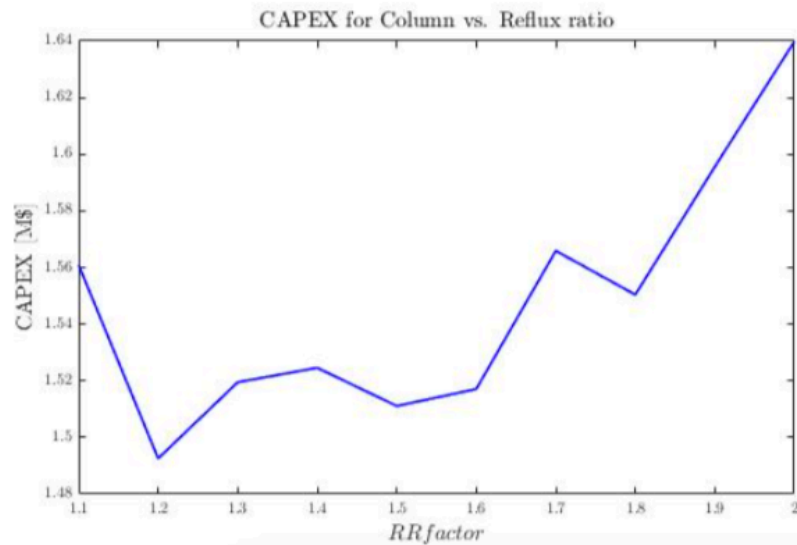
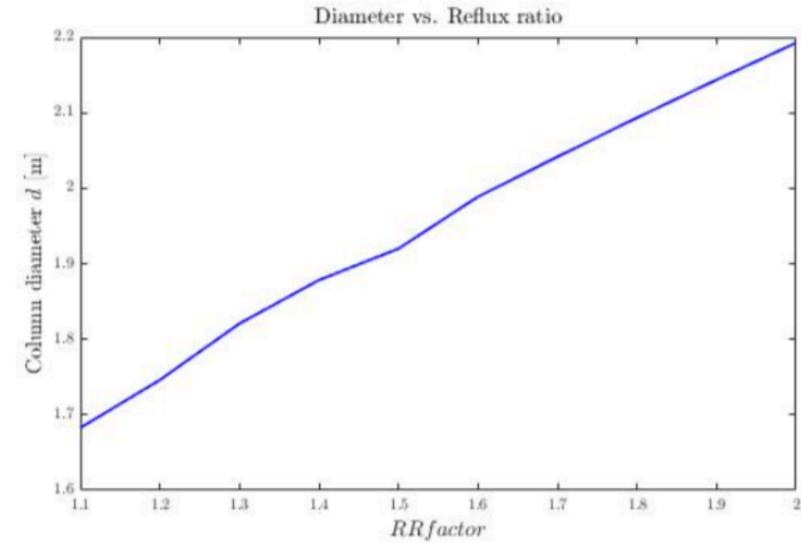
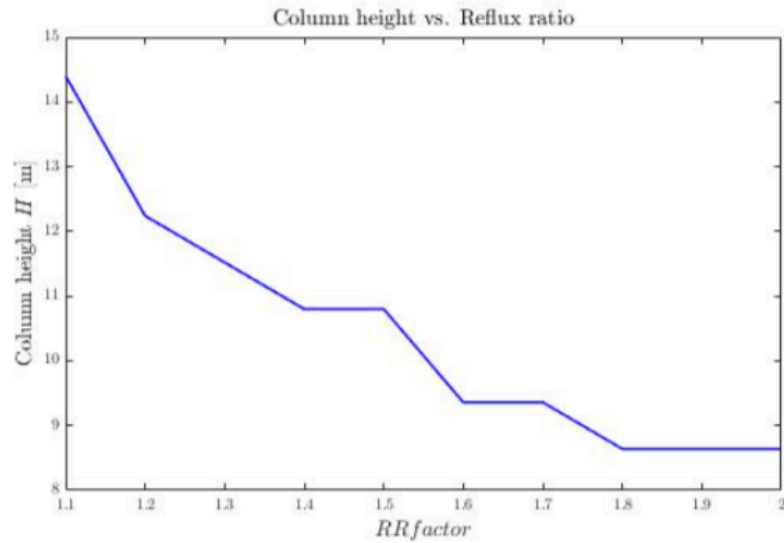
- Height of the column increases with decreasing inlet liquid flow
- Better absorption
- CAPEX decreases
- OPEX increases

HCN Distillation



- Ideal or non-ideal (NRTL for liquid phase)
- No $\Delta P / \Delta P$
- Fenske-Underwood-Gilliland method (justified since relative volatilities were pretty similar)
- $x_{HCN}^D = 0.995, x_{HCN}^B = 10 \text{ ppm}$ (going below these limits only makes the separation more difficult \rightarrow higher CAPEX/OPEX)
- total condenser, $q = 1$ (requires another heat exchanger before the distillation column)

HCN Distillation

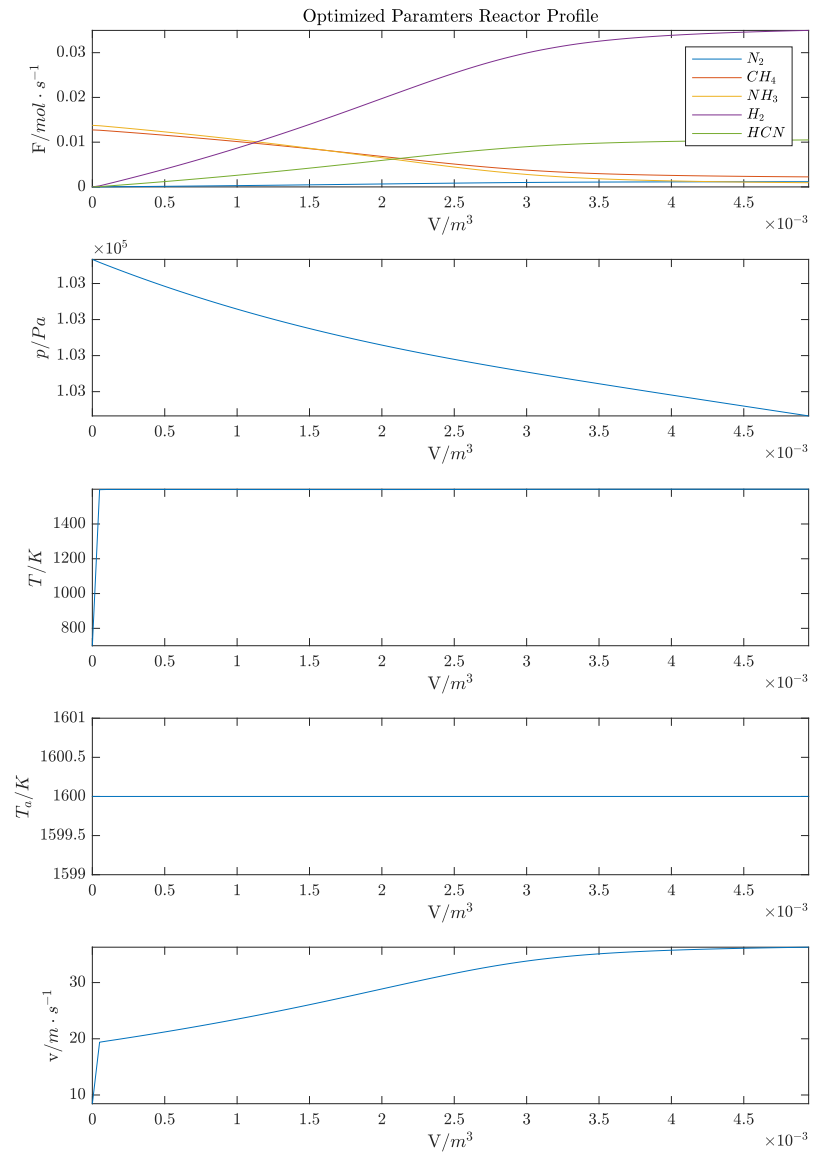


Results

Reactor:

- Ideal vs. real (case 1/case 3)
 - Length: 12 m / 28 m
 - Number of tubes: 1'342 / 17'149
- Cross- vs. co-current (case 3/case 4)
 - Number of tubes: 17'149 / 40'642
 - Heating medium flow: 0.10 / 0.05
 - Conversion CH_4 : 0.82 / 0.59

Reactor profile case 3



Results

NH₃ absorber:

- Ideal vs. real (case 1/case 3)
 - Column height: 2.44 m / 4.39 m
 - Diameter: 0.79 m / 0.79 m
 - Flow rate ratio 0.80 / 1.64
 - Temperature in the column
 - Volumetric fraction H₂SO₄ : 0.37 / 2.01

Results

HCN absorber:

- Ideal vs. real (case 1/case 3)
 - Column height: 9.73 m / 10.31 m
 - Diameter: 1.11 m / 1.10 m
 - Outlet temperature: 294.72 K / 293.57 K
 - Water

Results

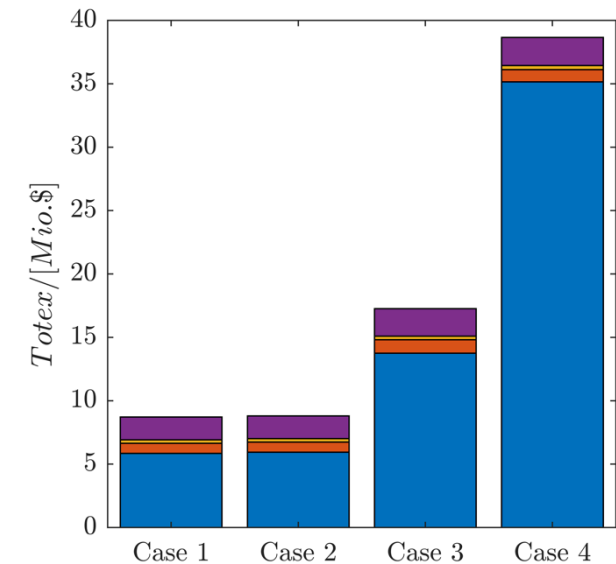
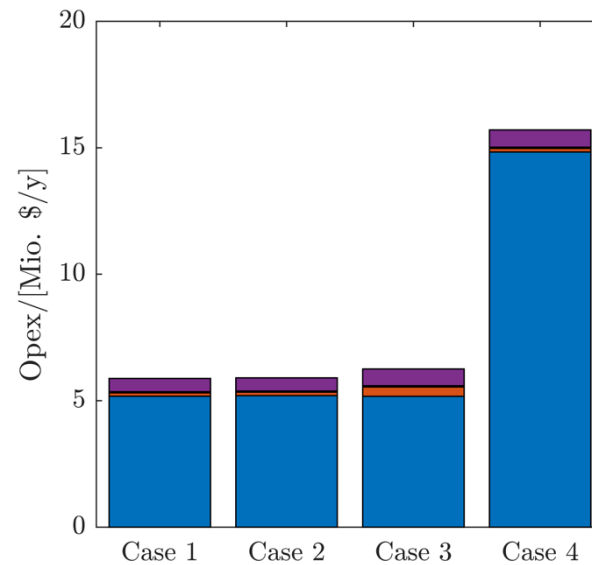
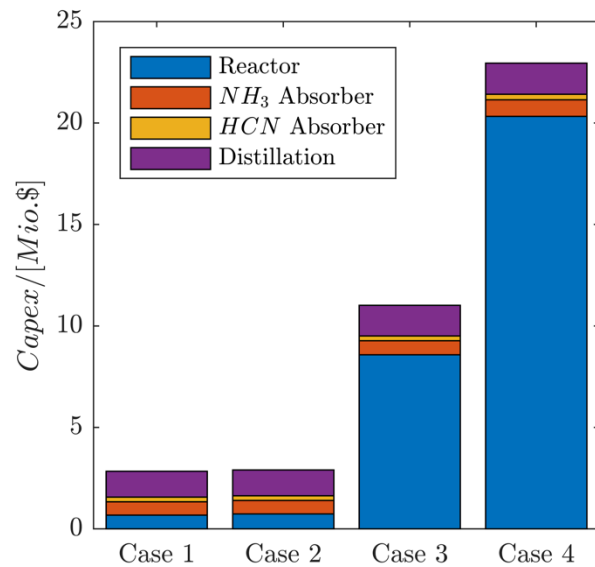
HCN distillation:

- Ideal vs. real (case 1/case 3)
 - Column height: 9.36 m / 10.08 m
 - Diameter: 1.73 m / 1.92 m
 - Heat exchanger area: 0.057 m² / 0.056 m²

Results

Costs

- Ideal vs. real (case 1/case 3)
 - Break-even price: 0.62 \$/kg / 0.76 \$/kg



Thank you for your attention

