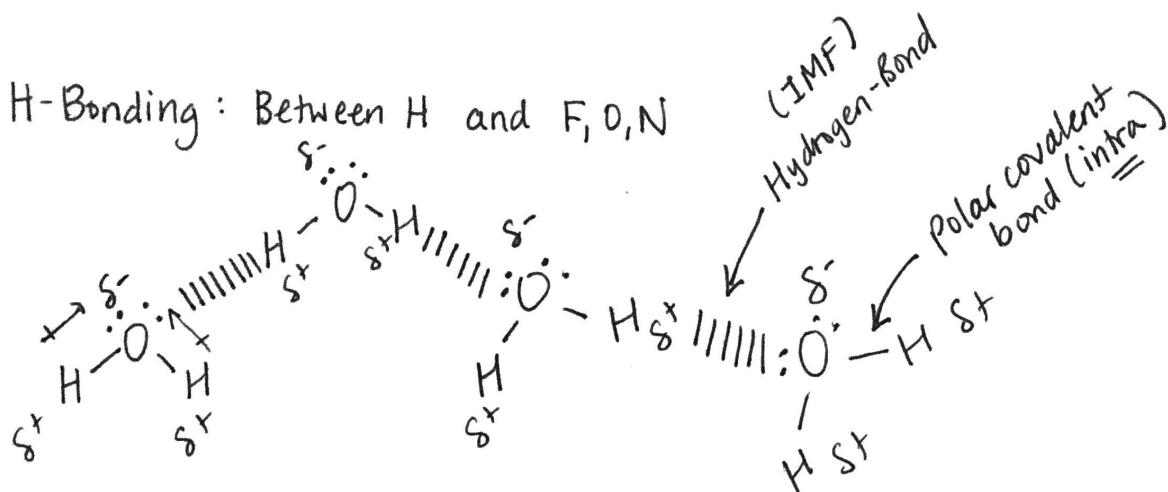


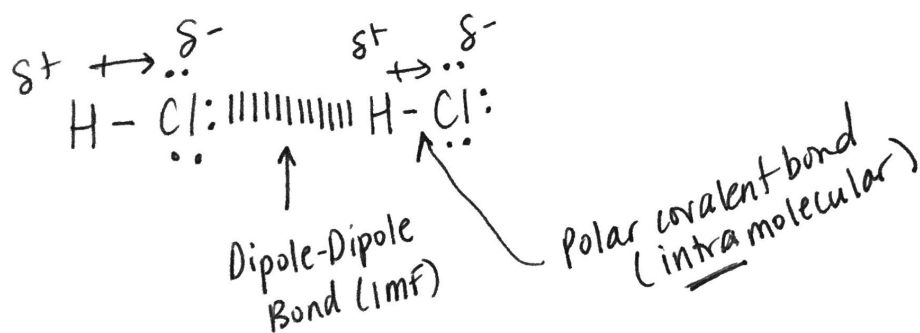
(Between)

Intermolecular Forces

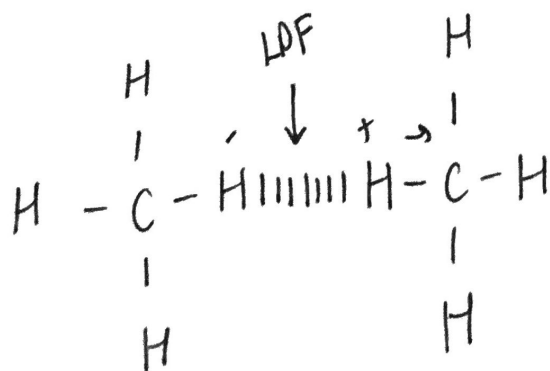
① H-Bonding: Between H and F, O, N



② Dipole-Dipole Bond: Between δ^- and δ^+



③ London Dispersion Force (LDF) OR : Between 2 non-polar molecules.
vanderwaals Force



in terms of strength: $\text{H-Bonds} > \text{Dipole-Dipole} > \text{LDFs}$
Strongest IMF Weakest IMF