Common Polyatomic Ions

1- .

C ₂ H ₃ O ₂ ¹⁻	Acetate	. 2	<u>.</u>
BrO ₃ ¹⁻	Bromate	CO ²⁻	Hypocarbonite
BrO_3^{-1}		CO ₃ ²⁻	Carbonate
	Bromite	CO ₄ ²⁻	Percarbonate
BrO ¹ -	Hypobromite	CrO ₄ ²⁻	Chromate
BrO ₄ ¹ -	Perbromate	Cr ₂ O ₇ ²⁻	Dichromate
ClO ₃ ¹ -	Chlorate	HPO ₄ ²⁻	Hydrogen phosphate (biphosphate)
ClO ₂ 1-	Chlorite	MnO ₄ ²⁻	Manganate
ClO 1-	Hypochlorite	$C_2 O_4^{2-}$	Oxalate
ClO ₄ ¹⁻	Perchlorate	SO ₄ ²⁻	Sulfate
CN 1-	Cyanide	SO ₃ ²⁻	Sulfite
$H_2PO_4^{1-}$	Dihydrogen phosphate	SO ₂ ²⁻	Hyposulfite
FO ₃ ¹ -	Fluorate	SO ₅ ²⁻	Persulfate
FO ₂ ¹⁻	Fluorite	$C_4H_4O_6^{2-}$	Tartrate
FO ¹⁻	Hypofluorite	$S_2O_3^{2-}$	Thiosulfate
FO ₄ ¹⁻	Perfluorate	SiO ₃ ² -	Silicate
HCO ₃ ¹⁻	Hydrogen carbonate (bicarbonate)	B ₄ O ₇ ²⁻	Tetraborate
HSO ₄ ¹⁻	Hydrogen sulfate (bisulfate)		Tetraborate
HSO ₃ ¹⁻	Hydrogen sulfite (bisulfite)	3-	
OH 1-	Hydroxide		
IO ₃ 1-	Iodate	BO ₃ ³⁻	Borate
IO ₂ 1-	Iodite	PO ₄ ³⁻	Phosphate
IO 1-	Hypoiodite	PO ₃ ³⁻	Phosphite
IO ₄ 1-	Periodate		
NO ₃ 1-	Nitrate	. 1+	<u> </u>
NO ₂ 1-	Nitrite	$NH_4^{\ 1+}$	Ammonium
NO^{1-}	Hyponitrite	H_3O^{1+}	Hydronium
NO ₄ 1-	Pernitrate		
MnO ₄ 1-	Permanganate		
SCN 1-	Thiocynate		

Common Monatomic Ions

Positive Ions (Cations)

Negative Ions (Anions)

•	<u>1</u> + .	<u>. 3+</u> .	•	1
Cs 1+	Cesium	Al ³⁺ Aluminum	Br ¹⁻	Bromide
Cu 1+	Copper (I) (cuprous)	Cr ³⁺ Chromium (III) (chromic)	Cl ¹⁻	Chloride
Au 1+	Gold (I)	Co ³⁺ Cobalt (III) (cobaltic)	F 1-	Fluoride
H 1+	Hydrogen	Ga ³⁺ Gallium	H 1-	Hydride
Li 1+	Lithium	Au ³⁺ Gold (III)	I 1-	Iodide
K 1+	Potassium	Fe ³⁺ Iron (III) (ferric)	N_3 1-	Azide *
Rb 1+	Rubidium	Mn ³⁺ Manganese (III) (manganic)		
Ag 1+	Silver *	Ni ³⁺ Nickel (III)		
Na 1+	Sodium	Ti ³⁺ Titanium (III) (titanous)		
	2+ .	. 4+ .		2
Ba ²⁺	Barium	Pb ⁴⁺ Lead (IV) (plumbic)	O 2-	Oxide
Be ²⁺	Beryllium	Sn ⁴⁺ Tin (IV) (stannic)	$O_2^{\ 2-}$	Peroxide *
Cd ²⁺	Cadmium *	Mn ⁴⁺ Manganese (IV)	S ²⁻	Sulfide
Ca ²⁺	Calcium	Ti ⁴⁺ Titanium (IV) (titanic)		
Cr ²⁺	Chromium (II) (chromous)			
Co ²⁺	Cobalt (II) (cobaltous)			
Cu ²⁺	Copper (II) (cupric)			
Fe ²⁺	Iron (II) (ferrous)			
Pb ²⁺	Lead (II) (plumbous)		•	3-
Mg ²⁺	Magnesium	<u>. 5+</u> .	N ³⁻	Nitride
Mn ²⁺	Manganese (II) (manganous)	Sb ⁵⁺ Antimony (V)	P ³⁻	Phosphide
Hg ₂ ²⁺	Mercury (I) (mercurous) *	As 5+ Arsenic (V)		
Hg ²⁺	Mercury (II) (mercuric) *			
Ni ²⁺	Nickel (II)			
Sr ²⁺	Strontium			
Sn ²⁺	Tin (II) (stannous)		<u>•</u>	4- <u>.</u>
Zn ²⁺	Zinc *		C 4-	Carbide