

Virtual Lab – Creating a Metal Activity Series

Complete the following doc using the link below.

The metal activity link is:

<https://teachchemistry.org/classroom-resources/metals-in-aqueous-solutions-simulation>

Activity 1 – Reaction of Magnesium, Zinc, Copper, and Silver

Observations:

Did Mg react with Zn^{2+} Yes, Cu^{2+} Yes, Ag^+ Yes?

Did Zn react with Mg^{2+} No, Cu^{2+} Yes, Ag^+ Yes?

Did Cu react with Mg^{2+} No, Zn^{2+} No, Ag^+ Yes

Did Ag react with Mg^{2+} No, Zn^{2+} No, Cu^{2+} No?

Based on these observations, create a metal activity series with the four metals, with the most reactive metal at the top.

Mg - Magnesium

Zn - Zinc

Cu - Copper

Ag - Silver

Activity 2 – Reaction of Iron, Tin, Lead, Nickel

Observations:

Did Fe react with Sn^{2+} Yes, Pb^{2+} Yes, Ni^{2+} Yes?

Did Sn react with Fe^{2+} No, Pb^{2+} Yes, Ni^{2+} No?

Did Pb react with Fe^{2+} No, Sn^{2+} No, Ni^{2+} No?

Did Ni react with Fe^{2+} No, Sn^{2+} Yes, Pb^{2+} Yes?

Based on these observations, create a metal activity series with the four metals, with the most reactive metal at the top.

Fe - Iron

Ni - Nickel

Sn - Tin

Pb - Lead

Activity 3 – Reaction of Zinc, Copper, Iron, Lead

Observations:

Did Zn react with Cu^{2+} Yes, Fe^{2+} Yes, Pb^{2+} Yes?

Did Cu react with Zn^{2+} No, Fe^{2+} No, Pb^{2+} No?

Did Fe react with Zn^{2+} No, Cu^{2+} Yes, Pb^{2+} Yes?

Did Pb react with Zn^{2+} No, Cu^{2+} Yes, Fe^{2+} No?

Based on these observations, create a metal activity series with the four metals, with the most reactive metal at the top.

Zn - Zinc

Fe - Iron

Pb - Lead

Cu - Copper

Create a metal activity series with the eight metals examined so far, with the most reactive metal at the top.

Magnesium (Mg)

Zinc (Zn)

Iron (Fe)

Nickel (Ni)

Tin (Sn)

Lead (Pb)

H⁺ - Hydrogen

Copper (Cu)

Silver (Ag)

Part 4 – Reaction of Tin, Zinc, Nickel, Copper, Lead, Iron with H⁺

Observations:

Did Sn react with H⁺ Yes?

Did Zn react with H⁺ Yes?

Did Ni react with H⁺ Yes?

Did Cu react with H⁺ No?

Did Pb react with H⁺ Yes?

Did Fe react with H⁺ Yes?

Place H⁺ in the appropriate spot in your activity series, based on the above observations.