# FYS3110 Oblig 3

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# 1 Problem 3

### 1.1 a)

The Hamiltonian of the system is:

$$\hat{H} = -g \sum_{i=0}^{N-1} \left( |i\rangle \langle i+1| + |i+1\rangle \langle i| \right) - V |0\rangle \langle 0|$$
 (1)

We set  ${\cal N}=4$  and write down the matrix representation of the Hamiltonian

in the basis where 
$$\langle 0| \simeq \begin{pmatrix} 1\\0\\0\\0 \end{pmatrix}$$
 and  $\langle 1| \simeq \begin{pmatrix} 0\\1\\0\\0 \end{pmatrix}$  etc.

First we write it out with our value for N

$$\hat{H} = -g \sum_{i=0}^{3} \left( |i\rangle \langle i+1| + |i+1\rangle \langle i| \right) - V |0\rangle \langle 0|$$
 (2)

$$\hat{H} = -g\Big(\left|0\right\rangle\left\langle1\right| + \left|1\right\rangle\left\langle0\right| + \left|1\right\rangle\left\langle2\right| + \left|2\right\rangle\left\langle1\right| + \left|2\right\rangle\left\langle3\right| + \left|3\right\rangle\left\langle2\right| + \left|3\right\rangle\left\langle4\right| + \left|4\right\rangle\left\langle3\right|\Big) - V\left|0\right\rangle\left\langle0\right| + \left|4\right\rangle\left\langle1\right| + \left|$$

Writing this out using the basis vectors from earlier we get:

Which gives us:

$$\hat{H} = -g \begin{pmatrix} V/g & 1 & 0 & 0 \\ 1 & 0 & 1 & 0 \\ 0 & 1 & 0 & 1 \\ 0 & 0 & 1 & 0 \end{pmatrix}$$
 (5)

#### 1.2 b)

For this problem we need to find a matrix representing the position operator  $\hat{X}$ . We know that the eigenvalues represent the atom at which our electron is localized. We choose N=4 giving us the eigenvalues  $\Lambda=0,1,2,3$ . From this we can calculate our matrix using the orthonormality of the eigenkets.

$$\langle 0|\hat{X}|0\rangle = 0, \ \langle 1|\hat{X}|1\rangle = 1, \ \langle 2|\hat{X}|2\rangle = 2, \ \langle 3|\hat{X}|3\rangle = 3$$
 (6)

from  $\left\langle i\right|\left|j\right\rangle =1$  if i=j and  $\left\langle i\right|\left|j\right\rangle =0$  if  $i\neq j$  This results i the matrix

$$\hat{X} = \begin{pmatrix} 0 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 \\ 0 & 0 & 2 & 0 \\ 0 & 0 & 0 & 3 \end{pmatrix} \tag{7}$$

#### 1.3 c)

print eigvals(H)

For this exercise we need to numerically generate our matrix using the given values of N=32, g=1 and V=4 with our expression for the Hamiltonian (eq.(1)).

```
from numpy import *
from numpy.linalg import *
from matplotlib.pyplot import *
g = 1
V = 4
N = 32
basis = []
for i in range(N):
        b = zeros(N)
        b[i] = 1
        b = matrix(b).T
        basis.append(b)
H = zeros((N,N))
for i in range (N-1):
        H = H - g*(basis[i]*basis[i+1].T+basis[i+1]*basis[i].T) #summing values 0 to M
H = H - g*(basis[-1]*basis[0].T+basis[0]*basis[-1].T) #Making sure i+1=N=0
H = H-V*(basis[0]*basis[0].T) #Adding potential
plot(range(N), sort(eigvals(H)))
title('Eigenvalue diagram')
ylabel('Value eigenvalue')
show()
print H
```

print "The lowest eigenvalue is Lambda= %.3f" % min(eigvals(H))

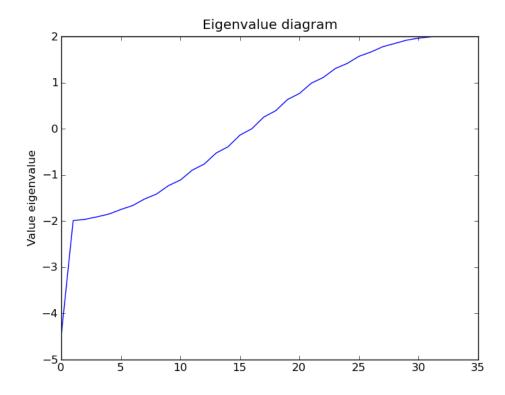


Figure 1: Eigenvalues from min to max

From figure 1 we can gather that the lowest energy is equal to the lowest eigenvalue  $\Lambda = -4.472$  as well as from the code printout:

```
-1.
            0.
                      0.
                          0.
 [-1.
       0.
           -1.
                      0.
                          0.
                               0.]
 [ 0.
            0.
                      0.
 [ 0.
       0.
            0.
                      0.
                         -1.
                               0.]
                          0. -1.]
 [ 0.
       0.
            0.
                    -1.
               . . . ,
                               0.]]
       0.
            0. ...,
                      0. -1.
[-4.47213595e+00]
                                         1.99094261e+00
                                                            1.91888827e+00
                     -1.98974034 e + 00
   1.77695753 e + 00
                     -1.90841083e+00
                                        -1.74964538e+00
                                                            1.56957610e+00
  -1.52082662e+00
                      1.30348176e+00
                                        -1.23218869e+00
                                                            9.87690606e-01
  -8.96133295e-01
                      6.33313168e-01
                                        -5.26587138e-01
                                                            2.53221011e-01
  -1.38402808e-01
                      1.96157056e+00
                                        -1.96157056e+00
                                                            1.84775907e+00
   1.66293922e+00
                     -1.84775907e+00
                                         1.41421356 e + 00
                                                           -1.66293922e+00
                     -1.11114047 e + 00
  -1.41421356e+00
                                         1.11114047 e + 00
                                                           -7.65366865e-01
   7.65366865 e - 01
                      2.10317280e-17
                                         3.90180644 e - 01
                                                           -3.90180644 e -01
The lowest eigenvalue is Lambda = -4.472
```

#### 1.4 d)

For this exercise we need to calculate the probability of finding the electron when N=0 and N=1. This is given by  $c_n = \langle f_n | \Psi \rangle$  Our probability is given by  $|c_n|^2$  which follows from the normalization of the wave function. Finding these probabilities using python and our previous script is done by adding:

```
eigenvalues, eigenvectors = eig(H)
print abs(eigenvectors[0,0])**2
print abs(eigenvectors[1,0])**2
Which prints:
0.894427191
0.0498447189992
```

So the probabilites corresponding to atom N=0 and N=1 are 89.44% and 4.98%

#### 1.5 e)

To find the ground state expectation value  $\langle \psi | \hat{X} | \psi \rangle$  we need to find the sum over all possible outcomes of the eigenvalue times the probability of getting that eigenvalue.

```
X = diag(range(N))
eigenvalues, eigenvectors = eig(H)
print eigenvectors[0]*X*eigenvectors[0].T
```

This prints our expectation value: [[ 1.20804923]]