



NATIONAL OPEN UNIVERSITY OF NIGERIA
14-16 AHMADU BELLO WAY, VICTORIA ISLAND LAGOS
MARCH/APRIL 2016 EXAMINATION

SCHOOL OF SCIENCE AND TECHNOLOGY

COURSE CODE:
COURSE TITLE:

CHM408
POLYMER CHEMISTRY

Duration: **2hrs**

Answer any four questions (each question carries a total $17\frac{1}{2}$ marks)

Question 1

- 1a) List any four initiators in an anionic polymerization (4 marks)
- b) Give the structure of the most commercially used initiator in an anionic polymerization (2 ½ marks)
- c) State with relevant equations the four ways by which initiation process of an ionic polymerisation can occur. (4 marks)
- d) List and explain the benefits of co-polymerization (7 marks)

Question 2

- a) Define polymer degradation and enumerate three broad classes of polymer degradation based on the agents causing degradation. (5 marks)
- b) Enumerate five agents of degradation and one each likely susceptible polymers.
- c) Write short notes on the following: (5 marks)
 - i) monomer
 - ii) homopolymer
 - iii) copolymer

Question 3

- a) What is meant by polyurethane? With equation show how polyurethane can be synthesized (5 ½ marks)
- b) State any five properties of polyurethane
- c) Explain briefly seven (7) advantages of polyurethane (7 marks)

Question 4

- a) Explain how stretching of rubber affects entropy. (5 marks)
- b) Discuss any seven (7) properties of thermoplastics which distinguish them from thermosetting polymers (7 marks)

- c) With the aid of example (and equations) describe the mechanism of condensation polymerization. (5 $\frac{1}{2}$ marks)

Question 5

Discuss any five instrumental techniques used for identification and characterization of polymers.

Question 6

- a) Describe the Industrial Important of any five Thermoplastics Polymers. (12 $\frac{1}{2}$ marks)
- b) Give the thermodynamic equation used for describing a given system defining each term (2 $\frac{1}{2}$ marks).
- c) What is the implication of negative Gibbs free energy change in solution process? (1 $\frac{1}{2}$ mark)
- d) What will be the effect of ΔH_m is positive, then $\Delta H_m < T\Delta S_m$ on solution process of polymer? (1 mark)