

## SCHOOL OF SCIENCE AND TECHNOLOGY OCTOBER 2013 EXAMINATION

Course Code: CHM 302 Course Title: Polymer Chemistry 1 Time Allowed: 2 Hours		
	RUCTION: Answer any four Questions	
1. a)	i. Discuss each of the following: Elastomers, Fibers, Thermosetting.	
	(7½ marks) ii. Give 5 differences between addition and condensation polymerization.	
b) 2.	(5 marks) Name the various types of synthetic rubber. marks)	(5
	1 2	(2½
b)	marks) i. Classify and explain briefly polymers based on their stereochemistry. (5 marks)	
	<ul><li>ii. Outline the established methods of producing polyethylene.</li><li>(5 marks)</li></ul>	
c)	Discuss the Effect of polarity on solubility of Polymers. (5 marks)	
3. a)	Discuss the 3 basic mechanisms by which addition- chain growth polymerization can occur. (6 marks)	
	at are condensation polymers and give 3 examples.  Discuss the effect of temperature and pressure respectively on the solubility of polymers. (6 marks)	ks)
of the	e structures of polymers and their functions depend on the following factors. List all esefactors. (2½marks)	
4. a)	List and explain briefly the different types copolymers based on the arrangements of the	f
b)	two monomers. (6 mar Explain the mechanism of the Ziegler-Natta process for substituted ethylenes. (3 marks)	ks)
c)	Discuss on the Applications of polymer solubility.	(4½
d)	marks) On which factors does the solubility of polymers depend? marks)	(4
5.		

(10

 $(7\frac{1}{2})$ 

a) Discuss on the following properties of polymers:

Glass transition temperature

Tensile strength

Young's modulus

Melting point

b) Explain the following:

marks)

i.

ii.

iii.

marks)

- i. Recycling of polymers
- ii. Chlorine attack (acetal resin plumbing joints)
- iii. Biodegradations in polymers

6.

- a) Describe briefly the 3 stages by which addition chain growth polymerization mechanisms occur. (6 marks)
- b) Differentiate between natural and synthetic rubber. (4 marks)
- Mention any 4 guidelines to be taken into cognizance in designing synthetic polymerization reactions to produce the desired products. marks)
- d) Explain the Ozone cracking in natural rubber tubing. (3½marks)

(4