

NATIONAL OPEN UNVERSITY OF NIGERIA 14/15 AHMADU BELLO WAY, VICTORIA ISLAND, LAGOS

SCHOOL OF SCIENCE AND TECHNOLOGY OCTOBER 2013 EXAMINATION

Course Code: CHM307

Course Title: ATOMIC AND MOLECULAR STRUCTURE AND SYMMETRY INSTRUCTION: ATTEMPT ANY FIVE QUESTIONS TIME: 2½ HOURS

1. (a) Highlight the conditions for the formation of symmetry point group. (5 marks)

(b) List the symmetry elements that a molecule can possess. (4 marks)

(c) With respect to symmetry, briefly write on the following;

(i) Mirror plane (ii) Inversion

(5 marks)

2. (a) Show the steps required to determine point groups for different molecules. (4 marks)

(b) List the symmetry elements of the following molecules:

(i) BCl₃(ii) NH₃(iii) BF₃

(6 marks)

(c)Discuss the implication of symmetry on the properties of molecules.(4 marks)

3. (a)What is an electronic spectrum?

(4 marks)

(b) Discuss briefly the Franck-Condon Principle.

(6 marks)

(c)Explainthe selection rules for a harmonic oscillator.

(4 marks)

4. (a) Write a short note on Vibration Spectroscopy.

(6 marks)

(b) What are internal coordinates?

(4 marks)

(c) What are the uses of microscope spectroscopy?

(4 marks)

5. (a) Write a short note onRotational spectroscopy.

(6 marks)

(b) Discuss the classes molecules based on rotational behavior.

(8 marks)

6. (a) Highlight the steps to writing resonance structures; and show resonance in Ozone and Benzene. (7 marks)

(b) What is Nuclear Coupling?(3 marks)

(c) State the usefulness of Quantum Mechanics.(4 marks)

7. (a) What is bond order?(3 marks)

- (b) State the relationship between bond order, bond length and bond strength. (3 marks)
- (c) What information could be obtained from bond order?

(3 marks)

(d) The ground state electron configuration of N_2 , with even valence electrons is $1\sigma_g^2 1\sigma_u^2 1\pi_u^4 2\sigma_g^2$.

(i) Calculate the bond order (ii) Is N₂ diamagnetic or paramagnetic?

 $(1\frac{1}{2} \text{ each}; = 3 \text{ marks})$

(iii) Is N₂stable?give reasons.

(2 marks)