

NATIONAL OPEN UNIVERSITY OF NIGERIA 14/16 AHMADU BELLO WAY, VICTORIA ISLAND, LAGOS SCHOOL OF SCIENCE AND TECHNOLOGY MARCH/APRIL 2014 EXAMINATION

COURSE CODE: CHM 307

COURSE TITLE: ATOMIC AND MOLECULAR SPECTRUM AND SYMMETRY

TIME ALLOWED: 2 1/2 HOURS

INSTRUCTION: ANSWER ANY FIVE QUESTIONS

- 1a) Write short note on electron shell.
- 1b) State the reason behind the electronic configuration of Lithium being written as $1s^22s^1$.
- 1c) DefineMadelung's rule and explains its contribution to electronic configuration of transition

metals.

1d) The Aufbau principle rests on a fundamental postulate that the order of orbital energies is fixed,

both for a given element and between different elements: neither of this is true discuss.

- 2a) Each electron in an atom is described by four quantum number, discuss.
- 2b) Write out the ground state electron configurations in the orbital box notation showing electron

spin for

- i) Chlorine
- ii) Manganese
- 2c) Highlight the appropriate combinations of orbital to form molecular orbital.
- 2d) Explain the magnetic behavior of oxygen molecule in relation to Molecular Orbital theory.
- 3a)State the fundamental Thermodynamic formula
- 3b) What is the effect, if density in the formula in 3a above is changed to molar density in an ideal

gas.
3c) Derive Heat Capacity using quantum mechanics.
3d) List two examples each of the below
 a) lons having sp² hybrid orbitals. b) lons having sp³ hybrid orbitals.
4a) Write on dsp³ Hybrid orbital
4b) write the shape of compounds having the below hybrid orbital;
 i) Sp ii) sp² iii) sp³ iv) dsp³ v) d²sp³
4c) Draw the energy levels in a Hydrogen Molecule.
4d) Highlight the steps to writing resonance.
5a) Show the molecular orbital diagram for a diatomic Oxygen molecule.5b) What is the wavelength of a 100Ev
6a) Define Rotational Spectroscopy
6b) Classify molecules based on Rotational behaviour
6c) Explain Centrifugal distortion.
7a) Enumerate the order of the effect of interaction responsible for small changes in spectra.
7b) What is Symmetry?
7c) Discuss the symmetry elements.