

NATIONAL OPEN UNIVERSITY OF NIGERIA UNIVERSITY VILLAGE, PLOT 91 CADASTRAL ZONE, NNAMDI AZIKIWE EXPRESS WAY, JABI - ABUJA. FACULTY OF SCIENCES DEPARTMENT OF PURE AND APPLIED SCIENCES FEBRUARY/MARCH2018 EXAMINATION

COURSE CODE: CHM 306

COURSE TITLE: INSTRUMENTAL METHODS OF ANALYSIS

TIME: 2 HOURS

INSTRUCTION: Question one is compulsory. Answer question one and

any other three questions.

QUESTIONONE

- 1a) Calculate the concentration of a sample solution whose absorbance and molar absorptivity at 270nm is 1.92 and 19400 respectively. 4 marks
- 1b) What happens when radiation and matter interact?6 marks
- 1ci) Discuss the basic concept of X-ray diffraction method.7 marks
- 1cii)Explain briefly Polarography.4 marks
- 1dii) State one use of each of the following
 - I. Infrared spectroscopy
 - II. X- ray diffraction method
 - III. Flame Emission and Flame Atomic Absorption Spectroscopy
 - IV. Nuclear Magnetic Resonance Spectroscopy 4 marks

QUESTION TWO

2ai Describe briefly the basic principle of Nuclear Magnetic Resonance (NMR) spectroscopy.

7marks.

- 2aii) What factor accounts for the difference, in the pattern of NMR spectrum of hydrogens in different organic molecules.4 marks
 - 2b) Enumerate on the function of the parts of a spectrophotometer.4 marks

QUESTION THREE

Explain how the concentration of a coloured sample can be estimated by colorimetry. 15 marks

QUESTION FOUR

- 4ai) What is infrared spectroscopy? $1^{1}/_{2}$ marks
- 4aii) How would you determine the functional groups present in an organic molecule using infrared spectroscopy? 8 marks
- 4b) Distinguish between Infrared spectrometer and Fourier Transformer Infrared spectrometer.
- $4^{1}/_{2}$ marks

QUESTION FIVE

- 5a) With the aid of a well labelled schematic diagram, expatiate on the working principle of Flame Atomic Absorption Spetroscopy.10¹/2 marks.
- 5b Distinguish between the following terms used in Flame Atomic Absorption Spectroscopy
 - i. Interference
 - ii. Sensitivity
 - iii. Detection Limit4¹/₂ marks