Question FBQ1 : A real gas behaves as an ideal gas when pressure approaches

Answer: zero

Question FBQ2 : What is a device that maintains a body at a temperature lower than the temperature of the surroundings

Answer: Refrigerator

Question FBQ3 : Nitrogen boils at -196 ^oC. What is this temperature

on the absolute scale?

Answer: 77K

Question FBQ4 : Work done in a free expansion process is equal to_____.

Answer: Zero

Question FBQ5 : The degree of hotness or coldness of a body is called_____.

Answer: Temperature

Question FBQ6 : When there is a flow of heat into a system, heat is said to be

Answer: Positive

Question FBQ7 : A Thermometer is named after the _____ property used for its

construction

Answer: Thermometric

Question FBQ8 : The instrument for measuring the temperature of a body is known

as____.

Answer: Thermometer

 ${\tt Question} \ {\tt FBQ9} \ : \ {\tt Electrical} \ {\tt conductivity} \ {\tt of} \ {\tt a} \ {\tt metal} \ {\tt depends} \ {\tt on} \ {\tt the} \ {\tt movement} \ {\tt of}$

_____ through its crystal lattice.

Answer: Electrons

Question FBQ10 : The relationship between the temperature and electrical

resistance is usually _____.

Answer: Non-linear

Question FBQ11 : The transfer of heat through solids is known as_____.

Answer: Conduction

Question FBQ12 : The transfer of heat through fluids is known as_____.

Answer: Convection

Question FBQ13 : When the thermal resistance of a slab is a high value, it

indicates a ____insulation

Answer: good thermal

Question FBQ14 : The process in which energy is transferred by means of

electromagnetic waves is known as_____.

Answer: Radiation

Question FBQ15 : If a fluid expands suddenly into vacuum through an orifice of

large dimension, then such a process is called_____ expansion.

Answer: free

Question FBQ16: The equation is

known as the____law. Answer: Stefan-Boltzmann

Question FBQ17: If x = zy find the partial derivative of x with respect to y.

Answer: z

Question FBQ18: The rate of energy loss per second that is proportional to the

as Answer: Newton's law of cooling
Question FBQ19 : The heating and expanding of a gas is called Answer: Thermodynamic cycle
Question FBQ20 : When work is being performed by a system it is said to be awork. Answer: Negative
Question FBQ21 : The transfer of heat between two bodies in direct contact is called Answer: Conduction
Question FBQ22 : The equation dQ=dE+dW holds good for process Answer: reversible
Question FBQ23 : The entropy always remains constant for a process that occurs in a closed system. Answer: Reversible
Question FBQ24 : Whose statement is this? ''No process is possible whose sole results is to the complete conversion of heat into work''. Answer: Kelvin's
Question FBQ25 : The most efficient heat engine is known as engine Answer: Carnot
Question FBQ26 : Total heat of a substance is also known as Answer: Enthalpy
Question FBQ27 : The Cycle consists of two adiabatic processes and two constant volume processes. Answer: Otto
Question FBQ28 : The purpose of the displacer mechanism is to move the working gas between the hot and cold ends of the machine through the Answer: Regenerator
Question FBQ29 : The is a device that uses work to transfer energy from a low temperature reservoir to a high temperature reservoir. Answer: Refrigerator
Question FBQ30 : When a carnot engine is reversed it is called a Answer: carnot refrigerator
Question FBQ31: The process in which the entropy of the system increases for irreversible process and remains unchanged for reversible process is law of thermodynamics. Answer: Second
Question FBQ32 : All gases and vapours approach ideal gas behaviour at pressure and low density Answer: low
Question FBQ33 : Water contained in a beaker can be made to boil by passing steam through it at a pressure greater than Answer: atmospheric pressure
Question FBQ34: If a system undergoes a reversible isothermal process without transfer of heat, the temperature at which this process takes place is called zero. Answer: Absolute

Question FBQ35 : The efficiency of all reversibleengines operating between the same heat reservoirs is Independent of the amount or nature of working substance Answer: heat
Question MCQ1 : A series of operations, which take place in a certain order and restore the initial condition, is known as $___$. Answer: Thermodynamic cycle
Question MCQ2 : In an irreversible process, there is a Answer: Loss of heat
Question MCQ3 : Which of the following is correct? Answer: Absolute pressure = gauge pressure + atmospheric pressure
Question MCQ4 : One watt is equal to Answer: 1 Nm/s
Question MCQ5 : An ideal gas as compared to a real gas at very high pressure occupies Answer: More volume
Question MCQ6 : When there is a flow of heat into a system heat is said to be Answer: Positive
Question MCQ7 : A Thermometer is named after the property used for its construction Answer: Thermometric
Question MCQ8 : The instrument for measuring the temperature of a body is known as $\underline{\hspace{1cm}}$. Answer: Thermometer
Question MCQ9 : Electrical conductivity of a metal depends on the movement of through its crystal lattice Answer: Electrons
Question MCQ10 : The relationship between the temperature and electrical resistance is usually $___$. Answer: Non-linear
Question MCQ11 : The transfer of heat through solids is known as Answer: Conduction
Question MCQ12 : The transfer of heat through fluids is known as Answer: Convection
Question MCQ13 : When the thermal resistance of a slab is a high value, it indicates ainsulation. Answer: Good thermal
Question MCQ14 : All are examples of Intensive properties except : Answer: Volume
Question MCQ15 : A system comprising a single phase is called a Answer: Homogeneous system
Question MCQ16 : The equation Q = $E6T < sup > 4 < / sup > At is known as thelaw$. Answer: Stefan-Boltzmann's
Question MCQ17: The ratio of radiant energy emitted by an object to the one it would have emitted if it were to be a perfect blackbody is known as

Question MCQ18: The rate of energy loss per second that is proportional to the temperature difference between the object and its surroundings is known as____ Answer: Newton's law of cooling Question MCQ19: When work is being performed on a system it is said to be a____work. Answer: Positive Question MCQ20 : The efficiency of carnot cycle may be increased by_____. Answer: Decreasing the highest temperature Question MCQ21: The work done in a system of wires is equal to_____. Answer: Tension X Length Question MCQ22: Water at standard atmospheric conditions_____. Answer: Is far below its critical state Question MCQ23: The entropy always remains constant for a _____ process that occurs in a closed system. Answer: Reversible Question MCQ24: No process is possible whose sole results is to the complete conversion of heat into work' is known as_____ Statement. Answer: Kelvin's Question MCQ25: The most efficient heat engine is known as_____ engine. Answer: Carnot Question MCQ26 : A _____ engine is a device used to convert thermal energy into mechanical work. Answer: Heat Question MCQ27 : The _____ Cycle consists of two adiabatic processes and two constant volume processes. Answer: Otto Question MCQ28: The purpose of the displacer mechanism is to move the working gas between the hot and cold ends of the machine through the_ Answer: Regenerator Question MCQ29 : The ____ is a device that uses work to transfer energy from a low temperature reservoir to a high temperature reservoir. Answer: Refrigerator Question MCQ30 : Change in internal energy in a closed system is equal to heat transferred if the reversible process takes place at constant Answer: Volume Question MCQ31: The temperature of an object that emits electromagnetic radiation must be higher than 0^oC converted from one form to other inferred from Answer: First law of thermodynamics Question MCQ32: Which of the following parameters is constant for a mole for most of the gases at a given temperature and pressure? Answer: Volume Question MCQ33 : The term N.T.P. stands for _____. Answer: Normal temperature and pressure Question MCQ34: The Celcius equivalent of 200 K is ____.

Answer: Emissivity

Answer: Negative 73^oC

Question MCQ35 : What is the standard fixed point of thermometry? Answer: The triple point of water $% \left(1\right) =\left(1\right) +\left(1\right)$