

NATIONAL OPEN UNIVERSITY OF NIGERIA 14-16 AHMADU BELLO WAY, VICTORIA ISLAND LAGOS MARCH/APRIL 2016 EXAMINATION

SCHOOL OF SCIENCE AND TECHNOLOGY

COURSE CODE: CHM417 CREDIT UNIT: 2

COURSE TITLE: INDUSTRIAL CHEMICAL PROCESSES II

DURATION: 2 HRS

INSTRUCTION: Answer any 4 questions

- 1. (a) What are minerals? 9mks
 - (b) List the physical properties of minerals. 4mks
 - (c) Using Sphalerite as an example, what does heating and roasting involve in mineral processing? $6\,\%$ MKS
- 2. (a) Enumerate the types of furnaces used in metallurgical operations and identify their uses. 10mks
 - (b) What constitutes an ore gangue or matrix? 1 ½ mks
- (c) With the aid of chemical equations, describe what happens in the blast furnace for the extraction of iron. 6mks
- 3. (a) Metals can be extracted by a universal method. How true is this statement? 4mks
 - (b) Explain in details the calcination process in the extraction of crude metal. 13 ½ mks
- 4. (a) Identify the two distinct steps in hydrometallurgy. 5mks
 - (b) Discuss briefly the characterization of cements. 12 ½ mks
- 5. (a) How is phosphoric acid produced? 11mks
- (b) List (with chemical equations where necessary) the methods of preparing potassium sulphate. $6 \frac{1}{2}$ mks
- 6. (a) Show clearly in a schematic diagram, the membrane cell process. 12 ½ mks
 - (b) State 2 advantages and 2 demerits of the above process. 5mks