

## ***Section 71 Ionic And Metallic Bonding Answers***

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**Section 71 Ionic And Metallic**

Section 7.3 – Bonding in Metals. The valence electrons of metal atoms can be modeled as a sea of electrons. Metallic bonds consist of the attraction of the free-floating valence electrons for the positively charged metal ions. Metals are good conductors and malleable because of their mobile electrons.

**Chapter 7 - Ionic and Metallic Bonding**

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Chapter 7 ionic and metallic bonding, section 73 bonding in metals the valence electrons of metal atoms can be modeled as a sea of electrons metallic bonds ... Section 71 ions chapter 7 “ionic and metallic bonding”, nonmetals will have many valence electrons (usually 5 or more) they will gain ...

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Essential Understanding Ionic compounds are the result of ionic bonds forming between oppositely charged ions. Lesson Summary Formation of Ionic Compounds An ionic compound is made up of anions and cations and has an overall charge of 0. The electrostatic attraction between an anion and a cation is an ionic bond.

**BONDING AND INTERACTIONS**

1 Chapter 7 “Ionic and Metallic Bonding” Click to add text 2 Section 7.1 - Ions

OBJECTIVES:-Determine the number of valence electrons in an atom of a representative element.

**Section 7.1 - Ions Chapter 7 “Ionic and Metallic Bonding”**

Chapter 7 Ionic and Metallic Bonding59 SECTION 7.1 IONS (pages 187–193) This section explains how to use the periodic table to infer the number of valence electrons in an atom and draw its electron dot structure.

**SECTION 7.1 IONS (pages 187–193)**

206 Chapter 7 • Ionic Compounds and Metals Section 7.1.1 Figure 7.1 As carbon dioxide dis-solves in ocean water, carbonate ions are produced. Coral polyps capture these carbonate ions, producing crystals of calcium carbonate, which they secrete as an exo-skeleton. Over time, the coral reef forms. A coral reef is a complex habitat that

**Chapter 7: Ionic Compounds and Metals**

Ionic Bonding Anions and cations are held together by opposite charges. Ionic compounds are called salts. Simplest ratio of elements in an ionic compound is called the formula unit. The bond is formed through the transfer of electrons. Electrons are transferred to achieve noble gas configuration.

### Chapter 7 Ionic and Metallic Bonding

Chapter 7 "Ionic and Metallic Bonding ... Section 7.2 Ionic Bonds and Ionic Compounds. OBJECTIVES: Explain the electrical charge of an ionic compound. Section 7.2 Ionic Bonds and Ionic Compounds. OBJECTIVES: Describe three properties of ionic compounds. Ionic Bonding. Anions and cations are held together by .

## Section 71 Ionic And Metallic Bonding Answers

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