

Solubility Temperature Graphs Chapter 14 Answers

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on the solubility of NaCl. 8. Explain how you might make a solution containing 42 g KCl dissolved in 100 g H₂O at a temperature of 40°C. What term describes this type of solution?

Solubility–Temperature Graphs TEACHING TRANSPARENCY WORKSHEET Use with Chapter 14, Section 14.3 42 Substance Solubility at 10°C Calcium chloride (CaCl₂)

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Solubility Temperature Graphs Chemistry Matter And Change ...

Solubility graphs represent the relationship between solubility (in grams of solid per volume of water) vs temperature. If the solution is above the solubility line it is supersaturated and below the solubility line it is unsaturated. Points along the line are points of saturation.

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This graph shows how the solubility of several solids changes with temperature. The temperature dependence of solubility can be exploited to prepare supersaturated solutions of certain compounds. A solution may be saturated with the compound at an elevated temperature (where the solute is more soluble) and subsequently cooled to a lower temperature without precipitating the solute.

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