

## *Sodium Azide Stock Solution*

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### **Sodium Azide Stock Solution**

sodium azide. To make a 10% stock solution of sodium azide, dissolve 10 g of sodium azide in 100 ml of distilled H<sub>2</sub>O. Store at room temperature.

### **sodium azide - CSH Protocols**

3) Add remaining sodium azide. 4) Stir continuously until all is in solution. 5) Filter with 0.45µm filter to remove particles Use this solution to protect protein containing solutions from contaminant growth. Use at 0.1% final concentration (thus add 5µl of 20% sodium azide solution to each 1ml of protein solution).

### **20% Sodium Azide (aqueous solution) - Creighton University**

Disposing of Sodium Azide Solutions: What You Need to Know. Sodium azide (NaN<sub>3</sub>) is extremely toxic and potentially explosive under certain conditions. This azide compound is frequently used as a preservative in many reagents and stock solutions that are utilized in healthcare facilities. Here's what you need to know about drain disposal of sodium azide as well as other solutions.

### **Disposing of Sodium Azide Solutions: What You Need to Know**

Sodium azide 0.1 M solution. Marek Wiergowski et. al Sodium azide (NaN<sub>3</sub>) is an inorganic matrix compound with a very high toxicity. Mechanism of action is not clarified, and it is assumed to interfere with the processes of oxidative phosphorylation. The acute intoxications with sodium azide are extreme... Halina Kostek et.

### **Sodium azide 0.1 M solution | Sigma-Aldrich**

Sodium azide is a useful probe reagent and a preservative. In hospitals and laboratories, it is a biocide ; it is especially important in bulk reagents and stock solutions which may otherwise support bacterial growth where the sodium azide acts as a bacteriostatic by inhibiting cytochrome oxidase in gram-negative bacteria; gram-positive ( streptococci , pneumococci , lactobacilli ) are intrinsically resistant.

### **Sodium azide - Wikipedia**

At concentrations below 2 mM sodium azide, cells were not observably detached from culture dish surfaces and cell metabolism was only partially inhibited. After 2 d cultured cells were treated with sodium azide for 60 min at 37° C, then incubated at 10° C for 30 min followed by additional incubation at 25° C for 5 min.

### **Sodium Azide - an overview | ScienceDirect Topics**

Many Abcam antibody products contain sodium azide and this information is provided on individual datasheets. If the antibody is to be used for cell culture assays or conjugation, sodium azide removal from the antibody solution is recommended. The following three procedures can be used to remove sodium azide.

### **Sodium azide removal protocol | Abcam**

A colorless salt, sodium azide is an inorganic chemical compound that can often be found in car airbags and many reagents or stock solutions used in healthcare facilities. When nearing its decomposition temperature, or when it comes into contact with some metals, sodium azide can be particularly explosive, especially when it hasn't been ...

### **What You Need to Know About Sodium Azide**

Sodium Azide: NaN<sub>3</sub>, A colorless & odorless white solid, CAS# 26628-22-8, Has a Density of 1.846 g/cm<sup>3</sup>, Has a Molar Mass of 65.0099 g/mol, Has a Melting Point of 275 °C (527 °F; 548 K) with violent decomposition, and is considered highly toxic by LD and LC-50. Sodium Azide is a useful probe reagent and a preservative.

### **Lab-Specific Standard Operating Procedure (LSOP) - Sodium ...**

Sodium azide can be removed from antibody solutions by dialysis or gel filtration. The molecular

weight of IgG is 150,000 daltons (IgM is ~600,000); the molecular weight of sodium azide is 65 daltons. A micro-dialysis unit with a cut off at 14,000 daltons will retain the antibody as the azide diffuses out.

### **Antibody storage guide | Abcam**

Purpose and Scope. Sodium azide is an odorless, colorless crystal that is commonly used in research laboratories as a preservative. As a preservative, sodium azide usually exists in a solution at 0.1 to 2.0%. However, of greater concern is the use and storage of pure sodium azide or a solution of 10% or greater.

### **Standard Operating Procedures - University of California ...**

Sodium Azide. PROPERTIES Sodium azide is a common preservative of samples and stock solutions in laboratories and a useful reagent in synthetic work. It is not explosive except when heated near its decomposition temperature (300°C) or reacted with metals; heating sodium azide should be avoided.

### **Sodium Azide | Environmental Health & Safety ...**

Sodium azide is the main explosive propellant used to initiate the deployment cycle in most air bag designs in use today. When sodium azide is ignited, the deploying air bag explodes, rapidly filling with nitrogen gas and carbon dioxide, and moves rearward toward the occupant at speeds of up to 210 miles h<sup>-1</sup> (336 km h<sup>-1</sup>).

### **Sodium Azide - an overview | ScienceDirect Topics**

By Tom Goss, Recycling Strategist. Sodium azide (NaN<sub>3</sub>) is extremely toxic and potentially explosive under certain conditions. This azide compound is frequently used as a preservative in many reagents and stock solutions that are utilized in healthcare facilities.

### **Disposing of Sodium Azide Solutions: Health Cautions**

Uses. Sodium azide (NaN<sub>3</sub>, CAS: 26628-22-8) is used in dilute solutions as a preservative of biological samples and stock solutions. Typical concentrations for use as a preservative are from 0.02 % to 0.1 %. Sodium azide is also used in laboratories as a reagent in synthesis work.

### **Fact Sheet: Sodium Azide | PennEHRS**

Sodium azide solution Convenient solution of 1% sodium azide minimizes exposure and contamination risks normally encountered while handling sodium azide powder. Features CAS #: 26628-22-8 Molecular Formula: NaN<sub>3</sub> Molecular Weight: 65.0099

### **Sodium Azide [1%] - G-Biosciences**

To prepare a stock solution, dissolve 1 g of desiccated salmon sperm DNA in 100 ml of H<sub>2</sub>O by stirring for at least 1 day. Add NaCl to a final concentration of 100 mM and extract with TE-saturated phenol (pH 8.0) (1:1). Shear the extracted DNA by sonication or by repeatedly passing (10–20 times) the DNA

### **01 LabRef Sect.1 p.1 42 - Cold Spring Harbor Laboratory Press**

Williams' Lab Recipes 2-log DNA ladder ... To make a 10% stock solution of sodium azide, dissolve 10 g of sodium azide in 100 ml of distilled H<sub>2</sub>O. Store at 4 degrees. \* Add Sodium Azide to a final concentration of 0.05% to prevent contamination: 5ul of 10% stock per 1 ml).

### **Williams' Lab Recipes**

Another example of this waste type is Sodium Azide solution, which is often used as a preservative in biological laboratories. Sodium Azide is a P-listed chemical, so all unused stock solutions containing Sodium Azide must be collected and disposed of as hazardous waste through EH&S.

### **EPA-Listed Hazardous Waste » Environmental Health & Safety ...**

azide solutions down the sink. Sodium azide can also ex-plode when heated. DO NOT dry sodium

azide in an oven. Prepare a 1000 mg/L stock standard solution of azide anion by dissolving 0.155 g of sodium azide in reagent grade water in a 100-mL volumetric flask. Bring to volume with reagent grade water and store the stock solutions in high ...

## Sodium Azide Stock Solution

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