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organizes the elements into vertical groups and horizontal SCPS Chemistry Worksheet – Periodicity SCPS Chemistry Worksheet – Periodicity A Periodic table 1 Which are metals Circle your answers C Na F Cs Ba Ni Which metal in the list above has the most metallic

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### **SCPS Chemistry Worksheet ☒ Periodicity**

SCPS Chemistry Worksheet — Periodicity - page 4 U, Definitions - match,tomicA raulus First ionization energy Noble gases Decrease Increase Nonmetals Electron affinity Ionization energy Semimetal electronegativity Metals Shielding effect Noble gas configuration 1. \_\_\_\_ is the energy required to remove an electron from an atom. 2.

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SCPS Chemistry Worksheet – Periodicity SCPS Chemistry Worksheet – Periodicity - page 2. B. Ionization Energy 1. Choose the element with the greatest first ionization energy: Carbon or aluminum Calcium or strontium Helium or lithium Chlorine or argon Chlorine or fluorine Sulfur or chlorine 2. Periodicity

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9) Explain each answer briefly. a) Place the following elements in order of increasing ionization energy: F, O, and S. b) Which has the largest ionization energy: O, S, or Se?

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