

Solutions Introduction Gpb Chemistry Answer Key

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Solutions Introduction Gpb Chemistry Answer

At the completion of this episode's lesson(s), you should be able to:

- Describe and explain the process of forming a solution using the terms solute and solvent.
- Compare the properties of suspensions, colloids, and solutions.
- List the factors that affect the rate at which a solid solute dissolves in a liquid solvent.

Chemistry 1001: Solutions: A Special Type of Mixture ...

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3. Solutions are said to be in a _____ phase even though the components may have been in different phases before the solution was formed. 4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) _____, which is a type of _____ solution. An example of a gaseous solution is

Worksheet: Solutions Introduction Name K

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ANSWER KEY Worksheet – Introduction to Solutions Part I- The Basics: 1. Define the following terms: a. Solute – substance being dissolved (lower concentration) b. Solvent – substance doing the dissolving (higher concentration) c. Solution – homogenous mixture of 2 or more substances (solute + solvent) d.

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