

## *Section 2 3 Carbon Compounds Answer Key*

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2. What gives carbon the ability to form chains that are almost unlimited in length?

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Section 2-3 Carbon Compounds. carbon compounds flashcards. STUDY. PLAY. What is organic chemistry the study of? the study of all compounds that contain bonds between carbon atoms. What is a monomer? small unit that can join together with other small units to form polymers. What is a polymer?

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Section 2-3 Carbon Compounds (pages 44-48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things. The Chemistry of Carbon (page 44) 1.

**Macromolecules The Chemistry of Carbon**

Section 2-3 Carbon Compounds This section explains how the element carbon able to form millions of carbon, or organic, compounds. It describes the four groups of organic compounds found in living things. The Chemistry of Carbon How many valence electrons does each carbon atom have? Each carbon atom has four electrons.

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44 Chapter 2 1 FOCUS Objective 2.3.1 Describe the functions of each group of organic compounds. Vocabulary Preview As students read, have them make a concept map using the section's Vocabulary terms, excluding the words

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have?

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Section 2.3 Carbon Compounds; Yara C. • 40 cards. Organic vs Inorganic Chemistry. Organic chemistry is the study of compounds that contain bonds between carbon atoms; Inorganic chemistry is the study of all other compounds; Importance of Carbon. 4 valence electrons allowing it to form strong ...

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Section 2.3. Lipids Fats/Oils/Steroids/Wax 4 Categories of Organic Molecules Molecules of Life Biochemicals ... 2. Alkynes: compounds that contain a triple bond ... 2. Number the carbon atoms of this chain sequentially, beginning at the end nearer the double bond. If the

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? 2. What gives carbon the ability to form chains that are almost unlimited in length? Macromolecules (page 45) 3.

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**Biology 2.3 Carbon Compounds by Roxanne Price on Prezi**

Carbon Compounds Ch 2.3. 2 2.3 Carbon Compounds A. The Chemistry of Carbon Organic any substance that contains carbon except CO and CO<sub>2</sub> Carbon is very important to living things 1. Carbon has 4 Valence electrons Makes it very versatile Will easily bond with ...

**Carbon Compounds Ch 2 - Northern Highlands**

2.3 Carbon Compounds Lesson Objectives Describe the unique qualities of carbon. Describe the structures and functions of each of the four groups of macromolecules. Lesson Summary The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between carbon atoms. Carbon atoms have four valence electrons, allowing them to form

**The Chemistry of Life - Weebly**

2-3 Carbon Compounds A. The Chemistry of Carbon B. Macromolecules Mr. M. Varco St. Joseph High School • The study of “organic” chemistry involves the study of compounds containing bonds between carbon (C) atoms • Why is carbon worth studying? – Carbon atoms have four valence

electrons allowing it to form strong covalent bonds

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