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Thanks for your answers. My question was mainly about the method that Aspen Plus applies in calculating the vapor pressure of the compounds in the case we are using an equation of state.

Where does ASPEN Plus calculate the vapor pressure of ...

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25 000 00 (2): 1983, 1984 KAIST 0000 A gas cylinder of 10 ft³ volume containing nitrogen initially at a pressure of 10 atm and a temperature of 230K, is connected to another cylinder of 10 ft³ volume which is evacuated. A valve between the two cylinders is opened until the pressures in both cylinders equalize. Find the final temperature and pressure in each cylinder if there ...

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