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Solutions Intermediate Unit 10 Progress

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where R is the ideal gas constant, n the molar number, Φ the volume fraction, and the subscripts 1 and 2 denote the solvent and the polymer, respectively. With this equation, ΔS_{com} is always positive because $\Phi_1 + \Phi_2 = 1$ and $\partial \Delta S_{\text{com}} / \partial n_2$, the partial molar entropy of the polymer, is also positive when the amount of the solvent is much larger than the polymer.

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Radioactive waste - Wikipedia

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