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We're now posting original research! Yes, as of late November we are hosting our own original study titled An Examination of the Effect of Prior Experience, Age, and Gender in Non-Food Blending Predictions. Though this title sounds pretty scientific, it just refers to an experiment I did with putting rubber balls in a blender to see...

**The Cavalcade o' Chemistry | Celebrating 20 years of ...**

When I was young and taking chemistry for the first time, my teacher kept mentioning something called stoichiometry. Given that I had no interest in chemistry and that I was rebelling against my scientist father, I ignored everything my teacher had to say. This turned out to be good, because my teacher was an idiot....

**Solutions Stoichiometry | The Cavalcade o' Chemistry**

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For chemistry help, visit [www.chemfiesta.com](http://www.chemfiesta.com) © 2000 Cavalcade Publishing, All Rights Reserved 7) How many liters of a 0.88 M solution can be made with 25.5 grams of

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Practice Problems: Solutions (Answer Key) What mass of solute is needed to prepare each of the following solutions? a. 1.00 L of 0.125 M K<sub>2</sub>SO<sub>4</sub> 21.8 g K<sub>2</sub>SO<sub>4</sub> b. 375 mL of 0.015 M NaF 0.24 g NaF c. 500 mL of 0.350 M C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> 31.5 g C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>; Calculate the molarity of each of the following solutions:

**Practice Problems: Solutions (Answer Key)**

<http://www.chemfiesta.com> Moles Worksheet (Solutions) 1) Define "mole".  $6.02 \times 10^{23}$  of anything, usually atoms or molecules. 2) How many moles are present in 34 ...

**Moles Worksheet - Awesome Science Teacher Resources**

Answers - Naming Chemical Compounds . Name the following chemical compounds: 1) NaBr sodium bromide. 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate. 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide. 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate. 5) FePO<sub>4</sub> iron(III) phosphate. 6) K<sub>3</sub>N potassium nitride. 7) SO<sub>2</sub> sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) Zn(NO

**Answers - Naming Chemical Compounds**

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Stoichiometry with Solutions Name \_\_\_\_ 1.  $\text{H}_3\text{PO}_4 + 3 \text{NaOH} \rightarrow \text{Na}_3\text{PO}_4 + 3 \text{H}_2\text{O}$  How much 0.20 M  $\text{H}_3\text{PO}_4$  is needed to react with 100 ml. of 0.10 M NaOH? 2.  $2 \text{HCl} + \text{Zn} \rightarrow \text{ZnCl}_2 + \text{H}_2$  When you use 25 ml. of 4.0 M HCl to produce  $\text{H}_2$  gas, how many grams of zinc does it react with?

**Stoichiometry with Solutions Problems - Dameln Chemsite**

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