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The solution the scientist made is unsaturated (44 g of HCl in 200 g of water). According to Table G, 72 g saturates 100 g of water so 144 g of HCl are needed to saturate 200 g of water. 37.

Piersa, Amanda / Unit 10: Solutions Chemistry

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This section explains how to solve problems involving molarity of a solution, how to prepare dilute solutions from more concentrated solutions, and what is meant by percent by volume and percent by mass. Molarity (pages 480-482) 1. A measure of the amount of solute dissolved in a given quantity of solvent is the ____ of a solution. 2.

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