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Answers. 1 Solubility Curves . There are charts and tables available that we can use to get an idea of how soluble a certain solute is in a certain solvent. We will take a look at two of them in these next two sections. Solubility curves, like the one shown here, tell us what mass of solute will dissolve in 100g (or 100mL; see note

### **Solubility Curve Practice Problems Worksheet 1**

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SOLUBILITY CURVE WORKSHEET KEY Use your solubility curve graphs provided to answer the following questions. 1. What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water 2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent. 3.

### **SOLUBILITY CURVE WORKSHEET**

Solubility Curves. Print Answer Key PDF Take Now Schedule Copy. Print Answer Key (Only the test content will print) Solubility Curves Answer Key. Use the solubility curve graph to answer the questions. 1. Which of the following has a solubility that decreases as temperature increases?

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6. Use the graph to answer the following two questions: ... -Use a different color for the two different solubility curves. Sodium Chloride Solubility. Temperature Solubility (g of solute/100 mL of H<sub>2</sub>O) 0 35.7 10 35.8 20 35.9 30 36 40 36.4 60 37.1 80 38 90 38.5 100 39.2 Copper Sulfate Solubility ... Solubility Curve Practice Problems Worksheet ...

### **Solubility Curve Practice Problems Worksheet 1**

Solubility Curve: A graph of the solubility of a compound ... (Answers will vary.) ... I created a supersaturated solution of sodium acetate and water. The sodium acetate trihydrate solution was made up of 160 g of sodium acetate and 30 ml of water. It was heated until the solute (sodium acetate) dissolved completely into the solvent (water

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Part 1 – Solubility Curves - USE YOUR SOLUBILITY CURVE GRAPH TO ANSWER #1-4. 1. At what temp does 135 grams of KI dissolved in 100 grams of water form a saturated solution? 2. o How many grams of KNO<sub>3</sub> will dissolve in 400 grams of water at 60 C? 3. If 10 grams of KClO<sub>3</sub> are dissolved in 100 grams of water at 30 oC, is the solution saturated,

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On a solubility curve, the lines indicate the concentration of a saturated solution - the maximum amount of solute that will dissolve at that specific temperature. The molar concentration of the substance can be calculated, as shown by this example:

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Solubility Guidelines for Aqueous Solutions 0 102030405060708090100 Temperature (°C) Solute per 100 g of H<sub>2</sub>O (g) 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 KI NaNO<sub>3</sub> NH<sub>3</sub> HCl NaCl KNO<sub>3</sub> NH<sub>4</sub>Cl KCl SO<sub>2</sub> KClO<sub>3</sub> Table G Solubility Curves Ions That Form Soluble Compounds Exceptions Group 1 ions (Li<sup>+</sup>, Na<sup>+</sup>, etc.)

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