# Solution Stoichiometry Chem Worksheet 15 6 Answer Key

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L solution mol solute 1 L = 1000 mL The molarity of a solution is a ratio of the moles of solute per liters of solution. The units for molarity are written as mol/L or M. This measurement is used to perform stoichiometric calculations. The strategy used for solving these solution stoichiometry problems is to set up the problem so that the units cancel.

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#### AP Chem: Solution Calc review: M= mol or Liters = mol Page ...

Chemistry: Stoichiometry – Problem Sheet 2 KEY 9) 2 24 2 2 23 2 2 2 2 4.63 x 10molecules I 1 mol I  $6.02 \times 10$  molecules I 1 mol Cl 1mol 71 g Cl Cl x 546 g Cl 10) 292 g Ag 1 mol Ag 108 g Ag 1 mol Cu 1 mol Ag 63.5 g Cu

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