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Numerical analysis - Wikipedia

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1. a. $^{13}_{27}\text{Al}$. Mass no. (A) = 27. Atomic no. (Z) = 13. No. of protons = Z = 13. No. of neutrons (N) = A - Z = 14. b. $^{92}_{238}\text{U}$. No. of protons = 92. No. of neutrons ...

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What is the solution to this physics numerical? - Quora

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