

Section 161 Thermal Energy And Matter Answers

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Chapter 16 Thermal Energy and Heat ... Section 16.1 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained.

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Chapter 16 Thermal Energy and Heat Section 16.1 Thermal ...

Chapter 16 -Thermal Energy and Heat Section 16.1 -Thermal Energy and Matter •In the 1700's most scientists thought that heat was a fluid called caloric that flowed between objects. Count Rumford •Rumford studied the process of drilling holes in the barrels of cannons. •The process occurred in water so that the metal

Chapter 16 - Thermal Energy and Heat

Section 16.1 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. It also discusses thermal expansion and contraction of materials, and explains uses of a calorimeter. Reading Strategy (page

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Chapter 16 Thermal Energy and Heat Section 16.1 Thermal Energy and Matter (pages 474–478) Calculations Using Specific Heat Content and Vocabulary Support Heat and Temperature Heat is the transfer of thermal energy from one object to another because of a temperature difference. Heat flows spontaneously from hot objects to cold objects.

Chapter 16 Thermal Energy and Heat Section 16.1 Thermal ...

Work and Heat Temperature is the measure of how hot or cold something is compared to a reference point. The Celsius scale has reference points of freezing and boiling points of water On

the Kelvin scale the reference point is absolute zero Absolute Zero is the temperature at which molecules essentially stop (no kinetic energy)

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