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Molarity Practice Problems #1 – Answer Key 1) How many grams of potassium carbonate are needed to make 280 mL of a 2.5 M solution? Using the molarity equation ( $M = \text{mol/L}$ ), we can find that we'll need 0.70 mol of potassium carbonate. Given that the molar mass of  $\text{K}_2\text{CO}_3$  is 138.21 g/mol, this means that we'll require 97 grams

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Investigating the concentration of a solution. Purpose: To investigate the concept of molarity and to determine the concentration of an acid which ... Unit 1 Reinforcement Worksheet .

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Molarity Practice Problems How many grams of potassium carbonate are needed to make 200 ml- of a 2.5 M solution? How many liters of 4 M solution can be made using 100 grams of lithium bromide? What is the concentration of an aqueous solution with a volume of 450 ml- that contains 200 grams of iron (II) chloride?

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Molarity Worksheet W 331 Everett Community College Student Support Services Program What is the molarity of the following solutions given that: 1) 1.0 moles of potassium fluoride is dissolved to make 0.10 L of solution.

**Molarity Worksheet W 331 - Everett Community College**

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