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Modern Chemistry 110 Chapter Test Name Class Date Chapter Test B, continued _____ 7. Effervescence is the a. dissolving of a gas in a liquid. b. escape of a liquid from a liquid-liquid solution. c. escape of a solid from a solid-liquid solution. d. escape of a gas from a gas-liquid solution. _____ 8.

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Modern Chemistry 130 Chapter Test Name Class Date Chapter Test B, continued PART II Write the correct term (or terms) in the space provided. 9.A substance that ionizes almost completely in aqueous solutions, producing H 30 ions, is a(n) acid. 10. An acid that contains hydrogen and only one other element is called a(n) acid. 11.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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| Modern Chemistry 57 Chapter Test Chapter Test B, continued 7. In a reaction, the ions of two |
|--|
| compounds exchange places in aqueous solution to form two new compounds. This reaction is |
| called a a. synthesis reaction. b. decomposition reaction. c. single-displacement reaction. d. double- |
| displacement reaction. 8. |

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Answers 5 The Periodic Law TEST B. 1. a 2. c 3. d 4. d 5. a 6. a 7. c 8. a 9. lanthanides 10. 2 11. fourth 12. transition elements 13. 32 14. valence electrons 15. electron affinity 16.

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Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued ______ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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a way to describe a chemical reaction using chemical formulas.... a solid that forms from a solution during a chemical reaction. A number in front of a chemical formula in an equation that in.... an equation in which the reactants and products in a chemical....

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