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From Masterton and Hurley, Chemistry: Principles and Reactions, 5th ed., Chap 4 Homework 1: 1. How would you prepare the following from the solid and pure water a. 0.400 L of 0.155 M $\text{Sr}(\text{OH})_2$ $0.0400 \text{ L} \times 0.155 \text{ mol Sr}(\text{OH})_2 \times 121.63 \text{ g Sr}(\text{OH})_2 = 1 \text{ L soln}$ 1 mol $\text{Sr}(\text{OH})_2$ b. 1.75 L of 0.333 M $(\text{NH}_4)_2\text{CO}_3$ 3.

From Masterton and Hurley, Chemistry: Principles and ...

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Dr. Masterton is co-author of the all-time best-selling general chemistry textbook CHEMICAL PRINCIPLES, which has sold well over 1.5 million copies. Dr. Masterton's field of research, solution thermodynamics, prepared him well for making maple syrup each March at the family farmhouse in New Hampshire.

Chemistry: Principles and Reactions: William L. Masterton ...

Abstract: An introduction of chemistry to a college-level audience. It is well recommended for a one-semester or two-quarter introductory chemistry sequence. It is well recommended for a one-semester or two-quarter introductory chemistry sequence.

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