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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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One way to test the salinity of a water sample is to add a few drops of silver nitrate solution with a known concentration. As the solutions of sodium chloride and silver nitrate mix, a precipitate of

8 Chemical Equations and Reactions - David Brearley High ...

Modern Chemistry 68 Chapter Test Name Class Date Chapter Test A, continued ____21. Metal X replaces the ions of metal Y from solution, but it cannot replace the ions of metal Z from solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z b. z, x, y

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Accelerated Chemistry Anderson - MCHS Modern Chemistry 7 Chemical Equations & Reactions
CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

CHAPTER 8 Chemical Equations and Reactions

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HONORS CHEMISTRY -- CHAPTER 8-9 TEST - MULTIPLE CHOICE NAME: ____ 1. Which of the following has bonds which are the least ionic? a) CCl 4 b) SiO 2 c) KCl d) NH 3 e) H 2 2. What type of bonding occurs between phosphorus and hydrogen within a molecule of PH 3?

HONORS CHEMISTRY -- CHAPTER 8-9 TEST - MULTIPLE CHOICE NAME:

Multiple Choice. Identify the letter of the choice that best completes the statement or answers the question. ____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to

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Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued ____ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

Assessment Chapter Test A - Kettering City School District

Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued ____ 13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. ____ 14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

Assessment Chapter Test A - Kettering City School District

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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