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Although ions and elements have similar chemical symbols, they are entirely different substances with different physical properties. A. Monatomic Ions In order to determine the charge of monatomic ions, you can use the periodic table as a guide: Group # (Column) Ion Charge Examples 1 +These elements lose one electron to form +1 ions.

Monatomic Ions - Academic Computer Center

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Nomenclature Worksheet 1 : Monatomic Ions Use a periodic table to complete the table below: Ion Formula Element Name sodium bromine magnesium

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Answer Key for Worksheet 5 . Part A: Nomenclature of Ions and Ionic Compounds . 1. Complete the table below for the following monatomic metal and nonmetal ions.

Answer Key for Worksheet 5 - Santa Monica College

Alsccher Nomenclature ws ionic compounds Page 3 of 8 Table 1: Monatomic ions on the periodic table IVA VA VIA VIIA C4— Carbide N3— Nitride O2— Oxide F— Fluoride P3— Phosphide S2— Sulfide Cl— Chloride As3— Arsenide Se2— Selenide Br— Bromide Te2— Telluride I— Iodide Table 2: Monatomic and polyatomic ions by charge.

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