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NCERT Solutions for Class 12 Chemistry Chapter 2 Solutions. INTEXT Questions. Question 1. Calculate the mass percentage of benzene (C_6H_6) and carbon tetrachloride (CCl_4) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride. Solution: Mass of solution = Mass of benzene + Mass of carbon tetrachloride = 22 g + 122 g = 144 g

NCERT Solutions for Class 12 Chemistry Chapter 2 Solutions

NCERT EXERCISES. 2.1 Define the term solution. How many types of solutions are formed? Write briefly about each type with an example. Sol. A solution is a homogeneous mixture of two or more chemically non-reacting substances.

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www.ncerthelp.com (Visit for all ncert solutions in text and videos, CBSE syllabus, note and many more) Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis

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