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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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MODERN CHEMISTRY STATES OF MATTER 81 ... Name Date Class CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ... States of Matter Answer . STATES OF MATTER MODERN CHEMISTRY ...

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Modern Chemistry • CHAPTER 9 HOMEWORK 9-1 (pp. 275–279) VOCABULARY ... Modern Chemistry • CHAPTER 9 HOMEWORK 9-2 (pp. 280–282) VOCABULARY ... Circle the letter of the best answer. 1. For the general chemical equation $2A + 3B \rightarrow 4C + D$, how many moles of A are required to produce

HOMEWORK 9-1

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District

CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCl—In the compound, there is one atom of silver for each atom of chlorine. Ba(OH) 2 —The subscript 2 refers to everything inside the parentheses, so there are two hydroxide ions for

HOMEWORK 7-2 - Kettering City School District

Accelerated Chemistry Anderson - MCHS Modern Chemistry 11 Chemical Bonding CHAPTER 6 STUDY GUIDE Chemical Bonding SECTION 4 METALLIC BONDING SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be

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Section Summary Chapter 10 Answer Key

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