

## ***Mixed Mole Problems Chemistry If8766 Answer Keyzip***

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### **Mixed Mole Problems Chemistry If8766**

MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in  $1.5 \times 10^{25}$  molecules of CO? 2? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO 1mol 3 23 2 25!=!!!! b) What volume would the CO

### **KEY - CP - Mixed Mole Problems**

MIXED PROBLEMS Name What volume of NH<sub>3</sub> at STP is produced if 25.0 g of N<sub>2</sub> is reacted with an excess of ... Chemistry IF8766 OInstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I. N<sub>2</sub> + 3H<sub>2</sub> Name How many moles of hydrogen are needed to completely react with two moles

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Mixed Mole Problems Directions: Solve the following problems. 1) How many grams are there in  $1.5 \times 10^{25}$  molecules of CO? 2) What volume would the CO<sub>2</sub> in Problem 1 occupy at STP? 3) A sample of NH<sub>3</sub> gas occupies 75.0 liters at STP. How many molecules is this? 4) What is the mass of the sample of NH<sub>3</sub> in Problem 3?

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### **Stoichiometry Mixed Problems Instructional Fair Answers**

Stoichiometry: Mixed Problems (KEY) 1) N<sub>2</sub> + 3H<sub>2</sub> → 2NH<sub>3</sub> What volume of NH<sub>3</sub> at STP is produced if 25.0 of N<sub>2</sub> is reacted with an excess of H<sub>2</sub>? 3 3 3 2 3 2 2 2 40.0L NH 1mol NH 22.4L NH 1mol N 2mol NH 28.0g N 25.0g N 1mol N × × × = 2) 2KClO<sub>3</sub> → 2KCl + 3O<sub>2</sub> If 5.0g of KClO<sub>3</sub> is decomposed, what volume of O<sub>2</sub> is produced at STP? 2

### **Stoichiometry: Mixed Problems (KEY)**

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Mixed Mole Problems Directions: Solve the following problems. 1) How many grams are there in  $2.5 \times 10^{25}$  molecules of CO? ANSWER: 1800 g 2) What volume would the CO<sub>2</sub> in Problem 1 occupy at STP? ANSWER: 930 L 3) A sample of NH<sub>3</sub> gas occupies 57.0 liters at STP. How many molecules is this? ANSWER:  $1.53 \times 10^{24}$  4) What is the mass of the sample of NH<sub>3</sub>

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3. How many grams of ammonia are produced in the reaction in Problem 2?  $4 \text{ O}_2 + \text{AgNO}_3 + 0 \text{ Ag} +$   
How many grams of silver chloride are produced from 5.0 g of silver nitrate reacting with an excess of barium chloride? 5J 143B 5. How much barium chloride is necessary to react with the silver nitrate in Problem 4? ÜQCA vvtc Chemistry IF8766 64

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