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Modern Chemistry 109 Chapter Test Chapter: Solutions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. Molecules whose water solutions can carry electric current a. are nonpolar. b. ionize in water. c. do not dissolve in water. d. decompose in water ...

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any compound whose dilute aqueous solutions conduct electricity well; this is due to the presence of all or almost all of the dissolved compound in the form of ions ... Modern Chemistry Chapter 13 Test. 74 terms. Chemistry Chapter 13 Test. 38 terms. thanks kas. 54 terms. ... Chapter 10 Modern Chemistry Vocab. 18 terms. Modern Chemistry - Holt ...

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Modern Chemistry 130 Chapter Test Name Class Date Chapter Test B, continued PART II Write the correct term (or terms) in the space provided. 9. A substance that ionizes almost completely in aqueous solutions, producing  $H^+$  ions, is a(n) acid. 10. An acid that contains hydrogen and only one other element is called a(n) acid. 11.

### Assessment Chapter Test B - clarkchargers.org

Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

### Modern Chemistry Chapter 12- Solutions

Modern Chemistry 56 Chapter Test Assessment Chapter Test B Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The production of a slightly soluble solid compound in a double-

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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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Modern Chemistry 140 Chapter Test Name Class Date Chapter Test B, continued \_\_\_\_ 7. An acid-base titration determines the solution volumes that are a. chemically equivalent. b. of equal molarity. c. of equal mass. d. of equal molality. \_\_\_\_ 8. In acidic solutions, an indicator that is a

weak acid, HIn, is primarily in the form a. In . b. In ...

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Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe. \_\_\_\_15. When you conduct an ...

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### **Holt McDougal Modern Chemistry Chapter 12: Solutions ...**

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

### **Assessment Chapter Test A - Kettering City School District**

Modern Chemistry • CHAPTER 13 HOMEWORK 13-3 (pp. 401-404) VOCABULARY If the statement is true, write true. If the statement is false, replace the underlined term with the term that makes the statement true. 1. A solution that contains the maximum amount of solute is a(n) unsaturated solution. \_\_\_\_2.

### **HOMEWORK 13-1 - Los Banos Unified School District**

MODERN CHEMISTRY HR W material copyrighted under notice in this work. Name CHAPTER 3 TEST continued Date Class SHORT ANSWER Write the answers to the following questions in the space provided. 22, Give three of the main concepts in Dalton's atomic theory. 23. What is molar mass? How is it related to atomic mass?

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