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Read the Text Version. (See Clues and Hints above.)1. Nitric oxide (NO) reacts with ozone (O<sub>3</sub>) to produce nitrogen dioxide and oxygen gas.2. Iron burns in air to form a black solid, Fe<sub>3</sub>O<sub>4</sub>.3. Sodium metal reacts with chlorine gas to form sodium chloride.4. Acetylene, C<sub>2</sub>H<sub>2</sub>, burns in air to form carbon dioxide and water.Modeling Chemistry 3 U7 ws1 v2.0.

### **Chemistry - Unit 7 Reaction Equations Worksheet 1 Pages 1 ...**

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### **Chemistry - Unit 7 Reaction Equations Worksheet 1**

The American Modeling Teachers Association (AMTA) was created by teachers to continue and expand the mission after government funding for Modeling Instruction(TM) ended. The AMTA has expanded to a nationwide community of teachers dedicated to addressing the nation's Science, Technology, Engineering, and Mathematics (STEM) education crisis.

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Chemistry – Unit 6 Worksheet 1 We have observed evidence that when M-NM compounds are dissolved, the metal particles tend to form positively charged ions (cations), while non-metal particles tend to form negatively charged ions (anions). However, when these same ... ©Modeling Instruction – AMTA 2013 4 U6 ws1 v3.0.

### **Chemistry - Unit 6 Worksheet 1**

Modeling Chemistry 1 U7 ws Bal Eq v2.0 Name Date Pd Chemistry Unit 7 More practice in writing and balancing equations Write balanced chemical equations for the following reactions. Zinc reacts with hydrogen chloride to form zinc chloride.  $\text{Zn} + 2 \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$ . Sodium oxide reacts with water to form sodium hydroxide.

**06\_balance-key balancing equations word prob - Name Pd ...**

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©Modeling Instruction - AMTA 2014 1 U7 review v2.0 Name Date Pd Chemistry - Unit 7 Review Chemical Reaction Model 1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

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