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Spectroscopy Problems. In each of these problems you are given the IR, NMR, and molecular formula. Using this information, your task is to determine the structure of the compound. The best approach for spectroscopy problems is the following steps: Calculate the degree of unsaturation to limit the number of possible structures.

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Syllabus. A. Introduction - the electromagnetic spectrum B. Basic  $^1\text{H}$  NMR 1. Chemical Shift 2. Integration 3. J-coupling - first order multiplets . C. Combining NMR with other spectroscopic methods to solve structure problems - IR, UV, MS.

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### **NMR Problem Set**

Problems in NMR and IR Spectroscopy Welcome to WebSpectra - This site was established to provide chemistry students with a library of spectroscopy problems. Interpretation of spectra is a technique that requires practice - this site provides  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR, DEPT, COSY and IR spectra of various compounds for students to interpret.

### **WebSpectra - Problems in NMR and IR Spectroscopy**

Spectrometric Identification of Organic Compounds is written by and for organic chemists, and emphasizes the synergistic effect resulting from the interplay of spectra. This text is characterized by its problem-solving approach with numerous practice problems and extensive reference charts and tables.

**Spectrometric Identification of Organic Compounds, 8th ...**

View Homework Help - Spectroscopy\_Silverstein\_Problem\_8.31\_Solutions from CHEMISTRY Chemistry at University of Nottingham University Park Campus. This is a solution of problem 8.31 in Silversteins. ... 6 Chapter 9NMR and MS NMR NMR stands for Nuclear Magnetic Resonance Basically

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We hope that this solutions manual will be a useful adjunct to Understanding NMR Spectroscopy (2nd edition, Wiley, 2010) and will encourage readers to work through the exercises. The old adage that 'practice makes perfect' certainly applies when it comes to getting to grips with the theory of

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This text presents a unified approach to the structure determination of organic compounds based largely on mass spectrometry, infrared (IR) spectroscopy, and multinuclear and multidimensional nuclear magnetic resonance (NMR) spectroscopy. The key strength of this text is the extensive set of practice and real-data problems (in Chapters 7 and 8).

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Show NMR answer Here, the two E methyl groups have different chemical shifts because they are permanently locked into different positions with respect to the alkene. Previous Problem Problems list Next Problem

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