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### CHAPTER 5 REVIEW The Periodic Law - apcscience.com

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### 14 Ions in Aqueous Solutions and Colligative Properties

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the

following

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MODERN CHEMISTRY CHAPTER 5 MIXED REVIEW 39 HRW material copyrighted under notice appearing earlier in this work. Name Date Class . MIXED REVIEW continued g. In general, which has a stronger electron attraction, large atoms or small atoms? h. Which has greater electronegativity, O or Se? i.

### HOMEWORK 5-1

CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Identify whether the descriptions below describe an ideal gas or a real gas. ideal gas a. ... SECTION 5 MODERN CHEMISTRY STATES OF MATTER. STATES OF MATTER MODERN CHEMISTRY.

### 10 States of Matter - Ms. Agostine's Chemistry Page

1. 4.5 mol The following equation represents a laboratory preparation for oxygen gas:  $2\text{KClO}_3(\text{s}) \rightarrow 2\text{KCl}(\text{s}) + 3\text{O}_2(\text{g})$  How many moles of  $\text{O}_2$  form if 3.0 mol of  $\text{KClO}_3$  are totally consumed? 2. 200 g Given the following equation:  $\text{H}_2(\text{g}) + \text{F}_2(\text{g}) \rightarrow 2\text{HF}(\text{g})$  How many grams of HF gas are produced as 5 mol of fluorine react?

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