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Determine what the volume of gas would be at STP. 3. Determine the mass, then the moles of Mg used. Modeling Chemistry TN Modeling Curriculum Committee Pope John Paul II High School 307. 4. Write a balanced equation to describe the reaction between the Mg and HCl. Determine the number of moles of H<sub>2</sub> gas in your sample.

### **Unit 9, Worksheet 1— Dalton's Law of Partial Pressures**

Modeling Chemistry 1 U2 ws 3 v2.0 . Name . Date Pd . Unit 2 Worksheet 2 - Measuring Pressure . Problem 1 and 2 Calculate the pressure using the barometer . 1. 720 mmHg = 0.95 atm. 2. 760 mmHg = 1 atm. Problems 3 and 4. Calculate the pressure of the gas in the flask connected to the manometer. 3. a.  $127-84= 43\text{mmHg}$   $730 + 43 = 773 \text{ mmHg}$

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### **Date Pd Unit 9 Worksheet 3: Molar Concentration**

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©Modeling Instruction – AMTA 2013 1 U1 ws3 v2.0. Name Date Pd. Chemistry – Unit 1 Worksheet 3. Mass, Volume, and Density. 1. Study the matter shown in Figure 1. Each dot represents a particle of matter. [Assume the particles are uniformly distributed throughout each object, and particles of the same size have the same mass.]

### **Date Pd Chemistry Unit 1 Worksheet 3 - Quia**

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Name Pd Date Chemistry – Unit 7 Worksheet 3 Write balanced chemical equations for the following reactions. 1. Ammonia (NH<sub>3</sub>) reacts with hydrogen chloride to form ammonium chloride. 2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide. 3.

### **Date Pd Chemistry Unit 7 Worksheet 3 - Welcome to Orange ...**

Modeling Chemistry TN Modeling Curriculum Committee Pope John Paul II High School 341. Name Veritas. Chemistry – Unit 10 Worksheet 3. 1. Element X has two natural isotopes. The isotope with a mass number of 6 has a relative abundance of 7.5%. The isotope with a mass number of 7 has a relative abundance of 92.5%.

### **Chemistry - Unit 10 Worksheet 1 - JPII**

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