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In this lesson, we will describe what it means to have a freezing point depression. We will explain how we determine the new freezing point and show examples of when this is used in the world.

Freezing Point Depression: Definition, Equation & Examples ...

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Honors Chemistry - Dr. VanderVeen

The freezing point of t-butanol is 25.50°C and K_f is 9.1°C • kg/mol. Usually t-butanol absorbs water on exposure to air. If the freezing point of a 10.0-g sample of t-butanol is 24.59°C, how many grams of water are present in the sample?

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