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Ch 6 Study Guide answers

Modern Chemistry 2 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. _____ In the modern periodic table, elements are ordered (a) according to decreasing atomic mass.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions.

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Accelerated Chemistry Anderson - MCHS Modern Chemistry 11 Chemical Bonding CHAPTER 6 STUDY GUIDE Chemical Bonding SECTION 4 METALLIC BONDING SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be

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Sample Study Sheet 4.2: Abbreviated Electron Configurations To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

Chapter 4 Modern Atomic Theory - An Introduction to Chemistry

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