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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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water to produce a sulfuric acid solu- tion that falls to the ground as rain or snow. 4(aq) 37. Have a sour taste; change the color of acid-base indicators; some react with active metals to release hydrogen gas; react with bases to produce salts and water; conduct electric current 38. Acid-Base Titration and pH, pp.

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a. Meyer c. Stas b. Ramsay d. Moseley _____ 13. The most useful source of general information about the elements for anyone associated with chemistry is a. a. calculator. c. periodic table. b. table of metric equivalents. d. table of isotopes. ____ 14. The periodic table. a. permits the properties of an element to be predicted before the element ...

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Modern Chemistry: Chapter Tests with Answer Key, 2006, Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, Harcourt School Publishers, 2006 ... Michigan Holt Chemistry and Modern Chemistry Test Preparation Workbook: Help for the MME , Holt, Holt Rinehart & Winston, Jun 1, 2006, Juvenile Nonfiction, 126 pages. .

Modern Chemistry: Chapter Tests with Answer Key, 2006 ...

Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative.

d. zero.	2.
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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued _____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. _____14. When there is a small decrease in temperature, the average kinetic energy of the particles of a liquid a. decreases. b. becomes zero.

Assessment Chapter Test A - Kettering City School District

Modern Chemistry Chapter 11. Gases and Gas Laws. STUDY. PLAY. Ideal Gas Law. PV=nRT. P. Pressure. V. Volume. ... temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = \sqrt{Mb} / \sqrt{Ma} . Kinetic-Molecular Theory ... Chapter 10: Modern Chemistry vocab. Features. Quizlet Live. Quizlet ...

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CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9, If a particular ompound is composed of elements A and B, the ratio of the mass of B to the mass of A will alw ys be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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