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CHAPTER 16 REVIEW Reaction Energy SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. For the following examples, state whether the change in entropy favors the forward or reverse reaction:

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The reaction of rubidium, Rb, with water is faster and more violent than the reaction of Na with water. Use the atomic structure and radius of each metal to account for this difference.

8 Chemical Equations and Reactions

forward reaction to be favored. MODERN CHEMISTRY ACIDS AND BASES 121 ... Name Date Class . SECTION 3 continued 4. Table 6on page 485 of the text lists several amphoteric species, but only one other than water is neutral. NH 3 a. Identify that neutral compound. b. ... Name Date Class . CHAPTER 14 Acids and Bases) ...

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Modern Chemistry 139 Reaction Kinetics CHAPTER 17 REVIEW Reaction Kinetics SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Below is an energy diagram for a particular process. One curve represents the energy profile for the uncatalyzed reaction, and the other curve represents the

CHAPTER 17 REVIEW Reaction Kinetics

CHAPTER 16 REVIEW . Reaction Energy . SHORT ANSWER for this reaction? MODERN CHEMISTRY REACTION ENERGY . 131. ... SECTION 1 continued _____ b. Which is the more stable part of this system, the reactants or the . products? . 6. The enthalpy of combustion of acetylene gas is -1301.1 kllmol of C2H2. a. ...

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