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1. SOLID STATE HOMEWORK SOLUTION Solid State 1. Given : $a = 408.7 \text{ pm} = 408.7 \times 10^{-12} \text{ m} = 408.7 \times 10^{-12} \times 100 \text{ cm} = 408.7 \times 10^{-12} \text{ cm}$ Atomic mass (M)= 108 To Find : Density of silver (d)

1. SOLID STATE - MT Educare

A gas decolourises cold aqueous alkaline KMnO_4 solution but does not give a precipitate with ammonical CuCl_2 (a) Hydrogenation (c) Tautomerism (b) Resonance (d) Metamerism Acetone can be obtained by treating a following alkynes with (b) Pent-1-yne on treatment with sulphuric acid and mercuric solution. The gas is (a) Methane (c) Ethylene

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Solution: The atmospheric pressure in New York City on a certain day was 768 mm Hg. What was the pressure in kPa? a. $1.02 \times 10^5 \text{ kPa}$ b. 584 kPa c. $5.76 \times 10^{-3} \text{ kPa}$ d. 102 kPa e. none of the above

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