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6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 6 REVIEW Chemical Bonding

Modern Chemistry • CHAPTER 6 HOMEWORK 6-3 (pp. 167-169) VOCABULARY Write true or false for

each statement. 1. The	bond length is	s the distanc	e between t	two bonded	atoms at th	ieir maxii	mum
potential energy	2. The energy	required to	form a chen	nical bond b	etween two	neutral	atoms
is called bond energy.	3.						

HOMEWORK 6-1 - Los Banos Unified School District

Modern Chemistry 1 Chemical Bonding Assessment Chemical Bonding Section 6-1 Quiz: Introduction to Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. Ionic bonds form as a result of the electrostatic attraction between ...

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

42.1 L b. The oxygen gas produced in part a has a density of 1.43 g/L. Calculate the volume of this gas. 6. A car air bag requires 70. L of nitrogen gas to inflate properly. The following equation represents the production of nitrogen gas: $2NaN 3(s) \rightarrow 2Na(s) 3N 2(g) 81 g a$. The density of nitrogen gas is typically 1.16 g/L at room temperature.

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1 Chapter 6 Notes - Chemical Bonding Chemical bond - A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together 6-1 Introduction to Chemical Bonding I. Types of Chemical Bonding A. Ionic Bonding 1. Chemical bonding that results from the electrical attraction between large

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