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4 Notes on Units and Formulas A. Volume The units of volume commonly used in this manual are cm3 and dm3. The conversion between cm3 and dm3 is 1 dm3 = 1000 dm3.Note that a cm3 are the same as millilitres (ml) and dm3 are the same as litres (L).

Guide to Chemistry Practicals - TETEA

Recommended software downloads: Below are links to general freeware programs that I highly recommend for learning chemistry.

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1. The particulate nature of matter (Cambridge International IGCSE Chemistry syllabus 0620). describe the states of matter (gases, liquids and solids), distinguishing features, and explain their interconversion in terms of the kinetic particle theory

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States of matter, state changes and gas laws revision notes index. Particle models of gases-liquids-solids, explaining properties, state changes GCSE chemistry notes. Particle theory models, internal energy, heat transfer in state changes and latent heat and particle motion in gases and gas pressure (written more from a 'physics' point of view) gcse physics revision notes

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What's your thoughts on #5c? I don't like that they give you the rate, but don't say what substance it is the rate for. The rate of formation of the product is different than the rate of the disappearance of the reactant.

2016 AP Chemistry Exam Draft Answers and Comments

You could have the libertarian state, the green state, the clothing-optional state, the state with free public housing for all, the state where puns are outlawed, the state with a two-drink minimum, the fiercely pro-business state—even a state that guarantees free speech but requires that you sing your speech like a show tune.

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