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Explain your diagram. © Modeling Instruction-AMTA 2013 1 U6 – review v1 Because atoms are metallic their weak positive cores allow electrons to leave one atom and travel closer to the positive charge. Both are neutral (3 electrons balances out the positive core). If both are neutral then there's no force of atraction or repulsion.

U6 review answers - Unit6Review 1 Recall your ...

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Modeling Chemistry 2 U6 ws2 v1 Atoms involved 1 sodium 1 chlorine 1 calcium 2 bromine 1 aluminum 3 chlorine formula particle diagram 2. Write each formula from Question 1 in the boxes corresponding to its elements. For example, the compound formed from sodium and sulfur have been written in the box for sodium and in the box for sulfur. Now add ...

Modeling Chemistry 2 U6 ws2 v1 Atoms involved 1 sodium 1 ...

Unit 6 – Review 1. Recall your representations of the atoms in the Sticky Tape activity. ... © Modeling Instruction–AMTA 2013 1 U6 – review v1 © Modeling Instruction–AMTA 2013 2 U6 – review v1. 5. Describe how JJ Thomson concluded that the mobile charged particle in the atom ... Apart from making life difficult for beginning chemistry ...

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Unit 6 Sticky Tape Post-Lab - Buckeye Valley

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© Modeling Instruction – AMTA 2013 1 U6 Sticky Tape v1.0 Chemistry – Unit 6 Sticky Tape Post-Lab Thomson Model and Sticky Tape Let's see how we can use Thomson's model to explain the behavior of the sticky tape when we made our tape stacks. A few atoms from the top tape and the bottom tape are represented in the diagram below.

1 Sticky Tape - University of Kentucky

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