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review key concepts in Section 13.3. L2 3 L2 L1 Section 13.3 Assessment 15. Particles in solids are packed tightly together in an orderly arrangement. The locations of the particles are fixed. 16. Section 13.3 The Nature of Solids 399 Non-Crystalline Solids Not all solids are crystalline in form; some solids

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133. The Nature of Solids Key Questions How are the structure and properties of solids related? ... You will find out about fullerenes in this section. A Model for Solids ... Review the key questions and answers in Lessons 13.1 and 13.2. Then have

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Solutions to Chemistry (9780130543844) :: Free Homework Help and Answers :: Slader When a solid 3. is heated until its particles vibrate so rapidly that they are no longer 4. held in fixed positions, the solid . The is the 5. temperature at which a solid changes to a liquid. The melting and 6. of a substance are at the same temperature. In general, 7. ionic solids tend to have relatively melting points, while 8.

05 CTR ch13 7/12/04 8:12 AM Page 317 THE NATURE OF GASES 13 - Pequannock Township High School

Chapter 13 States of Matter137. SECTION 13.1 THE NATURE OF GASES (pages 385–389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386)

Name Date Class STATES OF MATTER 13

This section brings the chapter full circle back to the particle nature of solids, liquids, and gases, but after reading this section, you will know more about the particles that compose solid, liquid, and gaseous elements. The section helps you to visualize the particle nature of the elements instead of relating to them as just symbols on the ...

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