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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Modern Chemistry 31 Quiz Section Quiz: Introduction to Chemical Bonding ... write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. Ionic bonds form as a result of the electrostatic attraction between a. dipoles. b. electrons. ... Assessment Element Electronegativity Element ...

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Chemical Equations and Reactions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3 on page 286 of the text. Which one

### 8 Chemical Equations and Reactions - David Brearley High ...

Modern Chemistry v Study Guide ... SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Technological development of a chemical product often (a) lags behind basic research on the same substance. (b) does not involve chance discoveries. (c) is driven by curiosity.

### mc06se cFMs i-vi

States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) ... SECTION 5 MODERN CHEMISTRY STATES OF MATTER. STATES OF MATTER MODERN CHEMISTRY.

### 10 States of Matter - Ms. Agostine's Chemistry Page

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the

### 3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...

Modern Chemistry 3 Chemical Bonding Assessment Chemical Bonding Section 6-2 Quiz: Covalent Bonding and Molecular Compounds In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1.

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PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided. 25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton ...

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