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Modern Chemistry Chapter Test B

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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Modern Chemistry 34 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following questions in the space provided. 27. What two early twentieth-century observations involving the interaction of light and matter could not be explained soley by the wave theory of light? 28.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry 101 Chapter Test Name Class Date Chapter Test B, continued 18. The pressure exerted by each gas in a mixture is called the of that gas. 19. If the temperature and number of moles of a gas remain constant but the volume increases, the pressure of the gas will . 20. The lowest possible temperature, corresponding to zero on the ...

Assessment Chapter Test B

Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued _____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. _____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued _____ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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Modern Chemistry: Chapter Tests with Answer Key, 2006, Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, Harcourt School Publishers, 2006

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic

crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued _____14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe.

Assessment Chapter Test A - Baumapedia

CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9, If a particular ompound is composed of elements A and B, the ratio of the mass of B to the mass of A will alw ys be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

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Modern Chemistry 56 Chapter Test Assessment Chapter Test B Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. The production of a slightly soluble solid compound in a double-

Assessment Chapter Test B - Hatboro

(b) It will be more difficult to break the bonds in compound A than those in compound B. (c) Compound B has larger crystals than compound A. (d) Compound A has larger crystals than compound B. 5. b The forces of attraction between molecules in a molecular compound are generally (a) stronger than the attractive forces among formula units in ...

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Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued ______ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

Assessment Chapter Test A - Kettering City School District

a. Meyer c. Stas b. Ramsay d. Moseley _____ 13. The most useful source of general information about the elements for anyone associated with chemistry is a. a. calculator. c. periodic table. b. table of metric equivalents. d. table of isotopes. ____ 14. The periodic table. a. permits the properties of an element to be predicted before the element ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

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