# Section 3 Acid Precipitation Review Answers

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this may be necessary to control the acid-precipitation problem. Section 3 Acid Precipitation 317 1. Explain how acid precipitation forms. 2. Describe the harmful effects that acid precipitation can have on plants, soils, and aquatic ecosystems. 3. Describe three ways in which acid precipitation can affect humans. 4. Describe a way in which ...

### SECTION 3 Acid Precipitation - Mrs. Jana Young

Section 3 Acid Precipitation 317 1. Explain how acid precipitation forms. 2. Describe the harmful effects that acid precipitation can have on plants, soils, and aquatic ecosystems. 3. Describe three ways in which acid precipitation can affect humans. 4. Describe a way in which countries are working together to solve the problem of acid ...

#### **SECTION 3 Acid Precipitation - BEATS INFO**

Air Section 3 • Acid precipitation is precipitation, such as rain, sleet, or snow that contains a high concentration of acids, often because of the pollution of the atmosphere. • When fossil fuels are burned, they release oxides of sulfur and nitrogen. • When these oxides combine with water in the atmosphere they form sulfuric acid and nitric acid, which falls as acid precipitation.

# ch12\_sec3 revised - Air Section 3 Chapter 12 Air Section 3 ...

Air Section 3: Acid Precipitation Look at the picture of U.S. rainfall patterns below: major producers of Now look at a picture of the average acidity of precipitation across the U.S.: \_\_\_\_ Why do the regions directly downwind from particulate pollutants have greater rainfall amounts than the areas that are upwind?

#### Air Section 3: Acid Precipitation - Rochester City School ...

Section 4 Review SECTION VOCABULARY acid precipitation rain, sleet, or snow that contains a high concentration of acids air pollution the contamination of the atmosphere by the introduction of pollutants from human and natural sources 1. Identify Relationships How are fossil fuels related to air pollution and acid precipitation? 2.

#### **CHAPTER SECTION 4 Air Pollution - Southwest Middle School**

SE Canada's acid precipitation is formed in Ohio, Indiana, Pennsylvania, Illinois, Missouri, West Virginia, and Tennessee! Acid precipitation falls downwind from where it is formed. International Conflict Know For TEST! Section Review (p.317): 1-3, 6 Acid Precipitation Section 12.3 Objectives: Explain causes of acid precipitation.

#### Acid Precipitation Section 12.3 - calhoun.k12.al.us

Section 3 Solutions of Acids and Bases ... In the United States, most acid rain has a pH between 4 and 4.5, but some precipitation has a pH as low as 3. Salts When an acid neutralizes a base, a salt and water are produced. ... Section Summary

# **Section 3 Solutions of Acids and Bases**

9. acid precipitation 10. emissions regulations and catalytic converters 11. If the outer hull is punctured, the oil does not leak out. 12. measures to reduce everyday contami-nation of waterways from oil SECTION: NUCLEAR ENERGY 1. neutrons 2. boron and cadmium 3. Control rods containing

boron or cad-mium to absorb neutrons are lowered

# Skills Worksheet Active Reading - rvrhs.com

Section: What Causes Air Pollution? MATCHING In the space provided, write the letter of the term or phrase that best matches the description. Name Class Date ... SECTION: ACID PRECIPITATION Matching Multiple Choice 1. d 7. b 2. c 8. a 3. e 9. d 4. a 10. c 5. f 6. b Chapter Test General MATCHING MULTIPLE CHOICE 1. c 11. b 2. e 12. a

#### Section: What Causes Air Pollution?

SECTION 3 continued 4. Table 6on page 485 of the text lists several amphoteric species, but only one other than water is neutral. NH 3 a. Identify that neutral compound. b. Write two equations that demonstrate this compound's amphoteric properties. (Answers will vary, but one equation should form NH

# 14 Acids and Bases - mchsapchemistry.com

Section: Acid Precipitation. Read the passage below and answer the questions that follow. Thousands of lakes throughout the world are victims of acid precipitation, which is also known as acid rain. Acid precipitation is precipitation such as rain, sleet, or snow that contains a high concentration of acids.

#### 01 - Plain Local Schools

Chapter 12-3 Acid Precipitation Answer the following question using the Acid Rain handout. ... What proportion of acid precipitation does sulfur dioxide contribute? \_\_\_\_\_ 5. What are the major sources of nitrogen oxide in the atmosphere? ... 7. What is the pH value of normal rain? \_\_\_\_\_ 8. What pH value is considered acidic for precipitation? ...

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