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Solutions WS Answers Section 2-3 VOCABULARY REVIEW 1. A solvent is a substance in which a solute is dissolved. 2. An aqueous solution is a solution in which water is the solvent. 3. A hydroxide ion is a negatively charged ion consisting of one oxygen atom and one hydrogen atom, OH-. 4. A base is a solution that contains more hydroxide ions than

Solutions WS Answers VOCABULARY REVIEW 1. 2. 3.

Honors Chemistry Solutions Worksheet - Review REVIEW SOLUBILITY 1. ____ is the substance that is dissolved. 2. ____ is the substance that does the dissolving.

Honors Chemistry Solutions Worksheet - Review

Chemistry 20 – Solutions Review Worksheet page 4 13. Calculate the molar concentration of sodium ions in a solution containing 2.56 g of sodium carbonate in 500 mL of solution. 14. A student was instructed to prepare 125 mL of a 0.250 mol/L CuSO 4(aq) solution. Calculate

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