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A) lons can move through the salt bridge. B) The salt bridge must prevent the flow of charged particles. C) The solutions joined by the salt bridge can mix freely.

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statements is true? A. An equation is balanced by changing the subscripts as needed. B. Matter is
usually, but not always, conserved in a balanced chemical equation.
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Test Practice. 9. Given the balanced equation NaOH HCl 0 NaCl H2O, what is the total number of
grams of H2O produced when 116 g of the product NaCl is formed?
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