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b. a higher percentage of vapor. c. equal amounts of liquid and vapor. d. all liquid. \_\_\_\_ 6. If water molecules were nonpolar, water would probably a. be a solid at room temperature. b. have stronger hydrogen bonding. c. be a gas at room temperature. d. be flammable. Name Class Date Chapter Test B Assessment

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Nuclear Chemistry. Predict the mode of decay of (a) carbon-14, (b) (b) xenon-118. (b) Xenon has an atomic number of 54. Thus, xenon-118 has 54 protons and  $118 - 54 = 64$  neutrons, giving it a neutron-to-proton ratio of According to Figure 21.2 , stable nuclei in this region of the belt of stability have higher neutron-to-proton ratios than xenon-118.

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nuclear reactions, but these isotopes all have very similar chemical reactivities. Is nuclear chemistry a curse or a blessing? Like all new technology, it's both. The awesome destruction of atomic (fission) and hydrogen (fusion) weapons is only too well documented. While the nightmare of nuclear Armageddon is fading, the nightmare of

### Chapter 21

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The products of nuclear fission are variable; many possible nuclides can be created. In the feature "An Unexpected Finding," in the text, it was noted that

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