

## ***Solution Chemistry Definition***

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**Solution Chemistry Definition**

Definition of Solution in Chemistry. The amount of solute that can be dissolved in solvent is called its solubility. For example, in a saline solution, salt is the solute dissolved in water as the solvent. For solutions with components in the same phase, the substances present in lower concentration are solutes, while the substance present in highest abundance is the solvent.

**Solution Definition in Chemistry - ThoughtCo**

1. (Chemistry) a homogeneous mixture of two or more substances in which the molecules or atoms of the substances are completely dispersed. 4. (Chemistry) a mixture of two or more substances in which one or more components are present as small particles with colloidal dimension; colloid: a colloidal solution.

**Solution (chemistry) - definition of Solution (chemistry ...**

Generally, an aqueous solution of a nonvolatile substance called a solution or liquor; an aqueous solution of a volatile substance, a water (aqua); an alcoholic solution of a nonvolatile substance, a tincture (tinctura); and an alcoholic solution of a volatile substance, a spirit (spiritus).

**Solution (chemistry) | definition of Solution (chemistry ...**

Definition of Solution. What is a Solution? Mixture of a solid and a liquid where the solid never settles out, eg saltwater.

**Definition of solution - Chemistry Dictionary**

On the other hand, a solution differs from a compoundcompound, in chemistry, a substance composed of atoms of two or more elements in chemical combination, occurring in a fixed, definite proportion and arranged in a fixed, definite structure. A compound is often represented by its chemical formula. ....

**Solution (chemistry) | Article about Solution (chemistry ...**

The air we breathe, what we drink, as well as products in our household - these are a few examples of solutions we encounter every day. In this lesson, we will discuss what a solution is, the ...

**What is a Solution in Science? - Definition & Examples ...**

Chemistry Unit 10 Definitions. STUDY. PLAY. aqueous solution. water that contains dissolved substances. colloid. a mixture whose particles are intermediate in size between those of a suspension and a solute solution. electrolyte. ... Chemistry Chapter 15 and 16 Vocabulary. 37 terms. Chemistry Chapter 15-16 Vocab.

**Chemistry Unit 10 Definitions Flashcards | Quizlet**

The salt is the solute and the water the solvent. In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. The term aqueous solution is when one of the solvents is water. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent.

**Solution - Wikipedia**

Chemistry. the process by which a gas, liquid, or solid is dispersed homogeneously in a gas, liquid, or solid without chemical change. such a substance, as dissolved sugar or salt in solution. a homogeneous, molecular mixture of two or more substances. Pharmacology. Also called liquor. a liquid, usually water, in which a medication is dissolved.

**Solution | Definition of Solution at Dictionary.com**

To achieve a dilute solution, more solvent is simply added without adding any more solute into the original mixture. The solution is then stirred to mix the two ingredients thoroughly. This ensures that all parts of the mixture have the same composition. The chemicals that can be diluted are gases, vapors and liquids.

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