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In the solubility listed below, the amount of solvent (water) is kept constant at 100 grams (100 mL). The only variable that is changed is the temperature of the solvent. A solution that is able to hold more solute (than is presently dissolved) at a particular temperature is unsaturated.

Graphing Solubility WS - Quia

Describe the trend of the curve graph of temperature vs solubility decide whether the solution process is endothermic or exothermic? ... The process is called interpolation, which applies a ...

What is the value of a solubility graph - answers.com

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Solubility graphs represent the relationship between solubility (in grams of solid per volume of water) vs temperature. If the solution is above the solubility line it is supersaturate and below the solubility line it is unsaturated. Points along the line are points of saturation.

Solubility Graphs - Chemistry | Socratic

Used to determine the mass of solute in 100g (100 ml) of water at a given temperature. Below is Table G- This gives information based on 100 grams of water. Warning-the Regents doesn't always ask about 100g. Sometimes they ask about 50g (half of what is on the table) or 200g (double the amount on the table).

Solubility Curves - AP Chemistry

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FYI, when graphing, always put manipulated (independent) on the x axis (horizontal) and the responding (dependent) on the y. Don't forget a graph title and units of measure as well. Keep a graph name simple by creating a title using y axis vs. x axis (for example - Mass of Solute Dissolved vs. Temperature).

Chemistry Help!! Which is independent and dependent ...

Extrapolation and interpolation are both used to estimate hypothetical values for a variable based on other observations. There are a variety of interpolation and extrapolation methods based on the overall trend that is observed in the data. These two methods have names that are very similar.

The Difference Between Extrapolation and Interpolation

The last known point on the graph occurs when the nitrogen gas solubility is 9.33 mg per 100 g. We need to find pressure for a solubility of 14.6 mg per 100 mg, which occurs after the last known point on the graph so the technique of interpolation, as described above, cannot be used here.

Predictions from Graphs Chemistry Tutorial - AUS-e-TUTE

Interpolation is an estimation of a value within two known values in a sequence of values. Polynomial interpolation is a method of estimating values between known data points. When graphical data contains a gap, but data is available on either side of the gap or at a few specific points within the gap, interpolation allows us to estimate the values within the gap.

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