

Sodium Thiosulfate Saturated Solution

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Footnotes 1. Additional tint etches are listed in Ref 1 and 2. 2. Whenever water is specified, use distilled water. 3. Maximum solubility of anhydrous sodium thiosulfate ($\text{Na}_2\text{S}_2\text{O}_3$) is 50 g/100mL water at 20 C (70 F), while that of the crystal form ($\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$) is 79.4 g/100mL water at 0 C (32 F) or 291.1g/100 mL water at 45 C (115 F).

Table I - Selected Tint Etchants - Metallography

Sodium sulfite (sodium sulphite) is a soluble sodium salt of sulfurous acid with the chemical formula Na_2SO_3 . It is also used as a preservative to prevent dried fruit from discoloring, and for preserving meats, and is used in the same way as sodium thiosulfate to convert elemental halogens to their respective hydrohalic acids, in photography and for reducing chlorine levels in pools.

Sodium sulfite - Wikipedia

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SDS Catalog # Name Synonyms; LC10030 LC10030: Acetate buffer (for chlorine, for cyanide)
Acetate buffer pH 4 (for chlorine), Acetate buffer (for cyanide)

All MSDS | Material Safety Data Sheets (MSDS)

Preparation. Sodium metabisulfite can be prepared by evaporating a solution of sodium bisulfite saturated with sulfur dioxide: $2\text{HSO}_3^- \rightleftharpoons \text{H}_2\text{O} + \text{S}_2\text{O}_5^{2-}$. which yields a residue of colorless solid $\text{Na}_2\text{S}_2\text{O}_5$. Chemical structure. The anion metabisulfite is a hybrid of dithionite ($\text{S}_2\text{O}_4^{2-}$) and dithionate ($\text{S}_2\text{O}_6^{2-}$). The anion consists of an SO_2 group linked to an SO_3 ...

Sodium metabisulfite - Wikipedia

Avoidance of cyanide toxicity: As described in CLINICAL PHARMACOLOGY above, when more than 500 mcg/kg of sodium nitroprusside is administered faster than 2 mcg/kg/min, cyanide is generated faster than the unaided patient can eliminate it. Administration of sodium thiosulfate has been shown to increase the rate of cyanide processing, reducing the hazard of cyanide toxicity.

Nitropress - FDA prescribing information, side effects and ...

Maintenance and Storage of the pH Electrode QUESTION: How should pH probes be stored best? ANSWER: All Hach pH probes are delivered with a plastic protection cap that can also be used for storage. Pour a few drops of saturated KCl solution into the cap to ensure that the glass membrane is kept hydrated and ready to use.

Hach pH Meters

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Chemical Compatibility Guide - Harrington Plastics

Molar Solutions —Molar solutions are solutions that contain, in 1000 mL, 1 gram-molecule of the reagent. Thus, each liter of a molar solution of sulfuric acid contains 98.07 g of H_2SO_4 and each liter of a molar solution of potassium ferricyanide contains 329.25 g of $\text{K}_3\text{Fe}(\text{CN})_6$. Solutions containing, in 1000 mL, one-tenth of a gram-molecule of the reagent are designated "tenth-molar ...

usp31nf26s1_volumetric-solutions-1366-1405, Reagents ...

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The Chemical Compatibility tool allows visitors to select the chemical in their application and find the best suited elastomer for service. Visitors can also do a reverse look up and match the elastomer to the chemical.

Chemical Compatability | Rubber | Elastomer

Click on a chemical to jump to its respective page in the fiberglass piping (FRP/GRP) chemical resistance guide. The full guide can be viewed here: Fiberglass Pipe (FRP/GRP) Chemical Resistance Guide (PDF)

Fiberglass Pipe Chemical Resistance Guide | Corrosion Chart

SPECIFIC METHODS ENZYME PREPARATIONS α -Amylase Activity (Bacterial). Application and Principle. This procedure is used to determine the α -amylase activity, expressed as bacterial amylase units (BAU), of enzyme preparations derived from *Bacillus subtilis* var., *Bacillus licheniformis* var., and *Bacillus stearothermophilus*. It is not applicable to products that contain P-amylase.

SPECIFIC METHODS - Food and Agriculture Organization

Rubber O-Ring Chemical Compatibility Chart - O-Ring Material Resistance Guide. Basic O-Ring chemical resistance compatibility information is based on isolated generic O-Ring material testing in optimal conditions at room temperature and pressure.

Rubber O-Ring Chemical Compatibility Chart Resistance Guide

GENERAL METHODS APPEARANCE AND PHYSICAL PROPERTIES Boiling Point and Distillation Range. The following method employs 100 ml of sample. In cases where it is necessary or would be desirable to use a smaller sample, the method of McCullough et al. [J. Chem. Ed. 47, 57 (1970)], which employs only 50 μ l of sample, may be used.

GENERAL METHODS - Food and Agriculture Organization

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Withdrawn Standards. UOP1-87 Distillation Range of Heavy Petroleum Oils and Determination of Coke Residue. UOP6-82 Induction Period of Gasolines Using the UOP Oxygen Pressure Vessel

ASTM International - UOP Standards

THE DETERMINATION OF LACTIC ACID.* BY THEODORE E. FRIEDEMANN AND ARTHIJR I. KENDALL. (From the Department of Research Bacteriology, Northwestern University Medical School, Chicago.) (Received for publication, August 25, 1928.)

THE DETERMINATION OF LACTIC ACID.*

What chemicals can be used to remove dye? There are two different ways to attack a dye, chemically, to remove its color. One is by oxidation, in which electrons are removed, while the other is by reduction, in which electrons are added. On this page I describe each of the discharge chemicals that are used by hand dyers to discharge dye.

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