# States Of Matter Class 11 Solutions

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#### States Of Matter Class 11

CBSE Class 11 Chemistry Notes: States of Matter. (ii) Boiling point This is the temperature at which the vapour" pressure of a liquid becomes equal to the atmospheric pressure. It increases in the presence of impurity and with rise in pressure. Boiling point of water is 100°C. (iii) Freezing point At this temperature,...

## CBSE Class 11 Chemistry Notes: States of Matter - AglaSem ...

States of Matter- CBSE Notes for Class 11 Chemistry CBSE NotesCBSE Notes ChemistryNCERT Solutions Chemistry • Intermolecular Forces Intermolecular forces are the forces of attraction and repulsion between interacting particles have permanent dipole moments. This interaction is stronger than the London forces but is weaker than ion-ion interaction because only partial charges are involved.

# States of Matter- CBSE Notes for Class 11 Chemistry ...

States of Matter Class 11: Overview. In our everyday life, we can observe four different states of matter, namely solid, liquid, gas and plasma. However, there are numerous other states that can be seen to exist but only under extreme conditions. The ones worth mentioning are glass and Bose-Einstein condensates.

#### States of Matter for Class 11: Different States ...

States of Matter Notes Class 11 Chemistry Chapter 5 Download in Pdf. 4. Boyle's Law: The pressure of fixed msss of gas varies inversely with the volume at constant T. P  $\alpha$  1/V(n,T const.) P 1 V 1 = 2 V 2 5. Charle's Law: At constant P, the volume of fixed amount of gas varies directly with its absolute temperature.

## States of Matter Notes Class 11 Chemistry Chapter 5 ...

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