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We can find the enthalpies of R-134a from the property tables: hR, in = hR(1 MPa, 90oC) = 324.64 kJ/kg from superheat tables on page 930. The value of hR, out = h(1 MPa, 30oC) is the enthalpy of a compressed liquid. We can see this since the temperature of a saturated liquid at 1 MPa is 39.37oC, which is higher than the outlet temperature.

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Solutions Unit Exam Ms. Roman Page 2 Regents Chemistry (1)K2CO3 (2)KC2H3O2 (3)Ca3(PO4)2 (4)Ca(NO3)2 15.According to your Reference Tables, which of these compounds is the least soluble in water?

Solutions Unit Exam Name Date Period - .::* Ms. Roman's ...

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Alignment with New York State Science Standards & Performance Indicators 4 3.3c Atoms join together in well-defined molecules or may be arranged in

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Water table Coarse-grained soils Below the water table the soil can be considered to be saturated. In coarse-grained soils, water will drain from the pores and air will therefore be present in the soil between the ground surface and the water table. Consequently, pore pressures above the water table can usually be ignored.

Groundwater - Faculty of Environment and Technology

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