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9) Explain each answer briefly. a) Place the following elements in order of increasing ionization energy: F, O, and S. b) Which has the largest ionization energy: O, S, or Se?

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organizes the elements into vertical groups and horizontal SCPS Chemistry Worksheet – Periodicity SCPS Chemistry Worksheet – Periodicity A Periodic table 1 Which are metals Circle your answers C Na F Cs Ba Ni Which metal in the list above has the most metallic

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