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solution is available to use as the titrant. The student uses a 25.00 mL volumetric pipet to deliver the propanoic acid solution to a clean, dry flask. After adding an appropriate indicator to the flask, the student titrates the solution with the 0.173 M NaOH, reaching the end point after 20.52 mL of the base solution has been added.

AP Chemistry 2014 Scoring Guidelines - College Board

Questions 4–7 are short free-response questions that require about 7 minutes each to answer and are worth 4 points each. Write your response in the space provided following each question.

A P Chemistry 2014 Free-Response Questions

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Chemistry guide - ibchem.com

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Solution Definition in Chemistry - ThoughtCo

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