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Oxygen, carbon dioxide, argon, and other gases are solutes. You are probably most familiar with solutions in which water is the solvent. Carbonated drinks, hot tea, and seawater are just a few examples of the many water-based solutions you might have encoun- tered.

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Chapter 16 Solutions 167. SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471–472) Look at Figure 16.1 on page 471 to help you answer Ouestions 1 and 2.

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Chapter 8 Solutions, Acids, and Bases Section 8.1 Formation of Solutions (pages 228–234) This section explains the parts of a solution, the processes that occur when compounds dissolve, and how the properties of a solution compare with those of its solvent and solute. Reading Strategy (page 228)

Chapter 8 Solutions, Acids, and Bases Section 8.1 ...

Transcript of 8.1 Formation of Solutions. Three physical properties of a solution that can differ from those of its solute and solvent: Heat of Solution: when a solution is formed, energy is either lost or gained.

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Ch 8 Solutions, Acids and Bases Study Guide Answers 16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must ATTRACT ONE ANOTHER. 17. During the formation of a solution, energy is ABSORBED & RELEASED. 18. The solubility of solids increases as the temperature of the solvent ...

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Chapter 8 Solutions, Acids, and Bases ... Section 8.1 Formation of Solutions (pages 228–234) This section explains the parts of a solution, the processes that occur when compounds dissolve, and how the properties of a solution compare with those of its solvent and solute. ... Circle the correct answer.

Chapter 8 Solutions, Acids, and Bases Section 8.1 ...

Solution Formation Temperature At higher temperatures, the kinetic energy of water molecules is greater than at lower temperatures, so the ... solubility and supersaturated solutions to explain your answer. Crystal-growing kits usually begin with a supersaturated solution. When a seed crystal is added to the solution, crystals rapidly begin ...

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