

Standardized Solution

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Standardized Solution

Standard solution. Jump to navigation Jump to search. In analytical chemistry, a standard solution is a solution containing an accurately known concentration of an element or a substance, A known weight of solute is dissolved to make a specific volume. It is prepared using a standard substance, such as a primary standard.

Standard solution - Wikipedia

Standardized solution. Also found in: Medical, Encyclopedia . (Chem.) a solution which is used as a reagent, and is of a known and standard strength; specifically, a normal solution, containing in each cubic centimeter as many milligrams of the element in question as the number representing its atomic weight; thus,...

Standardized solution - definition of Standardized ...

A standard solution is any chemical solution which has a precisely known concentration. Similarly, a solution of known concentration has been standardized. To prepare a standard solution, a known mass of solute is dissolved and the solution is diluted to a precise volume.

Standard Solution Definition - Chemistry Glossary

standard solution one that contains in each liter a definitely stated amount of reagent; usually expressed in terms of normality (equivalent weights of solute per liter of solution) or molarity (moles of solute per liter of solution).

Standardized solution | definition of Standardized ...

standard solution. n. A solution of known concentration, used as a standard of comparison or analysis.

Standard solution | Definition of Standard solution at ...

There are multiple ways to prepare a standard solution, but we'll focus on one example commonly encountered in a laboratory. Say you're in the lab and need to make a standard solution of 1.0 M sodium chloride. You know that you want the volume of this solution to be 400 ml.

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