

Text S1: Recipe for variant-Hypho medium

Concentrations in Final Medium

Chemical	Concentration	Purpose
K ₂ HPO ₄	14.5 mM	Buffer/Nutrient
NaH ₂ PO ₄	18.8 mM	Buffer/Nutrient
(NH ₄) ₂ SO ₄	3.8 mM	Nutrient
MgSO ₄	0.8 mM	Nutrient

General Comments

This medium uses phosphates as the buffer and the pH of the final media is determined by the relative concentration of the monobasic and dibasic phosphate components. A media equivalent to this one can also be made by switching the cation used in the phosphates as long as the relative concentration stays the same (such that $K_2HPO_4 \rightarrow Na_2HPO_4$ and $NaH_2PO_4 \rightarrow KH_2PO_4$). This simple recipe does not include calcium or trace metals. Historically, 1000X of a modified Vishniac trace metal mix has been added, as shown below.

Preparation

This media can be prepared by combining two stock solutions.

P-solution (10x): K₂HPO₄ 25.3 g (or 33.1 g K₂HPO₄ • 3 H₂O)
NaH₂PO₄ 22.5 g (or 25.9 g NaH₂PO₄ • H₂O)
in 1 L of deionized H₂O

S-solution (10x): $(\text{NH}_4)_2\text{SO}_4$ 5 g
 $\text{MgSO}_4 \bullet 7 \text{ H}_2\text{O}$ 2 g (or 0.98 g MgSO_4)
in 1 L of deionized H_2O

Vishniac Trace Elements (1000X):

Add in the following order, adjusting the pH to 5.0 with each addition.

dH ₂ O	500 mL
EDTA	5 g (or 6.37 g of EDTA •2 H ₂ O)
ZnSO ₄ •7 H ₂ O	2.2 g
CaCl ₂ •2 H ₂ O	0.733 g
MnCl ₂ •4H ₂ O	0.506 g
FeSO ₄ •7 H ₂ O	0.499 g
(NH ₄)MO ₇ O ₂₄ •4 H ₂ O	0.110 g
CuSO ₄ •5 H ₂ O	0.157 g
CoCl ₂ •6 H ₂ O	0.161 g

Note: This recipe is based on a trace metal formula given in the fourth footnote of a review paper by Vishniac and Saunter (Vishniac W & Santer M (1957) *The Thiobacilli. Bacteriological Reviews* 21:195.). However, the original formula had 50 fold higher concentrations and used a pH of 6.0.