

Text S1: Recipe for variant-Hypho medium

Concentrations in Final Medium

Chemical	Concentration	Purpose
K_2HPO_4	14.5 mM	Buffer/Nutrient
NaH_2PO_4	18.8 mM	Buffer/Nutrient
$(\text{NH}_4)_2\text{SO}_4$	3.8 mM	Nutrient
MgSO_4	0.8 mM	Nutrient

General Comments

This medium uses phosphates as the buffer and the pH of the final media is determined by the relative concentration of the monobasic and dibasic phosphate components. A media equivalent to this one can also be made by switching the cation used in the phosphates as long as the relative concentration stays the same (such that $\text{K}_2\text{HPO}_4 \rightarrow \text{Na}_2\text{HPO}_4$ and $\text{NaH}_2\text{PO}_4 \rightarrow \text{KH}_2\text{PO}_4$). This simple recipe does not include calcium or trace metals. Historically, 1000X of a modified Vishniac trace metal mix has been added, as shown below.

Preparation

This media can be prepared by combining two stock solutions.

Recipe (for 1 L): 100 mL 10X P-solution
 100 mL 10X S-solution
 800 mL deionized H_2O

P-solution (10x): K_2HPO_4 25.3 g (or 33.1 g $\text{K}_2\text{HPO}_4 \cdot 3 \text{H}_2\text{O}$)
 NaH_2PO_4 22.5 g (or 25.9 g $\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$)
 in 1 L of deionized H_2O

S-solution (10x): $(\text{NH}_4)_2\text{SO}_4$ 5 g
 $\text{MgSO}_4 \cdot 7 \text{H}_2\text{O}$ 2 g (or 0.98 g MgSO_4)
 in 1 L of deionized H_2O

Vishniac Trace Elements (1000X):

Add in the following order, adjusting the pH to 5.0 with each addition.

dH ₂ O	500 mL
EDTA	5 g (or 6.37 g of EDTA • 2 H ₂ O)
ZnSO ₄ • 7 H ₂ O	2.2 g
CaCl ₂ • 2 H ₂ O	0.733 g
MnCl ₂ • 4H ₂ O	0.506 g
FeSO ₄ • 7 H ₂ O	0.499 g
(NH ₄)MO ₇ O ₂₄ • 4 H ₂ O	0.110 g
CuSO ₄ • 5 H ₂ O	0.157 g
CoCl ₂ • 6 H ₂ O	0.161 g

Note: This recipe is based on a trace metal formula given in the fourth footnote of a review paper by Vishniac and Saunter (Vishniac W & Santer M (1957) The Thiobacilli. *Bacteriological Reviews* 21:195.). However, the original formula had 50 fold higher concentrations and used a pH of 6.0.