## Unit 12: Nitrogen and sulfur

### Subunit 12.1: Nitrogen and sulfur

#### Topical Question No: 1

- 13 Which reagent, when mixed with ammonium sulfate and then heated, liberates ammonia?
  - A aqueous bromine
  - B dilute hydrochloric acid
  - **C** limewater
  - **D** potassium dichromate(VI) in acidic solution

#### Topical Question No: 2

- 36 In which different forms does nitrogen exist in compounds?
  - 1 bonded by a triple covalent bond
  - 2 as part of a cation
  - 3 in an oxidation state of +5

#### Topical Question No: 3

**19** Ammonia, NH<sub>3</sub>, and hydrazine, NH<sub>2</sub>NH<sub>2</sub>, are two compounds of nitrogen, N<sub>2</sub>.

Which statement is correct?

- **A** The N–N bond in  $NH_2NH_2$  is polar.
- **B** NH<sub>3</sub> and NH<sub>2</sub>NH<sub>2</sub> have lone pairs of electrons but N<sub>2</sub> does not.
- C The oxidation number of each nitrogen in NH<sub>2</sub>NH<sub>2</sub> is +2.
- **D** The reaction of nitrogen with hydrogen has a high activation energy.

# **Answer Key**

- 1. Error
- 2. Error
- 3. Error