Unit 16: Hydroxy compounds

Subunit 16.1: Alcohols

Topical Question No: 1

26 Propan-2-ol undergoes many reactions.

Which row is correct?

	reagent added to propan-2-ol	product
Α	acidified KMnO₄	CH₃CH₂CHO
В	Cl ₂	CH₃CHC <i>1</i> CH₃
С	conc. H ₂ SO ₄	CH₃CHCH₂
D	methanoic acid	HCO ₂ CH ₂ CH ₂ CH ₃

Topical Question No: 2

24 Structural isomerism and stereoisomerism should be considered when answering this question.

The molecular formula of compound X is $C_5H_{12}O$.

Compound X:

- reacts with alkaline aqueous iodine
- can be dehydrated to form two alkenes only.

What could be the identity of compound X?

- A CH₃CH₂CH(CH₃)CH₂OH
- **B** CH₃CH₂CH(OH)CH₂CH₃
- C (CH₃)₂CHCH(OH)CH₃
- D CH₃CH₂CH₂CH(OH)CH₃

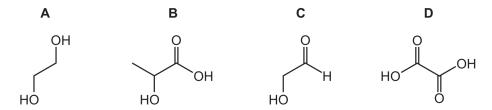
Topical Question No: 3

- 25 Which product is formed when 3-methylpentane-1,3,4-triol is heated under reflux with an excess of acidified potassium dichromate(VI)?
 - A HO₂CCH₂C(CH₃)(OH)COCH₃
 - **B** HO₂CCH₂COC(OH)(CH₃)₂
 - C OHCCH₂C(CH₃)(OH)COCH₃
 - **D** HO₂CCH₂CO(CH₃)COCH₃

Topical Question No: 4

- 29 1 mole of each of the following four compounds is reacted separately with:
 - an excess of sodium
 - an excess of sodium carbonate.

Which compound produces the same volume of gas with each of the two reagents?



Topical Question No: 5

- **38** What can be produced when an aqueous solution of butan-2-ol is oxidised under suitable conditions?
 - 1 butanone
 - 2 butanoic acid
 - 3 butanal

Topical Question No: 6

- **24** Which sequence of reagents may be used in the laboratory to convert propan-1-ol into 2-bromopropane?
 - A concentrated sulfuric acid, followed by bromine
 - B concentrated sulfuric acid, followed by hydrogen bromide
 - **C** ethanolic sodium hydroxide, followed by bromine
 - **D** ethanolic sodium hydroxide, followed by hydrogen bromide

Topical Question No: 7

- 26 Which compound gives an organic product with a lower boiling point when it is heated under reflux with an excess of acidified potassium dichromate(VI)?
 - A 2-methylbutan-1-ol
 - B 2-methylbutan-2-ol
 - C pentan-1-ol
 - D pentan-2-ol

Topical Question No: 8

39 The compound pentan-1,4-diol has two OH groups per molecule and can be oxidised.

Which statements about pentan-1,4-diol or its oxidation products are correct?

- 1 When one mole of pentan-1,4-diol reacts with an excess of sodium metal, one mole of hydrogen molecules is produced.
- **2** At least one of the possible oxidation products of pentan-1,4-diol will react with 2,4-dinitrophenylhydrazine reagent.
- 3 Dehydration of pentan-1,4-diol could produce a compound with empirical formula C₅H₈.

Answer Key

- 1. Error
- 2. Error
- 3. Error
- 4. Error
- 5. Error
- 6. Error
- 7. Error
- 8. Error