Unit 11: Group 17

Subunit 11.1: Physical properties of the Group 17 elements

Topical Question No: 1

16 Some uses of chlorine and bromine are given.

Which is a use of bromine?

- A making bleaches for textiles and the paper industry
- B making CFCs
- **C** making flame retardants and fire extinguishers
- **D** making the polymer PVC

Topical Question No: 2

16 The volatility of the Group 17 elements, chlorine, bromine and iodine, decreases down the group.

What is responsible for this?

- A bond length in the halogen molecule
- B bond strength in the halogen molecule
- C electronegativity of the halogen
- **D** number of electrons in the halogen molecule

Topical Question No: 3

- 18 What trend is observed on descending Group VII?
 - A The colours of the elements become lighter.
 - **B** The elements become more volatile.
 - **C** The hydrides of the elements become more thermally stable.
 - **D** The reactions of the elements with hydrogen become less vigorous.

Answer Key

- 1. Error
- 2. Error
- 3. Error