

# VE320 Introduction to Semiconductor Physics and Devices

## Recitation Class 1

VE320 Teaching Group SU2022

University of Michigan-Shanghai Jiaotong University  
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# Course Overview

- Textbook

- Semiconductor Physics and Devices (4th edition).
- A translated edition in Chinese also available.
- If you have time, practice some problems.

- Homework

- Solutions are mostly available online.
- Problems requiring drawing a graph does not have an online answer. Ask TAs if finding it difficult.
- Review homework before exams.
- We will try to cover some difficult questions in RC.

# Course Content

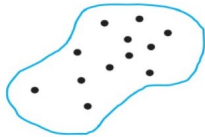
- Semiconductor physics: Ch. 1-6
  - Ch. 1 Crystalline structure of solids
  - Ch. 2 Quantum theory
  - Ch. 3 Quantum theory of solids
  - Ch. 4 Carriers concentration of semiconductors in thermal equilibrium
  - Ch. 5 Carrier flows in thermal equilibrium, and the current
  - Ch. 6 Behaviour of carriers under non-equilibrium
- Semiconductor devices: Ch. 7-12
  - pn junction, Schottky junction, MOSFET, BJT

# Semiconductor Definition

- Semiconductors are the materials that have resistivity between  $10^{-3} - 10^9 \Omega \cdot \text{cm}$ .
- Resistivity are influenced heavily by light, temperature, electric field, magnetic field, impurities ...
- Examples include Si, Ge, GaAs, InP ...
- Doping
  - n-type semiconductors: Charge carriers are negative, i.e. electrons doped by donor-type of dopants.
  - p-type semiconductors: Charge carriers are positive, i.e. holes doped by acceptor-type of dopants.

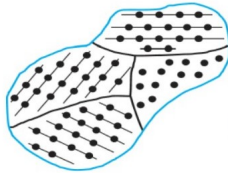
# Types of Solids

▪ Amorphous



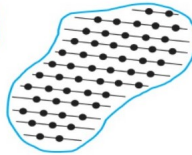
All atoms or ions are periodically ranged in a short range (a few atoms)

▪ Polycrystals



Multiple crystalline grains randomly packed

▪ Single crystals



All atoms or ions are periodically ranged in a long range ( $\mu\text{m}$  scale)

Figure: Types of solids

All semiconductors covered in this course are assumed to be single crystalline.

- Unit cell
  - Small volume of crystal that can reproduce the entire crystal.
  - Have multiple choices.
- Primitive cell
  - The smallest unit cell.
  - May also have multiple choices.
- volume density =  $\frac{\text{\#atom per cell}}{\text{volume of cell}}$
- surface density =  $\frac{\text{\#atom per lattice plane}}{\text{area of lattice plane}}$

# Lattice Types

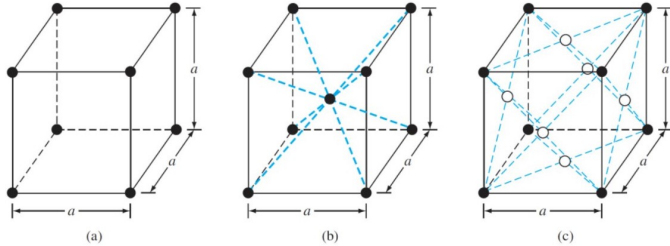


Figure: (a) Simple cubic; (b) Body-centered; (c) Face-centered.

$a$  in the figures is called the lattice constant.



# Diamond/Zincblende Structure

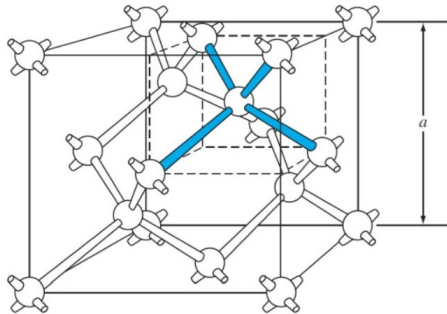
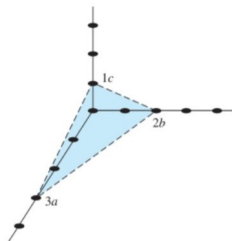


Figure: Diamond structure.

- In a diamond structure, all atoms are of the same type.
- In a zincblende structure, two different types of atoms are present, e.g., GaAs.

# Miller Index



$$(3, 2, 1) \xrightarrow{\text{Reciprocal}} \left(\frac{1}{3}, \frac{1}{2}, 1\right) \xrightarrow{\text{multiply lcd}} (2, 3, 6)$$

Figure: Miller index.

- All parallel planes are entirely equivalent.
- In a cubic,  $[hkl]$  directions are perpendicular to  $(hkl)$  planes.

# Preliminaries

- For matters
  - $p = mv$ ,  $E = \frac{1}{2}mv^2$ ,  $p = \frac{h}{\lambda}$
- For photons
  - $p = mv$ ,  $E = h\nu$ ,  $\nu = \frac{c}{\lambda}$
- Wave number  $k = \frac{2\pi}{\lambda}$
- Uncertainty principle  $\Delta p \Delta x \geq \hbar$  and  $\Delta E \Delta t \geq \hbar$  where  $\hbar = \frac{h}{2\pi}$

# Solutions of Differential Equations

- $\frac{\partial^2 y}{\partial x^2} = k^2 y \rightarrow y = Ae^{kx} + Be^{-kx}$
- $\frac{\partial^2 y}{\partial x^2} = -k^2 y \rightarrow y = Ae^{jkx} + Be^{-jkx} = C \sin(kx) + D \cos(kx)$

# Time-Independent Schrodinger Equation

- $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi = E\Psi$
- Physical meaning:  $|\Psi(x)|^2 = \Psi(x) \cdot \Psi^*(x)$  is the probability density function.
- Solution when  $V(x) = V_0$ 
  - If  $E > V_0$ ,  $\Psi(x) = Ae^{-jkx} + Be^{jkx}$  where  $k = \sqrt{\frac{2m(E-V_0)}{\hbar^2}}$ .
  - If  $E < V_0$ ,  $\Psi(x) = Ae^{-kx} + Be^{kx}$  where  $k = \sqrt{\frac{2m(V_0-E)}{\hbar^2}}$ .
  - The first term towards  $-x$ , the second term towards  $+x$ .
- Boundary condition
  - $\Psi(x)$  is continuous.
  - $\frac{d\Psi(x)}{dx}$  is continuous when  $V < \infty$ .
  - $\int_{-\infty}^{\infty} \Psi(x)\Psi^*(x)dx = 1$
  - $|\Psi(x)| < \infty$

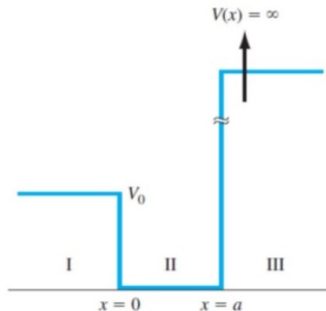
# Infinite Quantum Well

- Formulation  $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi = E\Psi$ , 
$$\begin{cases} V(x) = +\infty, & x \leq 0 \text{ or } x \geq a \\ V(x) = 0, & 0 < x < a \end{cases}$$
- General solution  $\Psi(x) = Ae^{-ikx} + Be^{ikx}$
- Boundary condition
  - $\Psi(x)|_{x=0,a} = 0$
  - $\int_0^a \Psi(x)\Psi^*(x)dx = 1$
- Results
  - $k = \frac{n\pi}{a}$ ,  $n = 1, 2, \dots$  ( $\pm n$  have the same physical meaning; a particle must have non-zero energy)
  - $E = \frac{k^2 \hbar^2}{2m} = \frac{n^2 \pi^2 \hbar^2}{2ma^2}$
  - $\Psi(x) = \sqrt{\frac{2}{a}} \sin(kx)$

# Example

Consider the one-dimensional potential function shown in the figure below. Assume the total energy of an electron is  $E < V_0$ .

- Write the wave solutions that apply in each region.
- Write the set of equations that result from applying the boundary conditions.
- Show explicitly why, or why not, the energy levels of the electron are quantized.



Questions?