2022-01-04

1.The x and y coordinates of a particle at any time t are given by $x=7t+4t^2$ and y=5t, where x and y are in metre and t in seconds. The acceleration of particle at t=5s is

0

 $8m/s^2$

 $20m/s^{2}$

 $40m/s^{2}$

2. Four wires of equal length and of resistances 10 ohms each are connected in the form of a square. The equivalent resistance between two opposite corners of the square is

10 ohm

20 ohm

40 ohm

2.5 ohm

3.The work function of a metal is

The energy for the electron to enter into the metal

The energy for producing X-ray

The energy for the electron to come out from metal surface

None of these

4.The image distance of an object placed 10 cm in front of a thin lens of focal length + 5 cm is

6.5 cm

8.0 cm

9.5 cm

10.0 cm

5.A particle is executing simple harmonic motion with a period of T seconds and amplitude a metres

. The shortest time it takes to reach a point $\frac{a}{\sqrt{2}}m$ from its mean position in seconds is

Τ

T/4

T/8

T/16
6.Two-point charges +8q and -2q are located at $x=0$ and $x=L$ respectively. The location of a point on the x-axis at which the net electric field due to these two-point charges is zero is
8L
4L
2L
$\frac{L}{4}$ 7.A ball is dropped from a height h. If the coefficient of restitution be e, then to what height will it rise after jumping twice from the ground
eh/2
2eh
eh
e4h
8.In Millikan oil drop experiment, a charged drop of mass 1.8*10-14 kg is stationary between its plates. The distance between its plates is 0.90 cm and potential difference is 2.0 kilo volts. The number of electrons on the drop is
5
6
7
8
9.A neutron having mass of $1.67\times 10^{-27}kg$ and moving at 108m/s collides with a deutron at rest and sticks to it. If the mass of the deutron is 3.34 x 10-27kg then the speed of the combination is
2.56 x 103m/s
2.98x 105m/s
3.33 x 107m/s
5.01 x 109m/s
10.A charged particle is suspended in equilibrium in a uniform vertical electric field of intensity 20000

V/m. If mass of the particle is $9.6\times10^{-16}{\rm kg}$, the charge on it and excess number of electrons on the particle are respectively $(g=10m/s^2)$

$$4.8\times10^{-19}\mathrm{C}, 3$$

$$5.8 \times 10^{-19} C, 4$$

$$3.8 \times 10^{-19}$$
C, 2

$$2.8 \times 10^{-19}$$
C, 1

11.A boat carrying steel balls is floating on the surface of water in a tank. If the balls are thrown into the tank one by one, how will it affect the level of water

It will remain unchanged

It will rise

It will fall

First it will first rise and then fall

12.Two identical cylindrical vessels with their bases at same level each contains a liquid of density r. The height of the liquid in one vessel is and that in the other vessel is . The area of either base is A. The work done by gravity in equalizing the levels when the two vessels are connected, is

$$(h_1 - h_2) \, g \rho$$

$$(h_1 - h_2) gA\rho$$

$$\frac{1}{2}{(h_1-h_2)}^2\,gA\rho$$

$$\frac{1}{4}{(h_1-h_2)}^2\,gA\rho$$

13.If a, b, c are three vectors such that a=b+c and the angle between b and c is $\pi/2$, then

$$a^2 = b^2 + c^2$$

$$b^2 = c^2 + a^2$$

$$c^2 = a^2 + b^2$$

$$2a^2 - b^2 = c^2$$

14.The intersection of two lines represented by $\left(x-1\right)^2-\left(y+2\right)^2+2\left(x-1\right)\left(y+2\right)=0$ is:

(0,0)

$$(-1,2)$$

$$(1, -2)$$

none of these

15.The vectors a, b and a + b are

Collinear

Coplanar

Non-coplanar

None of these

16.A bag contains 5 red and 6 blue balls. The number of ways to choose 8 balls is:

165

426

336

none of the above

17.

$$\lim_{x\to 0}\cos\frac{1}{x}$$

Is continuous at x = 0

Is differentiable at x = 0

doesn't exist

none of these

18. The roots of equation $x^2 + 2xi - 1 = 0$ are:

real and distinct

complex and distinct

real but identical

complex but identical

19. The area bounded by $y = \log x$, the ordinates y = 0, y = 2 and y axis is :

$$e^2 - e$$

$$e^2 - 2$$

$$e^2 - 1$$

$$e^{2} - 2e$$

20.Unit vector in the direction of 2i - 2j + k is:

$$\frac{2}{3}i - \frac{2}{3}j + \frac{1}{3}k$$

$$-\frac{2}{3}i + \frac{2}{3}j - \frac{1}{3}k$$

both (a) and (b)

none of the above

21.The equations of tangents to the circle $x^2+y^2-22x-4y+25=0$ which are perpendicular to the line 5x+12y+8=0 are

$$12x - 5y + 8 = 0, 12x - 5y = 252$$

$$12x - 5y = 0, 12x - 5y = 252$$

$$12x - 5y - 8 = 0, 12x - 5y + 252 = 0$$

none of these

22.For, |x| < 1, the infinite series $1 - 2x + 3x^2 - 4x^3 + \dots$ evaluates to:

$$\frac{1}{1+x}$$

$$\frac{1}{1-x}$$

$$\frac{1}{(1+x)^2}$$

$$\frac{1}{(1-x)^2}$$

23.A circle passes through (0, 0) and (1, 0) and touches the circle $x^2+y^2=9$, then the centre of circle is:

$$\left(\frac{3}{2},\frac{1}{2}\right)$$

$$\left(\frac{1}{2}, \frac{3}{2}\right)$$

$$\left(\frac{1}{2},\frac{1}{2}\right)$$

$$\left(\frac{1}{2}, \pm \sqrt{2}\right)$$

24. The equation of the plane which bisects the line joining (2, 3, 4) and (6, 7, 8) is:

$$x + y + z - 15 = 0$$

$$x - y + z - 15 = 0$$

$$x - y - z - 15 = 0$$

$$x + y + z + 15 = 0$$

25.If
$$\lim_{x \to 2} \frac{x^n - 2^n}{x - 2} = 80$$
, where n is a positive integer, then $n =$

3

5

2

none of these

26. The greatest coefficient in the expansion of $(1+x)^{2n+2}$ is

$$\frac{(2n)!}{(n!)^2}$$

$$\frac{(2n+2)!}{\{(n+1)!\}^2}$$

$$(2n+2)!$$

$$\frac{(2n+2)!}{n!(n+1)!}$$

$$\frac{(2n)!}{n!(n+1)!}$$

27.If
$$x\sqrt{1+y} + y\sqrt{1+x} = 0$$
, then $\frac{dy}{dx} =$

$$1+x$$

$$(1+x)^{-2}$$

$$-(1+x)^{-1}$$

$$-(1+x)^{-2}$$

28. r^{th} term in the expansion of $(a+2x)^n$ is

$$\frac{n(n+1)\dots(n-r+1)}{r!}a^{n-r+1}(2x)^r$$

$$\frac{n(n-1)\dots(n-r+2)}{(r-1)!}a^{n-r+1}(2x)^{r-1}$$

$$\frac{n(n+1)\dots(n-r)}{(r+1)!}a^{n-r}(x)^r$$

None of these

29.If an extremely hot copper wire is subjected to steam, the product formed is:

CuO

 Cu_2O Cu_2O_2 CuO_2 30.The number of molecules in 100 $\rm mL$ of each of $\rm O_2, NH_3$ and $\rm CO_2$ at STP are $CO_2 < O_2 < NH_3$ $NH_3 < O_2 < CO_2$ $NH_3 = CO_2 < O_2$ $NH_3 = O_2 = CO_2$ 31.In the cell $Zn\mid Zn^{2+}\mid \mid Cu^{2+}\mid Cu,$ the negative electrode is Cu Cu^{2+} Zn Zn^{2+} 32. The correct structure of benzene was proposed by Faraday Davy Kekule Wohler 33.pH values of HCl and NaOH solutions each of strength $\frac{N}{100}$ will be respectively 2 and 2 2 and 12 12 and 2 2 and 10 34.At 25°C, the solubility product of $Mg(OH)_2$ is 1.0×10^{-11} . At which pH, will Mg^{2+} ions start precipitating in the form of $Mg(OH)_2$ from a solution of 0.001 M Mg^{2+} ions 8 9

10

11
35.The largest number of molecules is in
34g of water
$28g$ of CO_2
$46g \ { m of} \ CH_3OH$
$54g \ {\rm of} \ N_2O_5$
36.Which of the following compound is formed when a gas obtained by reacting H_2SO_4 with excess of P_4O_{10} is treated with anhydrous HCl ?
Chlorosulphonic acid
Hypochlorous acid
Sulphur
Phosphine
37.She wants to know where you live ()
(?)
(!)
(;)
(.)
38.We took a taxi home.
to
for
by
with
39.I never work Fridays.
in
on
by
at
40.Everybody by the terrible news yesterday.

shocked

were shocked

be shocked

was shocked.