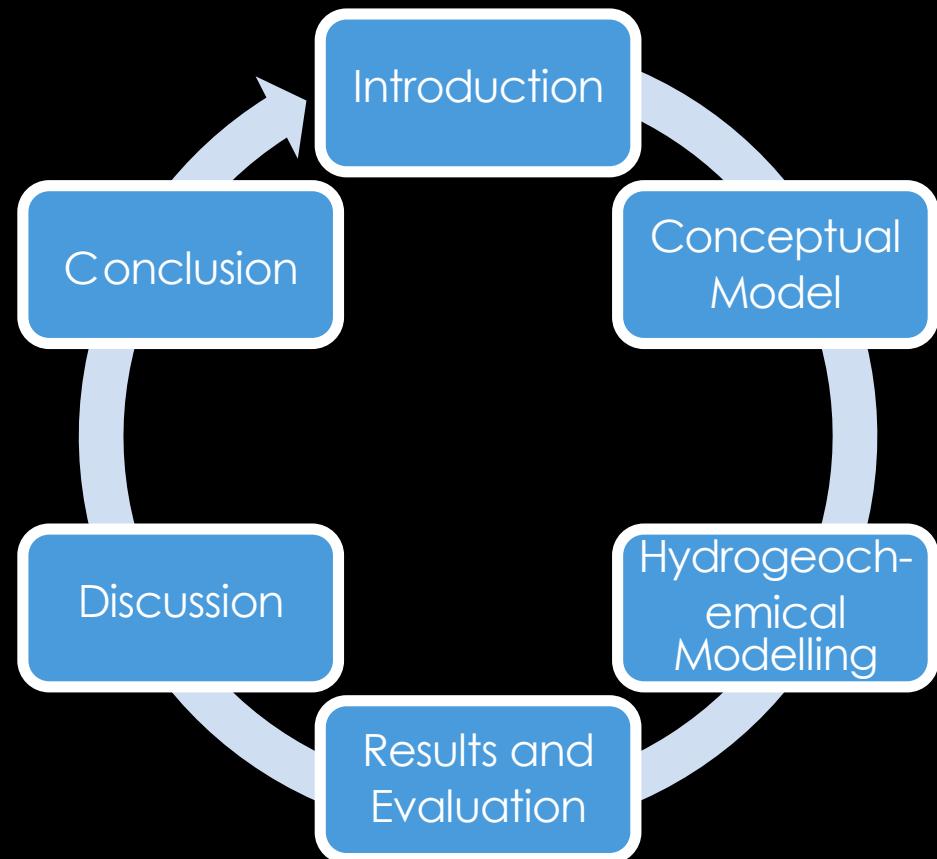


# EFFECT OF MINERAL SCALING ON GEOTHERMAL WELLS

Ullas Rajvanshi  
Bachelor Thesis Defense

# CONTENT



Effect of Mineral Scaling on Geothermal Wells

Ullas Rajvanshi

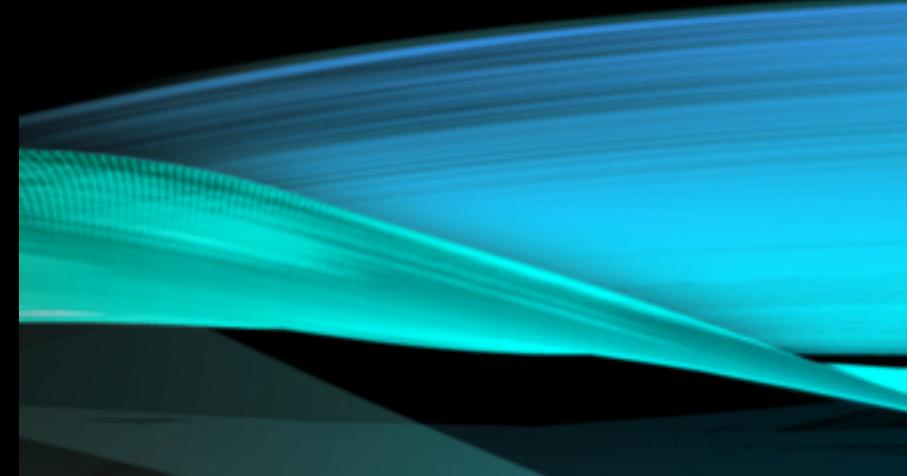
Carbonate Scaling

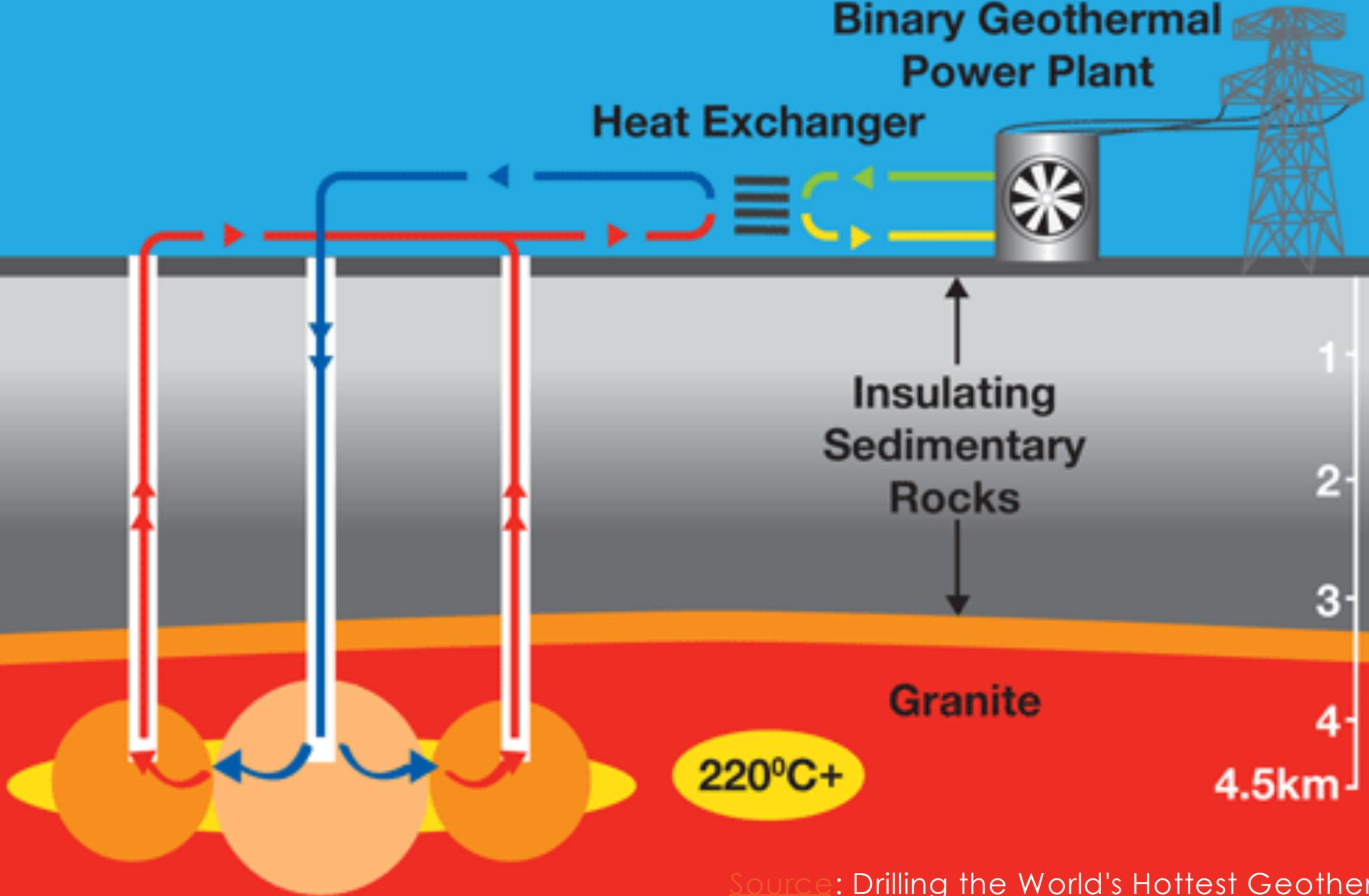


Available here: <https://goo.gl/SUATq1>



## INTRODUCTION





Source: Drilling the World's Hottest Geothermal Well

A large, dark-colored industrial pipe with a prominent valve at the top. The pipe is situated outdoors, with some greenery visible in the background. The lighting is dramatic, highlighting the texture of the pipe and the metallic sheen of the valve.

# SCALING

“Although the output rate of Dieng geothermal power plant [in Indonesia] is 60MW, recently it operates around 42MW. The prime cause of this low output is the stoppage of injection pipe caused by scaling”

Source: Geodipa

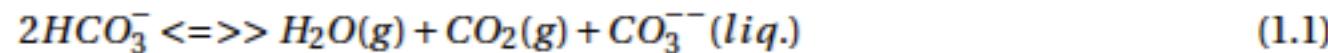
## TYPES OF SCALING

Carbonate

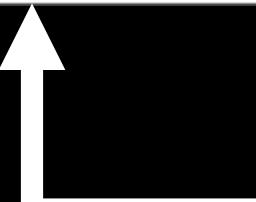
Silica and  
Silicates

Sulphates

# CARBONATE SCALING



$$(Ca^{2+}).(CO_3^{2-}) = K_p \quad (1.2)$$



Ca<sup>2+</sup> concentration depends upon

- Temperature
- Partial Pressure of CO<sub>2</sub>
- Ionic Strength

# Langelier Saturation Index

## SATURATION INDEX

$$SI = \text{pH}_{\text{as}} + Tf + Cf + Af - TDS_f$$

Saturation pH as Temperature Calcium Alkalinity TDS

An index showing whether a water will tend to dissolve or precipitate a particular mineral. Its value is negative when the mineral may be dissolved, positive when it may be precipitated, and zero when the water and mineral are at chemical equilibrium.

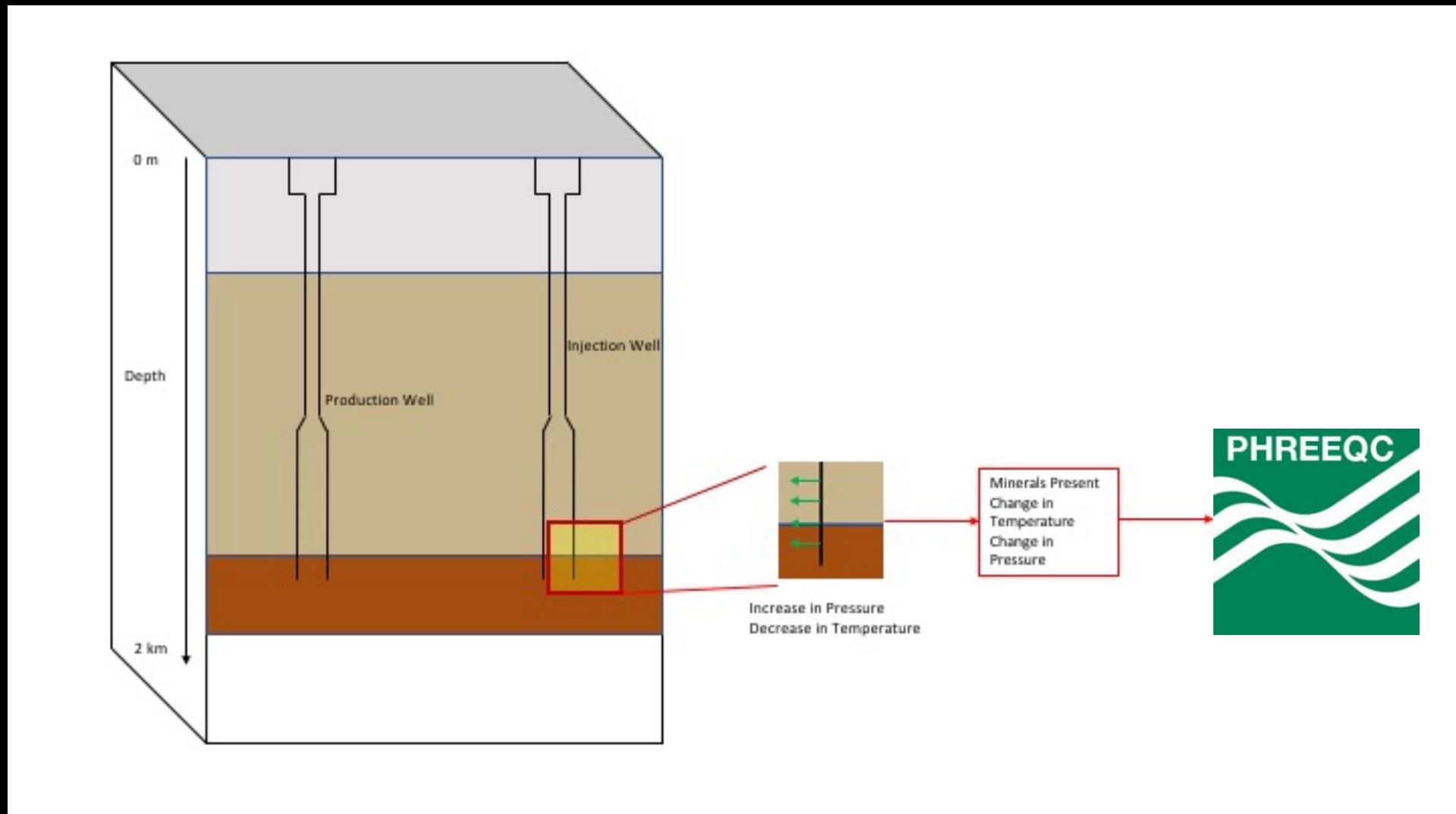
The *ideal* range for balanced water is between 0.3 to +0.3

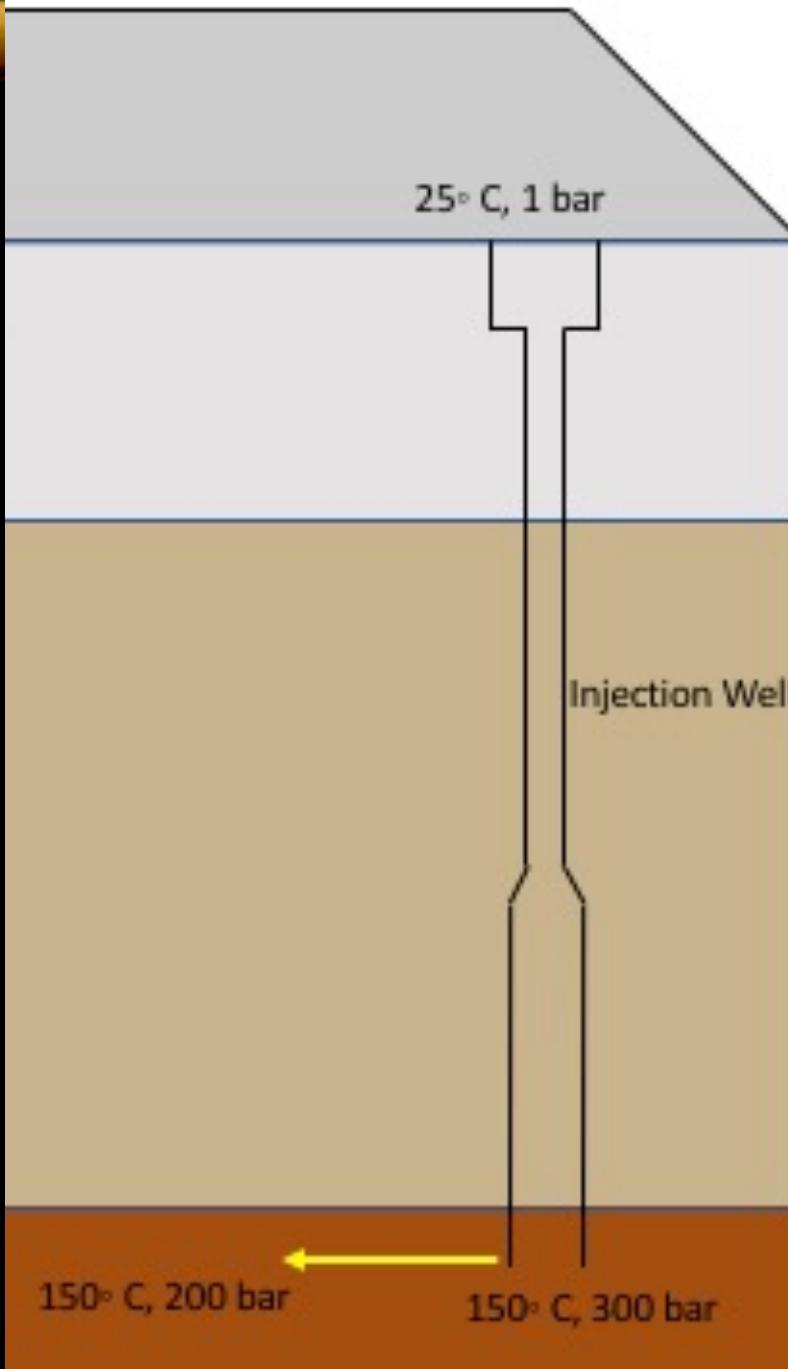
# PREVENTION

- Acting on CO<sub>2</sub> partial pressure
- Acting on pH of the solution
- Using chemical additives



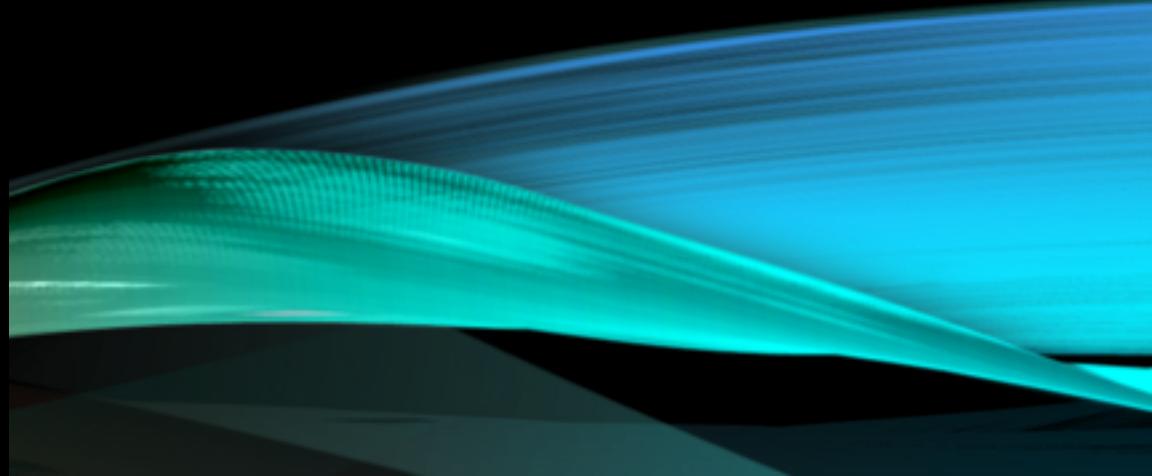
# CONCEPTUAL MODEL





Sample	Concentration (Mg/kg)
Ba	0.25
C	249.59
Ca	1350
Cl	4713 charge
K	74
Fe(2)	0.49
Li	2.44
Mg	18
Mn	1.28
Na	1670
S(6)	95.32
Sr	22
Zn	1.53
Si	1.3

## HYDROGEOCHEMICAL MODELLING



# PHREEQC SIMULATION

```
SELECTED_OUTPUT
-file resultsobtained.sel
-step false
-reaction true
-temperature true
-pressure true
-saturation_indices Aragonite Calcite
Dolomite
```

```
SOLUTION 1 pure water
```

```
pH 7.0
```

```
temp 25.0
```

```
EQUILIBRIUM_PHASES
```

```
Aragonite 0.0
```

```
Calcite 0.0
```

```
Dolomite 0.0
```

```
SAVE solution 1
```

```
END
```

```
SOLUTION 2
```

```
temp 25
```

```
pressure 1
```

```
pH 5.7
```

```
pe 4
```

```
redox pe
```

```
units mg/kgw
```

```
density 1
```

```
Ba 0.25
```

```
C 249.59
```

```
Ca 1350
```

```
Cl 4713 charge
```

```
K 74
```

```
Fe(2) 0.49
```

```
Li 2.44
```

```
Mg 18
```

```
Mn 1.28
```

```
Na 1670
```

```
S(6) 95.32
```

```
Sr 22
```

```
Zn 1.53
```

```
Si 1.3
```

```
-water 1 # kg
```

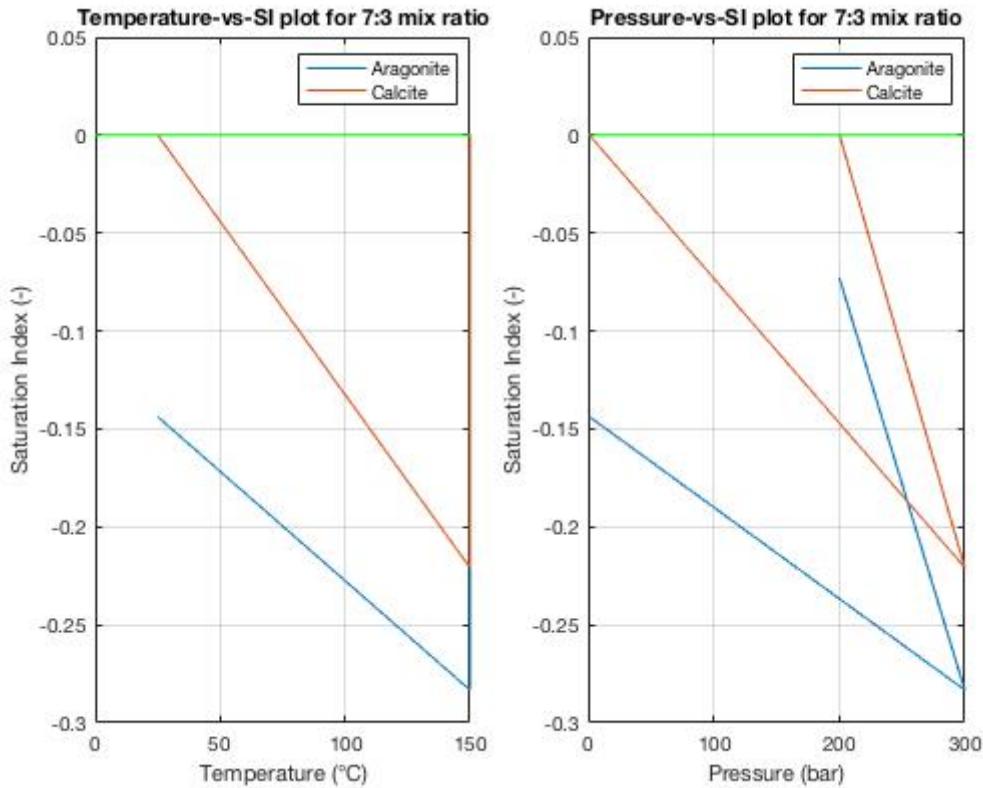
```
SAVE solution 2
```

```
END
```

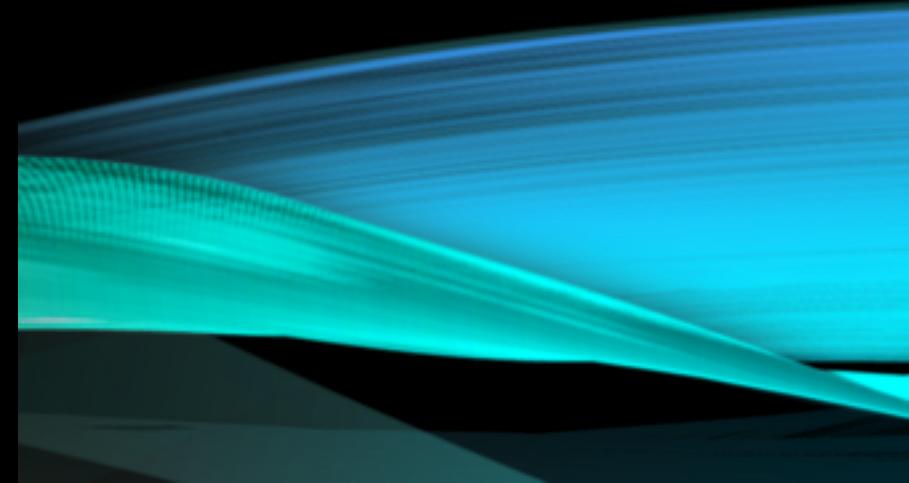
```
MIX 1
1 0.7
2 0.3
SAVE solution 3
END
EQUILIBRIUM_PHASES 1
USE solution 3
REACTION_TEMPERATURE 1
150 150 150 25
REACTION_PRESSURE 1
300 200 200 1
SAVE solution 1
END
```

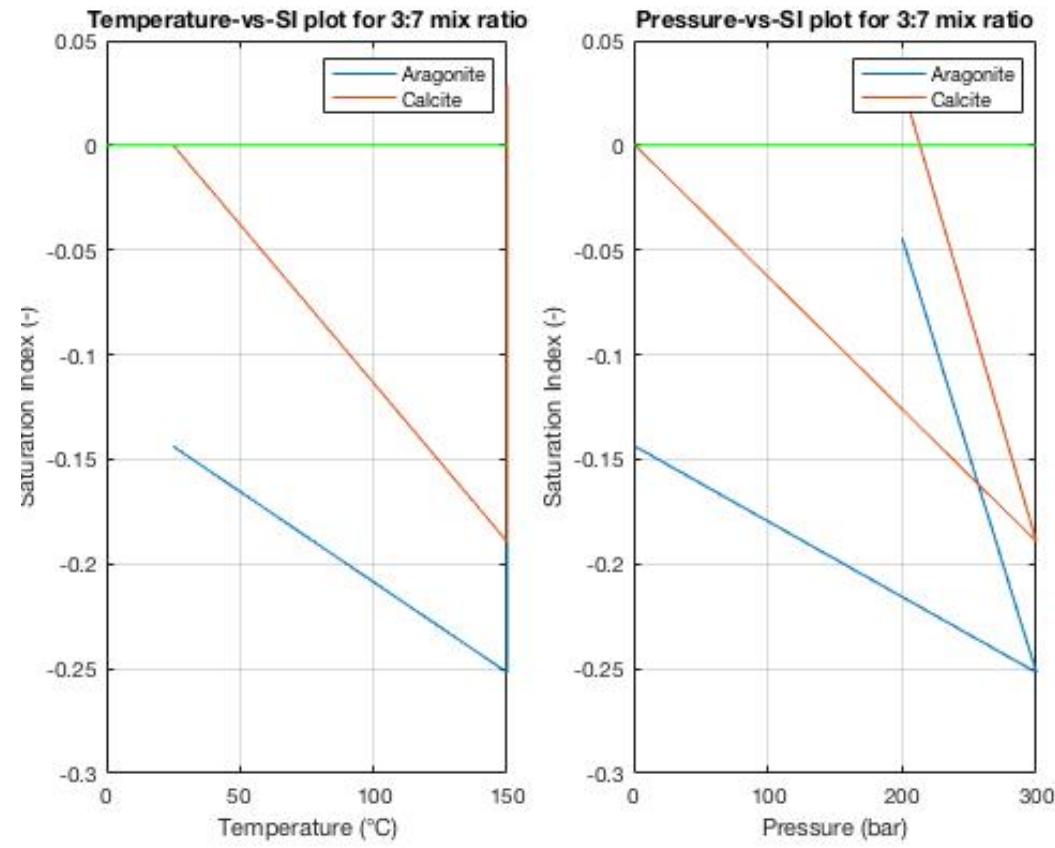
# RESULTS

Mix #	Pure Water (%)	Brine Fluid (%)
1	70	30
2	30	70
3	50	50
4	99	1
5	1	99

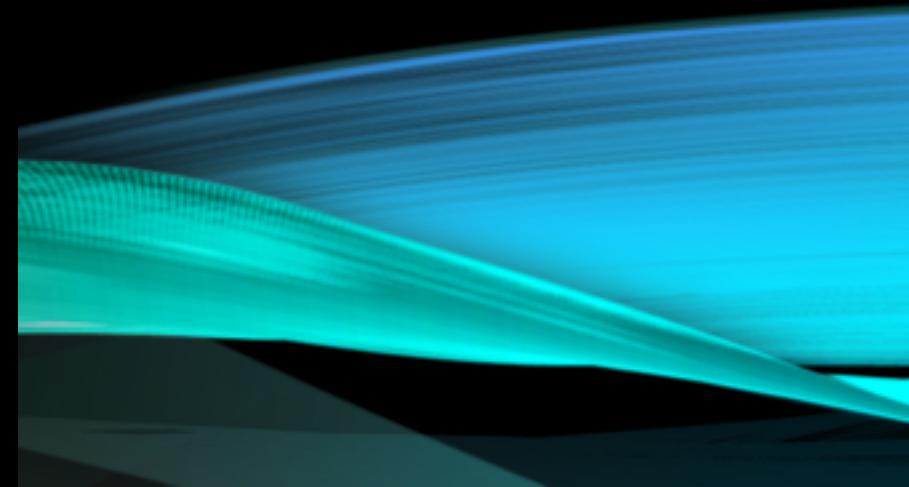


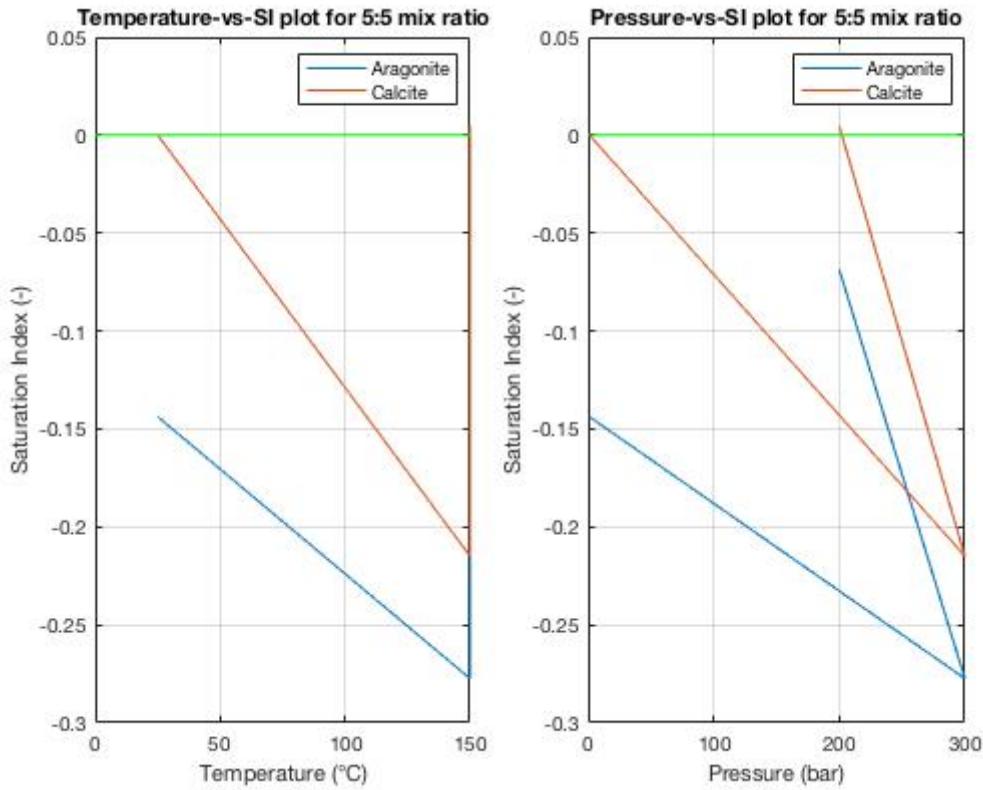
70% PURE  
WATER  
30% BRINE



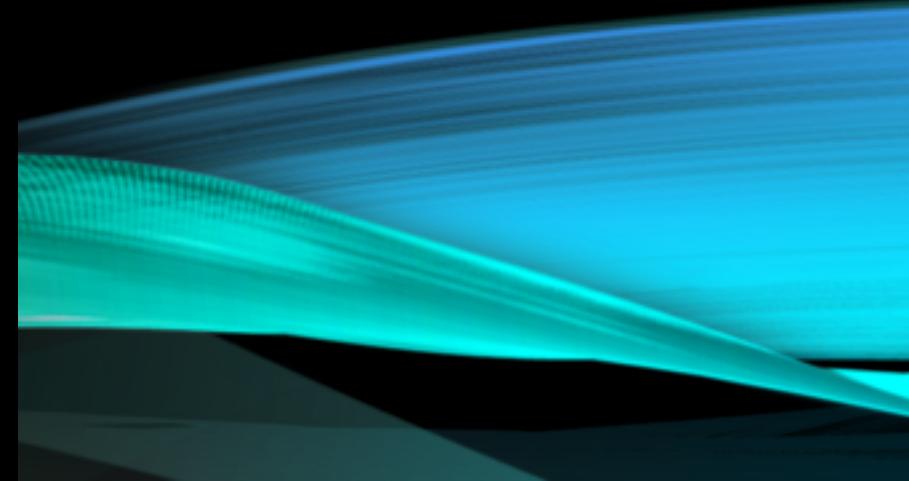


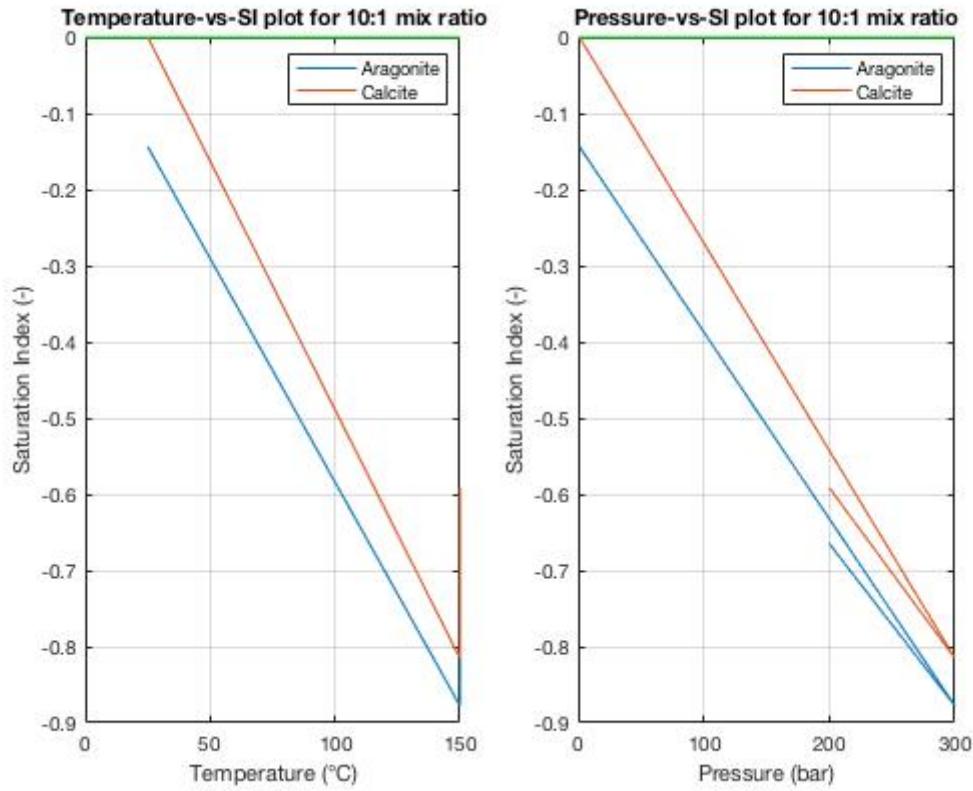
30% PURE  
WATER  
70% BRINE



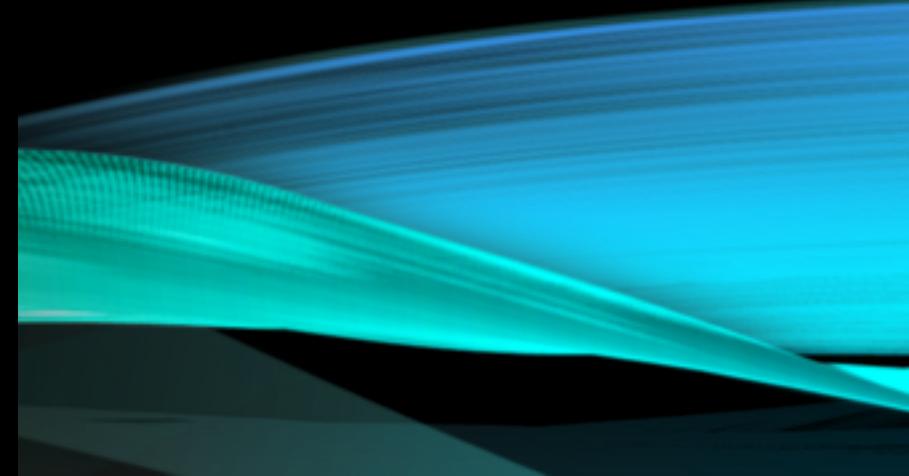


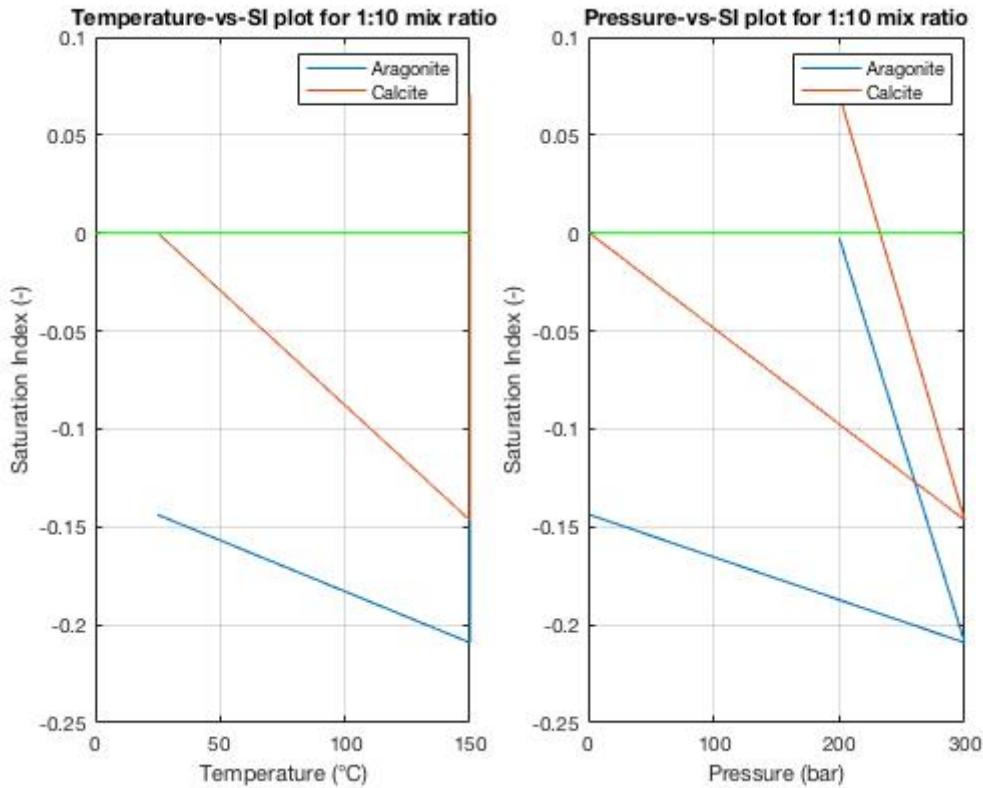
50% PURE  
WATER  
50% BRINE



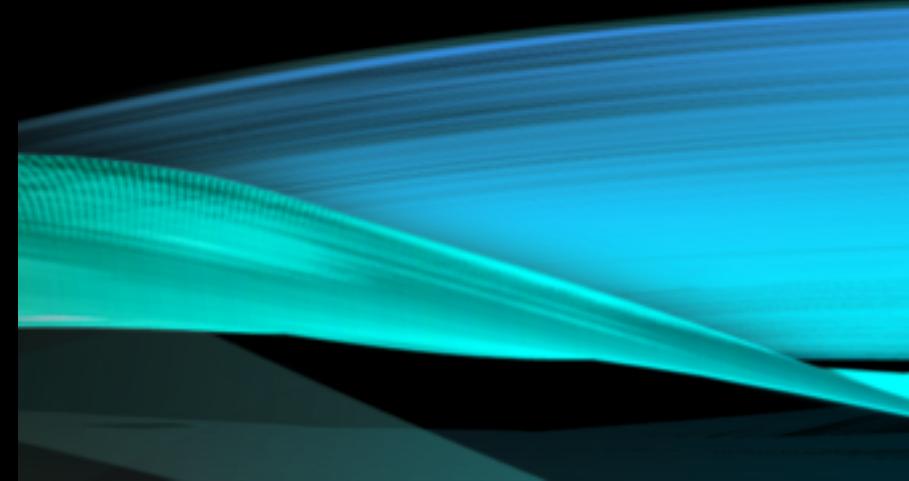


99% PURE  
WATER  
01% BRINE





01% PURE  
WATER  
99% BRINE



# DISCUSSIONS

- Chemical Composition of the fluid.
- Considering scaling in Production Well
- Kinetics of the reaction
- Other types of scaling



*That's all Folks!*

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RATIO- 7:3							
soln	pH	pe	Pressure (bar)	Temperature (Celsius)	si_Aragonite	si_Calcite	si_Dolomite
1	7	4	1	25	-999.999	-999.999	-999.999
1	9.97699	-4.71083	1	25	-0.1438	0	0
2	5.7	4	1	25	-1.4078	-1.2641	-4.0313
1	6.06874	-0.960122	1	25	-1.6342	-1.4904	-4.4183
3	6.12786	-3.39132	300	150	-0.2833	-0.2208	-3.1787
3	6.25578	-3.50899	300	150	-0.0727	0.0006	-2.7522
3	6.25578	-3.50899	200	150	-0.0727	0.0006	-2.7522
3	6.06874	-0.960122	1	25	-1.6342	-1.4904	-4.4183
RATIO 3:7							
soln	pH	pe	pree	temp	si_Aragonite	si_Calcite	si_Dolomite
1	7	4	1	25	-999.999	-999.999	-999.999
1	9.97699	-4.71083	1	25	-0.1438	0	0
2	5.7	4	1	25	-1.4078	-1.2641	-4.0313
1	5.79214	-0.545017	1	25	-1.5035	-1.3597	-4.2121
3	5.81555	-2.9751	300	150	-0.2518	-0.1893	-3.1702
3	5.94492	-3.09472	300	150	-0.0443	0.0289	-2.7489
3	5.94492	-3.09472	200	150	-0.0443	0.0289	-2.7489
3	5.79214	-0.545017	1	25	-1.5035	-1.3597	-4.2121
RATIO 5:5							
soln	pH	pe	pree	temp	si_Aragonite	si_Calcite	si_Dolomite
1	7	4	1	25	-999.999	-999.999	-999.999
1	9.97699	-4.71083	1	25	-0.1438	0	0
2	5.7	4	1	25	-1.4078	-1.2641	-4.0313
1	5.89043	-0.716187	1	25	-1.5735	-1.4297	-4.3359
3	5.93083	-3.12651	300	150	-0.2774	-0.2149	-3.2054
3	6.0596	-3.24568	300	150	-0.0687	0.0046	-2.7821
3	6.0596	-3.24568	200	150	-0.0687	0.0046	-2.7821
3	5.89043	-0.716187	1	25	-1.5735	-1.4297	-4.3359

soln	pH	pe	Pressure (bar)	Temperature (Celsius)	SI_Aragonite	SI_Calcite	SI_Dolomite
RATIO 10:1							
1	7	4	1	25	-999.999	-999.999	-999.999
1	9.97699	-4.71083	1	25	-0.1438	0	0
2	5.7	4	1	25	-1.4078	-1.2641	-4.0313
1	9.97699	-4.71083	1	25	-0.1438	0	0
3	7.59185	-2.69464	300	150	-0.8777	-0.8152	-3.0374
3	7.70296	-2.75269	300	150	-0.6649	-0.5916	-2.6396
3	7.70296	-2.75269	200	150	-0.6649	-0.5916	-2.6396
3	9.97699	-4.71083	1	25	-0.1438	0	0
RATIO 1:10							
soln	pH	pe	pree	temp	si_Aragonite	si_Calcite	si_Dolomite
1	7	4	1	25	-999.999	-999.999	-999.999
1	9.97699	-4.71083	1	25	-0.1438	0	0
2	5.7	4	1	25	-1.4078	-1.2641	-4.0313
1	5.7	-0.198378	1	25	-1.4078	-1.2641	-4.0313
3	5.70234	-2.82956	300	150	-0.2091	-0.1466	-3.0949
3	5.83248	-2.94956	300	150	-0.0026	0.0707	-2.6752
3	5.83248	-2.94956	200	150	-0.0026	0.0707	-2.6752
3	5.7	-0.198378	1	25	-1.4078	-1.2641	-4.0313