Lecture 2

Free particle wave functions

momentum operator in I D

$$\hat{p} = -i\hbar \frac{\partial}{\partial x}$$

wave function

$$\psi_p(x) = |p\rangle = \frac{1}{\sqrt{2\pi}} \exp[i\frac{p}{\hbar}x] = \frac{1}{\sqrt{2\pi}} \exp[ikx] = |k\rangle$$
 note: $\langle k| = \frac{1}{\sqrt{2\pi}} \exp[-ikx]$

superposition principle: wave packet

$$\psi(x) = |x\rangle = \int_{-\infty}^{\infty} f(k) |k\rangle dk = \int_{-\infty}^{\infty} f(k) \exp[ikx] dk$$

time dependent wave packet (in lecture 4)

$$\psi_k(x,t) = |x,t\rangle = \int_{-\infty}^{\infty} f(k) |k\rangle \exp\left[-i\frac{E_k}{\hbar}t\right] dk = \int_{-\infty}^{\infty} f(k) \exp[ikx - i\omega(k)t] dk$$

lectures 5 & 6:

Harmonic oscillator in Schrödinger representation

$$F(x) = -kx$$
 $V(x) = \frac{1}{2}kx^2$ $\omega = \sqrt{\frac{k}{m}}$

important model in chemistry

Infra-red spectroscopy

thermodynamics & kinetics

boundary conditions

wave function needs to be zero at infinity

two step solution strategy

find solution for $x \to \infty$

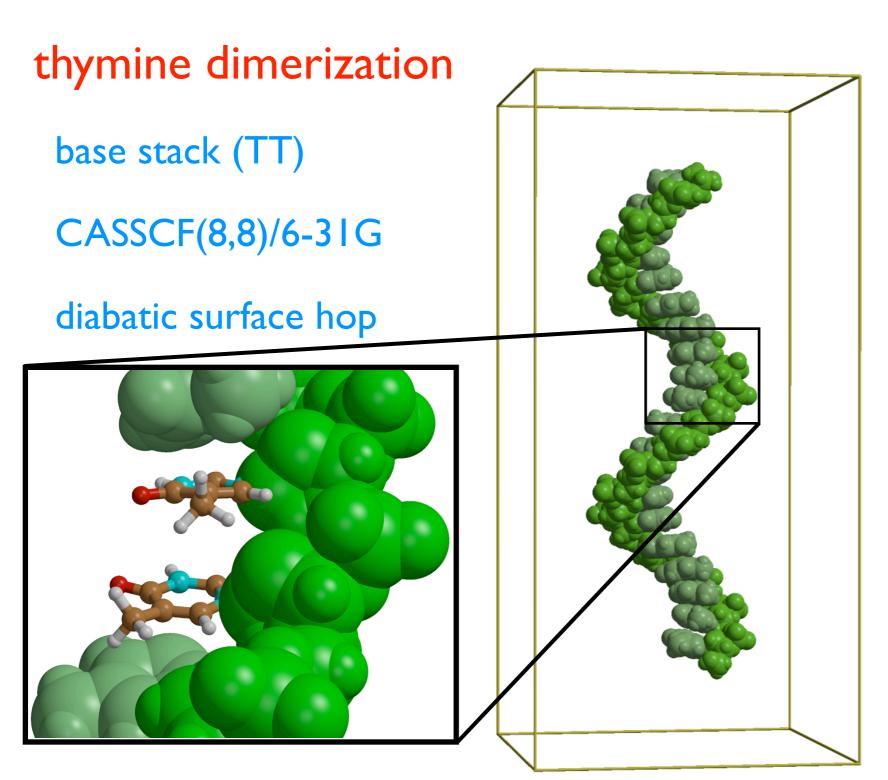
multiply by a polynomial

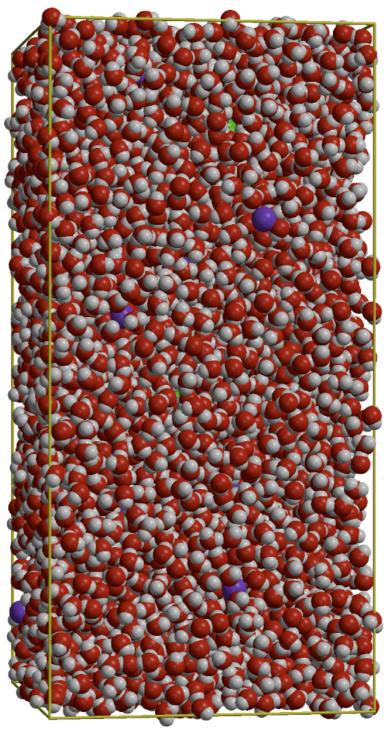
use boundary conditions to set limits on the polynomial

$$E_n = (n + \frac{1}{2})\hbar\omega$$

photochemistry

example: radiation damage in DNA



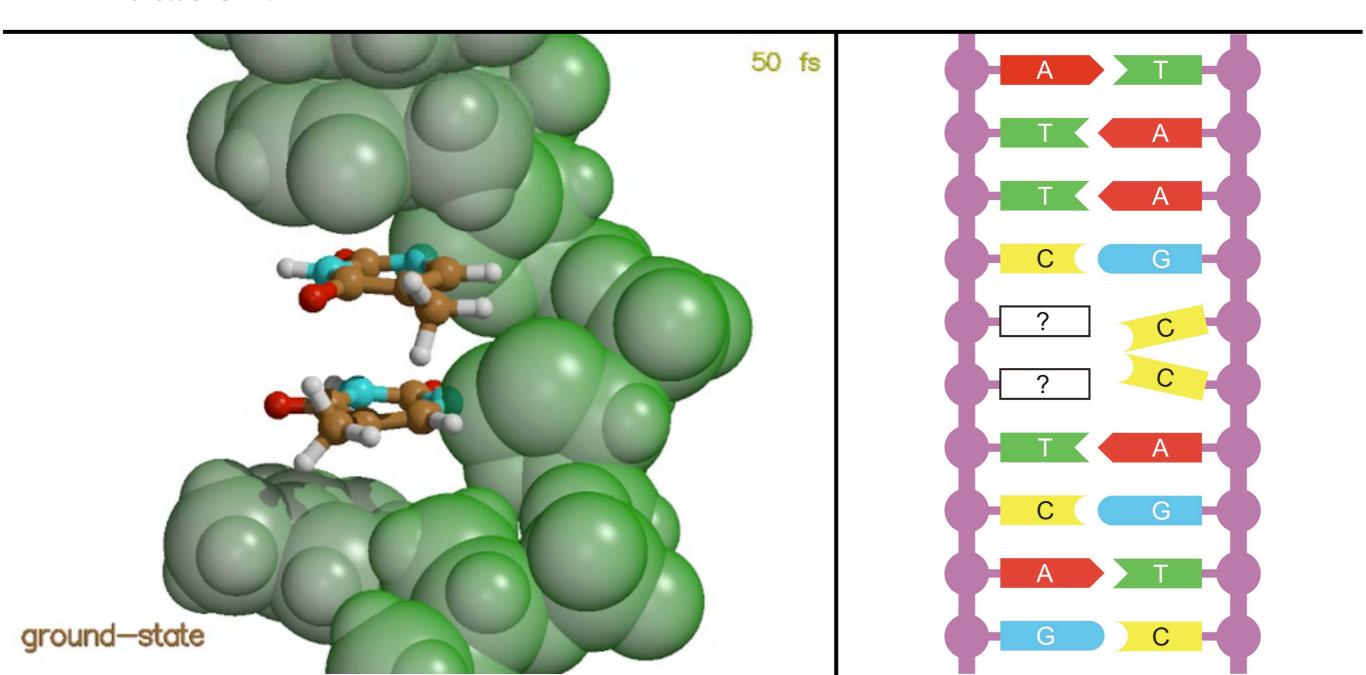


Radiation damage: UV absorption in DNA

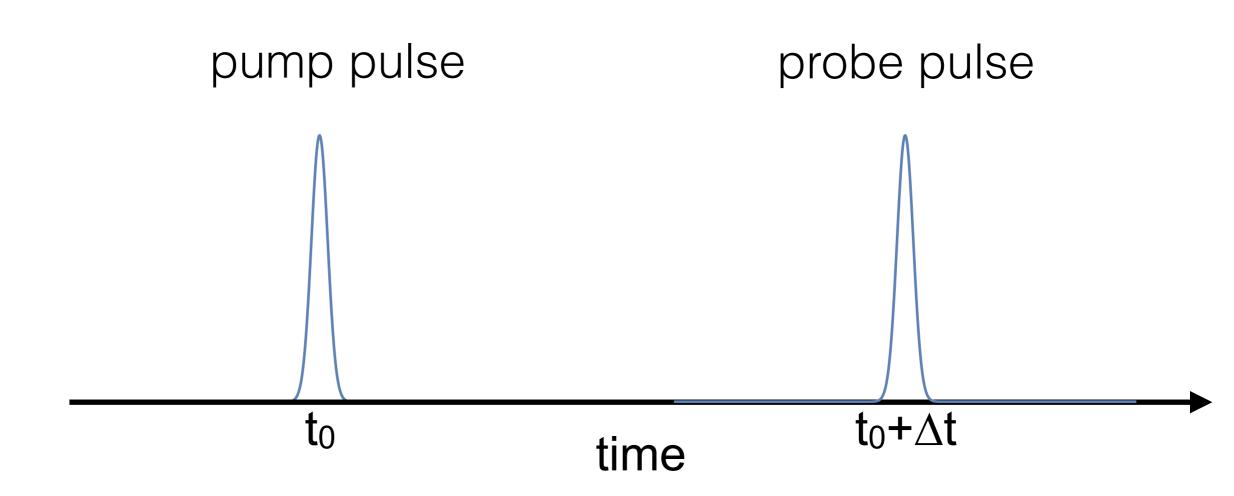
thymine dimerization

cell dead?

mutation?

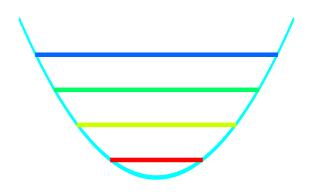


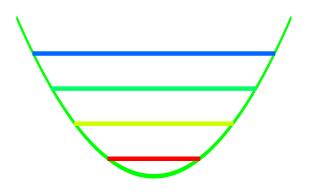
linear pump-probe spectroscopy



some issues (out of many)

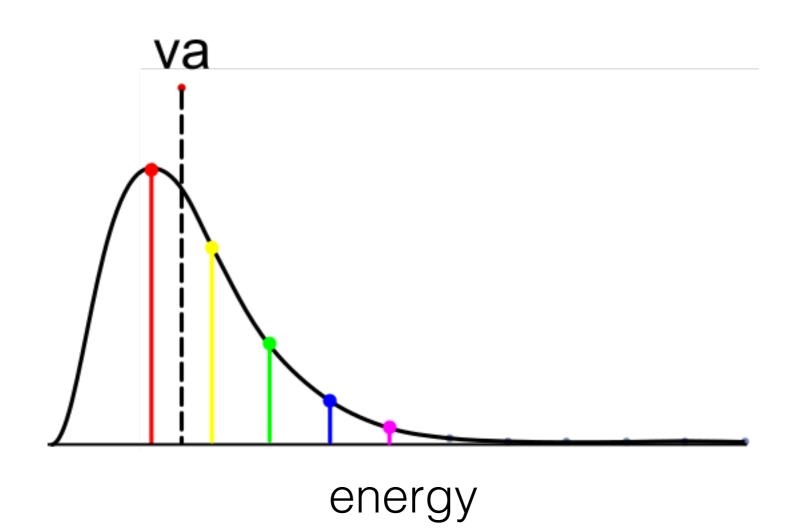
fs pump pulses have large bandwidth

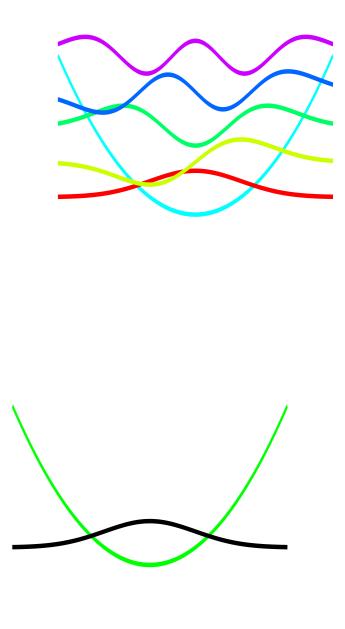




some issues (out of many)

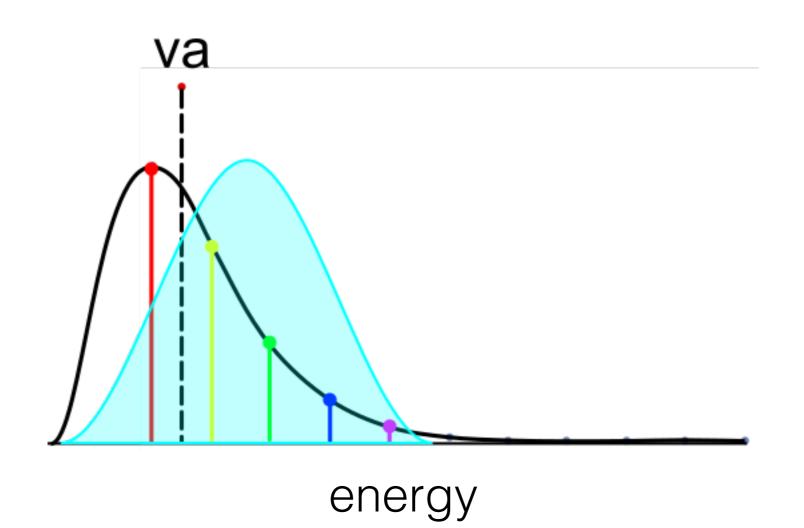
fs pump pulses have large bandwidth

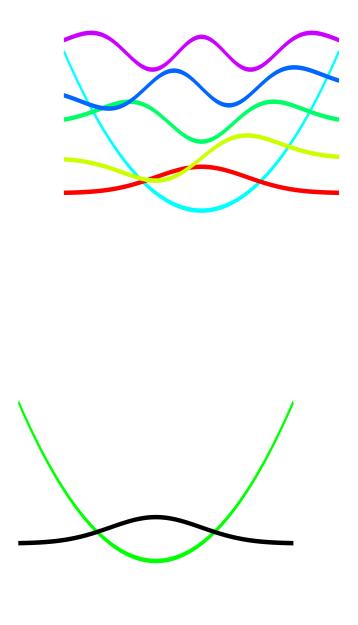




some issues (out of many)

fs pump pulses have large bandwidth

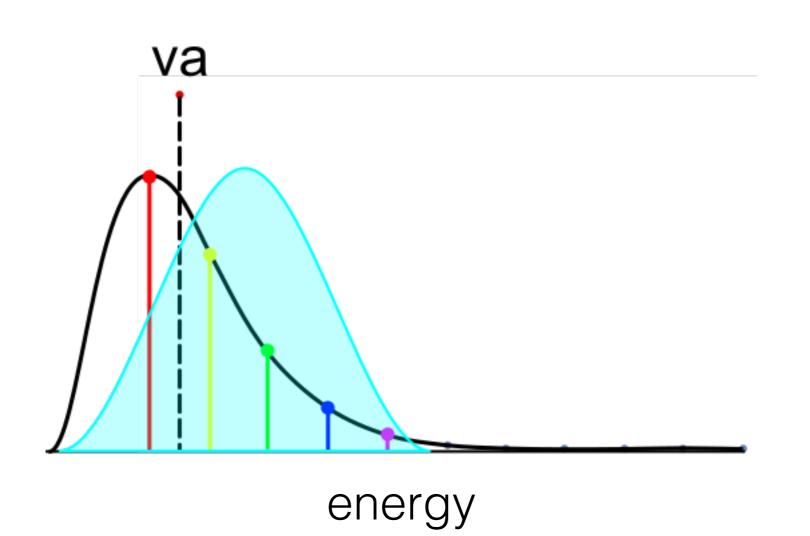


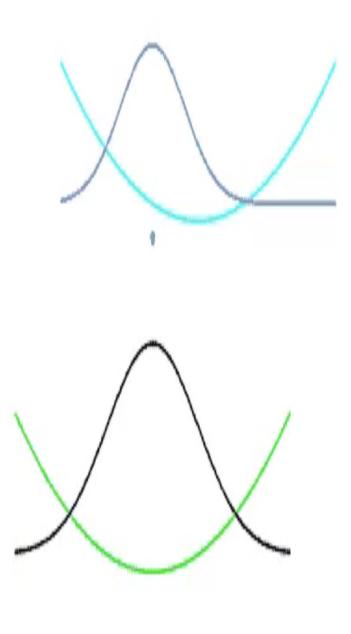


some issues (out of many)

fs pump pulses have large bandwidth

coherent vibrations in ground and excited state

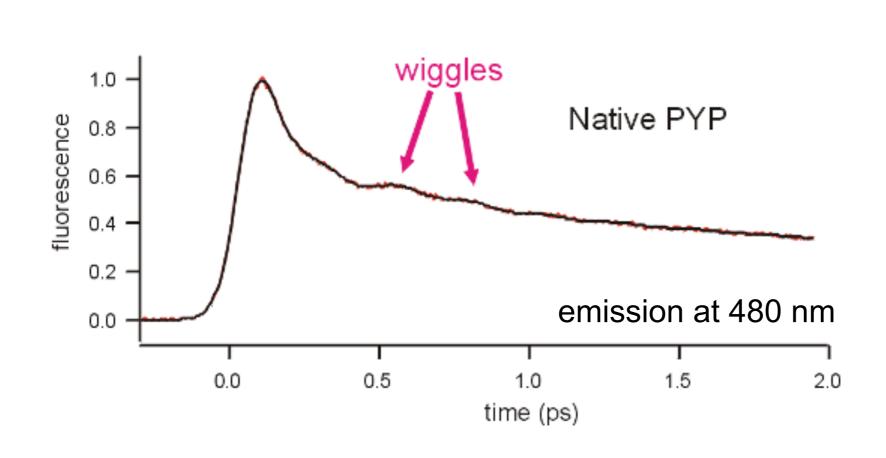


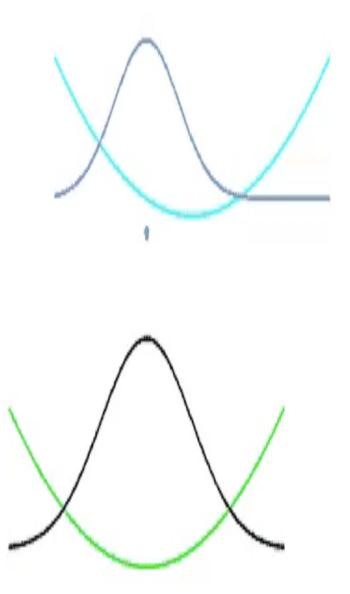


some issues (out of many)

fs pump pulses have large bandwidth

coherent vibrations in ground and excited state





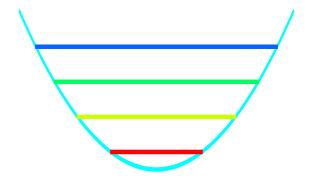
Chem. Phys. Lett. 352 (2002) 220-225

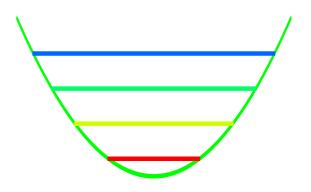
some issues (out of many)

fs pump pulses have large bandwidth

coherent vibrations in ground and excited state

stimulated emission limits population transfer



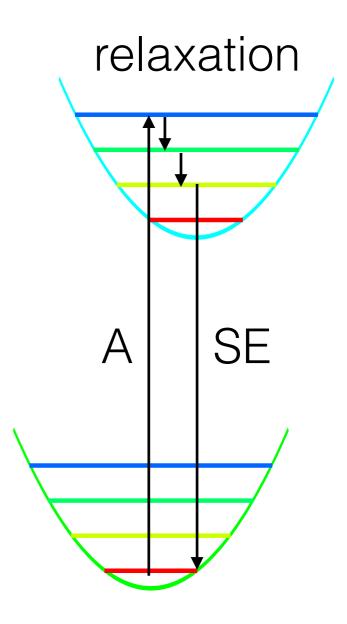


some issues (out of many)

fs pump pulses have large bandwidth

coherent vibrations in ground and excited state

stimulated emission limits population transfer



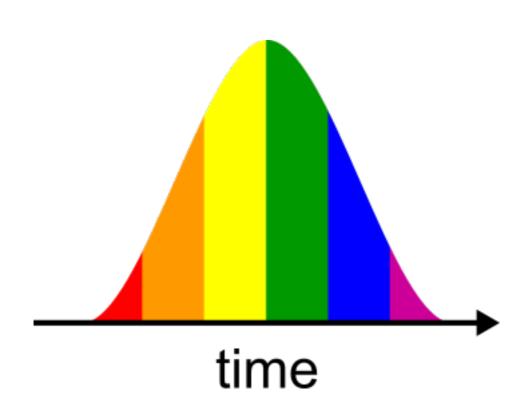
some issues (out of many)

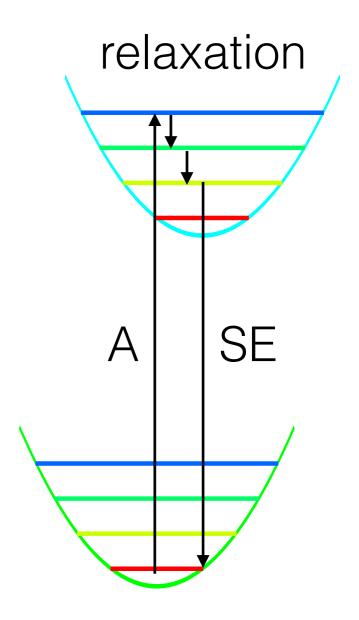
fs pump pulses have large bandwidth

coherent vibrations in ground and excited state

stimulated emission limits population transfer

chirped pulses





lecture 6:

Harmonic oscillator in Dirac Representation

ladder operators

atomic units: $\hbar = m = k = 1$

$$\hat{a} = \frac{i}{\sqrt{2}} \left(\hat{p} - i\hat{x} \right) \qquad \hat{a}^{+} = \frac{1}{i\sqrt{2}} \left(\hat{p} + i\hat{x} \right) \qquad \hat{a}\hat{a}^{+} = \hat{H} + \frac{1}{2} \qquad \hat{a}^{+}\hat{a} = \hat{H} - \frac{1}{2}$$

$$\hat{H} = \frac{1}{2} \left(\hat{a}\hat{a}^{\dagger} + \hat{a}^{\dagger}\hat{a} \right)$$

commutation relations

$$[\hat{a}, \hat{a}^+] = 1$$
$$[\hat{a}, \hat{H}] = \hat{a}$$
$$[\hat{a}^+, \hat{H}] = \hat{a}^+$$

raising and lowering operator

$$\hat{a}^{+}|n\rangle = \sqrt{n+1}|n+1\rangle$$
 $\hat{a}|n\rangle = \sqrt{n}|n-1\rangle$

lecture 6:

Harmonic oscillator in Dirac Representation

ladder operators

$$\hat{a}^{+}|n\rangle = \sqrt{n+1}|n+1\rangle$$
 $\hat{a}|n\rangle = \sqrt{n}|n-1\rangle$

$$\hat{a} | n \rangle = \sqrt{n} | n - 1 \rangle$$

number operator

$$\hat{a}^{\dagger}\hat{a} | n \rangle = n | n \rangle$$

real units

$$\hat{H} = \frac{1}{2}\hbar\omega(\hat{a}\hat{a}^+ + \hat{a}^+\hat{a})$$

$$\hat{a} = \frac{i}{\sqrt{2\hbar\omega}} \left(\frac{1}{\sqrt{m}} \hat{p} - i\sqrt{k}\hat{x} \right)$$

$$\hat{a} = \frac{i}{\sqrt{2\hbar\omega}} \left(\frac{1}{\sqrt{m}} \hat{p} - i\sqrt{k}\hat{x} \right) \qquad \hat{a}^{+} = \frac{1}{i\sqrt{2\hbar\omega}} \left(\frac{1}{\sqrt{m}} \hat{p} + i\sqrt{k}\hat{x} \right)$$

$$\hat{x} = \sqrt{\frac{\hbar\omega}{2k}} \left(\hat{a} + \hat{a}^+ \right)$$

$$\hat{p} = -i\sqrt{\frac{\hbar\omega m}{2}} \left(\hat{a} - \hat{a}^{+}\right)$$

lecture 7:

Particle on ring

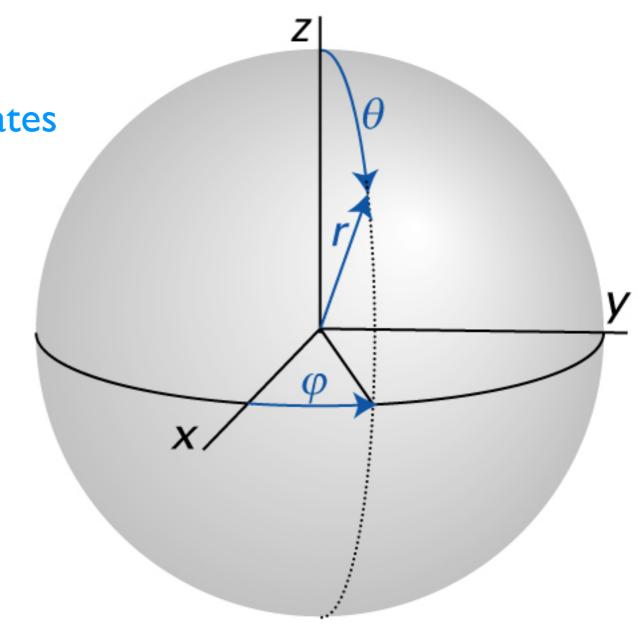
Hamiltonian in Cartesian coordinates

$$\hat{H} = -\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} \right)$$

polar coordinates (2D)

$$x = r \cos \varphi$$

$$y = r \sin \varphi$$



Hamiltonian in polar coordinates

$$\hat{H} = -\frac{\hbar^2}{2m} \left\{ \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \varphi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right\}$$

lecture 7/8:

Particle on sphere

Hamiltonian in Cartesian coordinates

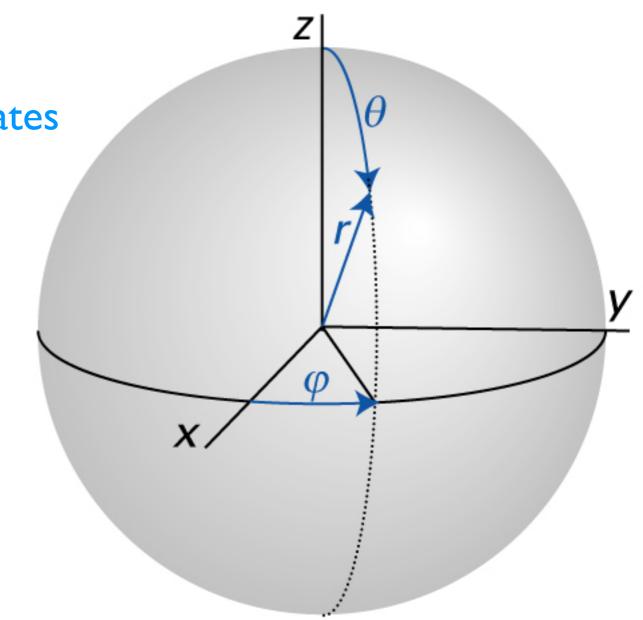
$$\hat{H} = -\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right)$$

spherical coordinates (3D)

$$x = r \sin \theta \cos \varphi$$

$$y = r \sin \theta \sin \varphi$$

$$z = r \cos \theta$$

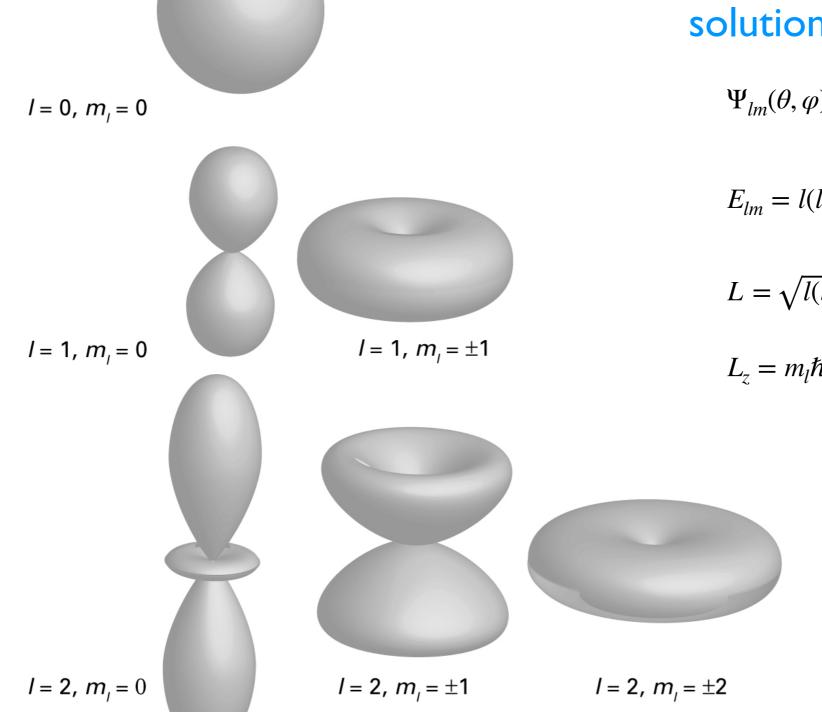


Hamiltonian in spherical coordinates

$$\hat{H} = -\frac{\hbar^2}{2m} \left\{ \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \varphi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right\}$$

lecture 8:

 $l = n' + |m_l|$



Particle on sphere

solutions: spherical harmonics

$$\Psi_{lm}(\theta, \varphi) = \Theta_{lm}(\theta)\Phi_{m}(\varphi)$$

$$E_{lm} = l(l+1)\frac{\hbar}{2I}$$
 2l+1 degeneracies

$$L = \sqrt{l(l+1)}\hbar$$

$$L_z = m_l \hbar \qquad m_l \in [-l, l]$$

lecture 8/9:

Hydrogen atom

Hamiltonian of electron and proton (6D)

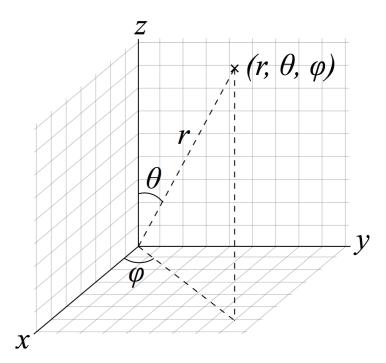
$$\hat{H} = -\frac{\hbar^2}{2m_p} \left(\frac{\partial^2}{\partial x_p^2} + \frac{\partial^2}{\partial y_p^2} + \frac{\partial^2}{\partial z_p^2} \right) - \frac{\hbar^2}{2m_e} \left(\frac{\partial^2}{\partial x_e^2} + \frac{\partial^2}{\partial y_e^2} + \frac{\partial^2}{\partial z_e^2} \right) - \frac{Ze^2}{4\pi\epsilon_0 \sqrt{(x_e - x_p)^2 + (y_e - y_p)^2 + (z_e - z_p)^2}}$$

center of mass coordinates (3D)

$$X = \frac{m_p x_p + m_e x_e}{m_p + m_e} \qquad Y = \frac{m_p y_p + m_e y_e}{m_p + m_e} \qquad Z = \frac{m_p z_p + m_e z_e}{m_p + m_e}$$

spherical coordinates (3D)

$$x = x_e - x_p = r \sin \theta \cos \phi$$
$$y = y_e - y_p = r \sin \theta \sin \phi$$
$$z = z_e - z_p = r \cos \theta$$



lecture 8/9:

Hydrogen atom

Hamiltonian of electron and proton (6D)

$$\hat{H} = \hat{H}_{COM}(X, Y, Z) + \hat{H}_{int}(x, y, z)$$

free particle: COM

$$\hat{H}_{COM} = -\frac{\hbar^2}{2(m_p + m_2)} \left(\frac{\partial^2}{\partial X^2} + \frac{\partial^2}{\partial Y^2} + \frac{\partial^2}{\partial Z^2} \right)$$

internal

$$\hat{H}_{\text{int}} = -\frac{\hbar^2}{2\mu} \left\{ \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right\} + V(r)$$

reduced mass

$$\mu = \frac{m_p m_e}{m_p + m_e}$$

lecture 8/9:

Hydrogen atom

free particle: COM

$$\hat{H}_{COM} = -\frac{\hbar^2}{2(m_p + m_2)} \left(\frac{\partial^2}{\partial X^2} + \frac{\partial^2}{\partial Y^2} + \frac{\partial^2}{\partial Z^2} \right)$$

$$\hat{H}_{COM}\Psi_{COM}(X, Y, Z) = E_{COM}\Psi_{COM}(X, Y, Z)$$

internal

$$\hat{H}_{\text{int}} = -\frac{\hbar^2}{2\mu} \left\{ \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2}{\partial \phi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right\} + V(r)$$

$$\hat{H}_{int}\Psi_{int}(r,\theta,\phi) = E_{int}\Psi_{int}(r,\theta,\phi)$$

total wave function

$$\Psi(x, y, z, r, \theta, \phi) = \Psi_{COM} \cdot \Psi_{int}$$

lecture 8/9:

Hydrogen atom

internal

$$\hat{H}_{\text{int}} = -\frac{\hbar^2}{2\mu} \left\{ \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial}{\partial \phi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) \right\} - \frac{Ze^2}{4\pi\epsilon_0 r}$$

$$\hat{H}_{int}\Psi_{int}(r,\theta,\phi) = E_{int}\Psi_{int}(r,\theta,\phi)$$

$$\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \Psi_{\text{int}}}{\partial r} \right) + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \Psi_{\text{int}}}{\partial \phi^2} + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \Psi_{\text{int}}}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) \Psi_{\text{int}} = 0$$

product of ID functions:

$$\Psi_{\text{int}}(r, \theta, \phi) = \Phi(\phi)\Theta(\theta)R(r)$$

lecture 8/9:

Hydrogen atom

product of ID functions:

$$\Psi_{\text{int}}(r, \theta, \phi) = \Phi(\phi)\Theta(\theta)R(r)$$

divide:

$$\frac{1}{\Psi_{\text{int}}} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial \Psi_{\text{int}}}{\partial r} \right) + \frac{1}{\Psi_{\text{int}}} \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \Psi_{\text{int}}}{\partial \phi^2} + \frac{1}{\Psi_{\text{int}}} \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \Psi_{\text{int}}}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) = 0$$

$$\frac{1}{R(r)} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R(r)}{\partial r} \right) + \frac{1}{\Phi(\phi)} \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \Phi(\phi)}{\partial \phi^2} + \frac{1}{\Theta(\theta)} \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \Theta(\theta)}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) = 0$$

lecture 8/9:

Hydrogen atom

product of ID functions:

$$\Psi_{\text{int}}(r, \theta, \phi) = \Phi(\phi)\Theta(\theta)R(r)$$

$$\frac{1}{R(r)} \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R(r)}{\partial r} \right) + \frac{1}{\Phi(\phi)} \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 \Phi(\phi)}{\partial \phi^2} + \frac{1}{\Theta(\theta)} \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial \Theta(\theta)}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) = 0$$

divide, multiply, re-arrange etc.

$$\sin^2\theta \frac{1}{R(r)} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R(r)}{\partial r} \right) + \frac{1}{\Phi(\phi)} \frac{\partial^2 \Phi(\phi)}{\partial \phi^2} + \frac{1}{\Theta(\theta)} \sin\theta \frac{\partial}{\partial \theta} \left(\sin\theta \frac{\partial \Theta(\theta)}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} r^2 \sin^2\theta \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) = 0$$

equation for $\Phi(\phi)$

$$\frac{1}{\Phi(\phi)} \frac{\partial^2 \Phi(\phi)}{\partial \phi^2} = -m^2$$

$$\Phi_m(\phi) = \frac{1}{\sqrt{2\pi}} e^{im\phi}$$
 $m = 0, \pm 1, \pm 2, \pm 3,...$

lecture 8/9:

Hydrogen atom

$$\sin^2\theta \frac{1}{R(r)} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R(r)}{\partial r} \right) - m^2 + \frac{1}{\Theta(\theta)} \sin\theta \frac{\partial}{\partial \theta} \left(\sin\theta \frac{\partial \Theta(\theta)}{\partial \theta} \right) + \frac{2\mu}{\hbar^2} r^2 \sin^2\theta \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) = 0$$

divide, multiply, re-arrange etc. once more

$$\frac{1}{R(r)}\frac{\partial}{\partial r}\left(r^2\frac{\partial R(r)}{\partial r}\right) - \frac{m^2}{\sin^2\theta} + \frac{1}{\Theta(\theta)}\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial\Theta(\theta)}{\partial\theta}\right) + \frac{2\mu}{\hbar^2}r^2\left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r}\right) = 0$$

equation for $\Theta(\theta)$

$$\frac{1}{\Theta(\theta)} \frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) - \frac{m^2}{\sin^2 \theta} = -\beta$$

$$\Theta_{lm}(\theta) = (1 - z^2)^{\frac{|m|}{2}} G(z) \qquad z = \cos \theta$$

$$\beta = l(l+1)$$
 $l = \nu + |m|$ $\nu = 0,1,2,3,4,...$

lecture 9:

Hydrogen atom

$$\frac{1}{R(r)}\frac{\partial}{\partial r}\left(r^2\frac{\partial R(r)}{\partial r}\right) - l(l+1) + \frac{2\mu}{\hbar^2}r^2\left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r}\right) = 0$$

divide, multiply, re-arrange etc. for the last time

$$\frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial R(r)}{\partial r} \right) + \left[-\frac{l(l+1)}{r^2} + \frac{2\mu}{\hbar^2} \left(E_{\text{int}} + \frac{Ze^2}{4\pi\epsilon_0 r} \right) \right] = 0$$

equation for R(r)

$$R_{nl}(r) = e^{-\frac{1}{2}\rho} \rho^l L(\rho) \quad \rho = 2\alpha r \qquad \alpha^2 = -\frac{2\mu E}{\hbar^2} \qquad E_n = -\frac{\mu Z^2 e^4}{8\epsilon_0^2 h^2 n^2}$$

energies

$$E_n = -\frac{\mu Z^2 e^4}{8\epsilon_0^2 h^2 n^2} \qquad n = 1, 2, 3, \dots \quad l = 0, 1, \dots, n - 2, n - 1 \quad m = -l, -(l - 1), \dots, 0, \dots (l - 1), l$$

$$n = s, p, d, \dots$$

lecture 9:

Hydrogen atom

$$\Phi_m(\phi) = \frac{1}{2\pi} e^{im\phi}$$

$$\Theta_{lm}(\theta) = \sqrt{\frac{(2l+1)}{2} \frac{(l-|m|)!}{(l+|m|)!}} P_l^{|m|}(\cos \theta)$$

$$R_{nl}(r) = -\sqrt{\left(\frac{2Z}{na_0}\right)^3} \frac{(n-l-1)!}{2n[(n+l)!]^3} e^{-\rho/2} \rho^l L_{n+l}^{2l+1}(\rho) \qquad \rho = 2\alpha r \qquad \alpha^2 = -\frac{2\mu E}{\hbar^2}$$

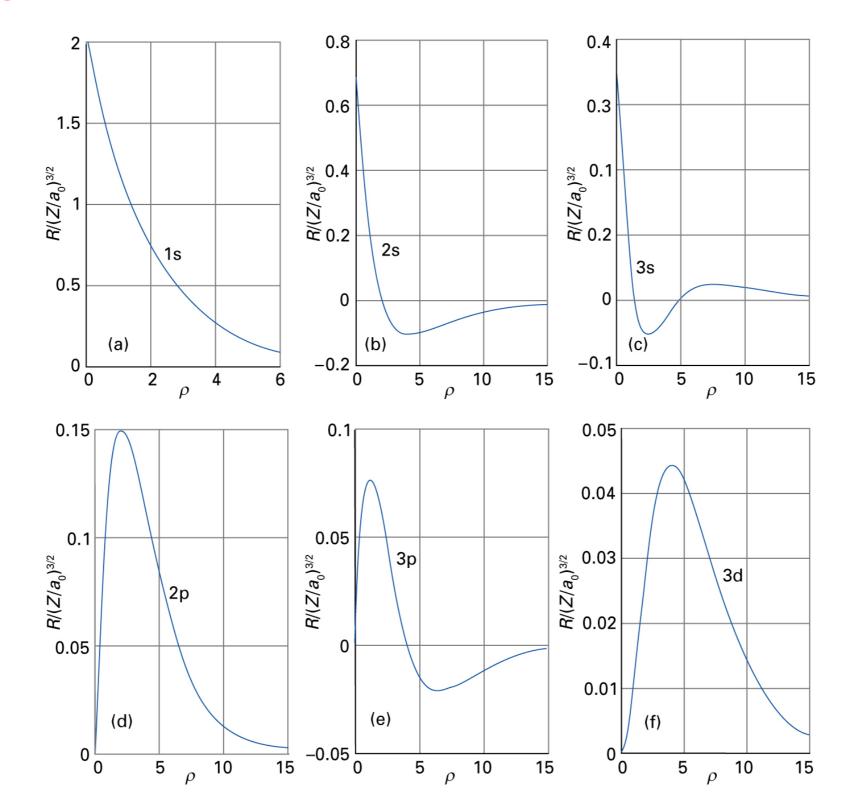
$$E_n = -\frac{\mu Z^2 e^4}{8\epsilon_0^2 h^2 n^2}$$

IS orbital

$$\psi_{100}(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$$

lecture 9:

Hydrogen atom



$$\rho = 2\alpha r$$

$$\alpha^2 = -\frac{2\mu E}{\hbar^2}$$

$$E_n = -\frac{\mu Z^2 e^4}{8\epsilon_0^2 h^2 n^2}$$

$$a_0 = \frac{\epsilon_0 h^2}{\pi \mu e^2}$$

lecture 9:

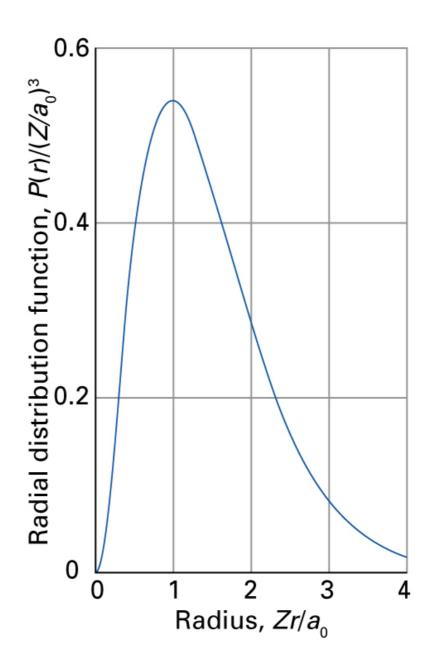
Hydrogen atom

Is radial distribution function

$$\psi_{100}(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$$

$$P(r)dr = 4\pi r^2 \psi_{100}^*(r) \psi_{100}(r)$$

$$P(r) = \frac{4r^2}{a_0^3} e^{-2r/a_0}$$



$$\rho = 2\alpha r$$

$$\alpha^2 = -\frac{2\mu E}{\hbar^2}$$

$$E_n = -\frac{\mu Z^2 e^4}{8\epsilon_0^2 h^2 n^2}$$

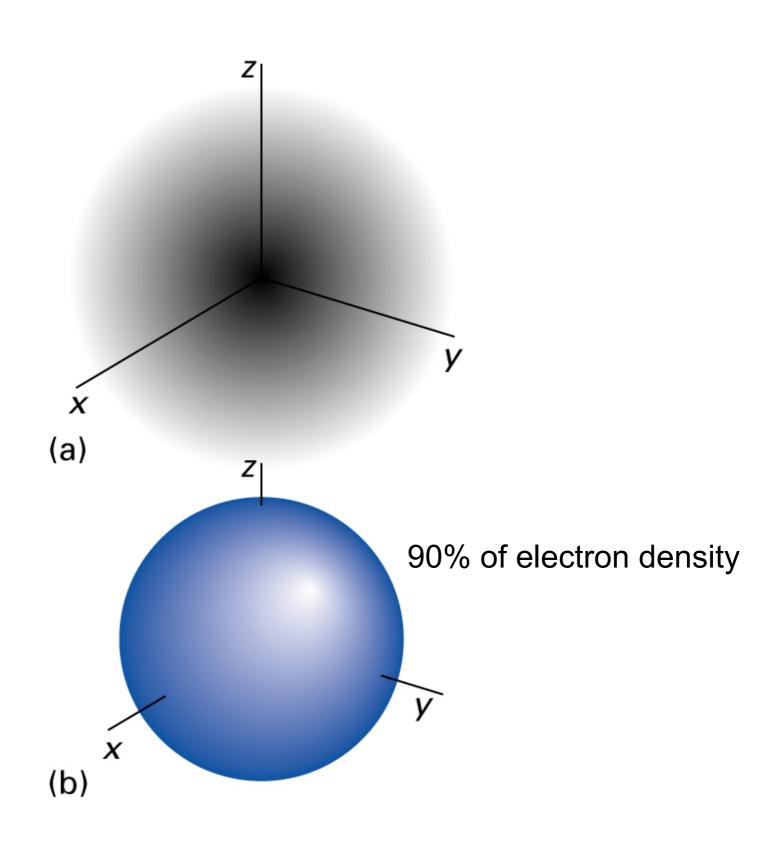
$$a_0 = \frac{\epsilon_0 h^2}{\pi \mu e^2}$$

lecture 9:

Hydrogen atom

Is orbital

$$\psi_{100}(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$$



lecture 9:

$x = x_e - x_p = r \sin \theta \cos \phi$

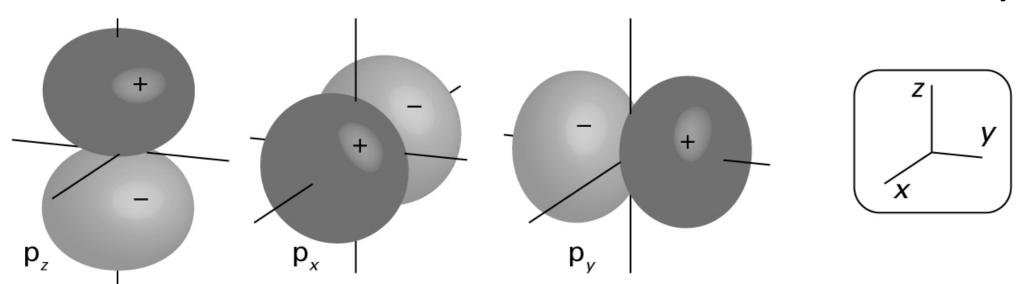
Hydrogen atom

$$y = y_e - y_p = r \sin \theta \sin \phi$$

$$z = z_e - z_p = r \cos \theta$$

p orbitals (3)
$$n > 1$$

90% of electron density



$$Y_{l=1,m=0}(\theta,\varphi)$$
 $p_0 = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} R_{n1}(r)\cos\theta$

$$p_0 = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} R_{n1}(r) \cos \theta$$
 $p_z = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} R_{n1}(r) \cos \theta$

$$Y_{l=1,m=1}(\theta,\varphi)$$
 $p_{+1} = -\left(\frac{3}{8\pi}\right)^{\frac{1}{2}} R_{n1}(r)\sin\theta e^{i\varphi}$

$$p_{+1} = -\left(\frac{3}{8\pi}\right)^{\frac{1}{2}} R_{n1}(r) \sin\theta e^{i\varphi} \qquad p_x = \frac{1}{\sqrt{2}} (p_{-1} - p_{+1}) = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} R_{n1}(r) \sin\theta \cos\varphi$$

$$Y_{l=-1,m=-1}(\theta,\varphi) \qquad p_{-1} = \left(\frac{3}{8\pi}\right)^{\frac{1}{2}} R_{n1}(r) \sin\theta e^{-i\varphi} \qquad p_{y} = \frac{i}{\sqrt{2}} (p_{-1} + p_{+1}) = \left(\frac{3}{4\pi}\right)^{\frac{1}{2}} R_{n1}(r) \sin\theta \sin\varphi$$

lecture 9:

$x = x_e - x_p = r \sin \theta \cos \phi$

Hydrogen atom

$$y = y_e - y_p = r \sin \theta \sin \phi$$

 $z = z_e - z_p = r \cos \theta$

d orbitals (5)
$$n > 2$$
 $Y_{l=2,m=-2}(\theta,\varphi) \dots Y_{l=2,m=2}(\theta,\varphi)$
$$d_{z^2} = \left(\frac{5}{16\pi}\right)^{\frac{1}{2}} R_{n2}(r) (3\cos^2\theta - 1) = \left(\frac{5}{16\pi}\right)^{\frac{1}{2}} R_{n2}(r) (3z^2 - r^2)/r^2$$

$$d_{x^2-y^2} = \frac{1}{\sqrt{2}}(d_{+2} + d_{-2}) = \left(\frac{15}{16\pi}\right)^{\frac{1}{2}} R_{n2}(r)(x^2 - y^2)/r^2$$

$$d_{xy} = \frac{1}{i\sqrt{2}}(d_{+2} - d_{-2}) = \left(\frac{15}{4\pi}\right)^{\frac{1}{2}} R_{n2}(r)xy/r^2 \qquad d_{yz} = \frac{1}{i\sqrt{2}}(d_{+} + d_{-1}) = -\left(\frac{15}{4\pi}\right)^{\frac{1}{2}} R_{n2}(r)yz/r^2$$

$$d_{yz} = \frac{1}{i\sqrt{2}}(d_{+} + d_{-1}) = -\left(\frac{15}{4\pi}\right)^{\frac{1}{2}} R_{n2}(r)yz/r^{2}$$

$$d_{zx} = \frac{1}{\sqrt{2}}(d_{+} - d_{-1}) = -\left(\frac{15}{4\pi}\right)^{\frac{1}{2}} R_{n2}(r)zx/r^{2}$$

