

The objective of this assignment is to

- numerically solve the radial part of Schrödinger Equation for "electron in H-atom" with Finite Difference method and determine the energy eigenvalues and corresponding normalised radial wavefunctions.

1. (10 marks) **Theory**

- Write down the Schrödinger Equation for an electron in H-atom potential in spherical polar coordinates.
- Use separation of variable method to separate this into angular and radial part. (Use $\psi_{nlm}(r, \theta, \phi) = \mathcal{R}_{nl}(r)\mathcal{Y}_{lm}(\theta, \phi)$ and take the separation constant as $\ell(\ell + 1)$.)
- Convert the Radial part of the Schrödinger Equation dimensionless form. Take $\mathcal{R}_{nl}(r) = \mathcal{K}_{nl}(r)/r$ and write the equation satisfied by $\mathcal{K}_{nl}(r)$. For this rescale r by Bohr radius and the energies by $|E_1|$, E_1 being the ground state Bohr energy.
- Discuss $V_{\text{eff}}(x)$ and its implications.
- Write down the analytical expressions for Bohr radius, Energy Eigenvalues and Energy eigenfunctions in this dimensionless form.
- Discuss the boundary conditions for numerical solution using finite difference method.

2. (10 marks) **Programming**

- Write a Python code to
 - Plot $V(r)$ and $V_{\text{eff}}(r)$ as a function of r for $\ell = 1, 2, 3$ on the same plot. Take range of r to be $[r_{\min} : r_{\max}]$ with $r_{\min} = 10^{-14}$ and $r_{\max} = 50$, r being the dimensionless variable.
 - Determine the first ten energy eigenvalues and normalised eigenfunctions for $\ell = 0$ using finite difference method with $r_{\max} = 10$
 - plot the first four radial wavefunctions (as points) along with the corresponding analytical wavefunctions (as continuous curves).
- Extend the code to determine the first ten energy eigenvalues and normalised eigenfunctions for $\ell = 1, 2$
- Extend the code to plot all radial probability densities (as scatter plots) along with the corresponding analytical wavefunction (as continuous curves) for all ℓ corresponding to a given n . i.e. the following graphs
 - radial probability density for $n = 0, \ell = 0$
 - radial probability density for $n = 1, \ell = 0, 1$
 - radial probability density for $n = 2, \ell = 0, 1, 2$