SOLUBAG REUSABLE BAG

MANUFACTURER SPECIFICATIONS

1.0. MATERIAL

This particular bag is made of 100% Polyvinyl Alcohol (PVA).

Polyvinyl Alcohol commonly known as PVA which is a water-soluble synthetic polymer with excellent film-forming, emulsifying and adhesive properties.

PVA is colorless (white) and odorless; resistant to oil, grease and solvents. It has high tensile strength and flexibility. With its unique features of biocompatibility, it is widely used for various applications in textile, paper and even in the medical industry for eye contacts or eye drops.

With water solubility and biodegradable properties, it can be completely dissolved in water. In natural environment, microbes ultimately break the products into carbon dioxide and water. After returning to the natural environment, it is non-toxic to the plants and animals.

This particular bag can be used as secondary packaging if several of the primary packages are placed in the bags for the purpose of transporting primarily retail goods.

SOLUBAG material does not contain any percent of plastic materials such as PP, HDPE, LDPE, PS, PVC, PET.

TEST ITEM		UNIT	TEST METHOD	TEST RESULT
THICKNESS		mm	CNS 1479	0.22
WEIGHT		g/m2	CNS 1479	40
DENSITY		g/cm3	CNS 1479	0.181
BREAKING STRENGTH	(MD)	N	ISO 9073-18	120
	(CD)	N	ISO 9073-18	89
HANDLE STRENGTH		N	KNWA 2884	96
SEAM STRENGTH		N	ISO 9073-18	186
WATER SOLUBLE TEMPERATURE		ōС		+85
MASS PER UNIT AREA		g/m2	ISO 9073-1	27



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Bag With



Solubag® Gusset Bag With Handle

- Performance Features

- Performance Features

 100 % Solubag material

 Soluble in water over + 185° F

 With reinforced handle

 Sewing with thread

 100% Environment friendly

 Certificates ✓

 Nch 409/1

 APHA.AWWA-WPCF

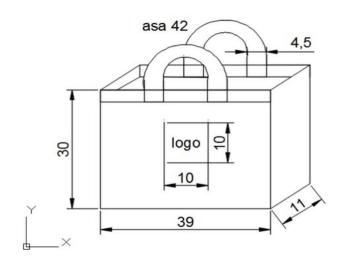
 UNE-EN-ISO-4121:2006

 NIXX-A-9073-INNTEX-2012

 Certificate of Conformity N*193-2019 ✓

 ASTM E 573

 ASTM E 1252



2.0 SHAPE AND DIMENSIONS Height = 30 cm.

Width = 39 cm.

Load Capacity = 8 kg.

GSM = 40.

Layers = 1.

Confection = Cotton Thread.

NB:

For a fast dissolve need water over 85°C.

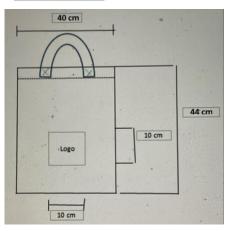


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2.0 SHAPE AND DIMENSIONS Height = 44 cm.

Height = 44 cm. Width = 40 cm.

Load Capacity = 6 kg.

GSM = 40.

Layers = 1.

Confection = Cotton Thread.

NB:

For a fast dissolve need water over 85°C.



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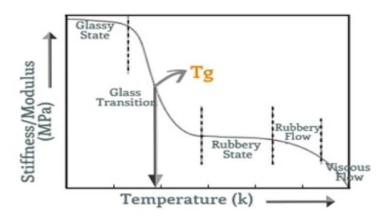
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UNDERSTANDING GLASS TRANSITION TEMPERATURE, Tg, OF A POLYMER

Glass transitionTemperature (Tg)

The temperature at which an amorphous polymer material (or plastic, a term used to generally refer to polymers) turns into a viscous liquid or rubbery form when heated is known as the glass transition temperature (Tg). A layman's definition of the glass transition temperature of a polymer is the temperature at which an **amorphous** polymer changes from a **hard or glassy state to a softer, often rubbery or viscous state**. For example, when Polystyrene or Polypropylene or any other polymer granules are heated, the material softens just before it melts. That temperature at which the polymer softens is called the **glass to rubber transition temperature** or simply, Tg.



Do all polymers have a Tg?

Yes, nearly all polymers will have a Tg, provided they have both **crystalline** and **amorphous** phases. Generally, crystalline polymers will have some **amorphous portions** and **crystalline phases**. This is why the same sample of a polymer can have both a glass transition temperature, Tg, and a melting temperature. Tm.

However, pure crystalline polymers do not have a glass transition temperature (Tg) because the glass transition temperature is only applicable to amorphous or non-crystalline polymers. Pure crystalline polymers do not have a glass transition temperature.

- In summary, Amorphous polymers only exhibit a Tg.
- Crystalline polymers exhibit a Tm (melt temperature) and typically a Tg if it has an amorphous ("semi"-crystalline). portion as well
- The value of Tg for most synthetic polymers lies between 170°K and 500°K (-103°C and 227°C). Pure crystalline polymers do not have a glass transition temperature because the glass transition temperature is only applicable to amorphous polymers



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