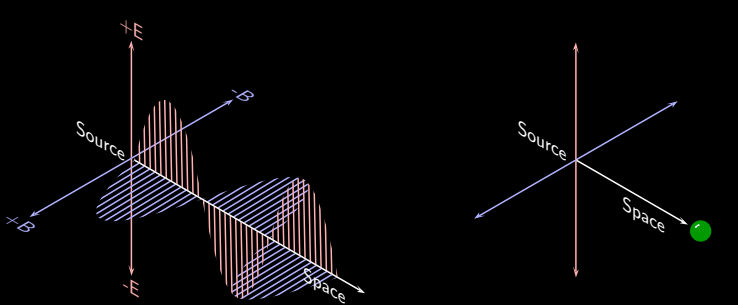


The Electromagnetic Radiation Spectrum



Electromagnetic Radiation (EMR)

- EMR is emitted in discrete units called photons but has properties of waves as seen by the images below. EMR can be created by the oscillation or acceleration of electrical charge or magnetic field. EMR travels through space at the speed of light ($2.997\,924\,58 \times 10^8$ m/s). EMR consists of an oscillating electrical and magnetic field which are at right angles to each other and spaced at a particular wavelength.



E = Electric Field Strength
B = Magnetic Field Strength
Wave Nature

- The particle nature of EMR is exhibited when a solar cell emits individual electrons when struck with very dim light.
- The wave nature of EMR is demonstrated by the famous double slit experiment that shows cancelling and addition of waves.
- Much of the EMR properties are based on theories since we can only see the effects of EMR and not the actual photon or wave itself.
- Albert Einstein theorized that the speed of light is the fastest that anything can travel. So far he has not been proven wrong.

- EMR can have its wavelength changed if the source is receding or approaching as in the red shift example of distant galaxies and stars that are moving away from us at very high speeds. The emitted spectral light from these receding bodies appears more red than it would be if the object was not moving away from us.
- We only have full electronic control over frequencies in the microwave range and lower. Higher frequencies must be created by waiting for the energy to be released from elements or atoms. We can either pump energy into the elements (ex. heating a rock with visible EMR and letting it release infrared EMR) or let it naturally escape (ex. uranium decay).
- We can only see the visible spectrum. All other bands of the spectrum are depicted as hatched colours.

Symbol	Name	Exp.	Multiplicator
Y	yotta	10 ²⁴	1,000,000,000,000,000,000,000,000
Z	zetta	10 ²¹	1,000,000,000,000,000,000,000,000
E	exa	10 ¹⁸	1,000,000,000,000,000,000,000,000
P	peta	10 ¹⁵	1,000,000,000,000,000,000,000,000
T	tera	10 ¹²	1,000,000,000,000,000,000,000,000
G	giga	10 ⁹	1,000,000,000,000,000,000,000,000
M	mega	10 ⁶	1,000,000,000,000,000,000,000,000
k	kilo	10 ³	1,000,000,000,000,000,000,000,000
m	milli	10 ⁻³	0.001
μ	micro	10 ⁻⁶	0.000 001
n	nano	10 ⁻⁹	0.000 000 001
p	pico	10 ⁻¹²	0.000 000 000 001
f	femto	10 ⁻¹⁵	0.000 000 000 000 001
a	atto	10 ⁻¹⁸	0.000 000 000 000 000 001
z	zepto	10 ⁻²¹	0.000 000 000 000 000 000 001
y	yocto	10 ⁻²⁴	0.000 000 000 000 000 000 000 001

Symbol	Name	Value
h	Speed of Light	2.997 924 58 × 10 ⁸ m/s
h	Planck's Constant	6.626 1 × 10 ⁻³⁴ J · s
f	Planck's Constant (freq)	1.054 592 × 10 ⁻³⁴ J · s
f	Frequency (cycles / second)	Hz
λ	Wavelength (meters)	m
E	Energy (Joules)	J

Formulas	Conversions
$E = h \cdot f$	1A = 0.1nm
$\lambda = \frac{c}{f}$	1nm = 10A
$f = \frac{c}{\lambda}$	1eV = 6.24 × 10 ¹⁸ eV

Gamma Rays

- Gamma radiation is the highest energy radiation (up to $\approx 10^{20}$ eV) that has been measured. At this energy, the radiation could be from gamma-rays, protons, electrons, or something else.
- Alpha, beta, and delta radiation are not electromagnetic but are actually parts of the atom being released from a radioactive element. In some cases this can cause gamma radiation. These are not to be confused with brain waves of similar names.

Visible Spectrum

- The range of EMR visible to humans is called "light". The visible spectrum also closely resembles the range of EMR that filters through our atmosphere from the sun.
- Other creatures see different ranges of visible light; for example bumble-bees can see ultraviolet light and dogs have a different response to colours than do humans.
- The sky is blue because our atmosphere scatters light and the shorter wavelength blue gets scattered the most. It appears that the entire sky is illuminated by a blue light but in fact that light is scattered from the sun. The longer wavelengths like red and orange move straight through the atmosphere which makes the sun look like a bright white ball containing all the colours of the visible spectrum.
- Interestingly, the visible spectrum covers approximately one octave.
- Astronomers use filters to capture specific wavelengths and reject unwanted wavelengths. The major astronomical (visual) filter bands are depicted as

Infrared Radiation

- Infrared radiation (IR) is sensed by humans as heat and is below the range of human vision. Humans (and anything at room temperature) are emitters of IR.
- IR remote control signals are invisible to the human eye but can be detected by most camcorders.
- Night vision scopes/goggles use a special camera that senses IR and converts the image to visible light. Some IR cameras employ an IR lamp to help illuminate the view.
- IR LASERS are used for burning objects.
- A demonstration of IR is to hold a metal bowl in front of your face. The IR emitted by your body will be reflected back using the parabolic shape of the bowl and you will feel the heat.
- Fiber-optic based infrared communication systems are sometimes amplified with Erbium-Doped Fiber Amplifiers [\[30\]](#).

LASER

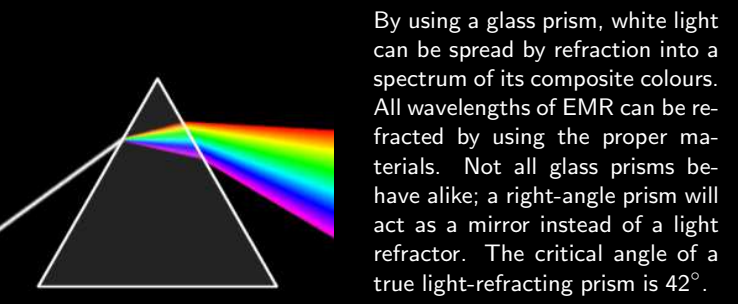
- LASER is an acronym for Light Amplification by Stimulated Emission of Radiation.
- A LASER is a device that produces monochromatic EMR of high intensity.
- With proper equipment, any EMR can be made to operate like a LASER. For example, microwaves are used to create a MASER.

Polarization

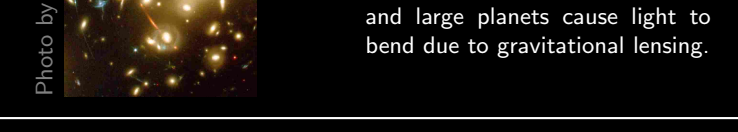
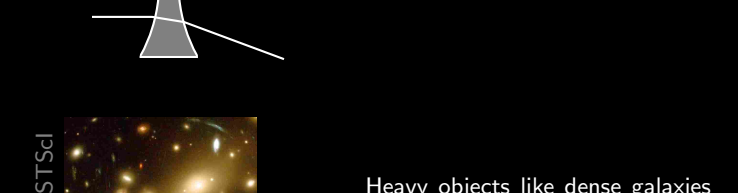
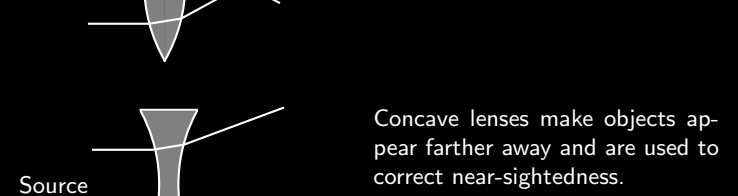
- As a photon (light particle) travels through space, its axis of electrical and magnetic fluctuations does not rotate. Therefore, each photon has a fixed linear polarity of somewhere between 0° to 360°. Light can also be circularly and elliptically polarized.
- Some crystals can cause the photon to rotate its polarization.
- Receivers that expect polarized photons will not accept photons that are in other polarities. (ex. satellite dish receivers have horizontal and vertical polarity positions).
- A polarized filter (like Polaroid™ sunglasses) can be used to demonstrate polarized light. One filter will only let photons that have one polarity through. Two overlapping filters at right angles will almost completely block the light that goes; however, a third filter inserted between the first two at a 45° angle will rotate the polarized light and allow some light to come out the end of all three filters.
- Light that reflects off an electrical insulator becomes polarized. Conductive reflectors do not polarize light.
- Perhaps the most reliably polarized light is a rainbow.
- Moonlight is also slightly polarized. You can test this by viewing the moonlight through a Polaroid™ sunglasses lens, then rotate that lens, the moonlight will dim and brighten slightly.

Refraction

- Refraction of EMR is dependent on wavelength as can be seen by the prism example below.



By using a glass prism, white light can be spread by refraction into a spectrum of its composite colours. All wavelengths of EMR can be refracted by using the proper materials. Not all glass prisms have alike; a right-angle prism will act as a mirror instead of a light refractor. The critical angle of a true light-refracting prism is 42°.



Reflection

- Reflection of EMR is dependent on wavelength as demonstrated when visible light and radio waves bounce off objects that X-Rays would pass through. Microwaves, which have a large wavelength compared to visible light, will bounce off metal mesh in a microwave oven whereas visible light will pass through.
- EMR of any wavelength can be reflected, however, the reflectivity of a material depends on many factors including the wavelength of the incident beam.
- The angle of incidence (θ_i) and angle of reflection (θ_r) are the same.

How to read this chart

- This chart is organized in octaves (frequency doubling/halving) starting at 1Hz and going higher (2,4,8, etc) and lower (1/2, 1/4, etc). The octave is a natural way to represent frequency.
- Frequency increases on the vertical scale in the upward direction.
- The horizontal bars wrap around from far right to far left as the frequency increases upwards.
- There is no limit to either end of this chart; however, due to limited space, only the "known" items have been shown here. A frequency of 0Hz is the lowest possible frequency but the method of depicting octaves used here does not allow for ever reaching 0Hz, only approaching it. Also, by the definition of frequency (Cycles per second), there is no such thing as negative frequency.
- Values on the chart have been labelled with the following colours: [Frequency] measured in Hertz, [Wavelength] measured in meters, [Energy] measured in electronVolts.

Ultraviolet Light

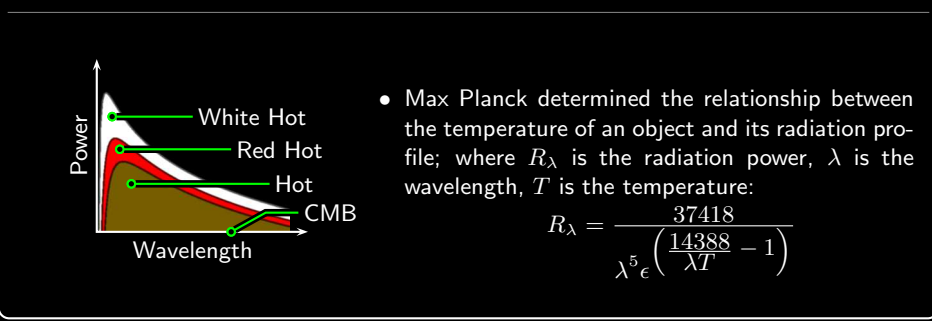
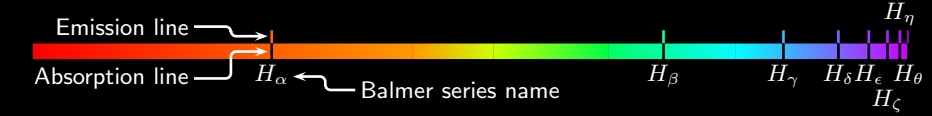
- Ultraviolet light is beyond the range of human vision.
- Physicists have divided ultraviolet light ranges into Vacuum Ultraviolet (VUV), Extreme Ultraviolet (EUV), Far Ultraviolet (FUV), Medium Ultraviolet (MUV), and Near Ultraviolet (NUV).
- UV-A, UV-B and UV-C were introduced in the 1930's by the Commission Internationale de l'Eclairage (CIE, International Commission on Illumination) for photobiological spectral bands.
- Short-term UV-A exposure causes sun-tanning which helps to protect against sunburn. Exposure to UV-B is beneficial to humans by helping the skin produce vitamin D. Excessive UV exposure causes skin damage. UV-C is harmful to humans but is used as a germicide.
- The CIE originally divided UVA and UVB at 315nm, later some photo-dermatologists divided it at 300nm.
- UVA is subdivided into UVA1 and UVA2 for DNA altering effects at the 340nm.
- The sun produces a wide range of frequencies including all the ultraviolet light; however, UVA is partially filtered by the ozone layer and UVB is totally filtered out by the earth's atmosphere.
- A bumblebee can see light in the UVA range which helps them identify certain flowers.

Emission and Absorption

- As EMR passes through elements, certain wavelength bands get absorbed and some new ones get emitted. This absorption and emission produces characteristic spectral lines for each element which are useful in determining the makeup of distant stars. These lines are used to prove the red-shift amount of distant stars.
- When a photon hits an atom it may be absorbed if the energy is just right. The energy level of the electron is raised – essentially holding the radiation. A new photon of specific wavelength is created when the energy is released. The jump in energy is a discrete step and many possible levels of energy exist in an atom.
- Johann Balmer created this formula defining the photon emission wavelength (λ), where m is the initial electron energy level and n is the final electron energy level.

$$\lambda = 364.50nm \left(\frac{m^2}{m^2 - n^2} \right)$$

- Much of the interstellar matter is made of the simplest atom hydrogen. The hydrogen visible-spectrum emission and absorption lines are shown below:



Max Planck determined the relationship between the temperature of an object and its radiation profile, where R_λ is the radiation power, λ is the wavelength, T is the temperature:

$$R_\lambda = \frac{2\pi^5 k^4}{15 h^3 c^2} \frac{1}{\lambda^5} \left(\frac{1}{e^{\frac{hc}{\lambda k T}} - 1} \right)$$

CMB

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

400 MJ/sr

Intensity

600 GHz

65 GHz

</