

Week 4: Metabolism and Metabolic Networks

Lecturer: Jeffrey Varner

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1 Introduction

Metabolism is the set of chemical reactions that occur in living organisms to maintain life. These chemical reactions are organized into metabolic pathways, which are sequences of chemical reactions that convert a substrate into a product. Metabolic pathways are interconnected to form metabolic networks, which are the focus of this week in lecture. Metabolic networks are complex systems that are responsible for the synthesis and degradation of metabolites, which are small molecules that are intermediates or products of metabolism. Metabolic networks are also responsible for the generation of energy, which is required for the maintenance of life. This week, we will discuss the structure and function of metabolic networks, and how they can be analyzed using mathematical models. Let's get started by discussing catabolism and anabolism, which are the two main types of metabolic pathways.

2 Catabolism and Anabolism

Metabolic pathways can be divided into two main types: catabolic pathways and anabolic pathways. Catabolic pathways are responsible for the breakdown of complex molecules into simpler molecules, which releases energy. On the other hand, anabolic pathways are responsible for the synthesis of complex molecules from simpler molecules, which requires energy.

The main purpose of catabolic pathways is to generate energy and small molecules that can be used as building blocks for anabolic pathways. Here we will discuss three important catabolic pathways (which we'll collectively call central carbon metabolism): glycolysis, the pentose phosphate pathway and the citric acid cycle. Glycolysis is a catabolic pathway that converts sugars, e.g., glucose into **pyruvate**, which can then be further metabolized to generate energy. The first step of glycolysis is the conversion of glucose into glucose-6-phosphate, which is then converted into fructose-6-phosphate, and so on. The pentose phosphate pathway is a catabolic pathway that converts glucose-6-phosphate into ribose-5-phosphate, which is a precursor for the synthesis of nucleotides, and some aromatic amino acids. It is also an important source of **NADPH**, which is a reducing agent that is required for biosynthetic reactions. Finally, the citric acid cycle is a catabolic pathway that converts acetyl-CoA (generated from pyruvate) into **CO₂**, **NADH**, **FADH₂** and **GTP**. The **NADH** and **FADH₂** generated by the citric acid cycle are used to generate energy through another pathway called **oxidative phosphorylation**.

Anabolic pathways are responsible for the synthesis of complex molecules from simpler molecules, which requires energy. For example, the synthesis of macromolecules such as proteins, nucleic acids, lipids and polysaccharides are examples of anabolic pathways, each of which requires energy in the form of high-energy phosphate bonds and reducing power in the form of **NADH** (**NADPH**). Anabolic pathways are often regulated by the availability of substrates and energy, as well as the demand for the end products of the pathway.

The details of the central carbon metabolism and other metabolic pathways can be found in many textbooks, such as (1) or Chapter 3 of Wüschiers et al (2). Thus, we will not discuss them in detail here, but instead abstract the structure and function of metabolic networks into a mathematical model.

3 Generic Metabolic Networks

Suppose we have a metabolic network with M metabolites and R reactions, where the set of reactions denoted by \mathcal{R} has dimension $|\mathcal{R}| = R$ and the set of metabolites \mathcal{M} , has dimension $|\mathcal{M}| = M$. Each reaction $j \in \mathcal{R}$ is catalyzed by an enzyme, $e_j \in \mathbb{E}$, where \mathbb{E} denotes the set of enzymes in the network. Let's not worry about the details of enzyme kinetics for now, or where the enzymes come from, but instead focus on the stoichiometry of the network, which describes the relationships between metabolites and reactions. The stoichiometry of the network can be represented by a stoichiometric matrix $\mathbf{S} \in \mathbb{R}^{M \times R}$, where s_{ij} is the stoichiometric coefficient of metabolite i in reaction j . If $s_{ij} > 0$, then metabolite i is a product of reaction j , and if $s_{ij} < 0$, then metabolite i is a substrate of reaction j . Finally, if $s_{ij} = 0$, then metabolite i is not involved in reaction j .

3.1 Material Balance Equations

Each metabolite in the network, as well as each enzyme, is subject to a material balance equation, which describes the rate of change of the concentration of the metabolite (enzyme) in the network. Imagine that we have a **abstract biophase**, which is a well-mixed compartment where our metabolic reactions are occurring. This biophase can be a single cell, a compartment within a single cell, such as the cytoplasm or mitochondria, a cell-free system or a even collection of cells in a bioreactor. Let the volume of the biophase be V (depending on the context, V could be the volume of a single cell, the volume of the cytoplasm, cell number, cell mass, thus it will units specific to the case).

Dynamic Species Mole Balances

If we are interested in the number of moles of each chemical species, we can write the open species mole balance. Let n_i denote the number of moles of chemical component i in the system (units: \star moles, e.g., mmol or μ mol). Further, denote the number of chemical components in the system we want to track as \mathcal{M} , and the set of streams flowing into (or from) the system as \mathcal{S} , where each stream $s \in \mathcal{S}$ has a direction parameter $\nu_s \in [-1, 1]$. If stream s enters the system $\nu_s = +1$, however is stream s exits the system then $\nu_s = -1$. Then, the number of moles of chemical component i in the system as a function of time is described by an open species mass balance equation:

$$\sum_{s \in \mathcal{S}} \nu_s \dot{n}_{s,i} + \dot{n}_{G,i} = \frac{dn_i}{dt} \quad \forall i \in \mathcal{M} \quad (1)$$

where $\dot{n}_{s,i}$ denotes the mole flow rate of component i in stream s (units: \star mol i /time), $\dot{n}_{gen,i}$ denote the rate of generation of component i in the system (units: \star mol i /time), and dn_i/dt denotes the rate of accumulation of the number of moles of component i in the system (units: \star mol i /time). Finally, let's discuss the generation terms. Generation terms describe the impact of chemical reactions; theoretically, we can describe reactions using a mass or mole basis. However, typically we operate on a mole (or concentration) basis when dealing with chemical reactions. Then, the species generation rate $\dot{n}_{G,i}$ can be written in terms of the open extent of reaction:

$$\dot{n}_{G,i} = \sum_{r \in \mathcal{R}} \sigma_{ir} \dot{\epsilon}_r \quad (2)$$

where σ_{ir} denotes the stoichiometric coefficient of species i in reaction r , and $\dot{\epsilon}_r$ denotes the open extent of reaction r (units: \star mol/time). Putting these ideas together, we can rewrite the dynamic species mole balance as:

$$\sum_{s \in \mathcal{S}} \nu_s \dot{n}_{s,i} + \sum_{r \in \mathcal{R}} \sigma_{ir} \dot{\epsilon}_r = \frac{dn_i}{dt} \quad \forall i \in \mathcal{M} \quad (3)$$

Dynamic Species Concentration Balances

When describing systems with chemical reactions where we write reaction rate expressions in terms of concentration, e.g., mole per unit volume basis, it is inconvenient to use mole or mass based units. In these cases, we need a new type of balance equation, the concentration balance. Let n_i denote the number of moles of chemical component i in the system (units: \star moles, e.g., mmol or μ mol). Further, denote the set of chemical species as \mathcal{M} , the set of streams flowing into (or from) the system as \mathcal{S} , and the set of chemical reactions as \mathcal{R} . Then, the number of moles of chemical component i in the system is described by an open species mass balance equation:

$$\sum_{s \in \mathcal{S}} \nu_s \dot{n}_{s,i} + \dot{n}_{G,i} = \frac{dn_i}{dt} \quad \forall i \in \mathcal{M} \quad (4)$$

However, we can re-write the number of moles of species i as:

$$n_i = C_i V \quad \forall i \in \mathcal{M} \quad (5)$$

where C_i denotes the concentration of species i (units: mole per volume), and V (units: volume) denotes the volume of the system. Thus, we can re-write the species mole balance in concentration units as:

$$\sum_{s \in \mathcal{S}} \nu_s C_{s,i} \dot{V}_s + \dot{C}_{G,i} V = \frac{d}{dt} (C_i V) \quad \forall i \in \mathcal{M} \quad (6)$$

where \dot{V}_s denotes the volumetric flow rate for stream s (units: volume/time), $C_{s,i}$ denotes the concentration of component i in stream s (units: concentration), and $\dot{C}_{G,i}$ denotes the rate of generation of component i by chemical reaction (units: concentration/time). The generation terms for species i in the concentration balance equation can be written as:

$$\dot{C}_{G,i} V = \sum_{j \in \mathcal{R}} \sigma_{ij} \hat{r}_j V \quad \forall i \in \mathcal{M} \quad (7)$$

where σ_{ij} denotes the stoichiometric coefficient of species i in reaction j , and \hat{r}_j denotes the rate of reaction j (units: concentration/time). Putting these ideas together, we can rewrite the dynamic species concentration balance as:

$$\sum_{s \in \mathcal{S}} \nu_s C_{s,i} \dot{V}_s + \sum_{j \in \mathcal{R}} \sigma_{ij} \hat{r}_j V = \frac{d}{dt} (C_i V) \quad \forall i \in \mathcal{M} \quad (8)$$

3.2 Properties of the Stoichiometric Matrix

The stoichiometric matrix \mathbf{S} is the digital representation of the metabolic network, and it can be used to analyze some of the structural properties of the network. There is a large body of literature on the analysis of the structure of metabolic networks using various decomposition approaches of stoichiometric matrices. Let's look at one particular approach to analyzing the structure of metabolic networks, using singular value decomposition (SVD) (3).

Singular Value Decomposition

Suppose we have a matrix $A \in \mathbb{R}^{m \times n}$. The SVD of \mathbf{A} is a factorization of the form: $\mathbf{A} = \mathbf{U} \mathbf{\Sigma} \mathbf{V}^T$, where $\mathbf{U} \in \mathbb{R}^{n \times n}$ and $\mathbf{V} \in \mathbb{R}^{m \times m}$ are orthogonal matrices, i.e., $\mathbf{U} \cdot \mathbf{U}^T = \mathbf{I}$ and $\mathbf{\Sigma} \in \mathbb{R}^{n \times m}$ is a diagonal matrix

containing the singular values $\sigma_i = \Sigma_{ii}$. The matrix $\mathbf{A} \in \mathbb{R}^{n \times m}$ can be decomposed as:

$$\mathbf{A} = \sum_{i=1}^{r_{\mathbf{A}}} \sigma_i \cdot (\mathbf{u}_i \otimes \mathbf{v}_i)$$

where $r_{\mathbf{A}}$ is the rank of matrix \mathbf{A} and \otimes denotes the outer product. The outer product $\hat{\mathbf{A}}_i = \mathbf{u}_i \otimes \mathbf{v}_i$ is a rank-1 matrix with elements:

$$\hat{a}_{jk} = u_j v_k \quad j = 1, 2, \dots, n \text{ and } k = 1, 2, \dots, m$$

The vectors \mathbf{u}_i and \mathbf{v}_i are the left (right) singular vectors, and σ_i are the singular values (ordered). Singular value decomposition is a special sort of eigendecomposition, thus, we could use QR-iteration to compute the SVD. The columns of \mathbf{U} are eigenvectors of $\mathbf{A}\mathbf{A}^T$, the columns of \mathbf{V} are eigenvectors of $\mathbf{A}^T\mathbf{A}$ and the singular values σ_i are the square roots of the eigenvalues of the matrix products $\mathbf{A}\mathbf{A}^T$ or $\mathbf{A}^T\mathbf{A}$.

4 Estimating Metabolic Fluxes

Metabolic fluxes are the rates of the reactions in a metabolic network, which are the unknowns that we want to estimate from various types of experimental data. You can think of metabolic fluxes as the flow of metabolites through the metabolic network, which is analogous to the flow of water through a pipe. Thus, estimating the metabolic fluxes is important for understanding the structure and operation of a metabolic network as a function of the environment. Metabolic fluxes can be estimated using a variety of mathematical methods, such as metabolic flux analysis (MFA) and flux balance analysis (FBA). These two methods are based on different assumptions about the network, and they have different strengths and weaknesses. Let's start by discussing metabolic flux analysis, and conclude with a discussion of flux balance analysis.

4.1 Metabolic Flux Analysis

Metabolic flux analysis is a powerful tool for analyzing the structure and function of metabolic networks. The goal of metabolic flux analysis is to estimate the rates of the reactions in the network, which are called metabolic fluxes, from experimental data, in particular, isotopic labeling data or exchange flux data (rates of uptake or secretion of metabolites). The key assumption of metabolic flux analysis is that the network is at (or near) a steady state, which means that the concentrations of the metabolites and enzymes in the network are approximately constant over time. Under this assumption, the rate of accumulation of each metabolite (enzyme) in the network is zero, which allows us to write a set of linear algebraic equations that describe the network flux. Traditional metabolic flux analysis uses a stoichiometric model of the network, which is a set of linear equations that describe the relationships between the metabolites and reactions in the network. Alternatively, labeled isotopes can be used to estimate the metabolic fluxes in the network, which is called isotopic flux analysis.

Traditional Metabolic Flux Analysis

One of the first examples of what we might recognize as traditional metabolic flux analysis was published by Vallino and Stephanopoulos (4) (although there were several earlier studies from Papoutsakis et al (5, 6) that used similar ideas). Vallino and Stephanopoulos used a stoichiometric model of the metabolism of *Corynebacterium glutamicum* in combination with exchange rate measurements to estimate the intracellular fluxes occurring during growth and lysine production.

4.2 Flux Balance Analysis

Flux balance analysis (FBA) is another mathematical method for analyzing the structure and function of metabolic networks (7). Flux balance analysis is arguably the most widely used computational tool to estimate the intracellular or cell-free reaction rates (which are called fluxes) throughout a steady state reaction network (units: $\star\text{mol}/\text{volume}\cdot\text{time}$). However, while we are introducing flux balance analysis in the context of metabolic network analysis, FBA is a general tool that can be used to estimate flows through many different types of networks and graphs, e.g., social graphs, communication networks, or other types of problems that can be represented as a network of graphs.

Like metabolic flux analysis, the goal of FBA is to estimate the rates of the reactions in the network from experimental data. However, FBA makes some additional simplifying assumptions about the network, which allows for the use of linear programming to solve for the metabolic fluxes. Suppose we have a system, with a species set \mathcal{M} , stream set \mathcal{S} and a reaction set \mathcal{R} . Further, suppose that the system (or at least part of it) is at or near a steady state. Then, the objective of the flux balance analysis is to estimate the value of the reaction rates operating in the system using linear programming. The flux balance analysis problem, whose solution provides estimates for the values of the unknown fluxes is the linear program: that maximizes (minimizes) the objective function:

$$\text{minimize/maximize} \quad \sum_{i \in \mathcal{R}} c_i \hat{v}_i \quad (9)$$

$$\text{subject to} \quad \sum_{s \in \mathcal{S}} \nu_s C_{s,i} \dot{V}_s + \sum_{j \in \mathcal{R}} \sigma_{ij} \hat{v}_j V = \frac{d}{dt} (C_i V) \quad \forall i \in \mathcal{M} \quad (10)$$

$$\text{subject to} \quad \mathcal{L}_j \leq \hat{v}_j \leq \mathcal{U}_j \quad \forall j \in \mathcal{R} \quad (11)$$

The matrix \mathbf{S} is the $\mathcal{M} \times \mathcal{R}$ stoichiometric matrix, c_i denote the objective coefficients, \mathbf{v} denotes the unknown flux vector (the unknown that we are trying to estimate), and $\mathcal{L}(\mathcal{U})$ denote the permissible lower (upper) bounds on the unknown fluxes. The first set of constraints are material constraints, while the second imparts thermodynamic and kinetic information into the calculation. Finally, there are potentially other types of constraints (both linear and nonlinear) not shown that may be necessary for specific applications encountered in practice.

5 Timescales in Metabolic Networks

Fill in here.

6 Summary

In this lecture, we discussed the structure and function of metabolic networks, and how they can be analyzed using mathematical models.

References

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