Stat Mech I HW4

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1 Problem 1

a) The multiplicity W(n,N) to put n quanta into N harmonic oscillators is $\binom{n}{N} = \frac{(N-1)!}{n!(N-1)!}$. Planck proved this using the generating function $g(t) = t^n$. We start with W(n,1), which is clearly equal to 1. Using the generating function:

$$\sum_{n=0}^{\infty} W(n,1)t^n = \sum_{n=0}^{\infty} t^n = \frac{1}{1-t}$$
 (1.1)

For N harmonic oscillators we take the generating function to the Nth power.

$$\left(\frac{1}{1-t}\right)^N = \sum_{n=0}^{\infty} W(n,N)t^n = a_0 + a_1t + a_2t^2 + \dots$$
 (1.2)

We can now solve for the coefficients of the series which are the W(n,N). To find W(n,N) we differentiate n times and set t=0.

$$\frac{d^n}{dt^n} \sum_{n=0}^{\infty} W(n,N)t^n = \frac{1}{n!}W(n,N)$$
(1.3)

$$W(n,N) = \lim_{t \to 0} \frac{1}{n!} \frac{d^n}{dt^n} \left(\frac{1}{1-t}\right)^n \tag{1.4}$$

$$W(n,N) = \lim_{t \to 0} \frac{1}{n!} N(N+1)(N+2)...(N+n-1)(1-t)^{-N-m}$$
 (1.5)

$$W(n,N) = \frac{N(N+1)...(N+n-1)}{n!} \frac{(N-1)!}{(N-1)!} = \frac{(N+n-1)!}{n!(N-1)!}$$
(1.6)

b) Using the microcanonical ensemble approach we can now calculate the entropy.

$$S = k \ln W = k \ln \left(\frac{(N+n-1)!}{n!(N-1)!} \right)$$
 (1.7)

Using the Stirling approximation this simplifies to:

$$S = k((N+n)\ln(N+n) - n\ln n - N\ln N)$$
 (1.8)

(1.9)

c) We now use Planck's expression for the total energy $U=n\hbar\omega$. Substituting $n=\frac{U}{\hbar\omega}$ we then have:

$$S = k((N+n)\ln(N+n) - n\ln n - N\ln N)$$
(1.10)

$$S = k\left(\left(N + \frac{U}{\hbar\omega}\right)\ln\left(N + \frac{U}{\hbar\omega}\right) - \frac{U}{\hbar\omega}\ln\frac{U}{\hbar\omega} - N\ln N\right)$$
 (1.11)

d) The temperature is $\frac{1}{T} = \frac{\partial S}{\partial U}$. We take $\frac{\partial S}{\partial n} \frac{\partial n}{\partial U}$, with $\frac{\partial n}{\partial U} = \frac{1}{\hbar \omega}$.

$$\frac{\partial S}{\partial n} = k \left(\ln \left(N + n \right) + 1 - \ln n - 1 \right) = k \ln \frac{N + n}{n} \tag{1.12}$$

$$\frac{1}{kT} = \frac{1}{\hbar\omega} \ln \frac{N + \frac{U}{\hbar\omega}}{\frac{U}{\hbar\omega}} \tag{1.13}$$

$$e^{\frac{\hbar\omega}{kT}} = \frac{N + \frac{U}{\hbar\omega}}{\frac{U}{\hbar\omega}} = \frac{N\hbar\omega}{U} + 1 \tag{1.14}$$

$$U = \frac{N\hbar\omega}{e^{\frac{\hbar\omega}{kT}} - 1} \tag{1.15}$$

2 Problem 2

- a) Call the number of quanta in SHO 1 by n_1 , then the number of quanta in SHO 2 is $n'-1-n_1$. We see that n_1 completely determines the system state, and can range from 0 to n'-1. There are therefore n' microstates available to the system, and the entropy is $S = k \ln n'$.
- b) We follow the same argument as part A, but now the total energy constrains the number of quanta in the second oscillator to $\frac{n''}{2} 1 n_1$. Now n_1 can range from 0 to $\frac{n''}{2}$ so there are $\frac{n''}{2}$ microstates (which is fine since n'' is even). The entropy is $S = k \ln \left(\frac{n''}{2}\right)$.
- c) S is an extensive parameter so we simply add the results from part A and part B. Expressed in terms of the energies:

$$S = k \left(\ln \frac{E'}{\hbar \omega} + \ln \frac{E''}{2(2\hbar \omega)} \right)$$
 (2.1)

$$S = k \left(\ln \frac{E'}{\hbar \omega} + \ln \frac{E''}{2\hbar \omega} - \ln 2 \right)$$
 (2.2)

$$S = k \left(\ln \frac{E'}{2\hbar\omega} + \ln \frac{E''}{2\hbar\omega} \right) \tag{2.3}$$

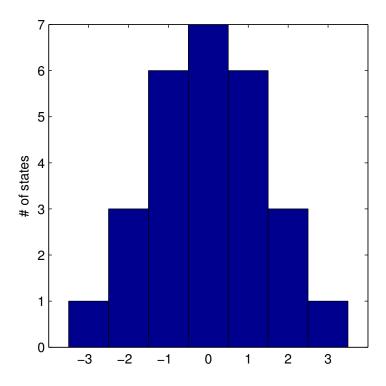
3 Problem 3

The complete enumeration of all states (M=1) is:

$$\begin{bmatrix} -1 & -1 & -1 \\ -1 & -1 & 0 \\ -1 & -1 & 1 \\ -1 & 0 & -1 \\ -1 & 0 & 0 \\ -1 & 0 & 1 \\ -1 & 1 & -1 \\ -1 & 1 & 0 \\ -1 & 1 & 1 \\ 0 & -1 & -1 \\ 0 & -1 & 1 \\ 0 & 0 & -1 & 1 \\ 0 & 0 & -1 & 1 \\ 0 & 0 & 0 & 0 \\ 0 & 0 & 1 & 0 \\ 0 & 1 & -1 & 0 \\ 0 & 1 & 1 & 1 \\ 1 & -1 & -1 & 1 \\ 1 & 0 & -1 & 1 \\ 1 & 0 & 0 & 1 \\ 1 & 0 & 1 & 1 \\ 1 & 0 & 1 & 1 \\ 1 & 0 & 1 & 1 \\ 1 & 1 & 0 & 1 \\ 1 & 1 & 1 & 0 \\ 1 & 1 & 1 & 1 \end{bmatrix}$$

$$(3.1)$$

The figure below shows the number of states with each possible value of M_z .



- a) With all state having probability $\frac{1}{27}$ the entropy is 3.2958.
- b) The entropy when only $M_z = 0$ is possible is 1.9549.
- c) The entropy when only $M_z = M$ is possible is 1.7918.
- d) There is only one state with $M_z = 3M$, so the entropy is 0.
- e) We minimize the entropy using Lagrange multipliers to incorporate the constraints $\sum f_i = 1$ and $\sum f_i M_{zi} = \gamma M$.

$$F = \sum f_i \ln f_i - \lambda_1 \left(\sum f_i - 1\right) - \lambda_2 \left(\sum f_i M_{zi} - \gamma M\right)$$
 (3.2)

$$\frac{\partial F}{\partial f_i} = 1 + \ln f_i - \lambda_1 - \lambda_2 M_{zi} = 0$$

$$f_i = e^{\lambda_1 + \lambda_2 M_{z_i} - 1}$$
(3.4)

$$f_i = e^{\lambda_1 + \lambda_2 M_{z_i} - 1} \tag{3.4}$$

$$f_{i} = e^{\lambda_{1} + \lambda_{2} M_{z_{i}} - 1}$$

$$\sum f_{i} = e^{\alpha} \sum e^{\lambda_{2} M_{z_{i}}} = 1, \alpha \equiv \lambda_{1} - 1$$

$$e^{-\alpha} = \sum e^{\lambda_{2} M_{z_{i}}}$$

$$(3.4)$$

$$(3.5)$$

$$e^{-\alpha} = \sum e^{\lambda_2 M_{zi}} \tag{3.6}$$

4 Problem 4

In class it was shown that $\frac{\partial U}{\partial \beta} = <(\Delta E)^2>$. To work out $<(\Delta E)^3>$ we find $\frac{\partial^2 U}{\partial \beta^2}$.

$$d\beta = -\frac{1}{kT^2}dT\tag{4.1}$$

$$\frac{\partial U}{\partial \beta} = \frac{\partial U}{\partial T} \frac{\partial T}{\partial \beta} = kT^2 \frac{\partial U}{\partial T} = kT^2 C_v \tag{4.2}$$

$$\frac{\partial^2 U}{\partial \beta^2} = \frac{\partial}{\partial T} \left(\frac{\partial U}{\partial \beta} \right) \frac{\partial T}{\partial \beta} \tag{4.3}$$

$$\frac{\partial^2 U}{\partial \beta^2} = \left(2kTC_v + kT^2 \frac{\partial C_v}{\partial T}\right) kT^2 \tag{4.4}$$

$$\frac{\partial^2 U}{\partial \beta^2} = k^2 \left(2T^3 C_v + T^4 \frac{\partial C_v}{\partial T} \right) \tag{4.5}$$

For an ideal gas $U = \frac{3}{2}NkT$ and $C_v = \frac{3}{2}Nk$. So we have:

$$<(\Delta E)^2> = kT^2C_v = \frac{3}{2}Nk^2T^2$$
 (4.6)

$$\langle U^2 \rangle = (\frac{3}{2}NkT)^2$$
 (4.7)

$$\langle (\frac{\Delta E}{U})^2 \rangle = \frac{2}{3N} \tag{4.8}$$

$$<(\Delta E)^3> = k^2(2T^3(\frac{3}{2}Nk) + T^4(0)) = 3Nk^3T^3$$
 (4.9)

$$\langle U^3 \rangle = \frac{3^3}{2^3} N^3 k^3 T^3$$
 (4.10)

$$<(\frac{\Delta E}{U})^3> = \frac{2^3}{3^3} \frac{1}{N^2} = \frac{8}{9N}$$
 (4.11)

5 Problem 5

For an ideal gas the entropy is $S = Nk \left(\ln \frac{V}{N\lambda^3} + \frac{5}{2} \right)$. We show that:

$$\frac{S}{Nk} = \ln\left(\frac{Q_1}{N}\right) + T\left(\frac{\partial \ln Q_1}{\partial T}\right)_P \tag{5.1}$$

The partition function for an ideal gas is $Q_n(V,T) = \frac{1}{N!} (\frac{V}{\lambda^3})^N$. Q_1 is $\frac{V}{\lambda^3}$. We expand Q_1 and find $(\frac{\partial \ln Q_1}{\partial T})_P$.

$$Q_1 = \frac{V}{h^3} (2\pi mkT)^{3/2} \tag{5.2}$$

$$\left(\frac{\partial \ln Q_1}{\partial T}\right)_P = \frac{1}{Q_1} \frac{\partial Q_1}{\partial T} \tag{5.3}$$

$$\left(\frac{\partial \ln Q_1}{\partial T}\right)_P = \frac{3}{2}\frac{1}{T} \tag{5.4}$$

We see that $\frac{S}{Nk} = \ln\left(\frac{Q_1}{N}\right) + T\left(\frac{\partial \ln Q_1}{\partial T}\right)_P = \ln\frac{V}{N\lambda^3} + \frac{3}{2}$.