## **COMMON IONS**

## **CATIONS**

Group I/II metals		$\mathrm{Au}^{1+}$	gold (I)	$\mathrm{Hg_2}^{2+}$	mercury (I)
		$\mathrm{Au^{3+}}$	gold (III)	$\mathrm{Hg}^{2+}$	mercury (II)
$\mathrm{NH_4}^+$	ammonium	$\mathrm{Co}^{2+}$	cobalt (II)	$\mathrm{Ni}^{2+}$	nickel (II)
$H_3O^+$	hydronium	$\mathrm{Co}^{3+}$	cobalt (III)	$\mathrm{Ni}^{3+}$	nickel (III)
		$\mathrm{Cu}^+$	copper (I)	$\mathrm{Pb}^{2+}$	lead (II)
$Al^{3+}$	aluminum	$Cu^{2+}$	copper (II)	$Pb^{4+}$	lead (IV)
$\mathrm{Zn}^{2+}$	zinc	$\mathrm{Fe}^{2+}$	iron (II)	$\mathrm{Sn}^{2+}$	tin (II)
$Ag^+$	silver	$\mathrm{Fe^{3+}}$	iron (III)	$\mathrm{Sn}^{4+}$	tin (IV)

## **ANIONS**

Group V/VI/VII nonmetals		${\rm MnO_4}^-$	permanganate	
	carbonate	$\mathrm{AsO_4}^{3-}$	arsenate	
$HCO_3^{2-}$	hydrogen carbonate			
	OR bicarbonate	$\mathrm{SO_4}^{2-}$	sulfate	
		$\mathrm{SO_3}^{2-}$	sulfite	
$\mathrm{NO_3}^-$	nitrate	$\mathrm{HSO_4}^-$	hydrogen sulfate	
$\mathrm{NO_2}^-$	nitrite	${ m HSO_3}^-$	hydrogen sulfite	
$\mathrm{N_3}^-$	azide	$S_2O_3^{2-}$	thiosulfate	
$CN^-$	cyanide			
$OCN^-$	cyanate	${ m ClO_4}^-$	perchlorate	
$SCN^-$	thiocyanate	${ m ClO_3}^-$	chlorate	
		${ m ClO_2}^-$	chlorite	
$\mathrm{O_2}^{2-}$	peroxide	$\mathrm{ClO}^-$	hypochlorite	
$\mathrm{OH}^-$	hydroxide			
$\mathrm{C_2H_3O_2}^-$	acetate	${\rm BrO_4}^-$	perbromate	
$C_2O_4^{\ 2-}$	oxalate	$\mathrm{BrO_3}^-$	bromate	
		$\mathrm{BrO_2}^-$	bromite	
$PO_4^{3-}$	phosphate	${\rm BrO^-}$	hypobromite	
	hydrogen phosphate			
	dihydrogen phosphate	$\mathrm{IO_4}^-$	periodate	
	phosphite	${\rm IO_3}^-$	iodate	
	_	$\mathrm{IO}_2^{-}$		
$\mathrm{CrO_4}^{2-}$	chromate		hypoiodite	
_	dichromate			