

COMMON IONS

CATIONS

Group I/II metals	Au^{1+}	gold (I)	Hg_2^{2+}	mercury (I)
	Au^{3+}	gold (III)	Hg^{2+}	mercury (II)
NH_4^+	Co^{2+}	cobalt (II)	Ni^{2+}	nickel (II)
H_3O^+	Co^{3+}	cobalt (III)	Ni^{3+}	nickel (III)
	Cu^+	copper (I)	Pb^{2+}	lead (II)
Al^{3+}	Cu^{2+}	copper (II)	Pb^{4+}	lead (IV)
Zn^{2+}	Fe^{2+}	iron (II)	Sn^{2+}	tin (II)
Ag^+	Fe^{3+}	iron (III)	Sn^{4+}	tin (IV)

ANIONS

Group V/VI/VII nonmetals	MnO_4^-	permanganate
CO_3^{2-}	AsO_4^{3-}	arsenate
HCO_3^{2-}	SO_4^{2-}	sulfate
	SO_3^{2-}	sulfite
	HSO_4^-	hydrogen sulfate
NO_3^-	HSO_3^-	hydrogen sulfite
NO_2^-	$\text{S}_2\text{O}_3^{2-}$	thiosulfate
N_3^-	ClO_4^-	perchlorate
CN^-	ClO_3^-	chlorate
OCN^-	ClO_2^-	chlorite
SCN^-	ClO^-	hypochlorite
O_2^{2-}	BrO_4^-	perbromate
OH^-	BrO_3^-	bromate
$\text{C}_2\text{H}_3\text{O}_2^-$	BrO_2^-	bromite
$\text{C}_2\text{O}_4^{2-}$	BrO^-	hypobromite
PO_4^{3-}	IO_4^-	periodate
HPO_4^{2-}	IO_3^-	iodate
H_2PO_4^-	IO_2^-	iodite
PO_3^{3-}	IO^-	hypoiodite
CrO_4^{2-}		
$\text{Cr}_2\text{O}_7^{2-}$		