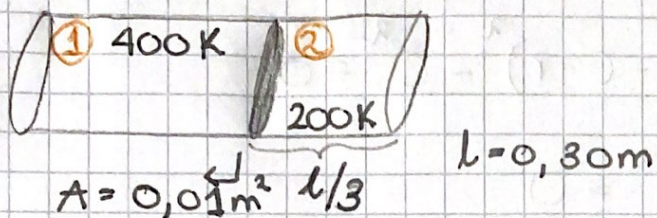


0.3 Termodinámica



$$a) U_{10} + U_{20} = U_{1f} + U_{2f}$$

$$\Rightarrow T_{1f} + T_{2f} = T_{10} + T_{20}$$

$$PV = nRT \Rightarrow T_{1f} = \frac{P_f V_{1f}}{nR}$$

$$P_f = \frac{nR(T_{10} + T_{20})}{V_{\text{total}}}$$

$$\Rightarrow T_{1f} = \frac{(T_{10} + T_{20}) V_{1f}}{V_{\text{total}}} = \frac{(200\text{ K} + 400\text{ K}) (0,1 \cdot 0,01\text{ m}^3)}{0,01\text{ m}^2 \cdot 0,30\text{ m}} = 200\text{ K}$$

$$T_{2f} = 600\text{ K} - 200\text{ K} = 400\text{ K}$$

b) Según la ley de Fourier, el flujo de calor se comporta de la forma

$$\frac{dQ}{dt} = -K A \frac{dT}{dx}$$

Como no hay trabajo, $dQ = dU = nC_V dT$

$$\Rightarrow nC_V \frac{dT}{dt} = -KA(T_1 - T_2)$$

$$c) T_1' = -C(T_1 - T_2)$$

$$T_2' = C(T_1 - T_2)$$

$$x'(t) = A \vec{x}(t)$$

$$\text{En nuestro caso, } A = \begin{pmatrix} -C & C \\ C & -C \end{pmatrix}; \vec{x} = \vec{v} e^{rt}$$

$$(A - rI)\vec{v} = 0$$

$$\det \begin{pmatrix} -C-r & C \\ C & -C-r \end{pmatrix} = 0 \quad (-C-r)^2 - C^2 = 0$$

$$\Rightarrow \cancel{C^2} - 2Cr + \cancel{r^2} - \cancel{C^2} = 0$$

$$r = 2Cr \Rightarrow r = 2C; r = 0$$

$$r = 2C$$

$$\begin{pmatrix} -3C & C \\ C & -3C \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \end{pmatrix} = 0, \quad \vec{v} = \begin{pmatrix} 1 \\ 1 \end{pmatrix}$$

$$-3C(v_1) + C(v_2) = 0 \Rightarrow -3v_1 = -v_2$$

$$v_1(C) - 3C(v_2) = 0 \Rightarrow v_1 = 3v_2$$

$$r = 0$$

$$\begin{pmatrix} -C & C \\ C & -C \end{pmatrix} \begin{pmatrix} v_1 \\ v_2 \end{pmatrix} = 0 \quad \begin{aligned} -Cv_1 + Cv_2 &= 0 \\ Cv_1 - Cv_2 &= 0 \end{aligned}$$

$$\vec{v} = \begin{pmatrix} -1 \\ 1 \end{pmatrix}$$

$$\vec{x} = C_1 \begin{pmatrix} 1 \\ 1 \end{pmatrix} e^{2ct} + C_2 \begin{pmatrix} -1 \\ 1 \end{pmatrix}$$

$$T_1 = C_1 e^{2ct} + C_2$$

$$T_2 = C_1 e^{2ct} + C_2$$

$$\frac{dT_1}{dt} = -C_1 (e^{2ct} \cdot 2c) = -C_1 (T_{10} - T_{20})$$

$$C_1 = -(T_{10} - T_{20}/2)$$

$$C_2 = (T_{10} - T_{20}/2) = -C_1$$