

## Section 1.6: Applications of Linear Systems:

Note: Here we explore applications of linear systems coming from economics, chemistry, & network flow.

\*Purpose: To demonstrate how real-world problems involving linear systems may have only one solution, no solution, &/or many solutions ::

### Application #1

## \*Homogeneous Systems in Economics; Equilibrium Prices\*

Here we explore Leontief's simple "exchange model".

- \$ a nation's economy is divided into many sectors

\*Ex: Industries (manufacturing, communication, etc.)

- For each sector, \$ we know we know its total output (per yr.) ~~AND~~ how this output is divided/exchanged amongst other sectors in the economy.

- Price of that output: Total dollar value of a sector's output.

- Conclusion:  $\exists$  'Equilibrium Prices' that can be assigned to the total outputs of the various sectors in such a way that the income of each sector exactly balances its expenses.

## Example (How to Find Equilibrium Prices):

①

\$ an economy consists of the Coal, Electric (power) & Steel sectors, and the output of each sector is distributed among the various sectors as shown in the table below (Note: The entries in a column represent the fractional parts of a sector's total output).

→ Denote the prices (\$) of the total annual outputs of the Coal, Electric, & Steel sectors by  $p_c$ ,  $p_e$ , &  $p_s$ , respectively. IF possible, find the equilibrium prices that make each sector's income match its expenditures.

### \* Distribution of Output Form \*

<u>Coal:</u>	<u>Electric:</u>	<u>Steel:</u>	<u>Purchased By:</u>
0.0	0.4	0.6	Coal
0.6	0.1	0.2	Electric
0.4	0.5	0.2	Steel

Note: Since all outputs are taken into account, the decimal fractions in each column must sum to 1.

### Answer:

\* A sector looks down the column to: See where its output goes

\* A sector looks across the row to: See what it needs as inputs

⇒ The sum of the rows entries, multiplied by their respective outputs, provides the total expenses per sector.



## Example (How to Find Equilibrium Prices) continued...

(2)

\*Step 1: Define the Total Expenses for each sector:

Note: Since the values of total outputs are  $p_c, p_e, p_s$

• Total Expenses of Coal:  $0.4p_e + 0.6p_s$

• Total Expenses of Electric:  $0.6p_c + 0.1p_e + 0.2p_s$

• Total Expenses of Steel:  $0.4p_c + 0.5p_e + 0.2p_s$

\*To find the Equilibrium, set "Total Income" = "Total Expenses"

$$\begin{cases} p_c = 0.4p_e + 0.6p_s \\ p_e = 0.6p_c + 0.1p_e + 0.2p_s \\ p_s = 0.4p_c + 0.5p_e + 0.2p_s \end{cases}$$

\*Step 2: Define the System of Equations

To define the System, we bring all the unknowns ( $p_c, p_e, p_s$ ) to the LHS:

$$\begin{cases} p_c = 0.4p_e + 0.6p_s \\ p_e = 0.6p_c + 0.1p_e + 0.2p_s \\ p_s = 0.4p_c + 0.5p_e + 0.2p_s \end{cases} \sim \begin{cases} p_c - 0.4p_e - 0.6p_s = 0 \\ -0.6p_c + 0.9p_e - 0.2p_s = 0 \\ -0.4p_c - 0.5p_e + 0.8p_s = 0 \end{cases}$$

\*Note (Optional): Since we are not allowed calculators in this course, I am going to multiply the entire system by "10" to remove the decimals  $\rightarrow$  Provides exact solution (verses approx.)

### Example (How to Find Equilibrium Price) Continued...

③

$$\sim \begin{cases} 10p_c - 4p_z - 6p_s = 0 \\ -6p_c + 9p_z - 2p_s = 0 \\ -4p_c - 5p_z + 8p_s = 0 \end{cases} \xrightarrow{\frac{1}{2}R_1} \begin{cases} 5p_c - 2p_z - 3p_s = 0 \\ -6p_c + 9p_z - 2p_s = 0 \\ -4p_c - 5p_z + 8p_s = 0 \end{cases}$$

\*Step 3: Solve the System:

Convert the system to its equivalent augmented matrix & row reduce as usual:

$$\Rightarrow [A \mid 0] = \left[ \begin{array}{ccc|c} 5 & -2 & -3 & 0 \\ -6 & 9 & -2 & 0 \\ -4 & -5 & 8 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot \frac{R_3}{+ R_1} \\ \hline \text{new } R_1 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & -7 & 5 & 0 \\ -6 & 9 & -2 & 0 \\ -4 & -5 & 8 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot \frac{6R_1}{+ R_2} \\ \hline \text{new } R_2 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & -7 & 5 & 0 \\ 0 & -33 & 28 & 0 \\ -4 & -5 & 8 & 0 \end{array} \right]$$



(4)

Example (How to Find Equilibrium Price) Continued...

$$\begin{array}{l} \bullet \quad 4R_1 \\ + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & -7 & 5 & 0 \\ 0 & -33 & 28 & 0 \\ 0 & -33 & 28 & 0 \end{array} \right]$$

$$\begin{array}{l} \bullet \quad -R_2 \\ + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & -7 & 5 & 0 \\ 0 & -33 & 28 & 0 \\ 0 & 0 & 0 & 0 \end{array} \right] \xrightarrow{-\frac{1}{33}R_2} \left[ \begin{array}{ccc|c} 1 & -7 & 5 & 0 \\ 0 & 1 & -\frac{28}{33} & 0 \\ 0 & 0 & 0 & 0 \end{array} \right] *$$

\*Note:  $x_3$  is a 'free variable'  $\rightarrow$  Nontrivial Solution 3.

$$\begin{array}{l} \bullet \quad 7R_2 \\ + R_1 \\ \hline \text{new } R_1 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & 0 & -\frac{31}{33} & 0 \\ 0 & 1 & -\frac{28}{33} & 0 \\ 0 & 0 & 0 & 0 \end{array} \right] \Leftrightarrow \begin{cases} p_c = \frac{31}{33} p_s \\ p_e = \frac{28}{33} p_s \\ p_s \text{ is free} \end{cases}$$

$$* \quad 7\left(-\frac{28}{33}\right) + 5 = -\frac{196}{33} + 5 = -\frac{196 + 165}{33} \quad \therefore$$

\*Step 4: Write the General Solution for Equilibrium price in Vector Form:

$$\vec{p} = \begin{bmatrix} p_c \\ p_e \\ p_s \end{bmatrix} = \begin{bmatrix} \frac{31}{33} p_s \\ \frac{28}{33} p_s \\ p_s \end{bmatrix} = p_s \begin{bmatrix} \frac{31}{33} \\ \frac{28}{33} \\ 1 \end{bmatrix}$$

\*Note: Any non-negative choice for  $p_s$  results in a choice of equilibrium prices  $\therefore$

Answer.

Example: \$ an economy has only 2 sectors: Goods & Services. Each year, Goods sells 75% of its outputs to Services & keeps the rest; Services sells 69% of its outputs to Goods & retains the rest. Find equilibrium prices for the annual outputs of the Goods & Services sectors that make each sector's income match its expenditures.

Answer:

\*Create a table for the Distribution of Output:

Recall: A sector looks...

i) Down the Column: To see where the output goes  
\*Sum of each column = 1\*

ii) Down the Row: To see what it needs as input

\*Sum of the entries, multiplied by their respective outputs, = total expenses (per sector) \*

<u>Goods:</u>	<u>Services:</u>	<u>Purchased By:</u>
0.25	0.69	Goods
0.75	0.31	Services

\*Let  $P_G$  = price of total annual output of "Goods"  
 (total income)

$P_S$  = price of total annual output of "Services"  
 (total income)



### Example Continued...

\* Define the Total Income (i.e. output) = Total Expenses:

• Good's Expenses:  $0.25p_G + 0.69p_S$

• Services's Expenses:  $0.75p_G + 0.31p_S$

$$\Rightarrow \boxed{\therefore \begin{cases} p_G = 0.25p_G + 0.69p_S \\ p_S = 0.75p_G + 0.31p_S \end{cases}}$$

\* Find the General Solution for the Good & Services System

$$\begin{cases} 0.75p_G - 0.69p_S = 0 \\ -0.75p_G + 0.69p_S = 0 \end{cases} \Leftrightarrow \left[ \begin{array}{cc|c} 0.75 & -0.69 & 0 \\ -0.75 & 0.69 & 0 \end{array} \right]$$

•  $\begin{matrix} R_1 \\ + R_2 \\ \hline \text{new } R_2 \end{matrix} \rightarrow \left[ \begin{array}{cc|c} 0.75 & -0.69 & 0 \\ 0 & 0 & 0 \end{array} \right] \xrightarrow[\sim]{\frac{R_1}{0.75}} \left[ \begin{array}{cc|c} 1 & -0.92 & 0 \\ 0 & 0 & 0 \end{array} \right]$

Ans

$$\Rightarrow \begin{cases} p_G = 0.92p_S \\ p_S \text{ is free} \end{cases}$$

$\therefore$  General Sol. for Equil. Price:

$$\vec{p} = \begin{bmatrix} p_G \\ p_S \end{bmatrix} = p_S \begin{bmatrix} 0.92 \\ 1 \end{bmatrix}$$

Notes: \* IF  $p_S = \$1000$ , then:  $p_G = 0.92(1000) = \$920$

Problem: \$ that an economy consists of Coal, Electric, & Steel sectors. Denote the prices (\$) of the total annual outputs of the coal, electric, & steel sectors by  $p_c$ ,  $p_e$ , &  $p_s$  respectively. \$ the general solution to find the equilibrium prices that make each sector's income match its expenditures is:

$$\begin{cases} p_c = 0.94 p_s \\ p_e = 0.9 p_s \\ p_s \text{ is free} \end{cases}$$

(a) \* One Set of Equilibrium Prices is  $p_c = \$94$ ,  $p_e = \$90$ , &  $p_s = \$100$ . Find another set.

(b) \* \$ the same economy used Japanese yen instead of dollars to measure the value of the various sector's output. Would this change the problem in any way?

Answer:

• Part (a): Since  $p_s$  is free  $\rightarrow$  Choose any  $\mathbb{Z}^+$   $\therefore$

$$\text{IF } p_s = \$300: \begin{cases} p_c = 0.94(300) = \$282 \\ p_e = 0.9(300) = \$270 \end{cases}$$

Not an exclusive ans.

• Part (b): Will need to convert dollars to yen

$\therefore$  Multiply the equilibrium prices in general solution by a scalar  $\Rightarrow$  IOW: Prices change, but ratio is the same.



## Application #2

### \*Balancing Chemical Equations\*

A chemical equation describes the quantities of substances consumed and produced by different chemical reactions.

Note: Since atoms can neither be destroyed nor created in a chemical reaction, chemical equations must be "balanced"  $\therefore$

#### To "Balance" a Chemical Equation:

- We need the total # of atoms on the LHS to match the total # of atoms on the RHS

$\rightarrow$  IOW: Find  $\vec{x} = \{x_1, x_2, \dots, x_n\} \in \mathbb{Z}^+$  st the entries (atoms) are whole #s (positive integers).

#### A Systematic Method For Balancing Chemical Equations:

- Set-up a vector equation that describes the # of atoms of each type present in a reaction

$\rightarrow$  IOW:  $A\vec{x} = [a_1 \ a_2 \ \dots \ a_n] \begin{bmatrix} x_1 \\ x_2 \\ \vdots \\ x_n \end{bmatrix} = x_1\vec{a}_1 + x_2\vec{a}_2 + \dots + x_n\vec{a}_n$

### Example (Balancing Chemical Equations):

①

When propane gas burns, the propane ( $C_3H_8$ ) combines with oxygen ( $O_2$ ) to form carbon dioxide ( $CO_2$ ) & water ( $H_2O$ ), according to an equation of the form:



Balance this equation.

$\Rightarrow$  find whole #s  $x_1, \dots, x_4$  such that the total # of carbon (C), hydrogen (H), & oxygen (O) atoms on the LHS match the RHS.

Answer:

\*Step 1: Construct a vector in  $\mathbb{R}^3$  for each reactant & product that lists the # of atoms per molecule:

• Propane,  $C_3H_8$ :  $\begin{bmatrix} 3 \\ 8 \\ 0 \end{bmatrix}$   $\leftarrow$  \* Carbon  
 $\leftarrow$  \* Hydrogen  
 $\leftarrow$  \* Oxygen

} same order for remaining vectors ::

• Oxygen,  $O_2$ :  $\begin{bmatrix} 0 \\ 0 \\ 2 \end{bmatrix}$

• Water,  $H_2O$ :  $\begin{bmatrix} 0 \\ 2 \\ 1 \end{bmatrix}$

• Carbon Dioxide,  $CO_2$ :  $\begin{bmatrix} 1 \\ 0 \\ 2 \end{bmatrix}$



## Example (Balancing Chemical Equations): Continued...

(2)

\*Step 2: Set-Up a vector equation:

• Chemical Equation:



• Vector Equation:

$$x_1 \begin{bmatrix} 3 \\ 8 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 0 \\ 2 \end{bmatrix} = x_3 \begin{bmatrix} 1 \\ 0 \\ 2 \end{bmatrix} + x_4 \begin{bmatrix} 0 \\ 2 \\ 1 \end{bmatrix}$$

$$x_1 \begin{bmatrix} 3 \\ 8 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 0 \\ 2 \end{bmatrix} - x_3 \begin{bmatrix} 1 \\ 0 \\ 2 \end{bmatrix} - x_4 \begin{bmatrix} 0 \\ 2 \\ 1 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix}$$

$$x_1 \begin{bmatrix} 3 \\ 8 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 0 \\ 2 \end{bmatrix} + x_3 \begin{bmatrix} -1 \\ 0 \\ -2 \end{bmatrix} + x_4 \begin{bmatrix} 0 \\ -2 \\ -1 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix}$$

\*Step 3: Solve the augmented matrix using Row Reduction:

$$[A \mid 0] = \left[ \begin{array}{cccc|c} 3 & 0 & -1 & 0 & 0 \\ 8 & 0 & 0 & -2 & 0 \\ 0 & 2 & -2 & -1 & 0 \end{array} \right] \xrightarrow{\frac{1}{2}R_2} \left[ \begin{array}{cccc|c} 3 & 0 & -1 & 0 & 0 \\ 4 & 0 & 0 & -1 & 0 \\ 0 & 2 & -2 & -1 & 0 \end{array} \right]$$
$$\xrightarrow[\frac{1}{4}R_3]{\frac{1}{3}R_1} \left[ \begin{array}{cccc|c} 1 & 0 & -\frac{1}{3} & 0 & 0 \\ 4 & 0 & 0 & -1 & 0 \\ 0 & 1 & -1 & -\frac{1}{2} & 0 \end{array} \right]$$

# Example (Balancing Chemical Equations) Continued...

③

$$\begin{array}{l} \bullet -4R_1 \\ + R_2 \\ \hline \text{new } R_2 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & -1/3 & 0 & 0 \\ 0 & 0 & 4/3 & -1 & 0 \\ 0 & 1 & -1 & -1/2 & 0 \end{array} \right] \xrightarrow{\frac{3}{4}R_2} \left[ \begin{array}{cccc|c} 1 & 0 & -1/3 & 0 & 0 \\ 0 & 0 & 1 & -3/4 & 0 \\ 0 & 1 & -1 & -1/2 & 0 \end{array} \right]$$

$$\bullet \text{Interchanging } R_2 \text{ \& } R_3 \rightarrow \sim \left[ \begin{array}{cccc|c} 1 & 0 & -1/3 & 0 & 0 \\ 0 & 1 & -1 & -1/2 & 0 \\ 0 & 0 & 1 & -3/4 & 0 \end{array} \right] \quad \begin{array}{l} * x_4 \text{ is free} \\ \rightarrow \text{Nontrivial} \\ \text{Solution } \exists \end{array}$$

$$\bullet \begin{array}{l} R_3 \\ + R_2 \\ \hline \text{new } R_2 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & -1/3 & 0 & 0 \\ 0 & 1 & 0 & -5/4 & 0 \\ 0 & 0 & 1 & -3/4 & 0 \end{array} \right]$$

$$\bullet \begin{array}{l} \frac{1}{3}R_3 \\ + R_1 \\ \hline \text{new } R_1 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & 0 & -1/4 & 0 \\ 0 & 1 & 0 & -5/4 & 0 \\ 0 & 0 & 1 & -3/4 & 0 \end{array} \right] \Leftrightarrow \begin{cases} x_1 = \frac{1}{4}x_4 \\ x_2 = \frac{5}{4}x_4 \\ x_3 = \frac{3}{4}x_4 \\ x_4 \text{ is free} \end{cases}$$

∴ The General Sol. for the Balanced Eq:

$$\vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{bmatrix} = x_4 \begin{bmatrix} 1/4 \\ 5/4 \\ 3/4 \\ 1 \end{bmatrix}$$

Note: Since  $\vec{x} \in \mathbb{Z}^+$ , the smallest coefficients are produced when  $x_4 = 4$  ∴  
BUT: This is NOT an exclusive solution ✓



Balance the following chemical equation. Assume that the coefficient on  $\text{H}_2\text{S}$  is '3'.



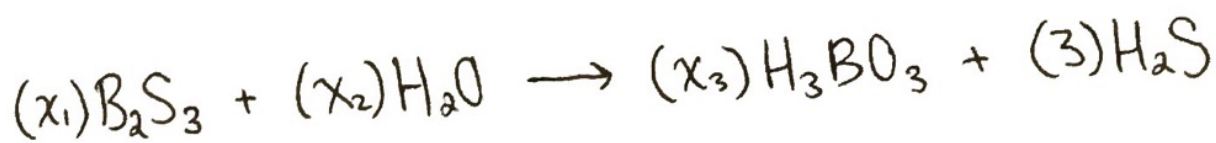
"Boron Sulfide reacts violently with water to form Boric Acid & Hydrogen Sulfide gas"... stinky!

Answer:

\* Let  $\vec{x}$  be a vector whose entries are the amount (whole number) of each compound needed to balance the given chemical eq:

$$\rightarrow \vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{bmatrix} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ 3 \end{bmatrix} \begin{array}{l} \rightarrow \text{Boron Sulfide} \\ \rightarrow \text{Water} \\ \rightarrow \text{Boric Acid} \\ \rightarrow \text{Hydrogen Sulfide Gas} \end{array}$$

So,



\* Set-up a vector that lists the number of atoms per molecule for each reactant/product in the eq:

$$\rightarrow \begin{bmatrix} 2 \\ 3 \\ 0 \\ 0 \end{bmatrix} : \text{B}_2\text{S}_3, \begin{bmatrix} 0 \\ 0 \\ 2 \\ 1 \end{bmatrix} : \text{H}_2\text{O}, \begin{bmatrix} 1 \\ 0 \\ 3 \\ 3 \end{bmatrix} : \text{H}_3\text{BO}_3, \begin{bmatrix} 0 \\ 1 \\ 2 \\ 0 \end{bmatrix} : \text{H}_2\text{S}$$

So,

$$(x_1) \begin{bmatrix} 2 \\ 3 \\ 0 \\ 0 \end{bmatrix} + (x_2) \begin{bmatrix} 0 \\ 0 \\ 2 \\ 1 \end{bmatrix} = (x_3) \begin{bmatrix} 1 \\ 0 \\ 3 \\ 3 \end{bmatrix} + (3) \begin{bmatrix} 0 \\ 1 \\ 2 \\ 0 \end{bmatrix}$$

\* To Balance the Chemical Eq., set-up a vector equation & then row-reduce the equivalent augmented matrix to produce a general solution:

$$x_1 \begin{bmatrix} 2 \\ 3 \\ 0 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 0 \\ 2 \\ 1 \end{bmatrix} = x_3 \begin{bmatrix} 1 \\ 0 \\ 3 \\ 3 \end{bmatrix} + \begin{bmatrix} 0 \\ 3 \\ 6 \\ 0 \end{bmatrix}$$

$$x_1 \begin{bmatrix} 2 \\ 3 \\ 0 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 0 \\ 2 \\ 1 \end{bmatrix} + x_3 \begin{bmatrix} -1 \\ 0 \\ -3 \\ -3 \end{bmatrix} = \begin{bmatrix} 0 \\ 3 \\ 6 \\ 0 \end{bmatrix} \iff \begin{bmatrix} 2 & 0 & -1 \\ 3 & 0 & 0 \\ 0 & 2 & -3 \\ 0 & 1 & -3 \end{bmatrix} \begin{bmatrix} x_1 \\ x_2 \\ x_3 \end{bmatrix} = \begin{bmatrix} 0 \\ 3 \\ 6 \\ 0 \end{bmatrix}$$

$$* [A | \vec{b}] = \left[ \begin{array}{ccc|c} 2 & 0 & -1 & 0 \\ 3 & 0 & 0 & 3 \\ 0 & 2 & -3 & 6 \\ 0 & 1 & -3 & 0 \end{array} \right] \xrightarrow{\frac{1}{3}R_2} \left[ \begin{array}{ccc|c} 2 & 0 & -1 & 0 \\ 1 & 0 & 0 & 1 \\ 0 & 2 & -3 & 6 \\ 0 & 1 & -3 & 0 \end{array} \right] \xrightarrow{\substack{\text{interchanging } \\ R_1, R_2, \\ \& R_4}} \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 2 & -3 & 6 \\ 2 & 0 & -1 & 0 \end{array} \right]$$

$$* \begin{array}{l} -2R_1 \\ + R_4 \\ \hline \text{new } R_4 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 2 & -3 & 6 \\ 0 & 0 & -1 & -2 \end{array} \right] \xrightarrow{-R_4} \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 2 & -3 & 6 \\ 0 & 0 & 1 & 2 \end{array} \right]$$

$$* \begin{array}{l} -2R_2 \\ + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 0 & 3 & 6 \\ 0 & 0 & 1 & 2 \end{array} \right] \xrightarrow{\frac{1}{3}R_3} \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 0 & 1 & 2 \\ 0 & 0 & 1 & 2 \end{array} \right]$$

$$* \begin{array}{l} -R_3 \\ + R_4 \\ \hline \text{new } R_4 \end{array} \rightarrow \left[ \begin{array}{ccc|c} 1 & 0 & 0 & 1 \\ 0 & 1 & -3 & 0 \\ 0 & 0 & 1 & 2 \\ 0 & 0 & 0 & 0 \end{array} \right]$$



Ex Continued...

\*  $3R_3$   
+  $R_2$   
-----  
new  $R_2$

$$\begin{bmatrix} 1 & 0 & 0 & 1 \\ 0 & 1 & 0 & 6 \\ 0 & 0 & 1 & 2 \\ 0 & 0 & 0 & 0 \end{bmatrix}$$

Note: While we have found a unique/trivial solution, this b/c we assumed  $x_4 = 3$  :

\*  $R_4$  shows us that " $x_4$ " is free!  
(the amount of  $H_2S$  defines the general solution; Now  $x_4$  determines how much of the other compounds we need :)

$$\Rightarrow \begin{cases} x_1 = 1 \\ x_2 = 6 \\ x_3 = 2 \\ x_4 \text{ is free} \end{cases}$$

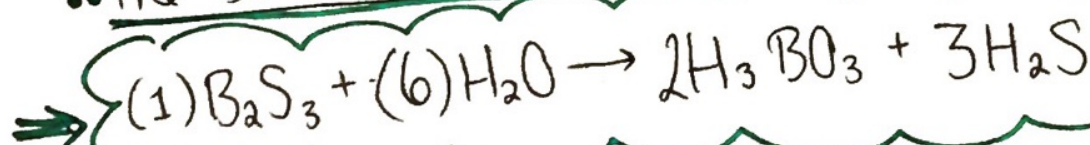
(specifically " $x_4 = 3$ " in this example)

∴ The "Particular" Solution Set of the Nonhomogeneous Equation  $A\vec{x} = \vec{b}$  is defined by the vector:

$$\vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{bmatrix} = \begin{bmatrix} 1 \\ 6 \\ 2 \\ 3 \end{bmatrix} \begin{array}{l} \rightarrow \text{Boron Sulfide} \\ \rightarrow \text{Water} \\ \rightarrow \text{Boric Acid} \\ \rightarrow \text{Hydrogen Sulfide} \end{array}$$

\*(i.e The # of each compound needed to balance eq.)

∴ The Balanced Eq. (Assuming "3" Hydrogen Sulfide):



Example: Balance the following equation:



Answer:

\*Set-up a vector in  $\mathbb{R}^4$  for each reactant & product that lists the # of atoms per molecule:

$$\underline{\text{H}_3\text{O}}: \begin{bmatrix} 3 \\ 1 \\ 0 \\ 0 \end{bmatrix} \begin{matrix} \text{H} \\ \text{O} \\ \text{Ca} \\ \text{C} \end{matrix}, \quad \underline{\text{CaCO}_3}: \begin{bmatrix} 0 \\ 3 \\ 1 \\ 1 \end{bmatrix}, \quad \underline{\text{H}_2\text{O}}: \begin{bmatrix} 2 \\ 1 \\ 0 \\ 0 \end{bmatrix}, \quad \underline{\text{Ca}}: \begin{bmatrix} 0 \\ 0 \\ 1 \\ 0 \end{bmatrix}, \quad \underline{\text{CO}_2}: \begin{bmatrix} 0 \\ 2 \\ 0 \\ 1 \end{bmatrix}$$

\*Set-up a vector eq:

$$(x_1) \begin{bmatrix} 3 \\ 1 \\ 0 \\ 0 \end{bmatrix} + (x_2) \begin{bmatrix} 0 \\ 3 \\ 1 \\ 1 \end{bmatrix} = (x_3) \begin{bmatrix} 2 \\ 1 \\ 0 \\ 0 \end{bmatrix} + (x_4) \begin{bmatrix} 0 \\ 0 \\ 1 \\ 0 \end{bmatrix} + (x_5) \begin{bmatrix} 0 \\ 2 \\ 0 \\ 1 \end{bmatrix}$$

$$x_1 \begin{bmatrix} 3 \\ 1 \\ 0 \\ 0 \end{bmatrix} + x_2 \begin{bmatrix} 0 \\ 3 \\ 1 \\ 1 \end{bmatrix} + x_3 \begin{bmatrix} -2 \\ -1 \\ 0 \\ 0 \end{bmatrix} + x_4 \begin{bmatrix} 0 \\ 0 \\ -1 \\ 0 \end{bmatrix} + x_5 \begin{bmatrix} 0 \\ -2 \\ 0 \\ -1 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 0 \\ 0 \end{bmatrix}$$

\*Solve the augmented matrix using row reduction:

$$[A | 0] = \left[ \begin{array}{ccccc|c} 3 & 0 & -2 & 0 & 0 & 0 \\ 1 & 3 & -1 & 0 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \end{array} \right] \begin{matrix} \text{inter-} \\ \text{change} \\ R_1, R_2 \text{ \& } \\ R_3 \text{ \& } \end{matrix} \left[ \begin{array}{ccccc|c} 1 & 3 & -1 & 0 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 3 & 0 & -2 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \end{array} \right]$$



## Example Continued...

②

$$\begin{array}{l} \cdot -3R_1 \\ \quad + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{ccccc|c} 1 & 3 & -1 & 0 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & -9 & 1 & 0 & 6 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot 9R_2 \\ \quad + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{ccccc|c} 1 & 3 & -1 & 0 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot -3R_2 \\ \quad + R_1 \\ \hline \text{new } R_1 \end{array} \rightarrow \left[ \begin{array}{ccccc|c} 1 & 0 & -1 & 3 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot -R_2 \\ \quad + R_4 \\ \hline \text{new } R_4 \end{array} \rightarrow \left[ \begin{array}{ccccc|c} 1 & 0 & -1 & 3 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} \cdot R_3 \\ \quad + R_1 \\ \hline \text{new } R_1 \end{array} \rightarrow \left[ \begin{array}{ccccc|c} 1 & 0 & 0 & -6 & 4 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{array} \right]$$

(3)

Ex. Continued...

$$\begin{array}{l} 6R_4 \\ + R_1 \\ \hline \text{new } R_1 \end{array} \rightarrow$$

$$\left[ \begin{array}{ccccc|c} 1 & 0 & 0 & 0 & -2 & 0 \\ 0 & 1 & 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} R_4 \\ + R_2 \\ \hline \text{new } R_2 \end{array} \rightarrow$$

$$\left[ \begin{array}{ccccc|c} 1 & 0 & 0 & 0 & -2 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \\ 0 & 0 & 1 & -9 & 6 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{array} \right]$$

$$\begin{array}{l} 9R_4 \\ + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow$$

$$\left[ \begin{array}{ccccc|c} 1 & 0 & 0 & 0 & -2 & 0 \\ 0 & 1 & 0 & 0 & -1 & 0 \\ 0 & 0 & 1 & 0 & -3 & 0 \\ 0 & 0 & 0 & 1 & -1 & 0 \end{array} \right]$$

$$\Rightarrow \begin{cases} x_1 = 2x_5 \\ x_2 = x_5 \\ x_3 = 3x_5 \\ x_4 = x_5 \\ x_5 \text{ is free} \end{cases}$$

 $\therefore$  General Sol:

$$\vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \\ x_5 \end{bmatrix} = x_5 \begin{bmatrix} 2 \\ 1 \\ 3 \\ 1 \\ 1 \end{bmatrix}$$

Answer ✓

\*Network Flow\*

Systems of Linear Equations arise when we study the Flow of some quantity through a network.

- A network consists of a set of points called 'junctions' or 'nodes', w/ lines/arcs called 'branches' connecting some or all of the junctions.

\* The direction of flow in each branch is indicated

\* The Flow amount/rate is either shown or denoted by a variable.

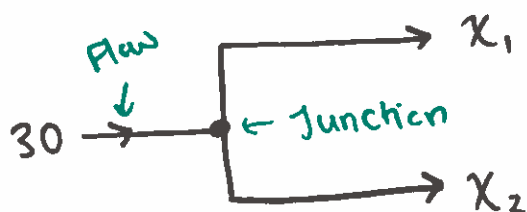
- The Basic Assumption of Network Flow is that:

i) Total Flow into Network = Total Flow out of Network

-AND-

ii) Total Flow into Junction = Total Flow out of Junction

- Illustration:



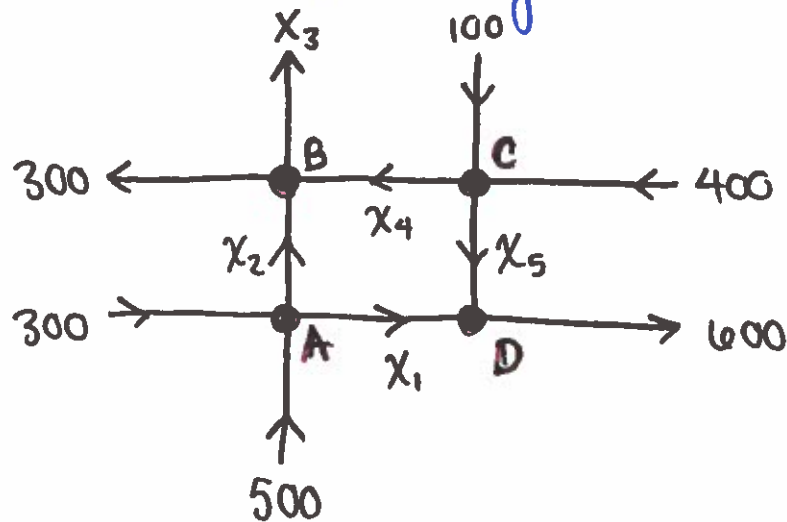
\* 30 units Flows into a junction through 1 branch.

\*  $x_1$  &  $x_2$  denote Flow out of junction through 2 branches.

\* Since "Flow" is conserved  $\Rightarrow 30 = x_1 + x_2$



Example (Network Flow): The network in the figure below shows the traffic flow (in vehicles per hour) over several one-way streets in a city during a typical early afternoon. Determine the general flow pattern for the network.



Answer:

\*Step 1: Write an equation that describes the 'Flow' @ each junction:

Note:

- LHS of  $E_j$  = Flow into the junction.
- RHS of  $E_j$  = Flow out of the junction.

- Junction A:  $300 + 500 = x_1 + x_2$
- Junction B:  $x_2 + x_4 = x_3 + 300$
- Junction C:  $100 + 400 = x_4 + x_5$
- Junction D:  $x_1 + x_5 = 600$

\*Step 2: Find the General Solution of the 'Flow' System:

- Bring all unknowns to LHS
- Use row reduction to solve the augmented matrix

# Example (Network Flow): Continued...

(2)

$$\begin{cases} x_1 + x_2 = 800 \\ x_2 - x_3 + x_4 = 300 \\ x_4 + x_5 = 500 \\ x_1 + x_5 = 600 \end{cases} \iff \begin{bmatrix} 1 & 1 & 0 & 0 & 0 & | & 800 \\ 0 & 1 & -1 & 1 & 0 & | & 300 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \\ 1 & 0 & 0 & 0 & 1 & | & 600 \end{bmatrix}$$

$\begin{matrix} -R_1 \\ +R_4 \end{matrix} \rightarrow \text{new } R_2$

$$\begin{bmatrix} 1 & 1 & 0 & 0 & 0 & | & 800 \\ 0 & 1 & -1 & 1 & 0 & | & 300 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \\ 0 & -1 & 0 & 0 & 1 & | & -200 \end{bmatrix} \xrightarrow[\sim]{\substack{\text{interchange} \\ R_2, R_3, \& \\ R_4}} \begin{bmatrix} 1 & 1 & 0 & 0 & 0 & | & 800 \\ 0 & -1 & 0 & 0 & 1 & | & -200 \\ 0 & 1 & -1 & 1 & 0 & | & 300 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \end{bmatrix}$$

$\begin{matrix} -R_2 \\ +R_1 \end{matrix} \rightarrow \text{new } R_1$

$$\begin{bmatrix} 1 & 0 & 0 & 0 & 1 & | & 600 \\ 0 & -1 & 0 & 0 & 1 & | & -200 \\ 0 & 1 & -1 & 1 & 0 & | & 300 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \end{bmatrix}$$

$\begin{matrix} R_2 \\ +R_3 \end{matrix} \rightarrow \text{new } R_3$

$$\begin{bmatrix} 1 & 0 & 0 & 0 & 1 & | & 600 \\ 0 & -1 & 0 & 0 & 1 & | & -200 \\ 0 & 0 & -1 & 1 & 1 & | & 100 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \end{bmatrix} \xrightarrow[\sim]{\begin{matrix} -R_2 \\ +R_3 \end{matrix}} \begin{bmatrix} 1 & 0 & 0 & 0 & 1 & | & 600 \\ 0 & 1 & 0 & 0 & -1 & | & 200 \\ 0 & 0 & 1 & -1 & -1 & | & -100 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \end{bmatrix}$$

$\begin{matrix} R_4 \\ +R_3 \end{matrix} \rightarrow \text{new } R_3$

$$\begin{bmatrix} 1 & 0 & 0 & 0 & 1 & | & 600 \\ 0 & 1 & 0 & 0 & -1 & | & 200 \\ 0 & 0 & 1 & 0 & 0 & | & 400 \\ 0 & 0 & 0 & 1 & 1 & | & 500 \end{bmatrix} \iff \begin{cases} x_1 = 600 - x_5 \\ x_2 = 200 + x_5 \\ x_3 = 400 \\ x_4 = 500 - x_5 \\ x_5 \text{ is free} \end{cases}$$

## Example (Network Flow) Continued...

(3)

∴ General Solution For the 'Flow' Pattern of the Network:

$$\vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \\ x_5 \end{bmatrix} = \begin{bmatrix} 600 - x_5 \\ 200 + x_5 \\ 400 \\ 500 - x_5 \\ x_5 \end{bmatrix} = \begin{bmatrix} 600 \\ 200 \\ 400 \\ 500 \\ 0 \end{bmatrix} + x_5 \begin{bmatrix} -1 \\ 1 \\ 0 \\ -1 \\ 1 \end{bmatrix}$$

Notes: A negative flow in a network branch corresponds to flow in the direction opposite to that shown in the model

\*For this specific example

• Since the streets are all one-way  $\Rightarrow$  NO variable can be negative!

$\Rightarrow$  Since Eq(1) would produce  $\ominus$  values, we can use Eq(4) to find restrictions/limitations on variables  $\therefore$

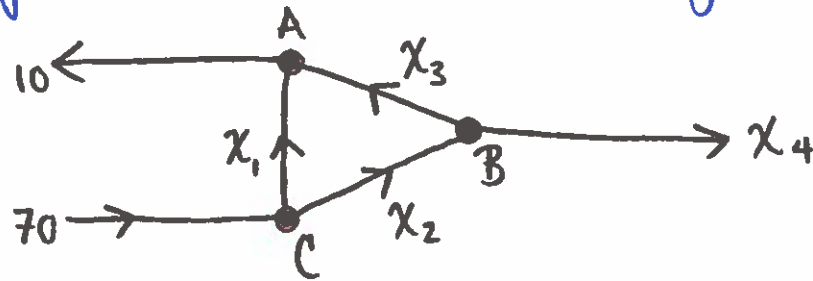
$$500 - x_5 \geq 0 \rightarrow \boxed{500 \geq x_5} \quad \checkmark$$

\*Caution: If we use Eq(1), then  $x_5 \leq 600 \dots$

This is OK. for Eq(1), but if  $x_5 \in (500, 600]$  then  $x_4 < 0 \therefore$



Example: Find the general Flow pattern for network shown in the figure below. Assuming that the flows are all non-negative, what is the largest possible value for  $x_3$ ?



Answer:

\*Write an equation that describes the Flow @ each node:

Note: Here we will let

i) LHS = the Flow into the node

ii) RHS = the Flow out of the node.

• Node A:  $x_1 + x_3 = 10$

• Node B:  $x_2 = x_3 + x_4$

• Node C:  $70 = x_1 + x_2$

$$\Rightarrow \begin{cases} x_1 + x_3 = 10 \\ x_2 - x_3 - x_4 = 0 \\ x_1 + x_2 = 70 \end{cases}$$

\*Find a General Solution for this Network Flow:

Note: Solve the equivalent augmented matrix using row reduction

$$[A : \vec{b}] = \left[ \begin{array}{cccc|c} 1 & 0 & 1 & 0 & 10 \\ 0 & 1 & -1 & -1 & 0 \\ 1 & 1 & 0 & 0 & 70 \end{array} \right]$$

### Example Continued...

$$\begin{array}{l} \cdot -R_1 \\ \quad + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & 1 & 0 & 10 \\ 0 & 1 & -1 & -1 & 0 \\ 0 & 1 & -1 & 0 & 60 \end{array} \right]$$

$$\begin{array}{l} \cdot -R_2 \\ \quad + R_3 \\ \hline \text{new } R_3 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & 1 & 0 & 10 \\ 0 & 1 & -1 & -1 & 0 \\ 0 & 0 & 0 & 1 & 60 \end{array} \right]$$

$$\begin{array}{l} \cdot R_3 \\ \quad + R_2 \\ \hline \text{new } R_2 \end{array} \rightarrow \left[ \begin{array}{cccc|c} 1 & 0 & 1 & 0 & 10 \\ 0 & 1 & -1 & 0 & 60 \\ 0 & 0 & 0 & 1 & 60 \end{array} \right]$$

\* Note:  $x_3$  is free;  
Nontrivial sol. }

\* Therefore: The General Solution is:

$$\left\{ \begin{array}{l} \cdot x_1 = 10 - x_3 \\ \cdot x_2 = x_3 + 60 \\ \cdot x_3 \text{ is free} \\ \cdot x_4 = 60 \end{array} \right. \iff \vec{x} = \begin{bmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{bmatrix} = \begin{bmatrix} 10 - x_3 \\ 60 + x_3 \\ x_3 \\ 60 \end{bmatrix}$$

\* Since  $\vec{x} \geq 0$ :

$$10 - x_3 \geq 0 \rightarrow 10 \geq x_3$$

\* Largest Possible value for  $x_3$  is 10.

$$\vec{x} = \begin{bmatrix} 10 \\ 60 \\ 0 \\ 60 \end{bmatrix} + x_3 \begin{bmatrix} -1 \\ 1 \\ 1 \\ 0 \end{bmatrix}$$