6. Ideal Gas

Kuan-Hsuan Yeh

November 14, 2017

- The ideal gas is a gas of noninteracting atoms in the limit of low concentration.
- An **orbital** is a state of the Schrodinger equation for only one particle.
- Any particle with half-integral spin is a **fermion**, and any particle with zero or integral spin is a **boson**.
- Occupancy rules of orbital model:
 - An orbital can be occupied by any integral number of bosons of the same species, including 0.
 - An orbital can be occupied by 0 or 1 fermion of the same species.
 (also known as the Pauli exclusion principle)
- The limit in which the orbital occupancy is small in comparison with unity is called the **classical regime**
- Fermic-Dirac distribution function gives the average number of fermions in a single orbital of energy ϵ (value lies between 0 and 1):

$$f(\epsilon) = \frac{1}{exp[(\epsilon - \mu)/\tau] + 1}$$

• Bose-Einstein distribution function:

$$f(\epsilon) = \frac{1}{exp[(\epsilon - \mu)/\tau] - 1}$$

• classical distribution function - when $f(\epsilon)$ is $\ll 1$:

$$f(\epsilon) = \lambda exp(-\epsilon/\tau)$$

• Any theory which contains \hbar cannot be a classical theory.

•