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The two isotopes of chlorine are and .Which isotope is the most abundant? Hint D.1 How to approach the problem Hint not displayed Hint D.2 Find the atomic mass Hint not displayed Hint D.3 Mass of each isotope Hint not displayed ANSWER: Correct Since the atomic mass of chlorine, 35.45 ,

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Express your answer using two significant figures. ANSWER: Chapter 10 Question 10 - Bimodal Description: (a) A sample of a gas originally at 25 °C and 1.00 atm pressure in a 2.5 L container is subject to a pressure of 0.85 atm and a temperature of 15 °C. The final volume of the gas is \_\_\_\_ L. Part A

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Make an outline of the materials you have been studying the chapter, without having the book open. How close does your chapter outline compare to the chapter summary and key terms, key skills, and key equations at the end of each chapter? Ensure you have the big picture and focus on meeting the chapter objectives. 5.) Practice.

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