

## *Modern Chemistry Assessment Chapter Test Reaction Kinetics*

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### Modern Chemistry Assessment Chapter Test

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

### Assessment Chapter Test B - clarkchargers.org

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1.

### Assessment Chapter Test A - mrbijlani.weebly.com

Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

### Assessment Chapter Test A - Kettering City School District

Modern Chemistry 68 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 21. Metal X replaces the ions of metal Y from solution, but it cannot replace the ions of metal Z from solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z b. z, x, y c. y, x, z d. z, y, x \_\_\_\_ 22.

### Assessment Chapter Test A - clarkchargers.org

Modern Chemistry 56 Chapter Test Assessment Chapter Test B Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The production of a slightly soluble solid compound in a double-

### Assessment Chapter Test B - Hatboro

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST HR W material copyrighted under notice appearing earlier in this work.

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Assessment Chapter 5 Pre-Test Chapter: The Periodic Law ... Modern Chemistry 2 The Periodic Law Chapter 5 Pre-Test, continued \_\_\_\_ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of ...

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

### **Assessment Chapter Test A - Kettering City School District**

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### **10 States of Matter - Ms. Agostine's Chemistry Page**

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

### **mc06se cFMsR i-vi**

Modern Chemistry 137 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. \_\_\_\_20. In the figure on the

previous page, the volume of titration standard

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