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Modern Chemistry • CHAPTER 9 HOMEWORK 9-4 (pp. 284–287) VOCABULARY Complete each sentence. 1. The sum of the masses on the reactant side must equal . 2.

HOMEWORK 9-1

1. Given the balanced equation $2C\ 4H10\ +\ 13O\ 2\ 8CO\ 2\ +\ 10H\ 2O$, how many moles of $CO\ 2$ are produced when $14.9\ g$ of $O\ 2$ are used? a. $0.29\ mol\ b$. $3.49\ mol\ c$. $12.61\ mol\ d$. $119.2\ mol\ 2$. Which statement is true? a. The inverted molar mass is the conversion factor used to convert from moles to grams. b. The mass of a substance is expressed in moles.

HOMEWORK 9-4 - apcscience.com

Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 (pp. 206–211) VOCABULARY Match the symbol with the name that describes it. 1. Br ... CHAPTER 7 • HOMEWORK 7-1 VOCABULARY 1. false 2. true 3. false 4. false 5. true SKILL BUILDER AgCI—In the compound, there is one atom of silver for each atom of chlorine.

HOMEWORK 7-2

Modern Chemistry • CHAPTER 6 HOMEWORK 6-9 VOCABULARY Complete each sentence. 1. Molecular polarity is determined by ______. 2. According to VSEPR theory, the repulsion between the sets of valence-level electrons surrounding an atom causes ...

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-10 (pp. 229–231) VOCABULARY Complete each sentence. 1. The symbols of the elements in a compound are represented in . 2.

HOMEWORK 7-1

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expressed in picometers -- 1 pm = 10-12 m = 10-10 cm) Homework: Section 3-2. III: Counting Atoms. Hard to measure atoms individually. Find out about basic properties of atoms. Use this information to count the number of atoms of an element in a sample whose mass you know. Mole is unit of measure we will use here and for the remainder of the year.

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-3 (pp. 211–213) VOCABULARY Write the prefix that corresponds to each number. Try not to use the book. Number Prefix Number Prefix 16 27 38 49 510 GRAPHIC ORGANIZER Complete the chart by supplying the missing information. Name of Compound Formula of Compound CO 2 Nitrogen dioxide S 2 Cl 2 Carbon ...

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