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Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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12. A photon having an energy corresponding to the energy difference between the two states is emitted by an atom in an excited state when it returns to its ground state.

Chapter 11 Modern Atomic Theory

Modern Chemistry • CHAPTER 12 HOMEWORK 12-3 (pp. 273–375) VOCABULARY ... Refer to the

phase diagram for water in Figure 12-14 on page 381 of the text to answer the ... 104 SECTION 12-3 REVIEW MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work.

HOMEWORK 12-3 - Los Banos Unified School District

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

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CHAPTER 12 REVIEW Liquids and Solids SECTION 12-2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the following descriptions on the right to the crystal type on the left. ionic crystal (a) has mobile electrons in the crystal covalent molecular crystal (b) is hard, brittle, and nonconducting metallic crystal (c) typically has the lowest melting point of the four

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