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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### 10 States of Matter - Ms. Agostine's Chemistry Page

CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

## 6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

## 5 The Periodic Law - Office of Instructional Technology

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Modern Chemistry v Study Guide mc06se\_cFMsr\_i-vi.qxd 11/22/04 12:07 PM Page v. mc06se\_cFMsr\_i-vi.qxd 11/22/04 12:07 PM Page vi. CHAPTER 1 REVIEW Matter and Change SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ... CHAPTER 1 REVIEW Matter and Change MIXED REVIEW

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

## CHAPTER 12 REVIEW Solutions - Weebly

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> 3 a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> 3 b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> 4 c. ammonium phosphate SnS<sub>2</sub> 2 d. tin(IV) sulfide HNO<sub>2</sub> 2 e. nitrous acid 2.

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