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The two isotopes of chlorine are and .Which isotope is the most abundant? Hint D.1 How to approach the problem Hint not displayed Hint D.2 Find the atomic mass Hint not displayed Hint D.3 Mass of each isotope Hint not displayed ANSWER: Correct Since the atomic mass of chlorine, 35.45 , is closer to 35 than it is to 37, must be the most abundant isotope. does in fact account for 75.53% of the ...

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At a certain temperature and pressure, 1.1 of reacts with 3.3 of . Part A If all the and are consumed, what volume of , at the same temperature and pressure, will be produced? Express your answer using two significant figures. ANSWER: Chapter 10 Question 10 - Bimodal

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