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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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164 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 11.1 The Mysterious Electron Goals To explain why it is very difficult to describe the modern view of the electron. To give you some understanding of the nature of the electron by describing how it is like a guitar string. To explain what atomic orbitals are.

Chapter 11 Modern Atomic Theory - An Introduction to Chemistry

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CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

CHAPTER 8 Chemical Equations and Reactions

Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms.(d) involved in covalent bonds. 2.

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