

## ***Modern Chemistry Chapter 9 Section 2 Review Answers***

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Let's try a real chemistry problem and use this concept to determine the limiting reagent and see how it determines how much product will be formed. Problem : Nitrogen gas reacts with hydrogen gas to produce gaseous ammonia ( $\text{NH}_3$ ). If we react 25.0 g of nitrogen with 25.0 g of hydrogen, how much ammonia will be produced.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following  
questions in the space provided. 1. Classify each of the following as a homogeneous or  
heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar  
2. For each type of investigation, select the most appropriate branch of chemistry from the  
following

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