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Ch 6 Study Guide answers

Modern Chemistry Chapter 6 Chemical Bonding . Chemical Bond •A link between atoms that results from the mutual attraction of their nuclei for electrons -Electrostatic attraction between proton and electron -Classified by the way the valence e- are distributed around nuclei of combined atoms .

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. _____ In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms.(d) involved in covalent bonds. 2.

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Modern Chemistry • CHAPTER 6 HOMEWORK 6-9 VOCABULARY ... Circle the letter of the best answer. 1. A molecule with 2 atoms bonded to the central atom generally has which shape? a. linear b. bent c. trigonal-planar d. either a or b 2. Which is true of an AB 4 molecule? a. The electron pairs separate to form a tetrahedron.

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HOMEWORK 6-10 - apcscience.com Modern Chemistry 46 Chapter Test Chapter: Chemical Bonding In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1. The charge on an ion is a. always positive. b. always negative. c. either positive or negative d. zero 2.

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Chapter 6 Notes - Chemical Bonding Chemical bond - A mutual electrical attraction between the nuclei and valence electrons of different atoms that binds the atoms together 6-1 Introduction to Chemical Bonding I. Types of Chemical Bonding A. Ionic Bonding 1. Chemical bonding that results from the electrical attraction between large

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Modern Chemistry • CHAPTER 6 HOMEWORK 6-3 (pp. 167-169) VOCABULARY Write true or false for
each statement. 1. The bond length is the distance between two bonded atoms at their maximum
potential energy 2. The energy required to form a chemical bond between two neutral atoms
is called bond energy 3.

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