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Modern Chemistry Chapter 10 States

change of state from gas directly to solid phase diagram A graph of pressure versus temperature that shows the conditions under which the phases of a substance exist.

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Modern Chemistry Chapter 10 - States of Matter Terms. States of Matter, Kinetic-Molecular Theory, Liquids, Solids, Changes in State, & Water. STUDY. PLAY. kinetic-molecular theory. ... Modern Chemistry Chapter 10. 28 terms. States of Matter. 48 terms. Prentice Hall Chemistry Chapter 13. 35 terms.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

CHEMISTRY NOTES - Chapter 10 States of Matter Goals : To gain an understanding of : 1.. The kinetic theory of matter. 2. Avogadro's hypothesis. 3. The nature of liquids, solids and gases. 4. Phase changes. NOTES: The kinetic theory of matter states that particles which make up all types of matter are in constant motion.

CHEMISTRY NOTES - Chapter 10 States of Matter

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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MODERN CHEMISTRY STATES OF MATTER 81 Copyright © by Holt, Rinehart and Winston. All rights reserved. Name Date Class CHAPTER 10 REVIEW States of Matter SECTION 1 ...

10 States of Matter - MARRIC

Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District

Title: Modern Chemistry Chapter 10 States of Matter 1 Modern Chemistry Chapter 10States of Matter 2 Kinetic-Molecular Theory. The kinetic-molecular theory of matter is based on the idea that

particles of matter are always in motion and this motion has consequences that affect its physical properties. 3 Kinetic-Molecular Theory of Gases

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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