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Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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_____1. Molecules whose water solutions can carry electric current a. are nonpolar. b. ionize in water. c. do not dissolve in water. d. decompose in water ...

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Modern Chemistry 130 Chapter Test Name Class Date Chapter Test B, continued PART II Write the correct term (or terms) in the space provided. 9.A substance that ionizes almost completely in aqueous solutions, producing H 30 ions, is a(n) acid. 10. An acid that contains hydrogen and only one other element is called a(n) acid. 11.

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Modern Chemistry 140 Chapter Test Name Class Date Chapter Test B, continued ______ 7. An acid-base titration determines the solution volumes that are a. chemically equivalent. b. of equal molarity. c. of equal mass. d. of equal molality. _____ 8. In acidic solutions, an indicator that is a weak acid, HIn, is primarily in the form a. In . b. In ...

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