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Study Guide Moles Show All Work What is the molar mass of CuNO_3 ? How many moles in 60 grams of KOH? How many grams in 5 moles of H_2O ? 1.5 moles of N_2 gas = ____ L of N_2 gas? 1 mole = 22.4 L 100 L of HF gas = ____ moles of HF gas? 1 mole = 22.4 L How many moles in 3.01×10^{23} atoms of Hg? 1 mole = 6.02×10^{23} How many atoms in 5 moles of Li?

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STUDY GUIDE FOR CONTENT MASTERY Section 12.2 Stoichiometric Calculations In your textbook,
read about mole-to-mole conversion. Read the following passage and then solve the problems. In
the equation that follows

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This guide can help keep you on track and make the most efficient use of your study time. ... The
Chemistry: Content Knowledge test is designed to measure the knowledge and competencies
necessary ... Mole concept and how it applies to chemical composition a. Avogadro's number, molar
mass, and mole ...

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62 Chemistry: Matter and Change • Chapter 11 Study Guide for Content Mastery Section 11.2 Mass
and the Mole In your textbook, read about the mass of a mole. For each statement below, write true
or false. 1. The isotope hydrogen-1 is the standard used for the relative scale of atomic masses. 2.
The mass of an atom of helium-4 is 4 amu. 3.

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- Describe how a mole is used in chemistry.
- Relate a mole to common counting units.
- Convert moles to number of representative particles and number of representative particles to moles.

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61 TCR Multimedia Resources Chemistry Interactive CD-ROM ...

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of Compounds In your textbook, read about chemical formulas and the mole, the molar mass of
com-pounds, and conversions among mass, moles, and number of particles. Study the table and
the diagram of a methane molecule and a trichloromethane molecule.

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