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Accelerated Chemistry Anderson - MCHS Modern Chemistry 7 Chemical Equations & Reactions CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

### CHAPTER 8 Chemical Equations and Reactions

Chemistry > Modern Chemistry > Pg. 290 Modern Chemistry Modern Chemistry ISBN:

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1)Temperature change 2)Precipitate formation 3)Color change 4)Release of gas 1 answ. 2. see explanantion 1 answ. 3.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Modern Chemistry • CHAPTER 8 HOMEWORK 8-3 (pp. 248–250) VOCABULARY Complete each

sentence. 1. A chemical equation is like an algebraic equation because they both \_\_\_\_\_. 2. The coefficients of a chemical reaction indicate \_\_\_\_\_. 3.

### HOMEWORK 8-1 - Los Banos Unified School District

e. electrolysis of molten potassium chloride decomposition;  $2\text{KCl(l)} \rightarrow 2\text{K(s)} + \text{Cl}_2\text{(g)}$  4.) (l 2 4((( ))) (((

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3.1 discusses some of the most fundamental assertions of chemistry, including the law of conservation of mass, the law of definite proportions, and the law of multiple proportions. It also gives a ...

### 3.1 The Atom

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

### mc06se cFMsr i-vi

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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Study Sections 8.4–8.11. Focus on Goals 3–8 as you study. Strategy Table 8.2 provides a summary of all the types of reactions and equations. As with Assignment A, practice is the key. Answer The Equation-Classification Exercises (3–12) and Questions, Exercises, and Problems 2–14. Check your answers with those at the end of the chapter.

### Chapter 8

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CHAPTER 8 REVIEW Chemical Equations and Reactions Teacher Notes and Answers Chapter 8 SECTION 1 SHORT ANSWER 1. a. d b. a c. b d. f e. e f. c 2. 8,4,9 3. a. 12 atoms ... Modern Chemistry 2 Chemical Equations and Reactions 6. Gold has a low reactivity and therefore does

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