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Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. \_\_\_\_14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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