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Chemistry 101: Chapter 1

Best Answer: here's the idea... The empirical formula is the lowest whole number MOLE ratio of the different elements in the compound. The %'s are very often given in MASS %'s.. as they are in this problem.. so to solve this, we will need to (1) convert mass %'s to grams to moles (2) simplify the mole ...

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Express your answer using two significant figures. ANSWER: Chapter 10 Question 10 - Bimodal Description: (a) A sample of a gas originally at 25 °C and 1.00 atm pressure in a 2.5 L container is subject to a pressure of 0.85 atm and a temperature of 15 °C. The final volume of the gas is ____ L. Part A

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