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The MO theory also works well for larger molecules such as  $N_2O$ ,  $CO_2$ , and  $BF_3$  and for ions such as  $NO_3^-$  and  $CO_3^{2-}$ , in which the bonding MOs are delocalized, involving three or more atoms. It is especially useful for aromatic molecules such as benzene.

### **Molecular Orbital Theory - Chemistry Encyclopedia ...**

In chemistry, a reaction mechanism is the step by step sequence of elementary reactions by which overall chemical change occurs.. A chemical mechanism is a theoretical conjecture that tries to describe in detail what takes place at each stage of an overall chemical reaction.

### **Reaction mechanism - Wikipedia**

The arrow in the equation means "changes to" or "forms." The reaction of gaseous nitrogen with hydrogen to produce ammonia,  $\text{NH}_3$ , is represented by the chemical equation Although there are thousands of chemical reactions, a significant number of them, especially those that are not organic reactions, can be classified according to four general patterns: combination, decomposition, displacement ...

#### **Chemical Equations - Chemistry Encyclopedia - reaction ...**

Many Different Molecules: Maintaining the Body's Chemistry. In order for blood to perform its essential functions of bringing nutrients and oxygen to the cells of the body, and carrying waste materials away from those cells, the chemical composition of the blood must be carefully controlled.

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Radioactive adipic acid was fed to fasted experimental rats and the metabolic products identified in the urine were urea, glutamic acid, lactic acid, beta-ketoadipic acid, and citric acid. The presence of beta-ketoadipic acid provided some evidence that adipic acid is metabolized by beta-oxidation in much the same fashion as fatty acids. Further evidence was provided by the appearance of ...

#### **Adipic acid | C<sub>6</sub>H<sub>10</sub>O<sub>4</sub> - PubChem**

Kaliumjodide is een wit kristallijn anorganisch zout met als brutoformule KI. Het wordt gebruikt in de fotografie en in de radiotherapie. Als bron van jodide-ionen wordt het vaker gebruikt dan natriumjodide, omdat het veel minder hygroscopisch is. De stof is opgenomen in de lijst van essentiële geneesmiddelen van de WHO

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