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This question was attempted by majority of the candidates and the performance was fair. In (a)(i)(ii) and (iii) most candidates correctly define standard electrode potential, state factors that affect standard electrode potential and gave uses of the values of standard potential.

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Mark Scheme (Results) Summer 2012 International GCSE Chemistry (4CH0) Paper 1C Science Double Award (4SC0) Paper 1C ... M2 salt /solution / water may spit out (when evaporating the salty water) / may get in your eye IGNORE references to hazards eg toxic / irritant OR

#### Mark Scheme (Results) Summer 2012 - Edexcel

(a) (i) What is a structural isomer? (ii) Write all the structural isomeric alkanols with the molecular formula C4H10O. (iii) Which of the isomers from 4(a)(ii) above does not react easily on heating with acidified K2Cr2O7?

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#### Mark Scheme (Results) Summer 2012

2012 HIGHER SCHOOL CERTIFICATE EXAMINATION Chemistry General Instructions • Reading time – 5 minutes • Working time – 3 hours • Write using black or blue pen Black pen is preferred • Draw diagrams using pencil • Board-approved calculators may be used • A data sheet and a Periodic Table are provided at the back of this paper

#### 2012 HSC Examination - Chemistry - Board of Studies

MAY/JUNE 2012 CHEMISTRY ... GENERAL COMMENTS Chemistry is a two-unit subject with each unit consisting of three modules. Both units are examined by three papers. Papers 01 and 02 are external examinations, while Paper 031 the School-Based Assessment (SBA), is examined internally by teachers and moderated by CXC. Private candidates write Paper 032

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