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The two isotopes of chlorine are and .Which isotope is the most abundant? Hint D.1 How to approach the problem Hint not displayed Hint D.2 Find the atomic mass Hint not displayed Hint D.3 Mass of each isotope Hint not displayed ANSWER: Correct Since the atomic mass of chlorine, 35.45 , is closer to 35 than it is to 37, must be the most abundant isotope. does in fact account for 75.53% of the ...

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Express your answer using two significant figures. ANSWER: Chapter 10 Question 10 - Bimodal Description: (a) A sample of a gas originally at 25 °C and 1.00 atm pressure in a 2.5 L container is subject to a pressure of 0.85 atm and a temperature of 15 °C. The final volume of the gas is \_\_\_\_ L. Part A

**My Courses Chemistry: The Central Science, 12e Brown/LeMay ...**

2. Associate Professor Michael Lufaso has been teaching General . Chemistry I since he started at the school in 2006. He has taught General Chemistry II since 2009. General Chemistry II is a three-credit course, and the second course in a two-semester sequence taken primarily by biology . and chemistry majors. The course covers the chemistry of

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