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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Modern Chemistry 84 Chapter Test Chapter: States of Matter In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

Assessment Chapter Test A - Kettering City School District

Modern Chemistry Chapter 10 States of Matter Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

Modern Chemistry Chapter 10 States of Matter

Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMs r i-vi

Chemistry Chapter Resources. Chapter One Matter & Change. Objectives: Chapter 1 Honors: ... CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1 Mixed Questions. ... CH 10 Liquids & Solid Test . CH 10 HC-Thermo chemistry 1A . CH 10 HC Thermo chemistry .

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CHEMISTRY NOTES - Chapter 10 States of Matter Goals : To gain an understanding of : 1.. The kinetic theory of matter. 2. Avogadro's hypothesis. 3. The nature of liquids, solids and gases. 4. Phase changes. NOTES: The kinetic theory of matter states that particles which make up all types of matter are in constant motion.

CHEMISTRY NOTES - Chapter 10 States of Matter

STANDARDIZED TEST PREP Circle the letter of the best answer. 1. What conditions are required so that the average pressure of the atmosphere supports a 760 mm column of mercury? ... Modern Chemistry • CHAPTER 10 HOMEWORK 10-8 (pp. 322-325) VOCABULARY Complete each sentence. 1.

HOMEWORK 10-1 - Los Banos Unified School District

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