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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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1.

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Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued _____ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

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CP Chemistry I, Chapter#6 "Chemical Bonding" Test Questions Bank Practice Quiz. This is a practice quiz of all the potential multiple choice questions that a student taking Mr. New's Chemistry I course should expect on the chapter exam for chapter #6 from textbook "Modern Chemistry" (Holt, 2006).

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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued _____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. _____14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

Assessment Chapter Test A - Kettering City School District

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. In his periodic table, Mendeleev did not list all of the elements in order

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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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