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| Microbiology \ Microbiology-Chapter 4. Mic | robiology-Chapter 4. Which of the following is NOT found |
|--|--|
| in all bacterial cellsa. cell membrane b. nu | cleoid c. ribosomes d. actin cytoskeleton: d. actin |
| cytoskeleton: Pili are tubular shafts in | _bacteria that serve as a means of |

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The _____ is outside the cell wall, usually sticky, and the slime layer is unorganized and loose. Its functions are to attach to surfaces and prevent phagocytosis.

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Microbiology Study Guide - Los Angeles Mission College

Microbiology Chapter 4 Study Guide 1. Prokaryotes and Eukaryotes 2. Contain nucleic acids, protein, lipids, and carbohydrates 2. Use the same kind of chemical reactions to metabolize food, build proteins, and store energy 1. Prokaryotes 2. DNA not enclosed within membrane 2. Lack of membrane bound organelles 2. Divide by binary fission 1. Eukaryotes 2. DNA found in nucleus 2.

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Microbiology Ch. 4 Created by klh0032 36 terms Biofilm Polysaccharide-encased community of microorganisms Chemically defined medium Bacteriaological growth medium composed of precise mixtures of known pure chemicals; generally used for specific experiments when nutrients must be precisely controlled Complex medium Bacteriological medium that contains an ingredient that can be changed by ...

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Viruses are typically described as obligate intracellular parasites, acellular infectious agents that require the presence of a host cell in order to multiply. Viruses that have been found to infect all

types of cells – humans, animals, plants, bacteria, yeast, archaea, protozoa...some scientists even claim they have found a virus that infects other viruses!

Introduction to Viruses | Microbiology - Open Oregon State

1. Which of the following structures contains genes for enzymes and antibiotic resistance? Plasmid Pilus Capsule Plasma Membrane; 2. Which of the following is the most important structure related to microbial attachment to cells?

Microbiology Practice Questions - Test Prep

The pH scale ranges from 0 to 14, with 0 representing an extremely acidic solution (1.0 M $_{\rm H+}$) and 14 representing an extremely alkaline solution (1.0 x 10-14 M $_{\rm H+}$). Each pH units represents a tenfold change in hydrogen ion concentration, meaning a solution with a pH of 3 is 10x more acidic than a solution with a pH of 4.

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