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Modern Chemistry Chapter 7. compounds composed of two different elements; (compounds of 2 elements between metal (positive, s-block and p-block) and nonmetal (negative); metal is always listed first; change nonmetal to -ide ending)

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Holt McDougal Modern Chemistry Chapter 7: Chemical Formulas and Chemical Compounds Chapter Exam. Exam Instructions: Choose your answers to the questions and click 'Next' to see the next set of questions. You can skip questions if you would like and come back to them later with the yellow "Go To First Skipped Question" button.

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-1 ... Answer the following questions in the space provided. 1. ... MODERN CHEMISTRY SECTION 7-1 REVIEW. 53.

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Modern Chemistry 91 Chapter Test Name Class Date Chapter Test B, continued 27. Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided. The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

### **Assessment Chapter Test B - clarkchargers.org**

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1.

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Modern Chemistry 85 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 6. Which factor is the most important in determining the average kinetic energy of gas particles? a. pressure b. temperature c. volume of the container d. mass of the container \_\_\_\_ 7. Which gas is most likely to deviate from ideal gas behavior? a. Ne b. CO<sub>2</sub> c. NH<sub>3</sub> d. H<sub>2</sub> \_\_\_\_ 8.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

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