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n in equation 5? 8. Write the nuclear symbol for the missing term in equation 5. 9. Write the nuclear symbol for the missing term in equation 6. Name Date Class Balancing Nuclear Equations Balancing Nuclear Equations MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 25, Section 25.2 41

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Chapter 25 Nuclear Chemistry 25.1 Nuclear Radiation 25.2 Nuclear Transformations 25.3 Fission and Fusion 25.4 Radiation in Your Life. ... • In a balanced nuclear equation, the sum of the mass numbers (superscripts) on the right must equal the sum on the left. • The same is true for the atomic numbers

Chapter 25

804 Chapter 25 Nuclear Chemistry CHAPTER 25 What You'll Learn You will trace the history of nuclear chemistry from dis-covery to application. You will identify types of radioactive decay and solve decay rate problems. You will describe the reac-tions involved in nuclear fission and fusion. You will learn about appli-cations of nuclear reactions

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Answer Keys - HONORS CHEMISTRY

The subscripts and superscripts are necessary for balancing nuclear equations, but are usually

optional in other circumstances. For example, an alpha particle is a helium nucleus (He) with a charge of +2 and a mass number of 4, so it is symbolized. This works because, in general, the ion charge is not important in the balancing of nuclear ...

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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

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