

## *Modern Chemistry Chapter Test B Answers*

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### Modern Chemistry Chapter Test B

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

### Assessment Chapter Test B - clarkchargers.org

Modern Chemistry 34 Chapter Test Name Class Date Chapter Test B, continued PART IV Write the answers to the following questions in the space provided. 27. What two early twentieth-century observations involving the interaction of light and matter could not be explained solely by the wave theory of light? 28.

### Assessment Chapter Test B - clarkchargers.org

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

### Assessment Chapter Test B - Wag & Paws

Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

### Assessment Chapter Test B - Baumapedia

Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_ 13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. \_\_\_\_ 14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

### Assessment Chapter Test A - Kettering City School District

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### 10 States of Matter - Ms. Agostine's Chemistry Page

Modern Chemistry: Chapter Tests with Answer Key, 2006, Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, Harcourt School Publishers, 2006 ... Michigan Holt Chemistry and Modern Chemistry Test Preparation Workbook: Help for the MME , Holt, Holt Rinehart & Winston, Jun 1, 2006, Juvenile Nonfiction, 126 pages. .

### Modern Chemistry: Chapter Tests with Answer Key, 2006 ...

Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued \_\_\_\_ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c. wave patterns repeat at regular intervals.

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