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if two or more different compounds are composed of the same two elements, then the ratio of the masses of the second element combined with a certain mass of the first element is always a ratio of small whole numbers; textbook: if two or more different compounds are composed of the same two elements with a certain mass of the first element is always a ratio of small whole numbers

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Modern Chemistry Chapter 3 Atoms: The Building Blocks of Matter law of conservation of mass-mass is neither created nor destroyed during ordinary chemical reactions or physical changes e.g. $20\text{ g A} + 20\text{ g B} \rightarrow 40\text{ g AB}$

Modern Chemistry Chapter 3 Atoms: The Building Blocks of ...

see figure 3-2 page 67. Dalton's theory and Law of Definite Proportion --see figure 3-3 page 67 . C: Modern Atomic Theory. Some of Dalton's postulates have been proven to be incorrect. His third postulate indicated that atoms cannot be subdivided. His second postulate indicates atoms of the same element have identical mass. Isotopes

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Unformatted text preview: Modern Chemistry Chapter 3 Atoms: The Building Blocks of Matter law of conservation of massmass is neither created nor destroyed during ordinary chemical reactions or physical changes e.g. $20\text{ g A} + 20\text{ g B} \rightarrow 40\text{ g AB}$ law of definite proportions- a chemical compound contains the same elements in exactly the same proportions by mass regardless of the size of the sample or ...

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