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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms In the space provided, write the letter of the term that best completes each ... Chapter 4 PRE TEST, continued _____ 8. What is the frequency of a photon whose energy is 3.4 10 19 J? ... Modern Chemistry 4 Electron Arrangement Chapter 4 PRE-TEST, continued

CHAPTER 4 PRE-TEST Arrangement of Electrons in Atoms

Taken from the book Modern Chemistry by Holt, Rinehart, and Winston on Chapters 4 and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things.

Modern Chemistry (Holt, Rinehart, and Winston): Chapter 4 ...

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. In his periodic table, Mendeleev did not list all of the elements in order

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Modern Chemistry 27 Chapter Test Name Class Date Chapter Test A, continued _____ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund's rule. d. Both (a) and (c) _____ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

4. What two factors determine whether or not a molecule is polar? electronegativity difference and molecular geometry or unshared electron pairs 5. Arrange the following types of attractions in order of increasing strength, with 1 being the weakest and 4 the strongest. 3 hydrogen bonding 4 ionic 2 dipole-dipole 1 London dispersion 6.

6 Chemical Bonding - Effingham County Schools / Overview

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1.

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