

Modern Chemistry Chapter 7 Mixed Review Answers

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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14 Ions in Aqueous Solutions and Colligative Properties

Modern Chemistry 2 Stoichiometry CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C ... Modern Chemistry 3 Stoichiometry MIXED REVIEW continued c. If 0.1 mol of N₂ combine with H₂, what must be true about the quantity of H

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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N_2 are mixed with 12.0 mol of H_2 according to the following equation: $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$...

mc06se cFMsr i-vi

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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MIXED REVIEW . continued. PROBLEMS Write the answer on the line to the left. 5. a. ... Modern Chemistry 62 Chemical Bonding. ... Modern Chemistry 61 Chemical Bonding. Title: CHAPTER 7 REVIEW Author: user Last modified by: amuermann Created Date: 1/4/2010 3:09:00 PM Company: arme

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Molecular Composition of Gases MIXED REVIEW ... MODERN CHEMISTRY CHAPTER 11 MIXED REVIEW 97 HRW material copyrighted under notice appearing earlier in this work. Name Date Class . MIXED REVIEW continued 6. 60.0 g/mol What is the molar mass of a gas that has a density of 2.68 g/L at STP? 7.

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electronegativity is greater than 1.7, the bond is considered ionic. 17. Describe the electron distribution in a polar-covalent bond and its effect on the partial charges of

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