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You would deduce which way you look at it gives you the smallest number on the carbon chain, from left to right it would be 4,4-DiChloro-3-MethylPentane. You then would look at it from the other way (right to left) and it would be 2,2-DiChloro-3-MethylPentane. As you can see the position of the functional groups on the 2nd Haloalkane is lower than the 1st Haloalkane so that would be it.

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