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Modern Chemistry Chapter 13 Ions in Aqueous Solutions and Colligative Properties Section 1-Compounds in Aqueous Solutions dissociation is the separation of ions that occurs when an ionic compound dissolves see sample problem A on page 436 do practice problem #1 on page 436 Solubitity precipitation reactions occur when combinations of ions in a solution have an extremely low solubility and a ...

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Modern Chemistry 113 Ions In Aqueous Solutions And Colligative Properties CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 2 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. ____ a. Predict the boiling point of a 0.200 m solution of glucose in water.

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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