

**EXPERIMENTAL ANALYSIS OF PERFORMANCE ,COMBUSTION AND
EMISSION CHARACTERISTICS OF SINGLE CYLINDER DIESEL
ENGINE USING BIODIESEL (DIESEL-TURPENTINE BLEND) WITH
HYDROGEN INDUCTION.**

A PROJECT REPORT

Submitted by

NIROSHINI K 2019502032

AKILAN S 2019502509

DHINESH BABU R 2019502517

HARIVIGNESH N S 2019502523

In partial fulfilment for the award of the degree

of

BACHELOR OF ENGINEERING

IN

AUTOMOBILE ENGINEERING



**DEPARTMENT OF AUTOMOBILE TECHNOLOGY MADRAS
INSTITUTE OF TECHNOLOGY ANNA UNIVERSITY**

December 2022

ANNA UNIVERSITY : CHENNAI 600 025

BONAFIDE CERTIFICATE

Certified that this project report “**EXPERIMENTAL ANALYSIS OF PERFORMANCE, COMBUSTION AND EMISSION CHARACTERISTICS OF SINGLE CYLINDER DIESEL ENGINE USING BIODIESEL (DIESEL-TURPENTINE BLEND) WITH HYDROGEN INDUCTION.**” is the bonafide work of

NIROSHINI K	2019502032
AKILAN S	2019502509
DHINESH BABU R	2019502517
HARI VIGNESH N S	2019502523

who carried out the project work under my supervision. Certified further that to the best of my knowledge, the work reported herein does not form part of any other thesis or laboratory report on the basis of which a degree or award was conferred on an earlier occasion on this or any other candidate

SIGNATURE

Dr.A.Sangeet Sahaya Jeyangel

SUPERVISOR

Teaching fellow

Department of Automobile Engineering
Madras Institute of Technology

SIGNATURE

Dr. K. Annamalai

HEAD OF THE DEPARTMENT

Professor

Department of Automobile Engineering
Madras Institute of Technology

ACKNOWLEDGEMENT

We are highly grateful to **Dr. K. Annamalai**, Head of the Department, Department of Automobile Engineering, Madras Institute of Technology, Chennai. We would like to record our sincere thanks to **Dr. J. Prakash**, Dean, Madras Institute of Technology, Chennai for having given consent to carry out the project work.

We are very thankful to our batch co-ordinator **Mr. G. Anandkumar**, Assistant Professor, Department of Automobile Engineering, Madras Institute of Technology, Chennai for his kind support for the project work. We extend our sincere thanks to the review panel faculty members **Mr. Arulselvan** and **Mr. S. Mariappa** for their valuable suggestions during all the reviews which took our project to greater heights.

We are very grateful to our guide **Dr. A. Sangeet Sahaya Jeyangel**, Teaching Fellow, Department of Automobile Engineering, Madras Institute of Technology, Chennai for his valuable time and guidance that made the project work a success. His extensive vision, creative thinking and ideas have been a source of inspiration and helped us gain immense amount of knowledge during the course of the project. We are very grateful to all staff members of the Department of Automobile Engineering and my friends and family for their valuable suggestions and help provided during this project.

ABSTRACT

Main aim of the project is to study the influence of hydrogen induction on the performance, combustion and emission characteristics of a compression ignition engine fueled with Turpentine oil biodiesel. The work is categorized into two phases. For the phase 1, The Turpentine oil biodiesel was mixed with diesel in various proportions. Mainly the B20, B30, B40 and B50 blends were taken into account. The test was conducted on a 5.2 kW four stroke, single cylinder diesel engine with compression ratio of 17.5:1. The results obtained are compared with neat diesel fuel at all the loads. The biodiesel blends show a significant reduction in the BTE when compared to the diesel fuel. The in cylinder pressure and the amount of heat released is also found to be lesser than diesel. The reduction in the CO and NO_x emissions was found. But still the smoke and HC emissions were found to be higher than the diesel. The BTE for B20, B30, B40, B50 blends were founded as 31.92%, 31.71%, 31.38%, 30.89%. The HC and CO emissions were 81, 83, 65, 68 ppm and for 0.15%, 0.27%, 0.21%, 0.26% for B20, B30, B40, B50 respectively. It was found that the B20 and B30 blends were promising to have higher BTE and lower emissions. so for the phase 2, B20 and B30 blends will be continued for further testing with hydrogen induction.

TABLE OF CONTENTS

CHAPTERS	TITLES	PAGE NO
	ABSRTACT	4
	LIST OF FIGURES	8
	LIST OF TABLES	9
1	INTRODUCTION	10
1.1	OVERVIEW	10
1.2	RELATED STUDY	10
1.2.1	FUELS	10
1.2.2	CONVENTIONAL FUELS	11
1.2.3	ADVANTAGES	11
1.2.4	DISADVANTAGE	11
1.2.5	NON CONVENTIONAL FUELS	12
1.2.6	ADVANTAGE	12
1.2.7	DISADVANTAGES	12
2	LITERATURE STUDY	
2.1	LITERARURE SUMMARY	13
3	ALTERNATE FUELS	
3.1	TURPENTINE OIL	19
3.2	COMPOSITON	20
3.3	OIL YIELD	20

3.4	PRODUCTION OF TURPENTINE OIL	20
4	METHODOLOGY	22
4.1	EXTRACTION OF TURPENTINE OIL	23
4.1.1	COLLECTION OF PINE SAPS	23
4.1.2	EXTRACTION OF RAW OIL	23
4.1.3	PREPARATION OF BIODIESEL BLEND	24
5	PROPERTIES	25
5.1	PROPERTIES OF TURPENTINE OIL BLENDS	25
5.1.1	KINEMATIC VISCOSITY	25
5.1.2	FLASH POINT	25
5.1.3	DENSITY	26
5.1.4	CALORIFIC VALUE	26
5.2	INSTRUMENTS USED TO MEASURE PROPERTIES	26
5.2.1	REDWOOD VISCOMETER	26
5.2.2	DIGITAL DENSITY METERS	27
5.2.3	OPEN CUP APPARATUS	28
6	SPECIFICATIONS	
6.1	ENGINE DESCRIPTION	31
6.2	FEATURES	31
6.3	ENGINE LAYOUT	32
6.4	UTILITIES REQUIRED	33
6.5	ENGINE SPECIFICATIONS	33
6.6	SOFTWARE	34
6.7	EXHAUST GAS ANALYSER	35

6.8	SMOKE METER	36
7	PERFORMANCE , COMBUSTION AND EMISSION CHARACTERISTICS	
7.1	PERFORMANCE PARAMETERS	37
7.1.1	BRAKE THERMAL EFFICIENCY	37
7.1.2	SPECIFIC FUEL CONSUMPTION	38
7.2	COMBUSTION PARAMETERS	
7.2.1	IN-CYLINDER PRESSURE	39
7.2.2	HEAT RELEASE RATE	40
7.3	EMISSION PARAMETERS	
7.3.1	HYDROCARBON EMISSION	41
7.3.2	CARBON MONOXIDE EMISSION	42
7.3.3	SMOKE OPACITY	43
7.3.4	OXIDES OF NITROGEN	44
8	CONCLUSION	45
9	REFERENCE	46

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE NO.
4.1	PINE SAPS	23
4.2	PINE TREE	23
4.3	BIODIESEL BLENDS	24
5.1	REDWOOD VISCOMETER	27
5.2	DIGITAL DENSITY METER	28
5.3	CLEVELAND OPEN CUP APPARATUS	29
5.4	PROPERTIES OF BLENDS	29
6.1	ENGINE LAYOUT	32
6.2	AVL GAS ANALYSER	35
6.3	AVL SMOKE METER	36
7.1	BTE vs BRAKE POWER	37
7.2	SFC vs LOAD	38
7.3	CRANK ANGLE vs PRESSURE	39
7.4	CRANK ANGLE vs HRR	40
7.5	LOAD vs HC	41
7.6	LOAD vs CO	42
7.7	LOAD vs SMOKE	43
7.8	LOAD vs NO _x	44
7.9	ZELDOVICH MECHANISM	44

LIST OF TABLES

TABLE NO.	TITLE	PAGE NO.
3.1	TURPENTINE OIL PRODUCTION IN INDIA	21
5.1	PROPERTIES OF BIODIESEL BLENDS	30
6.1	ENGINE SPECIFICATIONS	33

CHAPTER 1

INTRODUCTION

1.1 OVERVIEW

Transportation has been an important aspect of human life from the old ages. The modes of transport have been evolving and upgrading from horses and carts to hyperloops. Only after the invention of engines, the transportation became easier. The 2 and 4 stroke engines have made our trips effort less. Two wheelers and four wheelers have become more popular and user friendly among us. The usage of the vehicles is increasing day by day. Now, the fuels used for powering these engines must be taken in account. There are numerous fuels used for this purpose, which mostly fall under two categories, renewable and non-renewable resources. Both has its own advantages and disadvantages.

1.2 RELATED STUDY

1.2.1 FUELS

The evolution of fuels used in today's automotive industry is very well known to everyone. In the pre-automotive ages, people used firewood as fuel. No automotive was invented till then of course. With this, they started using the coal present on the ground and when they used up all of it, miners started digging the ground for more coal. This was the revolutionary point where Thomas Newcomen invented a coal-burning steam engine to pump out water and allow miners to dig deeper. Further, James Watt made some improvements in the engine and made it fit for other uses as well. Most of us know that trains used to run on steam-powered engines and therefore, this is one of the largest used automotive power sources from the past. The next important fuel is the petrol or gasoline. It became a commonly used automotive fuel. In 1876, Nicolaus Otto invented the petrol engine and its working cycle is named after him only. Another commonly used petroleum product as an automotive fuel is diesel. Developed in the 1890s, the diesel engines are still used to power many machines. It is the most commonly used automotive fuel used to power heavy vehicles like trucks and buses. These are the majorly used conventional fuels in Automobiles.

1.2.2 CONVENTIONAL FUELS

When a source of Energy cannot be reused after using it once, we call them a conventional Source of Energy. The conventional Sources of Energy are fixed and harmful to the environment. It can be further categorized into two divisions, such as

1. Commercial sources of Energy (such as coal, petroleum, nuclear energy, natural gas, etc.)
2. Non-commercial sources of Energy (such as firewood, straw, dried dung, etc.)

1.2.3 ADVANTAGES

1. The efficiency of the energy sources is high.
2. The production expenses are low.
3. The raw materials of conventional Sources of Energy are easy to transport. Raw materials such as coal, petroleum, natural gas can be transported easily through trains or ships from one place to another.
4. Generally, it doesn't need any specific place for installation. The government can easily set up a conventional plant according to their requirements. Such as if the government wants to install a thermal plant in Uttarakhand or in Jammu, they can easily install it.
5. Though it can generate energy instance, so there is no need to wait and it produces much energy as the requirements.

1.2.4 DISADVANTAGES

1. They are the main reason for the pollution. Because it is released carbon monoxide from polluters into the atmosphere. According to The International Energy Agency, only in 2018, India emitted 2,299 million tonnes of carbon monoxide. This report also said that India's per capita emissions were about 40% of the global average and contributed 7% to the global carbon dioxide burden.

2. Generating radioactive waste. We all know about the Fukushima Daiichi nuclear disaster which happened in 2011. Recently Japanese government has decided to release

radioactively contaminated water into the ocean. But the environmentalists warn that it should be harmful to ocean life.

3. High start-up cost. According to the government estimate, the cost may be about 50 to 70 crores INR for setting up a 10 MW thermal power unit. Whereas to set up a nuclear power plant, it is required ₹60,000 crores.

1.2.5 NON CONVENTIONAL FUELS:

These types of resources get renewed or replenished fast for that reason, it is also known as renewable energy or clean energy.

Example: solar energy, wind energy, hydro energy, tidal energy, etc.

1.2.6 ADVANTAGES:

1. Non-conventional sources of energy are usually inexhaustible. They do not pollute the environment.
2. Nuclear power is emitted in large amounts.
3. Most non-conventional sources of energy cost less.
4. These forms of energy are safe to use and clean.

1.2.7 DISADVANTAGES:

1. Windmills are costly to set up. So, using them to harness wind energy is costly, even though the electricity generated from it is cheap.
2. Setting up windmills disturbs radio and TV broadcasts.
3. Harnessing tidal energy destroys the natural habitats of wildlife.
4. Moreover, tidal energy is difficult to harness.
5. Obtaining nuclear energy from radioactive material generates radioactive

CHAPTER 2

LITERATURE SURVEY

2.1 LITERATURE SUMMARY

Analysis of emission and performance characteristics of compression

ignition engine using Mahua oil-based biodiesel

M. Kannan, G. Elavarasan, D. Karthikeyan, Vikas Goyat

- In this paper, Mahua oil-based bio-Diesel (MHD) has been extracted from the plants of Madhuca longifolia and tested its emission as well as performance characteristics using Compression Ignition (CI) engine test rig. Raw Mahua oil generally has fatty acid, it may cause maintenance problems in CI engine.

- Therefore, the transesterification process has been used to remove unwanted fatty acid.

- The transesterified Mahua oil has been added with diesel in the ratio of 0%, 5%, 10% and 15% to prepare MHD. The nanoparticles of Cerium Oxide (Ce_2O_3) also have been added in prepared MHD (in 100 ppm proportion) as an additive to improve chemical reaction between the fuel and air inside the CI combustion chamber. The Constant speed diesel engine performance and emission characteristics of MHD have been analyzed for the different brake powers i.e. 1.15, 1.71, 2.29 and 2.75 kW.

- The analysis shows that the Carbon monoxide (CO), Hydro Carbon (HC) and smoke emissions decreases with increase in the blend however, the NOX emissions increases with increase in blend.

- The performance characteristic of the mahua biodiesel blend is very impressive compare to diesel. Even though, Mahua's thermal efficiency is low and specific fuel consumption is good compare to diesel.

Combustion investigation of waste cooking oil (WCO) with varying compression ratio in a single cylinder CI engine

Tomesh Kumar Sahu, Sumit Sarkar, Pravesh Chandra Shukla

- Biodiesel is a popular alternative fuel of mineral diesel which is normally produced by using straight vegetable oils (SVO). Since the conversion of edible oils into biodiesel may affect its supply/availability as a food item, it is desirable to use non-edible oils for biodiesel production for a long time.
- Waste cooking oil (WCO) is one such non-edible oil, however superior in quality for biodiesel production through transesterification. The current study is mainly focused on the utilization of WCO as an alternative for mineral diesel in a typical agricultural compression ignition four stroke water-cooled engine.
- Effect of higher compression ratio (CR20) on various combustion and performance parameters such as in-cylinder pressure ($P-\theta$), heat release rate (HRR), peak pressure (P_{max}), maximum pressure rise rate $(dP/d\theta)_{max}$, combustion duration, combustion phasing etc. were investigated for WCO biodiesel and compared with mineral diesel.
- These experiments were performed for five different loading conditions (No load to full load) at 1500 rpm. Increase in compression ratio (CR) from 18 to 20 resulted in lower ignition delay for higher engine load condition which reduced by 6 CAD.

- Full load engine operation resulted in 32% brake thermal efficiency (BTE) for WCO biodiesel which was 18% higher than diesel.

A REVIEW ON NOVEL BIO-FUEL FROM TURPENTINE OIL

Mehmet Hakkı Alma, Tufan Salan

- In this study, a review of the usage of various turpentine oils, the volatile fraction of the resinous exudate or extract from the genus *Pinus* or obtained as a by-product (so called sulfate turpentine oil) of the Kraft pulp process, as fuel alone or fuel additive for gasoline and diesel was studied.
- Turpentine oil, consisting of cyclic monoterpenes ($C_{10}H_{16}$) have been used as thinner, antiseptic, drug, pesticide, insecticide, biofuels or fuels additives as well.
- The studies performed on the effects of turpentine oil blended with gasoline, diesel and gasoline-like distilled engine oil on the engines performance and exhaust emission demonstrated that the turpentine oil had a favourable influence on the engine performance and led to decrease CO content, however, slightly increased NO_x content.
- On the other hand, the results for turpentine oil fuel blended with diesel indicated that all the other performances and emission parameters were better than those of diesel fuel alone

- Turpentine oil as a fuel or fuel additive has a great potential reducing usage of the fossil fuels and led to decrease particulate residue in spark ignition engines. It can be produced from a wide range of natural agricultural and forestry sources as well as municipal and industrial wastes

Potential utilization of turpentine oil as an alternative fuel

Anil Singh Yadav , Padam Singh , Rahul Sahu , P. Thangamuthu , Renuka Shyam Narain , Y. Anupam Rao, Azmeera Balu , Bishnu Prasad Panda , Abhishek Sharma

- After the oil price rise in 1973, and increased global pollution; individual, social and national life and economy were largely affected by the increasing cost of energy. Rapidly increasing prices of petroleum
- Its crisis, and uncertainties regarding the availability of petroleum have serious effect on the development of economy all over the world.
- The fuel trends for transportation vehicles have been affected greatly by both availability of fuel and environmental factors. To address these energy and environmental issues, it is essential to find some potential alternative to fossil fuel.
- Biofuels derived from biological materials; basically, renewable sources have emerged as potential alternative to fossil fuel.

- Turpentine oil is one of the vital biofuels and biodegradable fuel source which has the potential to replace partially the diesel fuel.
- Commencing from the biofuel generations, the current paper discusses about the production methods of turpentine oil, its important physicochemical properties and application as diesel engine fuel.
- Turpentine oil holds a good future as an alternative fuel in automation sector but lot more research is required to be done for enhancing its properties.

Experimental study on the effect of cetane improver with turpentine oil on CI engine characteristics

A.K. Jeevanantham , D. Madhusudan Reddy , Neel Goyal , Devansh Bansal, Gopalakrishnan Kumar , Aman Kumar , K. Nanthagopal , B. Ashok

- The Turpentine oil has getting wider attention in recent times due to many significant benefits. In this present study, 20% Turpentine oil has been blended with 80% diesel fuel by volume.
- A novel cetane improver called SC5D has been doped with 20% biofuel at 2.5% and 5% concentrations. All the properties of all the blends are evaluated and it has been identified that the 5% of cetane improver has increases the fuel density.

- All the fuel samples have been tested in a single cylinder CRDI diesel engine under different pilot injection rate of 15% and 30% at 600 bar injection pressure. For the higher NO_x emission condition, the 10% EGR has also been applied.
- The experimental study revealed that the addition of cetane improver with biofuel blends has shown comparable performance behaviours at all concentrations.
- The unburned hydrocarbon and smoke emissions are remarkably lower for biofuel blends at both the injection rate with cetane improver addition. Significant reductions has been noticed by 10% EGR addition for PI15% rate at 2.5% and 5% concentrations without much defect in other emissions.
- All the combustion behaviours have shown comparable behaviours with cetane improver addition and 10% EGR implementation with biofuel blends.

CHAPTER 3

ALTERNATE FUELS

After analysing the pros and cons of using these fuels, many scientists have opted for the usage of various alternative sources of fuel for automobile. Alternative fuels are derived from sources other than petroleum. Most are produced domestically, reducing our dependence on imported oil, and some are derived from renewable sources. Often, they produce less pollution than gasoline or diesel. Alternative fuels include gaseous fuels such as hydrogen, natural gas, and propane; alcohols such as ethanol, methanol, and butanol; vegetable and waste-derived oils; and electricity. These fuels may be used in a dedicated system that burns a single fuel, or in a mixed system with other fuels including traditional gasoline or diesel, such as in hybrid-electric or flexible fuel vehicles.

3.1 TURPENTINE OIL:

Turpentine oil is used as medicine. In medicine, it is used in preparation that acts as a local stimulant, anesthetic, and antiseptic. It is used externally in ointments for neuralgia and myositis.

Turpentine oil works as a feedstock chemical in the manufacture of flavaring, pinenes, pine oil, polymer additives and fragrances. It is also used as a commercial source of fuel and as an alternative of fossil fuels in a broad range of applications such as furnaces, rocket fuels and industrial boilers. Turpentine oil has a low heating coefficient ranging between 16,000 & 18,000 Btu/lb as compared to other fossil fuels such as gasoline, diesel, butane and propane

3.2 COMPOSITON :

It contains

- 65% of rosin (solid residue)
- 18% of oil

3.3 OIL YIELD :

Pine trees are hardy trees that tolerate at low temperatures, high altitudes, and acidic, sandy soil. Typically, pine trees reach full maturity in 25 to 30 years with some species of pine tree growing to be 150 feet tall. All pine trees begin as small seeds, growing to saplings and then to a mature trees. The height of a mature tree can vary greatly depending on the species. For example ,a dwarf mugo pine might only grow to be 4 feet tall ,But under the right conditions, a white pine can grow to be over 150 feet tall . In pine trees, this age is considered “grown up” because this is when growth dramatically slows and its wood is harvested.

3.4 PRODUCTION OF TURPENTINE OIL :

Pines are naturally found almost exclusively in the Northern Hemisphere. They are found in much of North America, China, South-East Asia, Russia and Europe .Pine trees are the dominant plants which grow in many cold temperature and boreal forests. They are particularly successful in cold areas where broad-leaved plants are unable to survive such as the boreal forest and at high altitude. There is estimated to be around 250 species of pines throughout the world and It includes a total of 11 genera which include the spruces, cedars, firs, pines and more. According to the height, space between the tress it can be planted 620 or 435 trees per acer. In India pine trees are mostly cultivated in cold and hilly regions like Himachal Pradesh, Ooty , kodaikanal etc.

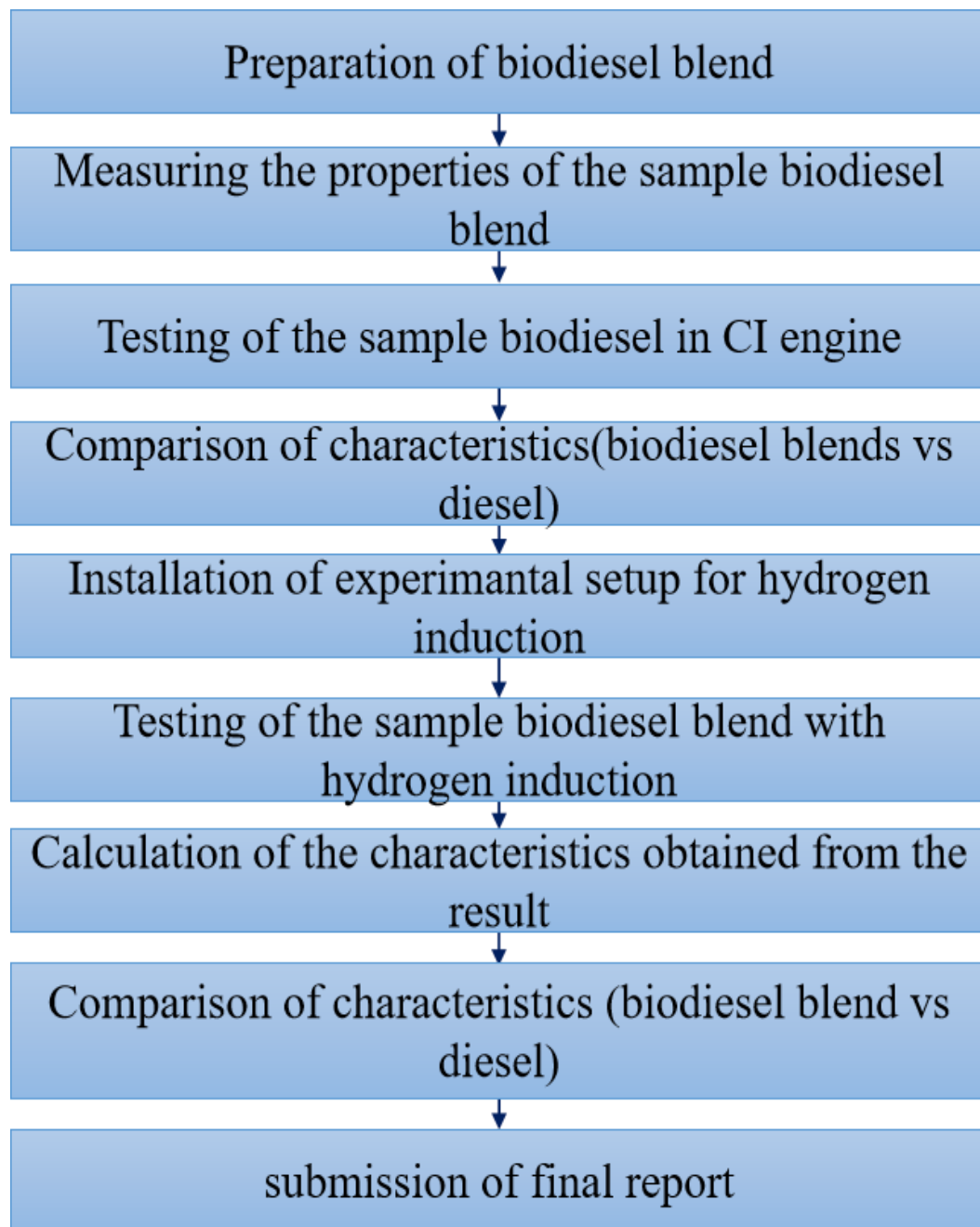
Pine species are also used widely for shelter and theinterlinked purposes of revegetation, soil stabilization, and soil conservation, as befits their tolerance of exposure and degraded soils in certain species. Most of the widely planted species of pines are used in some degree for one or more of these purposes, often with timber production as a major bonus. Ornamental and festive use is common, with *P. sylvestris* and *P. virginiana* being very widely grown for Christmas trees.

Table 3.1 TURPENTINE OIL PRODUCTION IN INDIA

Type of forest ^[21]	Area (in square kilometres)	Percentage of total forest
Tropical Wet Evergreen Forests	51,249	8.0
Tropical Semi Evergreen Forests	26,424	4.1
Tropical Moist Deciduous Forests	2,36,794	37.0
Littoral & Swamp Forests	4,046	0.6
Tropical Dry Deciduous Forests	1,86,620	28.6
Tropical Thorn Forests	16,491	2.6
Tropical Dry Evergreen Forests	1,404	0.2
Subtropical Broadleaved Hill Forests	2,781	0.4
Subtropical Pine Forests	42,377	6.6
Subtropical Dry evergreen Forests	12,538	2.5
Montane Wet Temperate Forests	23,365	3.6
Himalayan Moist Temperate Forests	12,012	3.4
Himalayan Dry Temperate Forests	312	0.0
Sub Alpine Forests	18,628	2.9
Total (Forest Cover + Scrub)	7,54,252	98.26
Grass land in different forest type groups (without forest cover)	13,329	1.74
Grand Total	7,67,581	100.00

CHAPTER 4

METHODOLOGY



EXTRACTION OF TURPENTINE OIL

4.1 STEP 1 – COLLECTION OF PINE SAPS

Turpentine were collected from the pine tree . pine trees are mostly seen in the hill stations . pine saps were collected by Drilling a hole into pine tree using a bit size recommended for your spile. Once your hole is started, insert the spile at a downward angle and gently tap the tapered end in with a hammer. If the sap is running, a drop will appear at the tip of the spile. Hang a covered collection bucket from the spile. turpentine is obtained by the steam distillation of dead, shredded bits of pine wood, while gum turpentine results from the distillation of the exudate of the living pine tree obtained by tapping



FIG 4.1 PINE SAPS

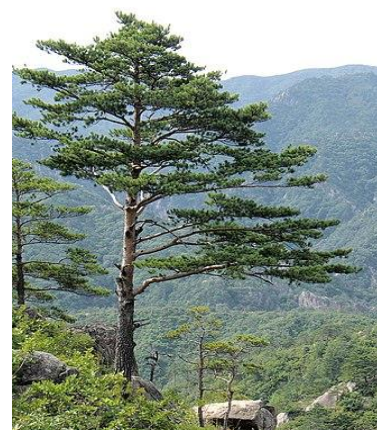


FIG 4.2 PINETREE

4.2 STEP 2 - EXTRACTION OF RAW OIL:

The collected oil was initially filtered and pre-treated for removal of suspended as well as unwanted/foreign particles. The collected saps are crushed using oil crusher to obtain raw oil.

4.3 STEP 3 - PREPARATION OF BIODIESEL BLENDS:

Biodiesel was blended in the ratio of 20%(D80T20),30%(D70T30), 40%(D60T40) , 50%(D50T50) with diesel and the engine test was carried out .



Fig 4.3 BIODIESEL BLENDS

CHAPTER 5

PROPERTIES

5.1 : PROPERTIES OF TURPENTINE BIO DIESEL:

5.1.1 : KINEMATIC VISCOSITY:

The kinematic viscosity of a fluid is the ratio of the viscosity of the fluid to the fluid's density. Mathematically, it is expressed as:

$$\nu = \mu / \rho$$

where ν is the kinematic viscosity, μ is the dynamic viscosity, and ρ is the density.

The kinematic viscosity is an important fuel property which affects the atomization of a fuel upon injection into the combustion chamber. The values of diesel and turpentine oil bio diesel are very close, when compared to other bio diesels. Hence, as the viscosity is less, the fuel is atomized properly, resulting in a complete combustion and good engine performance.

5.1.2 FLASH POINT:

The flash point is the lowest temperature at which a volatile substance evaporates to form an ignitable mixture with air in the presence of an igneous source and continues burning after the trigger source is removed. Flash point is a significant property not for the operability of a diesel fuel, but for its storage and handling. Diesel fuels are classified as nonvolatile fuels, and their storage does not need specific precautions. Flash point is an excellent indication of diesel fuel contamination with more volatile products.

5.1.3 DENSITY:

The density of material shows the denseness of that material in a specific given area. A material's density is defined as its mass per unit volume. Density is essentially a measurement of how tightly matter is packed together. It is a unique physical property for a particular object. The principle of density was discovered by the Greek scientist Archimedes. It is easy to calculate density if you know the formula and understand the related units. The symbol ρ represents density or it can also be represented by the letter D.

5.1.4 CALORIFIC VALUE:

Calorific value is defined as the amount of heat energy released during complete combustion of a unit mass of a fuel. It is expressed in kJ/kg.

5.2 INSTRUMENTS USED TO MEASURE PROPERTIES

5.2.1 REDWOOD VISCOMETER

The redwood viscometer consists of a vertical cylindrical oil cup with an orifice in the centre of its base. The orifice can be closed by a ball. A hook pointing upward serves as a guide mark for filling the oil. The cylindrical cup is surrounded by the water bath. The water bath maintains the temperature of the oil to be tested at a constant temperature. The oil is heated by heating the water bath by means of an immersed electric heater in the water bath. The provision is made for stirring the water, to maintain the uniform temperature in the water bath and to place the thermometer to record the temperature of oil and water bath. The cylinder is 47.625mm in diameter and 88.90mm deep. The orifice is 1.70mm in diameter and 12mm in length. This viscometer is used to determine the kinematic viscosity of the oil. From the kinematic viscosity the dynamic viscosity is determined.

Redwood viscometers multiple apparatus It consists; Two Red Wood No. 1 cups and Two Red Wood No. 2 cups are installed in the bath. The bath is made from S.S. Sheet. The temperature is Controlled by Electronic Digital Temperature Controller. The Stirring is done by FHP electric motor in the heating compartment. The unit has 4 sets of Oil Cup covers, thermometer Clips and ball valve. Suitable to operate on 220 volts AC Circuits.



Fig 5.1 REDWOOD VISCOMETER

5.2.2 DIGITAL DENSITY METERS

Density meters are instruments that measure the density of a sample liquid or gas. Digital density meters are used in the pharmaceutical, petroleum, chemical, and food and beverage industries for quality control and in research and development. Density meters work by measuring the oscillation of a glass tube that contains the sample. There are digital density meters that can measure specific gravity and refraction index as well as the density of a sample. There are density meters available that are portable or bench top models, that can measure liquid, gas, or both, and have a built in thermostat or need a water bath to maintain temperature. Some other variations include sample volume needed, speed of testing, and the ability to take and store images.

The density of a material is defined as its mass per volume. The most common units for density are the SI unit kilogram per cubic meter (kg/m^3) and the cgs unit gram per cubic centimeter (g/cm^3).

Formula to calculate density = mass/volume

$$\rho = m/v \text{ (kg/m}^3\text{)}$$



Fig 5.2 DIGITAL DENSITY METER

5.2.3 OPEN CUP APPARATUS

The Cleveland open-cup method is one of three main methods in chemistry for determining the flashpoint of a petroleum product using a Cleveland open-cup apparatus, also known as a Cleveland open-cup tester. First, the test cup of the apparatus (usually brass) is filled to a certain level with a portion of the product. Then, the temperature of this chemical is increased rapidly and then at a slow, constant rate as it approaches the theoretical flash point. The increase in temperature will cause the chemical to begin to produce flammable vapor in increasing quantities and density

Wash the test cup with an appropriate solvent to remove any oil or traces of gum or residue remaining from a previous test. If any deposits of carbon are present, they may be removed with steel wool. Flush the cup with cold water and dry for a few minutes over an open flame or a hot plate to remove the last traces of solvent and water. Cool

the cup to a point at least 56°C below the expected flash point before using. Support the thermometer in a vertical position with the bottom of the bulb 6 mm from the bottom of the cup and locate at a point half way between the center and side of the cup on a diameter perpendicular to the arc (or line) of the sweep of the test flame and on the side opposite to the test flame burner arm.



Fig 5.3 CLEVELAND OPEN CUP APPARATUS

The biodiesel blends and base diesel and turpentine were measured for their properties and the result were given below

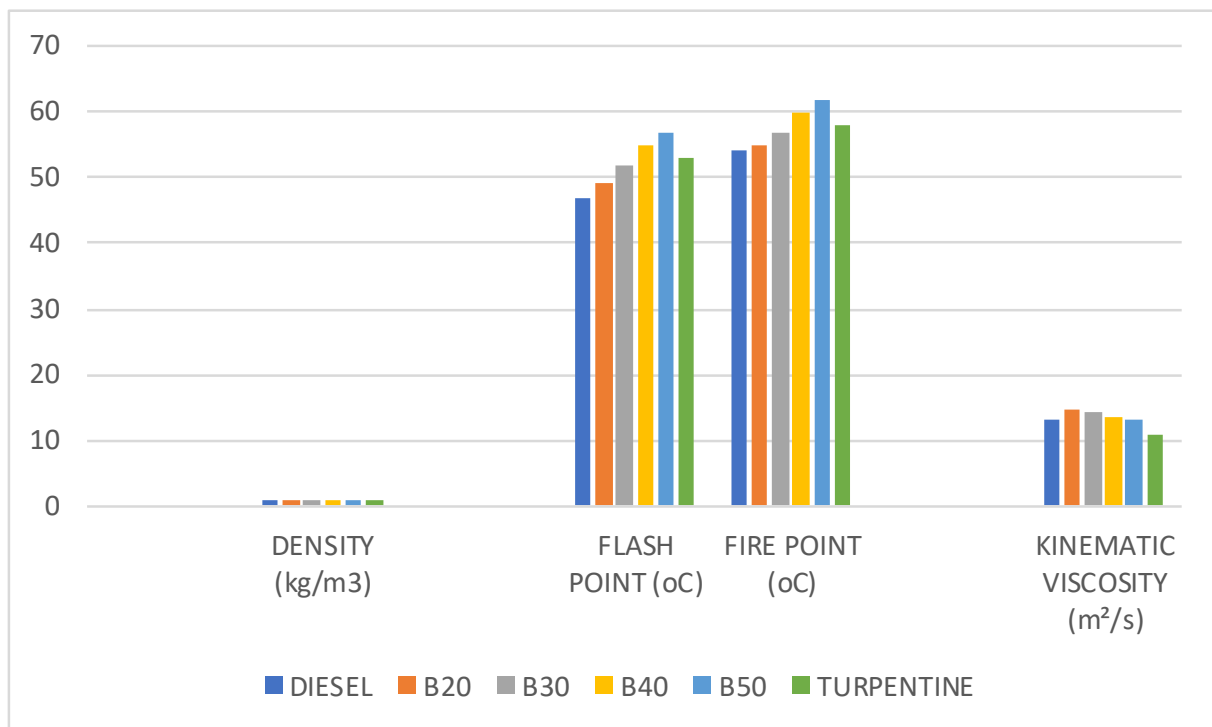


Fig 5.4 PROPERTIES OF BLENDS

Table 5.1 PROPERTIES OF FUEL

BLEND	DENSITY(kg/m³)	KINEMATIC VISCOSITY (m²/s)	FLASH POINT (°c)	FIRE POINT (°c)	CALORIFIC VALUE (kJ/kg)
DIESEL	0.827	13.28	47	54	42500
B20	0.831	14.567	49	55	42880
B30	0.835	14.147	52	57	43070
B40	0.839	13.360	55	60	43260
B50	0.844	13.055	57	62	43450
TURPENTINE	0.845	10.870	53	58	44400

CHAPTER 6

SPECIFICATIONS

6.1 ENGINE DESCRIPTION:

The setup consists of single cylinder, four stroke, Diesel engine connected to eddy current type dynamometer for loading. It is provided with necessary instruments for combustion pressure and crank-angle measurements. These signals are interfaced to computer through engine indicator for P θ –PV diagrams. Provision is also made for interfacing airflow, fuel flow, temperatures and load measurement. The setup has stand-alone panel box consisting of air box, fuel tank, manometer, fuel measuring unit, transmitters for air and fuel flow measurements, process indicator and engine indicator. Rotameters are provided for cooling the water and to measure the water flow. The setup enables study of engine performance for brake power, indicated power, frictional power, BMEP, IMEP, brake thermal efficiency, indicated thermal efficiency, Mechanical efficiency, volumetric efficiency, specific fuel consumption, A/F ratio and heat balance. LabVIEW based Engine Performance Analysis software package “Enginesoft” was used for performance evaluation. A computerized Diesel injection pressure measurement was also provided with the setup.

6.2 FEATURES:

- Online measurements and performance analysis
- P θ -PV plots, performance plots and tabulated results
- Configurable graphs
- IP, IMEP, FP indication
- Combustion analysis

6.3 ENGINE LAYOUT:

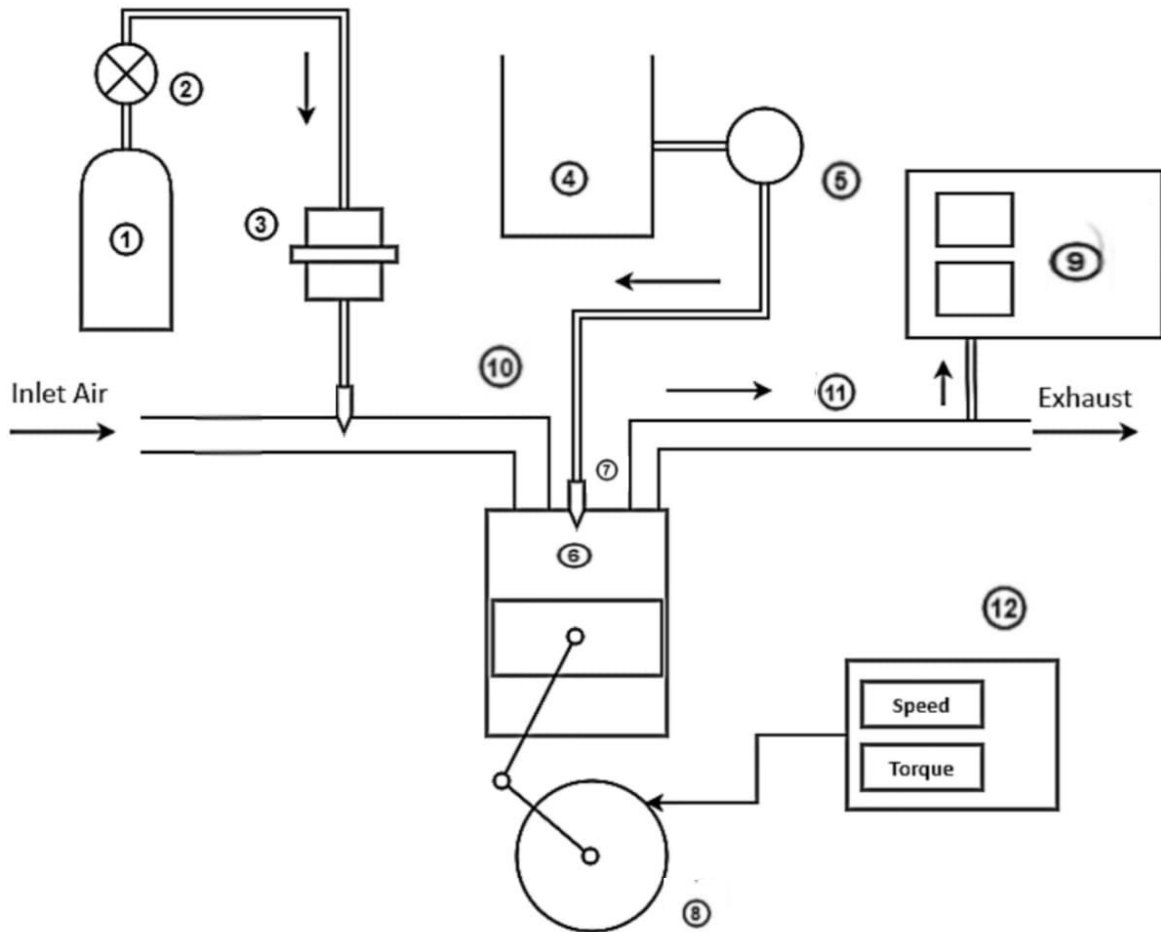


Fig 6.1 ENGINE LAYOUT

- | | |
|---------------------------------|--------------------------|
| ① - Hydrogen Storage Tank | ⑨ - Exhaust Gas Analyzer |
| ② - Pressure Regulator | ⑩ - Inlet Manifold |
| ③ - Flame Arrestor | ⑪ - Exhaust Manifold |
| ④ - biodiesel-Diesel blend tank | ⑫ - Dynamometer |
| ⑤ - Fuel Pump | |
| ⑥ - Diesel Engine | |
| ⑦ - Fuel Injector | |
| ⑧ - Crank Wheel | |

6.4 UTILITIES REQUIRED:

- Electric supply 230 +/- 10 VAC, 50 Hz
- 1Phase Computer IBM compatible with standard configuration
Water supply Continuous
- clean and soft water supply @ 1000 LPH, at 10 m. head
- Provide tap with 1" BSP size connection

6.5 ENGINE SPECIFICATIONS:

Table 6.1 ENGINEE SPECIFICATIONS

Product	Engine test setup 1 cylinder, 4 stroke, Diesel (Computerized)
Product code	224
Engine	Make Kirloskar, Model TV1, Type 1 cylinder, 4 stroke Diesel, water cooled, power 5.2 kW at 1500 rpm, stroke 110 mm, bore 87.5 mm. 661 cc, CR 17.5
Dynamometer	Type eddy current, water cooled
Propeller shaft	With Universal joints
Air box	M S fabricated with orifice meter and manometer
Fuel tank	Capacity 15 lit with glass fuel metering column
Calorimeter	Type Pipe in pipe
Piezo sensor	Range 5000 PSI, with low noise cable

Crank angle sensor	Resolution 1 Deg, Speed 5500 RPM with TDC pulse
Data acquisition device	NI USB-6210, 16-bit, 250kS/s.
Piezo powering unit	Model AX-409
Temperature sensor	Type RTD, PT100 and Thermocouple, Type K
Temperature transmitter	Type two wire, Input RTD PT100, Range 0–100 DegC, I/P Thermocouple, Range 0–1200 DegC, O/P 4–20mA
Load indicator	Digital, Range 0-50 Kg, Supply 230VAC
Load sensor	Load cell, type strain gauge, range 0-50 Kg
Fuel flow transmitter	DP transmitter, Range 0-500 mm WC
Air flow transmitter	Pressure transmitter, Range (-) 250 mm WC
Software	“Enginesoft” Engine performance analysis software

6.6 SOFTWARE:

EngineSoft is LabVIEW based software package developed by Apex Innovations Pvt. Ltd. for engine performance monitoring system. EngineSoft can serve most of the engine testing application needs including monitoring, reporting, data entry, data logging. The software evaluates power, efficiencies, fuel consumption and heat release. It is configurable as per engine set up. Various graphs are obtained at different operating condition. While on line testing of the engine in RUN mode necessary signals are scanned, stored and presented in graph. Stored data file is accessed to view the data graphical and tabular formats. The results and graphs can be printed. The data in excel format can be used for further analysis.

6.7 EXHAUST GAS ANALYSER:

The AVL-DI gas analyzer is an instrument used to measure HC, CO, NO_x, CO₂, and O₂ emissions in the diesel engine exhaust. It is based on the nondispersive infrared rays technique to measure pollutants. NDIR technique is adopted to measure the pollutants HC, CO, and CO₂ and the electrochemical measurement principle is adapted to measure O₂ concentration and NO_x in the exhaust. Before usage, the analyzer is allowed to warm up for a few hours. A leak check test is carried out by closing the inlet of the analyzer and avoiding the leakage of exhaust gases. This is followed by the HC residue test where all the HC and carbon particles and deposits inside the analyzer for accurate results. The compressed air is supplied over the exhaust probe to remove moisture, water droplets, carbon deposits, and blockage inside the exhaust probe before it is connected to the analyzer.



Fig 6.2 AVL-DI GAS ANALYZER

6.8 SMOKE METER:

AVL 437C smoke meters are used to analyze the smoke emission levels from the test engine. AVL smoke meter uses the light opacity technique, where light is passed through the exhaust sample and the opacity is measured by a photovoltaic cell. The opacity is a direct measure of smoke intensity.



Fig 6.3 AVL SMOKE METER

CHAPTER 7

PERFORMANCE, COMBUSTION AND EMISSION CHARACTERISTICS

7.1 PERFORMANCE PARAMETERS:

7.1.1 BRAKE THERMAL EFFICIENCY:

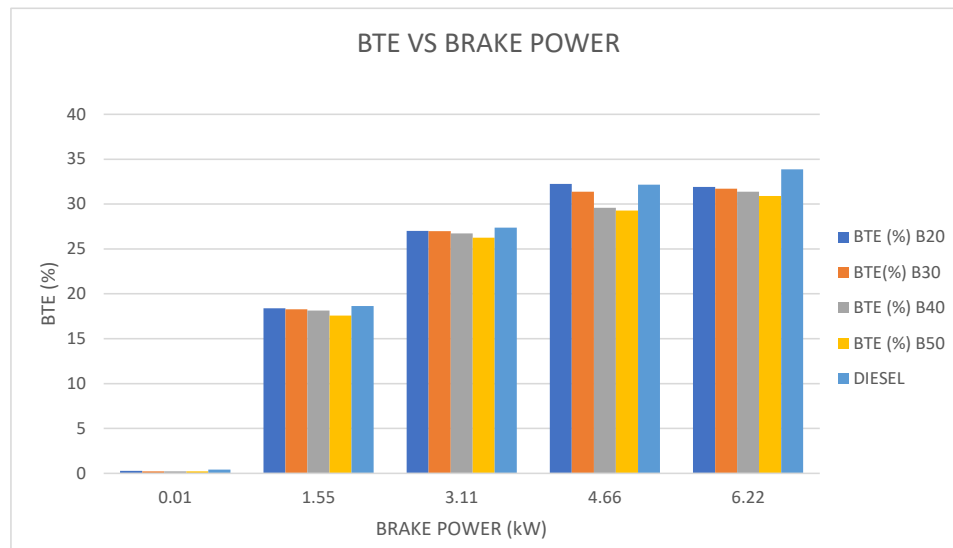


Fig 7.1 BTE vs BRAKE POWER

The Brake Thermal Efficiency of various fuels namely neat diesel, biodiesel blends namely B20,B30,B40 and B50 blends was plotted against Brake power and the results are analyzed. It is observed that brake thermal efficiencies of biodiesel blends decreases with increase in percentage of biodiesel because of poor volatility and inferior combustion characteristics of the biodiesel. At partial load, The brake thermal efficiency of B20,B30,B40,B50 blends were found to be 27.02%,26.98%,26.74%,26.25%. At full load condition, the BTE of the blends were found to 31.92%,31.71%,31.38%,30.89%.

7.12 SPECIFIC FUEL CONSUMPTION:

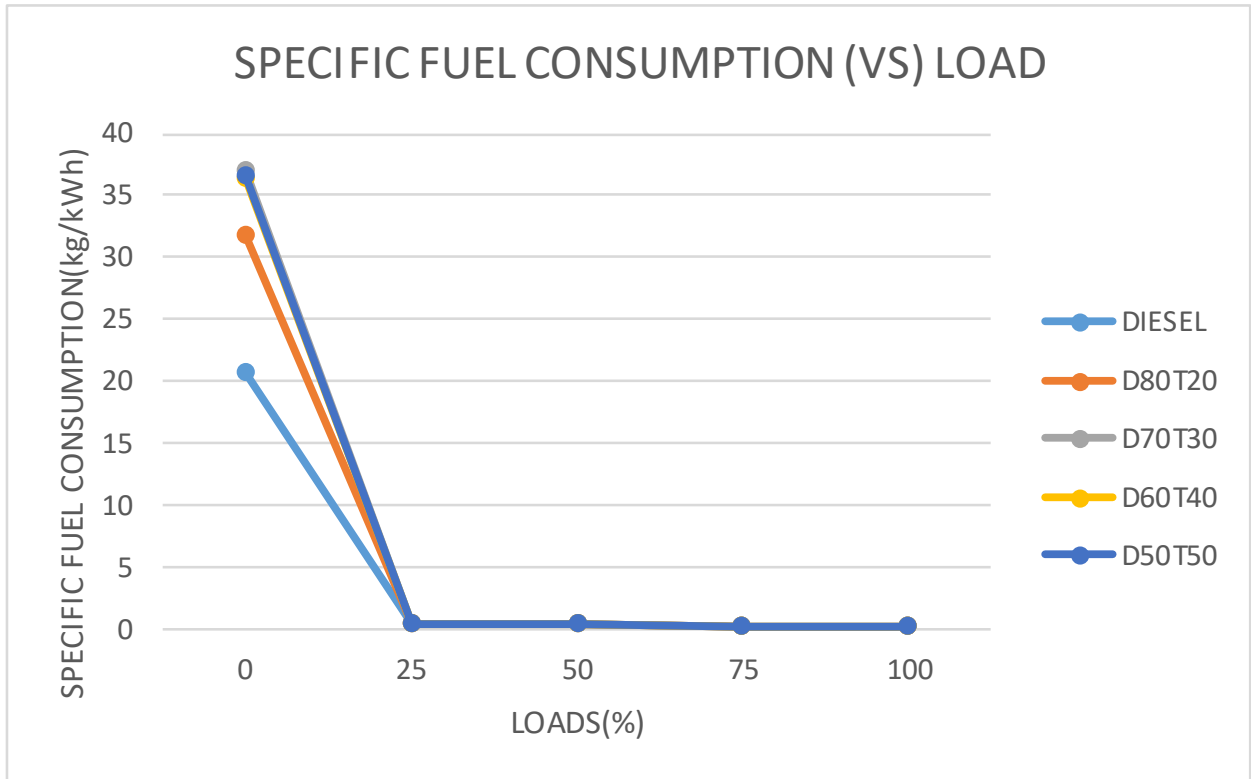


Fig 7.2 SFC vs LOAD

The specific fuel consumption of various bio diesel blends namely B20,B30,B40,B50 were plotted against various loads. It was found that the SFC of the bio diesel blends were found to increased than the diesel. This is due to the lower calorific values of the bio diesel blends than the diesel. At partial loads the SFC of B20,B30,B40,B50 blends were found to be 31%,32%,32%,33%.At full load condition, the SFC was found to be 26%,26%,27%,27%.

7.2 COMBUSTION PARAMETERS:

7.2.1 IN-CYLINDER PRESSURE:

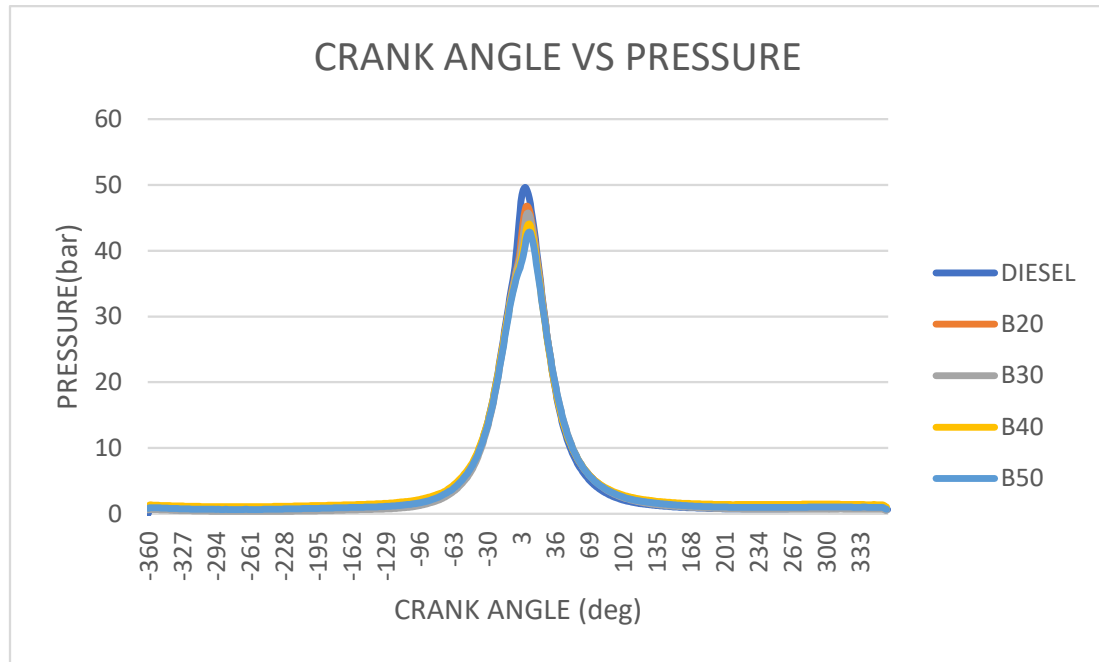


Fig 7.3 CRANK ANGLE vs PRESSURE

The cylinder pressure obtained using the pressure sensor at various crank angle is plotted. Pressure variations of neat diesel, biodiesel blends B20, B30, B40, B50 is seen in the above graph. There is a drop in the peak pressure of the biodiesel blend B20 when compared to the neat diesel fuel. The reduced pressure produced by the biodiesel was due to lower calorific value which means lower heat releasing capacity. Also, there is improper combustion of fuel samples due to lower volatility and higher viscosity. The combined effect of all these conditions, leads to reduction in the peak pressure produced during combustion by the biodiesel than neat diesel fuel.

7.2.2 HEAT RELEASE RATE:

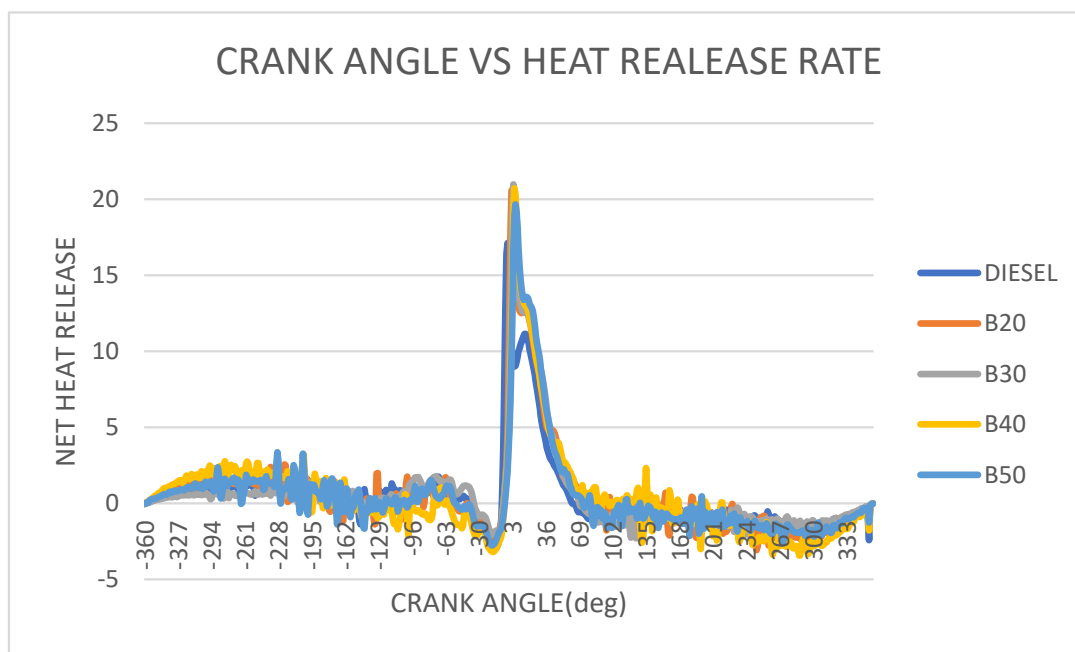


Fig 7.4 CRANK ANGLE vs HEAT RELEASE RATE

The heat released by various fuels namely neat diesel, biodiesel blends B20, B30, B40, B50 at various loading conditions are plotted and the results are analysed. It is seen that a negative HRR value was obtained before starting the combustion. This is due to the evaporation of the liquid fuel accumulated during the ignition delay period and the heat absorption from the environment. The HRR value move towards positive when the combustion started. The lesser HRR of the biodiesel B20 is mainly due to the lower calorific value of the biodiesel. Due to this, the amount of heat released on the combustion of this fuel also tends to be lesser.

7.3 EMISSION PARAMETERS:

7.3.1 HYDROCARBON EMISSION:

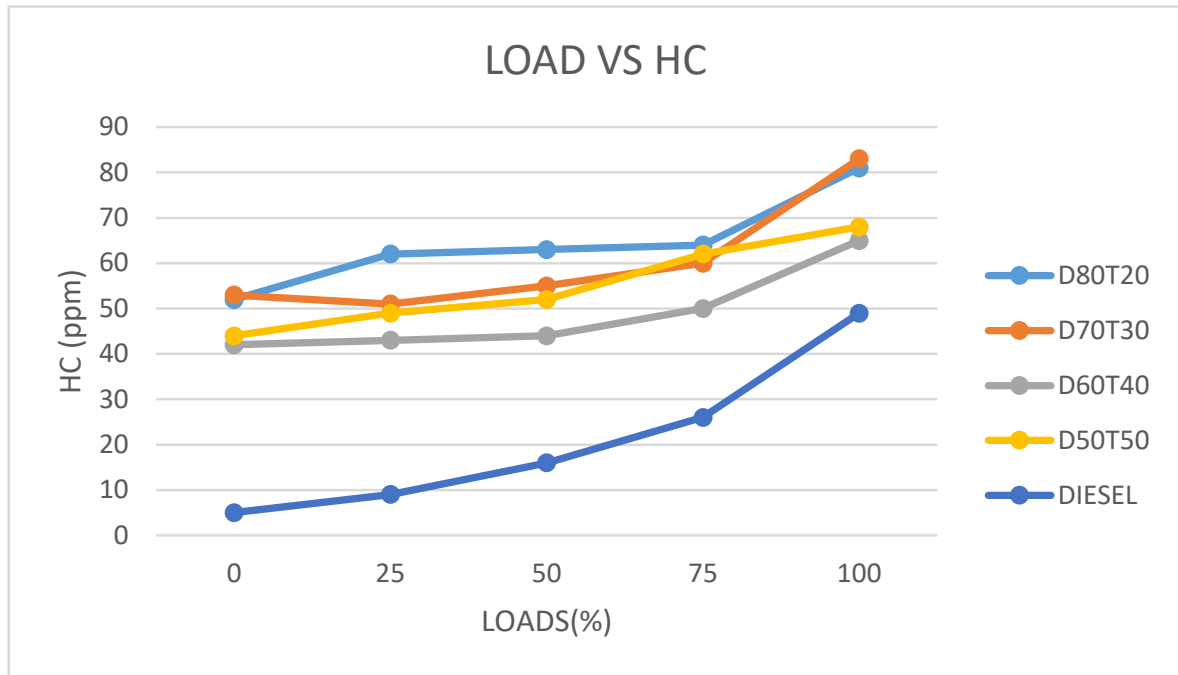


Fig 7.5 LOAD vs HC

HC emission is mainly due to the unburnt hydrocarbon molecules. Variation of HC with operating load conditions for the neat diesel, B20, B30, B40, B50 is shown in the above graph. Compared to the biodiesel fuel, diesel gives lesser HC emissions due the presence of more oxygen in it. Presence of oxygen enhances the rate of combustion. Hence the emission of unburnt hydrocarbons is automatically reduced. At partial loads, the percentage of HC of B20, B30, B40, B50 were 63%, 55%, 44%, 52%. At full load condition, the HC was found to be 81%, 83%, 65%, 68%.

7.3.2 CARBON MONOXIDE EMISSIONS:

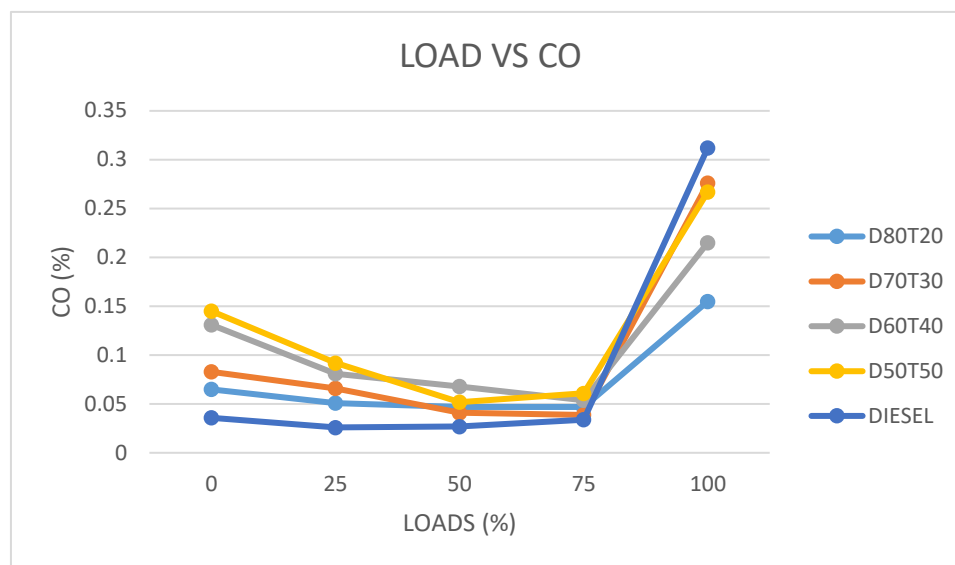


Fig 7.6 LOAD vs CO

Carbon monoxide is one of the compounds formed during the intermediate combustion stages of hydrocarbon fuels. As combustion proceeds to completion, oxidation of CO to CO₂ occurs through recombination reactions between CO and the different oxidants. If these recombination reactions are incomplete due to lack of oxidants or due to low gas temperatures, CO will be left. Variation of CO emissions with operating load conditions for the neat diesel, B20, B30, B40, B50 is shown in the above graph. The main reason for reduction in CO emission is the more amount of oxygen and lesser carbon content in biodiesel. The excess oxygen present in the biodiesel reacts with the CO and facilitates the oxidation of CO into CO₂. At partial loads, the percentage of CO for B20, B30, B40, B50 blends were found to be 47%, 41%, 68%, 52%. At full loads, the CO was found to be 155%, 276%, 215%, 267%..

7.3.3 SMOKE OPACITY:

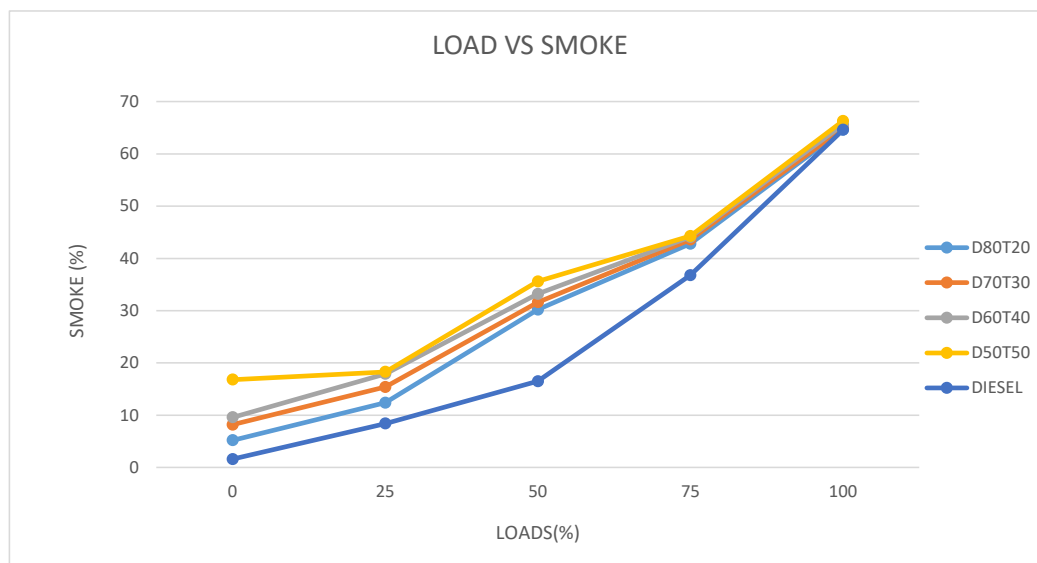


Fig 7.7 LOAD vs SMOKE

Smoke is also seen as a result of incomplete combustion. Variation of smoke opacity with operating load conditions for the neat diesel, B20, B30, B40 and B50 is shown in the above graph. It is observed that the smoke emissions increased due to the lower heating capacity of the turpentine biodiesel blends. when the load increases the percentage of smoke also increases. At low loads the percentage of smoke is lower and as the load increases the smoke emission increased. At partial load condition, the percentage of smoke for B20,B30,B40,B50 blends were 30.2%,31.6%,33.2%,35.6%. At full load condition, the smoke was found to be 64.6%,64.9%,65.6%,66.3%.

7.3.4 OXIDES OF NITROGEN:

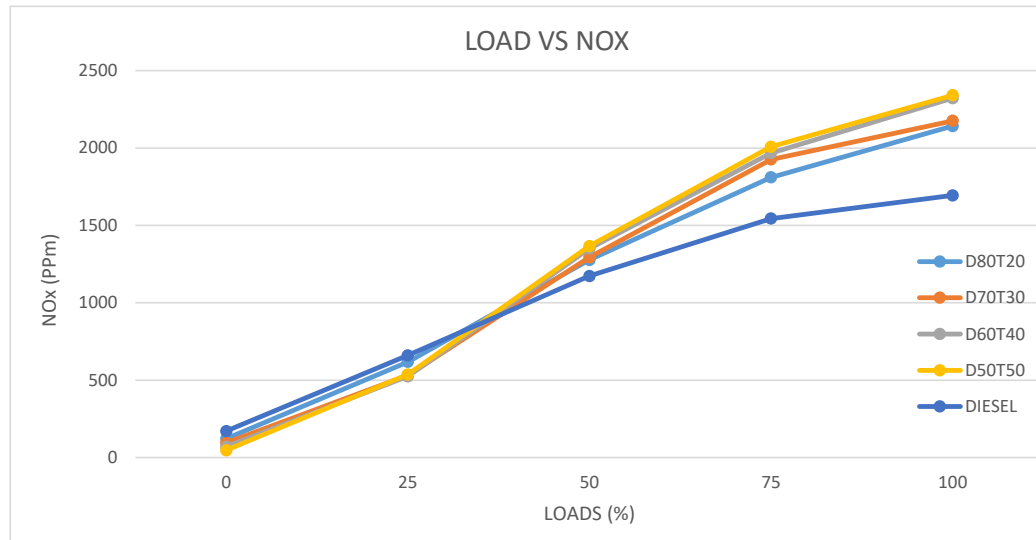


Fig 7.8 LOAD vs NOX

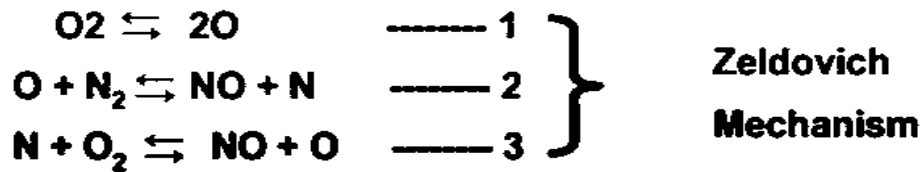


Fig 7.9 ZELDOVICH MECHANISM

According to the Zeldovich mechanism, the excess oxygen molecules will dissociate into atomic oxygen and these will react with the nitrogen molecules forming NO emissions. These take place during conditions of high temperature.

Variation of NOx emissions with operating load conditions for the neat diesel, B20, B30, B40 and B50 biodiesel fuel is shown in the above graph. At low loads, the emission of NOx decreases with the addition of turpentine biodiesel. AT high loads, the NOx emissions increases. At partial loads, the NOx was measured in ppm . for B20,B30,B40,B50 blends the NOx was found to be 1277,1294,1351,1366 ppm respectively. At full load condition, the NOx was found to be 2142,2176,2322,2341 ppm respectively.

CHAPTER 8

CONCLUSION

In this project, the effect of turpentine oil biodiesel in single cylinder diesel engine its effect on performance, combustion and emission characteristic is studied. The following conclusions are derived:

1. The brake thermal efficiency and fuel consumption in CI engine has decreased with the Turpentine biodiesel blend. This is due to the low heating capacity of the turpentine biodiesel blends. The BTE of B20 and B30 blend was found as 31.92% and 31.71% respectively.
2. The Diesel when compared to the biodiesel blend produces a higher in-cylinder pressure and heat release rate due to the higher flame velocity and good combustion rate .
3. The NO_x and CO emissions of the biodiesel blends were reduced. At full load, the CO was found to be 285% and 276% for B20 and B30 blends respectively.
4. At low loads , the biodiesel blends shows decrease in NO_x emissions. At low load, the NO_x was found to be 121 and 95 ppm for B20 and B30 blend.
5. The smoke and HC emissions of the biodiesel blends were increased when compared to the diesel fuel. At full loads, the smoke percentage for B20 and B30 were 64.6% and 64.9% respectively and the percentage of HC was 81% and 83% respectively.

REFERENCES:

1. M. Kannan, G. Elavarasan, D. Karthikeyan, Vikas Goyat.
Analysis of emission and performance characteristics of compression
ignition engine using Mahua oil-based biodiesel
2. Tomesh Kumar Sahu, Sumit Sarkar, Pravesh Chandra Shukla.
Combustion investigation of waste cooking oil (WCO) with varying
compression ratio in a single cylinder CI engine
3. Mehmet Hakkı Alma, Tufan Salan A review on novel bio-fuel from
Turpentine oil.
4. Anil singh yadav , Padam singh , Rahul sahu , p. Thangamuthu ,
Renuka shyam narain , y. Anupam rao, Azmeera balu , Bishnu prasad
panda , Abhishek sharma potential utilization of turpentine oil as
an alternative fuel
5. a.k. jeevanantham , d. madhusudan reddy , neel goyal , devansh
bansal, gopalakrishnan kumar , aman kumar , k. nanthagopal , b. ashok
experimental study on the effect of cetane improver
with turpentine oil onci engine characteristics