

# MD/GE - Molecular dynamics simulation and lattice energy of argon

Protocol for the Physical chemistry lab course PC 2 by  
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# 1 Theory

[1]

$$V(r_{ij}) = 4\epsilon_0 \left[ \left( \frac{\sigma_0}{r_{ij}} \right)^{12} - \left( \frac{\sigma_0}{r_{ij}} \right)^6 \right] \quad (1)$$

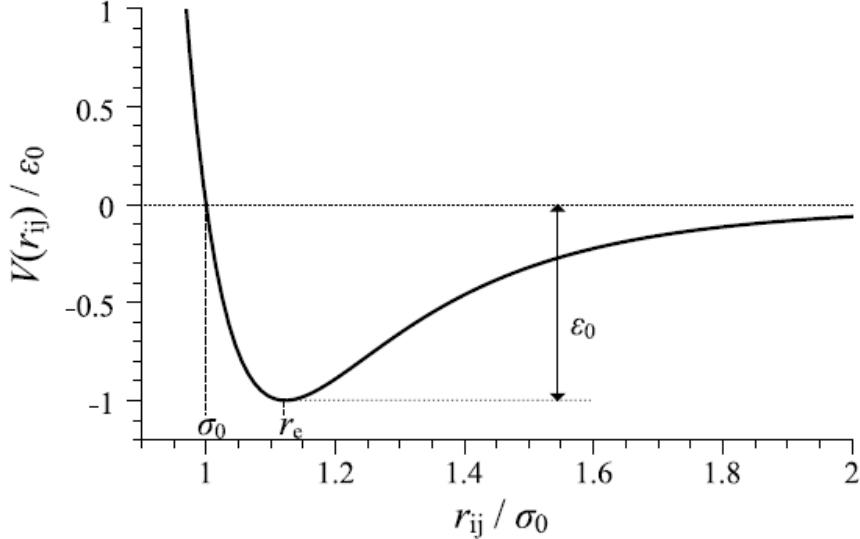


Abb. 1: .

$$T^* = \frac{k_B T}{\epsilon_0} \quad (2)$$

$$p^* = \frac{p \sigma_0^3}{\epsilon_0} \quad (3)$$

$$\ln p = -\frac{\Delta H_{\text{sub}}}{RT} + C \quad (4)$$

$$\Delta U_s = U_{\text{lattice}} + U_{\text{vib}} \quad (5)$$

$$U_s = \frac{5}{2}RT - \Delta H \quad (6)$$

$$U_{\text{vib}} = \frac{9}{8}R\Theta_D + 3RT \quad (7)$$

$$U_{\text{lattice}}(s) = 2N_A\epsilon_0 \left[ 12.132 \left( \frac{\sigma_0}{a} \right)^{12} - 14.454 \left( \frac{\sigma_0}{a} \right)^6 \right] \quad (8)$$

$$\log(p/\text{Torr}) = 7.781845 - \frac{341.619 \text{ K}}{T} - 0.0062649 \frac{T}{\text{K}} \quad (9)$$

## 2 Procedure

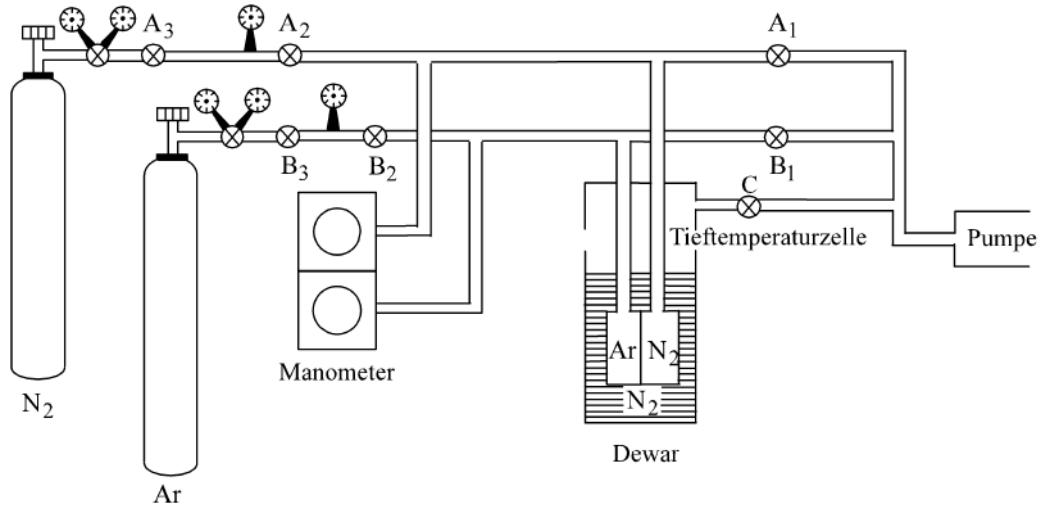


Abb. 2: Scheme of the measuring apparatus.

## 3 Evaluation

## 4 Error Analysis

## 5 Conclusion

## 6 References

- [1] H. Dilger, *2025-pc2-script-en*, 2025.