

CP203 Chemical Principles – Equation Sheet

$$V = -\frac{\mu_A \mu_B}{4\pi\epsilon r^3} [-2 \cos \theta_A \cos \theta_B + \sin \theta_A \sin \theta_B \cos(\phi_B - \phi_A)]$$

$$F_{\text{electrostatic}} = -k \frac{q_1 \cdot q_2}{r^2}$$

$$V = -\frac{\mu Q \cos \theta}{4\pi\epsilon r^2}$$

$$E_{\text{electrostatic}} = k \frac{q_1 \cdot q_2}{r}$$

$$V = -\frac{\alpha Q^2 e^2}{32\pi^2 \epsilon^2 r^4}$$

$$\epsilon_r = \frac{\epsilon}{\epsilon_o}$$

$$V = -\frac{\alpha_B \mu_A^2 (3 \cos^2 \theta_A + 1)}{32\pi^2 \epsilon^2 r^6}$$

$$k = \frac{1}{4\pi\epsilon_o}$$

$$V = -\frac{3h(\nu_A \nu_B) \alpha_A \alpha_B}{512(\nu_A + \nu_B) \pi^4 \epsilon^4 r^6}$$

$$K.E. = \frac{1}{2}mv^2$$

$$p = p^* e^{\frac{V_m \Delta P}{RT}}$$

$$\langle v^2 \rangle^{\frac{1}{2}} = \left(\frac{3RT}{M} \right)^{\frac{1}{2}}$$

$$p = p^* e^{\frac{2\gamma V_m}{rRT}}$$

$$\frac{\left(\frac{3RT}{M_A} \right)^{\frac{1}{2}}}{\left(\frac{3RT}{M_B} \right)^{\frac{1}{2}}} = \left(\frac{M_B}{M_A} \right)^{\frac{1}{2}}$$

$$p_{\text{in}} = p_{\text{out}} + \frac{2\gamma}{r}$$

$$d^2 = 2Dt$$

$$p_{\text{in}} = p_{\text{out}} + \frac{4\gamma}{r}$$

$$D = \frac{\lambda^2}{2\tau}$$

$$\gamma = \frac{F}{L}$$

$$D = D_0 e^{-\frac{E_a}{RT}}$$

$$\frac{2\gamma}{r} = \rho gh$$

$$D = \frac{k_B T}{6\pi\eta a}$$

$$\gamma = \frac{r h \rho g}{2 \cos \theta}$$

$$\eta = \eta_0 e^{\frac{E_a}{RT}}$$

$$\text{ratio} = \frac{V}{A \cdot L}$$

CP203 Chemical Principles – Equation Sheet

$$PV = nRT$$

$$\Delta G = -RT \ln K$$

$$\left(p + \frac{n^2 a}{V^2}\right)(V - nb) = nRT$$

$$\frac{p}{n(p^o - p)} = \frac{1}{n_m c} + \frac{(c - 1)}{n_m c} \cdot \frac{p}{p^o}$$

$$\Gamma = \frac{-1}{RT} \left(\frac{d\gamma}{d \ln c} \right)_{p,T}$$

$$c = \exp \left(\frac{|\Delta H_A| - |\Delta H_L|}{RT} \right)$$

$$\Gamma = \frac{-1}{2.303RT} \left(\frac{d\gamma}{d \log c} \right)_{p,T}$$

$$\theta = c_1 \ln(c_2 p)$$

$$A = \frac{1}{N_A \Gamma}$$

$$\theta = c_1 p^{\frac{1}{c_2}}$$

$$\Delta G = \Delta H - T \Delta S$$

$$\theta_A = \frac{K_A P_A}{1 + \sum_i K_i P_i}$$

$$Z = \frac{P}{(2\pi m k_B T)^{\frac{1}{2}}}$$

$$V_{ads} = \frac{(K \cdot p)^{\frac{1}{2}} \cdot V_m}{\left[1 + (K \cdot p)^{\frac{1}{2}}\right]}$$

$$T = \frac{\text{Number of sites}}{\text{Collision rate}}$$

$$\log w = \log w_o - D \log_{10}^2 \left(\frac{p^o}{p} \right)$$

$$\frac{P}{n_{ads}} = \frac{1}{K n_m} + \frac{P}{n_m}$$

$$D = B \left(\frac{T}{\beta} \right)^2$$

$$s = n_m A_m L$$

$$\frac{M_t}{M_e} = 1 - e^{-kt}$$

$$\frac{d \ln K}{dT} = \frac{\Delta_f H^\circ}{RT^2}$$

$$k = A e^{-\frac{E_a}{RT}}$$

$$\left(\frac{\partial \ln p}{\partial (1/T)} \right)_\theta = \frac{\Delta_{ad} H^\circ}{R}$$

$$\frac{M_t}{M_e} = A_1 \left[1 - e^{-(k_1 t)^{\beta_1}} \right] + A_2 \left[1 - e^{-(k_2 t)^{\beta_2}} \right]$$