

P.3

Problem statement:

Compound A diffuses through 4m tube and reacts as it diffuses. An equation given below, one end has large (0.1M) concentration of A and other absorbent that absorbs all A (0M).

Given,

$$D = 1.5 \times 10^{-6}$$

$$C_A(x=0) = 0.1 \text{ M}$$

$$k = 5 \times 10^{-6}$$

$$C_A(x=4) = 0 \text{ M}$$

concentration of A as function of dist in tube = ?
 $L = 4 \text{ m}$

equation,

$$D \frac{d^2 A}{dx^2} - kA = 0$$

$$\frac{d^2 A}{dx^2} = \frac{d}{dx} \left(\frac{dA}{dx} \right) = \frac{dT}{dx} \bigg|_{i+1} - \frac{dT}{dx} \bigg|_i$$

 Δx

$$\Delta x = x_{i+1} - x_i = L/N \longrightarrow N \text{ is step size}$$

$$\frac{d^2 A}{dx^2} = \frac{A_{i+1} - 2A_i + A_{i-1}}{\Delta x^2} \longrightarrow \text{by Finite diff. Method}$$

$$D(A_{i+1} - 2A_i + A_{i-1}) = (kA_i)(\Delta x^2)$$

$$i=1 \quad D(A_2 - A_1 + A_0) = (kA_1)(\Delta x^2)$$

$$i=2 \quad D(A_3 - A_2 + A_1) = (kA_2)(\Delta x^2)$$

$$i=3 \quad D(A_4 - A_3 + A_2) = (kA_3)(\Delta x^2)$$

 \vdots \vdots \vdots \vdots

$$i=20 \quad D(A_{21} - 2A_{20} + A_{19}) = (kA_{20})(\Delta x^2)$$

Concentration Profile of A in the Tube

