

# Cahn - Hilliard Equation

Vishal S

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## 1 Introduction

### Aim

To derive the Cahn-Hilliard equation to model spinodal decomposition.

### Strategy

- Treat Gibbs energy as a functional
- Derive the chemical potential for such a system
- Use Fick's law to understand the composition evolution

### Assumptions

- The system has cubic symmetry
- The gibbs energy is a quadratic in  $c$

### Derivation

#### Gibbs energy formulation

The total Gibbs' energy of the system is given by,

$$F(x, y, z, c, \nabla c, \nabla^2 c) = \int f(c, \nabla c, \nabla^2 c) dV, \quad (1)$$

Expanding Gibbs' function using Taylor's approximation, we get

$$\begin{aligned} f &= f(c) + \frac{\partial f}{\partial (\nabla c)_i} (\nabla c)_i + \frac{\partial f}{\partial (\nabla^2 c)_{ij}} (\nabla^2 c)_{ij} + \frac{1}{2!} \frac{\partial f}{\partial (\nabla c)_i \partial (\nabla c)_j} (\nabla c)_i (\nabla c)_j, \\ \alpha_i &= \frac{\partial f}{\partial (\nabla c)_i}, \\ \beta_{ij} &= \frac{\partial f}{\partial (\nabla^2 c)_{ij}}, \\ \gamma_{ij} &= \frac{\partial f}{\partial (\nabla c)_i \partial (\nabla c)_j}, \\ f &= f(c) + \alpha_i (\nabla c)_i + \beta_{ij} (\nabla^2 c)_{ij} + \frac{1}{2} \gamma_{ij} (\nabla c)_i (\nabla c)_j, \end{aligned}$$

Simplifying the third term in the above equation, we get,

$$\begin{aligned} &\int \beta_{ij} (\nabla^2 c)_{ij} dV, \\ &= \int \beta_{ij} \frac{\partial^2 c}{\partial x_i \partial x_j} dV, \\ &= \int \beta_{ij} \frac{\partial}{\partial x_i} \left( \frac{\partial c}{\partial x_j} \right) dV, \\ &= \left[ \beta_{ij} (\nabla c)_j \right]_s - \int \frac{\partial \beta_{ij}}{\partial x_i} (\nabla c)_j dV, \\ &= - \int \frac{\partial \beta_{ij}}{\partial c} (\nabla c)_i (\nabla c)_j dV, \end{aligned}$$

$$f = f(c) + \alpha_i (\nabla c)_i + \left[ \frac{\gamma_{ij}}{2} - \frac{\partial \beta_{ij}}{\partial c} \right] (\nabla c)_i (\nabla c)_j,$$

$$f = f(c) + \alpha_i (\nabla c)_i + \kappa_{ij} (\nabla c)_i (\nabla c)_j,$$

Since, we assumed that the material has cubic symmetry, we can assume  $\alpha_i = 0$  and  $\kappa_{ij} = \kappa \delta_{ij}$  by invoking Neumann's principle.

### Chemical Potential

$$\frac{\delta F}{\delta c} = \mu,$$

$$F = \int \left[ f(c) + \kappa (\nabla c)^2 \right] dV,$$

$$\frac{\delta F}{\delta c} = \frac{\partial f}{\partial c} - 2\kappa \nabla^2 c$$

### Fick's Law

$$J = -M \nabla \mu,$$

$$\frac{\partial c}{\partial t} = -\nabla J,$$

$$\frac{\partial c}{\partial t} = M \nabla^2 \mu,$$

$$\frac{\partial c}{\partial t} = M \nabla^2 \left( \frac{\partial f}{\partial c} - 2\kappa \nabla^2 c \right),$$

Thus simplifying the above equation gives, Cahn-Hilliard Equation,

$$\frac{\partial c}{\partial t} = M f'' \nabla^2 c - 2\kappa M \nabla^4 c \tag{2}$$