

### Free particle in 1 D box:

Here free particle refers to electron and 1D box can be an atom or a piece of metal wire. As the electron is completely free to move within the box, its potential energy is zero. And the electron moves only along x-axis, the following partial differential equation (from Schrodinger's time independent equation) will be written as complete differential equation.

$$\nabla^2 \Psi + \frac{2m(E - V)}{\hbar^2} \Psi$$

As  $V=0$  between the walls, the equation becomes

$$\frac{\partial^2 \Psi}{\partial x^2} + \frac{2m}{\hbar^2} E \Psi = 0$$

$\frac{d^2 \Psi}{dx^2} + k^2 = 0$  is the wave equation for electron's travel along one dimension.

$x=0$  and  $x=L$  are the boundaries of the one dimensional potential box. In the following figure the potential energy inside the box is zero; and the electron may face up to infinite potential by the walls of the box in such a way that the electron never gets outside the box. Thus, the potential function is defined in the following way:

$$\begin{aligned} V(x) &= 0 & \text{for } L > x > 0 \\ V(x) &= \infty & \text{for } x \geq L, x \leq 0 \end{aligned}$$

$$\therefore k^2 = \frac{2mE}{\hbar^2}$$

The solution of the second order differential equation is as follows,

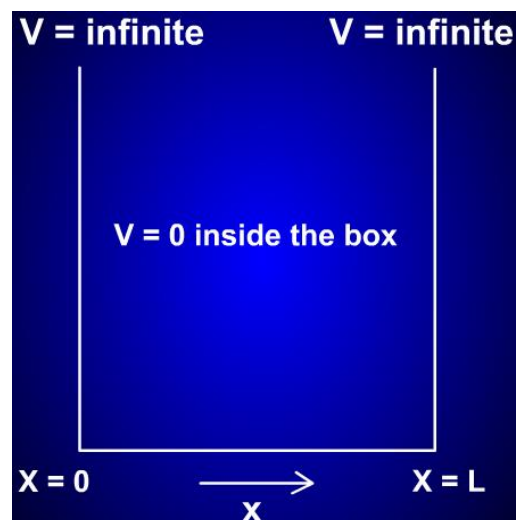
$$\Psi(x) = A \sin kx + B \cos kx$$

The wave function is zero at the boundaries so,

$$\Psi = 0 \text{ at } x = 0$$

After applying the first boundary condition,

$$0 = A \times 0 + B \times 1$$



$$= 0 + B$$

$$\Rightarrow B = 0$$

$$\text{Thus, } \Psi(x) = A \sin kx$$

$\Psi = 0$  at  $x = a$  (the second boundary condition)

After applying the second boundary condition

$$0 = A \sin ka$$

$$ka = n\pi, n = 1, 2, 3, \dots$$

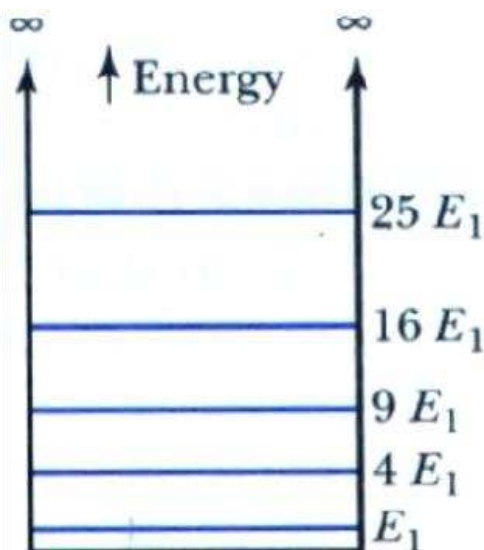
$$k = \frac{n\pi}{a}$$

$$k = \sqrt{\frac{2mE}{\hbar^2}}, \text{ and also } k = \frac{n\pi}{a}$$

$$k^2 a^2 = \frac{2mE_n a^2}{\hbar^2} = n^2 \pi^2$$

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2ma^2}, n = 1, 2, 3, \dots$$

Thus, for each value of  $n$ , the possible energy of the particle is given by the above equation



$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2}$$

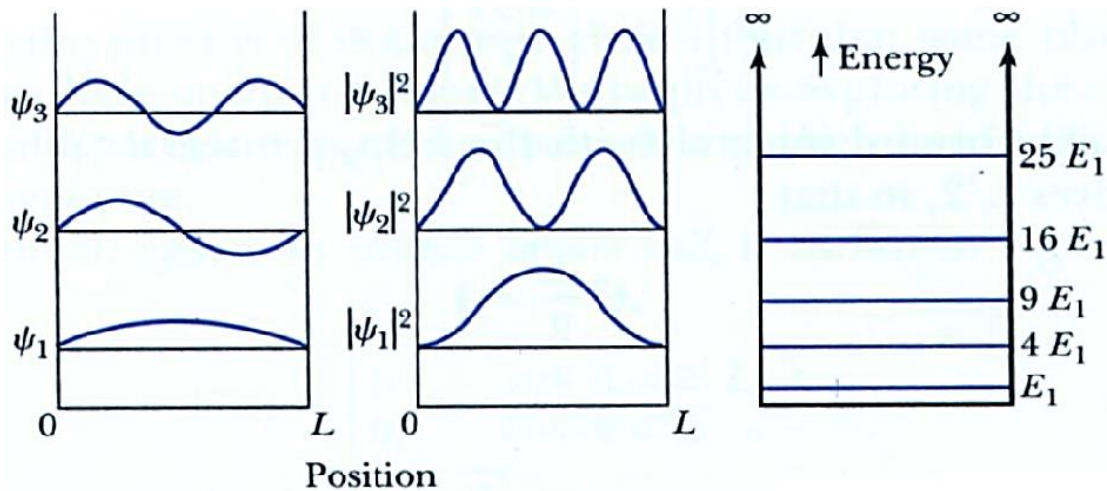
$$E_2 = 4 \frac{\pi^2 \hbar^2}{2mL^2} = 4E_1$$

$$E_3 = 9 \frac{\pi^2 \hbar^2}{2mL^2} = 9E_1$$

$$E_4 = 16 \frac{\pi^2 \hbar^2}{2mL^2} = 16E_1$$

$$E_5 = 25 \frac{\pi^2 \hbar^2}{2mL^2} = 25E_1$$

The following figure explains the wave functions, probability densities and energy values of the particle at different energy levels.



At a particular point, the probability of the particle being present is different for different quantum numbers. For example,  $|\Psi_1|^2$  has its maximum value at  $L/2$  in the middle. Thus, a particle in the lowest energy level of  $n=1$  is most likely to be in the middle while a particle in the next higher state of  $n=2$  is never there.

### Nodes and Antinodes:

- $n = 1$  corresponds to the ground state
- $n = 2$  corresponds to the first excited state, etc

$n = 3$  is the second excited state, 4 nodes, 3 antinodes

$n = 2$  is the first excited state, 3 nodes, 2 antinodes

$n = 1$  is the ground state (fundamental mode): 2 nodes, 1 antinode

Note that lowest possible energy for a particle in the box is not zero but  $E_0 (= E_1)$ , the zero-point energy.

This is a result consistent with the Heisenberg uncertainty principle

The diagram shows the wave functions  $\psi(x)$  for  $n=1$  to  $n=5$  in a box of length  $L$ . The energy levels are shown on the right, ranging from  $E=0$  to  $25E_1$ .

### Eigen functions:

Eigen means special or particular. After including particular energy value function, the wave function is called as Eigen function.

$$\Psi_n = A \sin kx = A \sin \sqrt{\frac{2mE_n}{\hbar^2}} x, \text{ where, } n = 1, 2, 3, \dots$$

$$\Psi_n = A \sin \frac{n\pi x}{a}$$

$\int_{-\infty}^{+\infty} |\Psi_n|^2 dx = 1$  (within the box, the hundred percentage of the particle is existing, since boundaries prevent them strongly from escaping outside)

$$\int_0^a |\Psi_n|^2 dx = \int_0^a A^2 \sin^2 \left( \frac{n\pi x}{a} \right) dx = 1$$

$$\int_0^a \sin^2 \left( \frac{n\pi x}{a} \right) dx = \frac{a}{2} \text{ (standard integral)}$$

$$\int_0^a |\Psi_n|^2 dx = A^2 \frac{a}{2} = 1$$

$$A^2 = \frac{2}{a}$$

$$A = \sqrt{\frac{2}{a}} \text{ is the normalised constant}$$

Thus by applying the normal condition (100% existence of the particle within the box) the constant A has been found out. So the constant A is called as normalised constant.

**Normalised Eigen functions:**

Eigen function with the normalised constant is called as normalized Eigen function.

$$\Psi_n(x) = A \sin kx = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a}x\right), \text{ where, } n = 1, 2, 3, \dots$$