



Engineering Physics

(PHY1701)

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Contents

- Laser Characteristics,
- Spatial and Temporal Coherence,
- Einstein Coefficient & its significance,
- Population inversion,
- Two, three & four level systems,
- Pumping schemes,
- Threshold gain coefficient,
- Components of laser,
- Nd-YAG, He-Ne, CO₂ and their engineering applications

❖ William Silfvast, Laser Fundamentals, 2008, Cambridge University Press.

Problem 1

Q) Find the ratio of population of the two states in He-Ne laser that produces light of wavelength 6328 Å at 27°C.

Given data:

Wavelength, $\lambda = 6328 \text{ Å}$

Temperature, $T = 27^\circ\text{C} = 300 \text{ K}$

Solution

$$E = \frac{N_2}{N_1} = e^{-(E_2 - E_1)/kT}$$

$$E_2 - E_1 = \frac{12400}{6328} \text{ eV} = 1.96 \text{ eV}$$

$$\begin{aligned} \therefore \frac{N_2}{N_1} &= \exp \left[\frac{-1.96 \text{ eV}}{(8.61 \times 10^{-5} \text{ eV/K}) 300 \text{ K}} \right] \\ &= e^{-75.88} = 1.1 \times 10^{-33} \end{aligned}$$

Q) The CO₂ laser is one of the most powerful lasers. The energy difference between the two lasers is 0.117 eV. Determine the frequency and wavelength of radiation.

$$\lambda = \frac{12400}{E \text{ (eV)}} = \frac{12400}{0.117}$$

$$\lambda = 105983 \text{ \AA} = 10.5 \text{ }\mu\text{m}.$$

$$\gamma = \frac{c}{\lambda} = \frac{3 \times 10^8}{10.5 \times 10^6} = 2.9 \times 10^{13} \text{ Hz}$$

Problem 3

Q) A He-Ne laser produces an output power of 5 mW. If it emits light of wavelength 632.8 nm, calculate the number of photons emitted by the laser in one second.

Given data:

$$\text{Output power, } P = 5 \text{ mW}$$

$$\text{Wavelength, } \lambda = 632.8 \text{ nm}$$

Solution

$$\begin{aligned} \text{Energy of one photon, } h\nu &= \frac{hc}{\lambda} = \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{632.8 \times 10^{-9}} \\ &= 3.141 \times 10^{-19} \text{ J} \\ &= 1.96 \text{ eV} \end{aligned}$$

$$\begin{aligned} \text{Number of photons emitted} &= \frac{\text{output power}}{\text{energy of one photon}} = \frac{5 \times 10^{-3}}{3.141 \times 10^{-19}} \\ &= 1.591 \times 10^{16} \text{ photons per second} \end{aligned}$$

Problem 4

Q) Calculate the relative population of the energy levels N_1 and N_2 at 300K, $\lambda=500$ nm.

From Maxwell and Boltzmann law, the relative population is given by

$$\frac{N_1}{N_2} = \frac{\exp\left(-\frac{E_1}{kT}\right)}{\exp\left(-\frac{E_2}{kT}\right)} = \exp\left(-\frac{E_1 - E_2}{kT}\right) = \exp\left(\frac{h\nu}{kT}\right)$$

Substituting the values of T and λ , we get

$$\begin{aligned}\frac{N_1}{N_2} &= \exp\left(\frac{h\nu}{kT}\right) = \exp\left(\frac{hc}{\lambda kT}\right) \\ &= \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{500 \times 10^{-9} \times 1.38 \times 10^{-23} \times 300} = \exp(96.029) \\ &= 5.068 \times 10^{41}\end{aligned}$$

The relative population between N_1 and N_2 is 5.068×10^{41} .

Problem 5

(a) The He-Ne system is capable of lasing at several different IR wavelengths, the prominent one being $3.3913 \mu\text{m}$. Determine the energy difference (in eV) between upper and lower levels for this wavelength.

Given data:

Wavelength, $\lambda = 3.3913 \mu\text{m}$

Solution

$$\begin{aligned} E &= \frac{12400 \text{ (eV)}}{\lambda \text{ (Å)}} = \frac{12400}{33913} \text{ (eV)} \\ &= 0.37 \text{ eV} \end{aligned}$$

Problem 6

(b) Calculate the efficiency of a He-Ne laser, if it produces an output power of 5 mW and if it is operated with a current of 10 mA at 3 kV.

Given data:

Output power, $P = 5 \text{ mW} = 5 \times 10^{-3} \text{ W}$

Current, $I = 10 \text{ mA} = 10 \times 10^{-3} \text{ A}$

Voltage, $V = 3 \text{ kV} = 3 \times 10^3 \text{ V}$

Solution

$$\text{Efficiency} = \frac{\text{output power}}{\text{input power}} \times 100 \%$$

$$= \frac{5 \times 10^{-3}}{10 \times 10^{-3} \times 3 \times 10^3} \times 100 \% = 0.016667 \%$$

The efficiency of the laser = 0.016667 %

Problem 7

Q) A transition between the energy level E_2 and E_1 produces a light of wavelength 632.8 nm, calculate the energy of the emitted photons..

Given data:

Wavelength, $\lambda = 632.8 \text{ nm}$

Solution

$$\begin{aligned}\text{Energy of the emitted photon, } E &= h\nu = \frac{hc}{\lambda} \\ &= \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{632.8 \times 10^{-9}} \\ &= 3.141 \times 10^{-19} \text{ J} = 1.96 \text{ eV} \\ \text{The energy of the photon} &= 1.96 \text{ eV}\end{aligned}$$

Problem 8

Q) A system has three energy levels E_1 , E_2 and E_3 . The energy levels E_1 and E_2 are at 0 eV and 1.4 eV respectively. If the lasing action takes place from the energy level E_3 to E_2 , and emits a light of wavelength $1.15 \mu\text{m}$, find the value of E_3 ?

Given data:

The value of first energy level, $E_1 = 0 \text{ eV}$

Value of second energy level, $E_2 = 1.4 \text{ eV}$

Wavelength, $\lambda = 1.15 \mu\text{m}$

Solution

$$\text{Energy of the emitted photon, } E = h\nu = \frac{hc}{\lambda}$$

$$= \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{1.15 \times 10^{-6}}$$

$$= 1.728 \times 10^{-19} \text{ J}$$

$$= 1.079 \text{ eV}$$

The energy value of $E_3 = E_2 + h\nu$

$$= 1.4 \text{ eV} + 1.079 \text{ eV}$$
$$= 2.479 \text{ eV}$$

The energy value of $E_3 = 2.479 \text{ eV}$

Problem 9

A laser transition takes place from an energy level at 3.2 eV to another level at 1.6 eV. Calculate the wavelength of the laser beam emitted.

Given data:

The value of higher energy level $E_1 = 3.2 \text{ eV}$

The value of lower energy level $E_2 = 1.6 \text{ eV}$

Solution

$$\begin{aligned}\text{Energy difference, } E_2 - E_1 &= 3.2 - 1.6 \\ &= 1.6 \text{ eV}\end{aligned}$$

$$\begin{aligned}\text{Wavelength, } \lambda &= \frac{hc}{E} = \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{1.6 \times 1.6 \times 10^{-6}} \\ &= 7.7648 \times 10^{-7} \text{ m}\end{aligned}$$

The wavelength of the photon, $\lambda = 7.7648 \times 10^{-7} \text{ m}$

Problem 10

The band gap of GaAs is 1.42 eV. What is the wavelength of the laser beam emitted by a GaAs diode laser?

Given data:

Band gap of GaAs = 1.42 eV

Solution

Wavelength of laser emitted by GaAs,

$$\begin{aligned}\lambda &= \frac{hc}{E} = \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{1.42 \times 1.6 \times 10^{-6}} \\ &= 8.749 \times 10^{-7} \text{ m}\end{aligned}$$

The wavelength of the laser emitted by GaAs, $\lambda = 8.749 \times 10^{-7} \text{ m}$

Problem 11

A laser beam emits an output power of 1 mW. If it is focused as a spot having a diameter of $1\mu\text{m}$, calculate the intensity of the laser beam.

Given data:

Output power, $P = 1\text{ mW} = 1 \times 10^{-3}\text{ W}$

Diameter = $1\mu\text{m}$

Radius, $r = 0.5\mu\text{m} = 0.5 \times 10^{-6}\text{ m}$

Solution

$$\begin{aligned}\text{Intensity of laser} &= \frac{\text{power}}{\text{area of cross section}} = \frac{1 \times 10^{-3}}{\pi(0.5 \times 10^{-6})^2} \\ &= 1.273 \times 10^9\text{ W m}^{-2}\end{aligned}$$

The intensity of the laser = $1.273 \times 10^9\text{ W m}^{-2}$