

Spontaneous Instability in Ultra Thin Polymer Films:-

25.03.2022
Lecture 31
Extra class, Fri 5 PM

Liquid Thin Film Hydrodynamics

- ↳ Film thk will keep on changing.
Dewetted Features → They will slowly evaporate
- Preferred model system is a Polymer Thin Film
 - Solid like at Room
 - Liquid State only when heated above T_g .
- All the evolution takes place in the Liquid state
 - cool down the sample → So the dewetted features can again be converted to solid and made permanent.



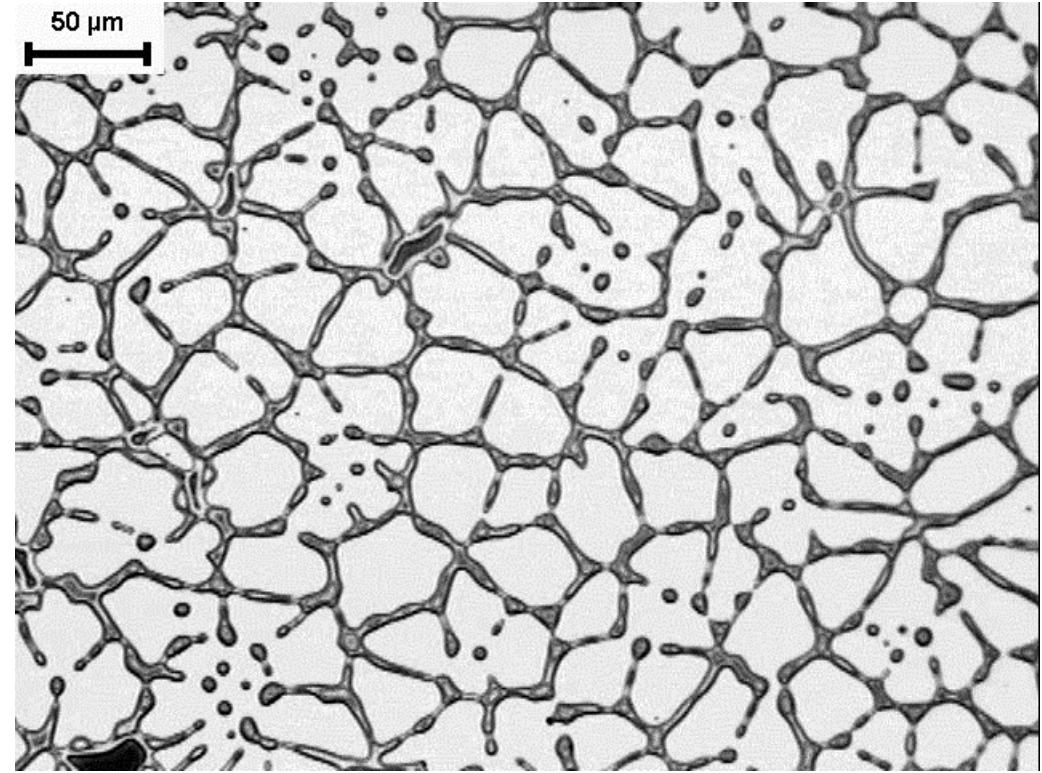
Spin Coated Film: $\sim 40\text{ nm}$

At Room Temp

$< T_g$

AE +ve

→ Film is solid like. Viscosity is ∞ .



Evolution starts only $T > T_g$.

↳ growth of fluctuation
formation of holes.
retraction of contact line

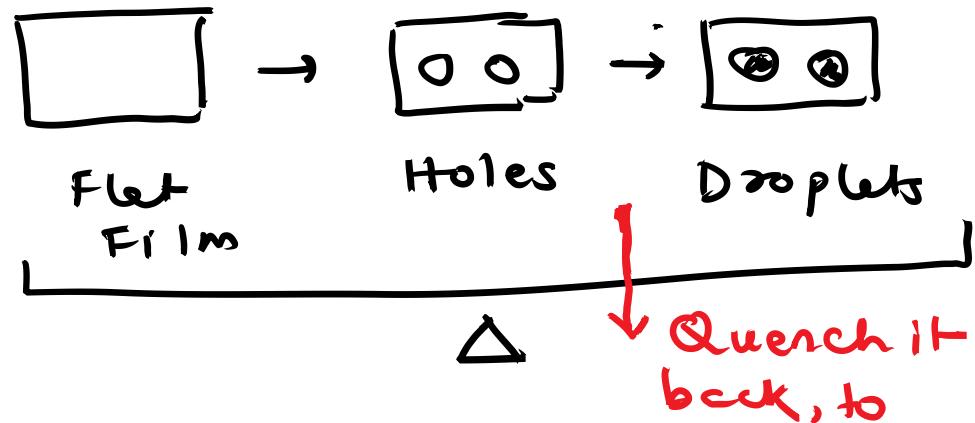
* At room temp (any $T < T_g$), the film despite being thermodynamically unstable, the film actually remains stable. → Thermodynamically unstable but **kinetically stable**

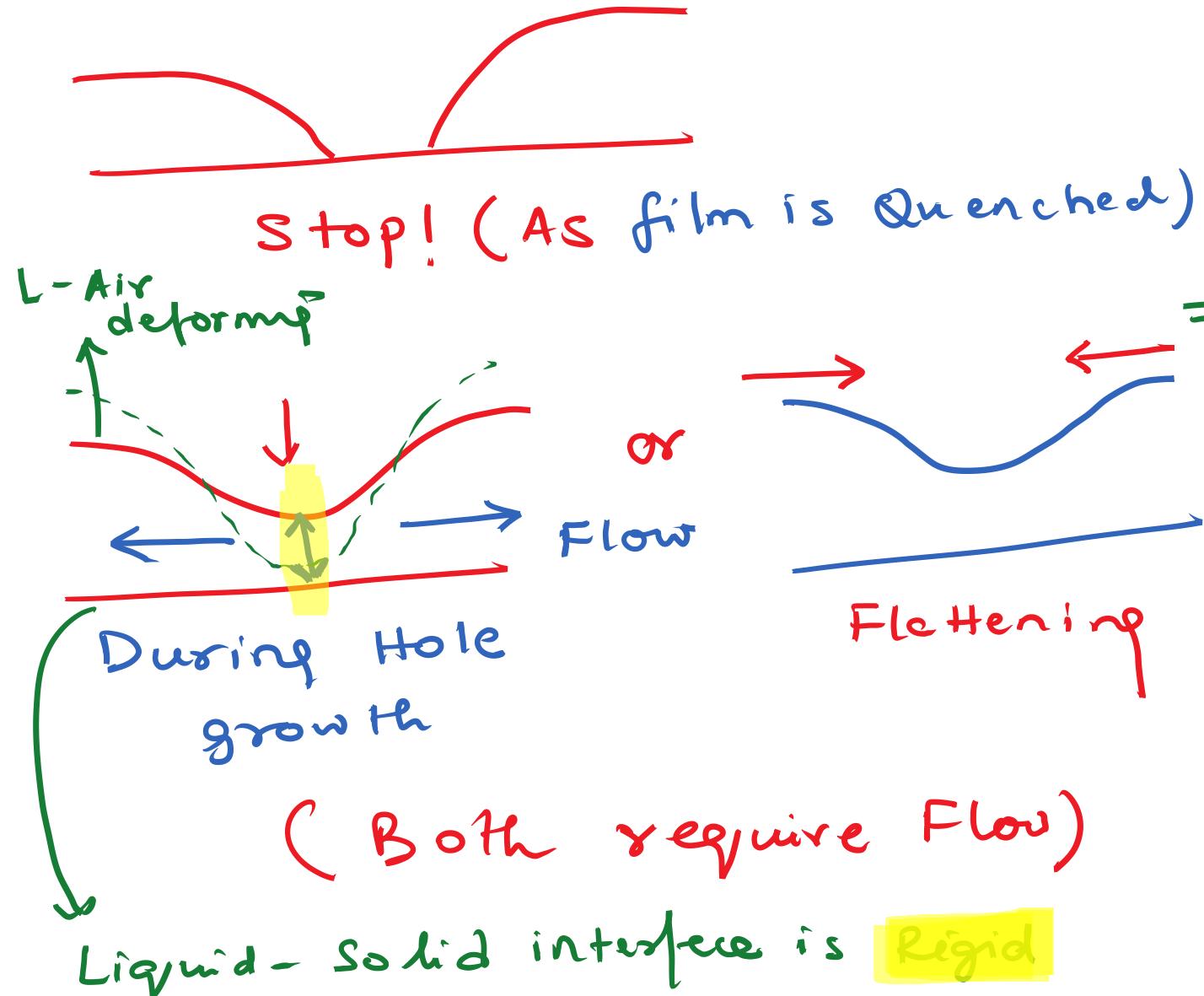
Conclusions from the video:

- (1) Film is unstable.
- (2) Film must be at $T > T_g$.

If complete dewetting is allowed → Film evolves \xrightarrow{RT} to droplets.

Can you control the extent upto which the film has evolved / dewetted? → By controlling the Temp / Quenching the Film → Arrest the evolution.





| Laplace Pr. mediated flattening and

| Π mediated growth

| both require Flow.

| Now if you bring

| $T < T_G \rightarrow$

| $\mu \rightarrow$ shoots up.

| (Flow gets stopped)

Instability in a Thin Polymer Bilayer :-



⇒ The system must be heated above T_G of both polymers.

⇒ How does one create such a sample?

Sequential Spin Coating -

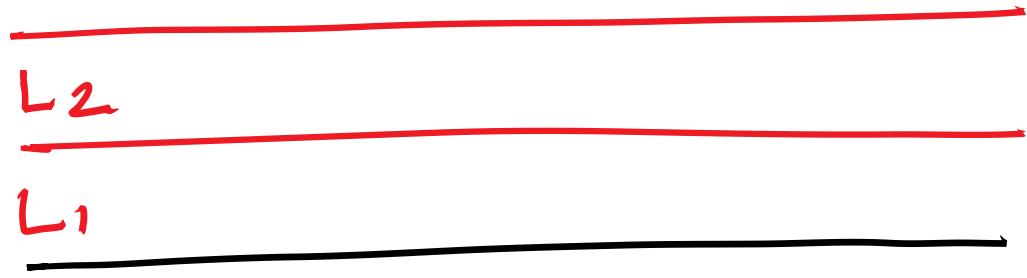
→ Coat L₁ → Allow all solvent to evaporate →
Then coat L₂ → allow all solvent to evaporate.

Any restriction during second spin coating operation?

↳ in terms of choice of solvent used for spin coating!!
mutually exclusive Solvent

Solvent used for Coating L_2 should not be a solvent for Layer L_1 .

In case solvent for L_2 layer can dissolve the polymer used for L_1 Layer \rightarrow then there will be insitu dissolution of film L_1 during the second spin Coating operation.



* L_1, L_2 are both thin
 $(L_1 + L_2) < 100 \text{ nm}$

$T > T_{G1}, T_{G2} \rightarrow$
 Stacked liquids.

Evolution Sequence:-

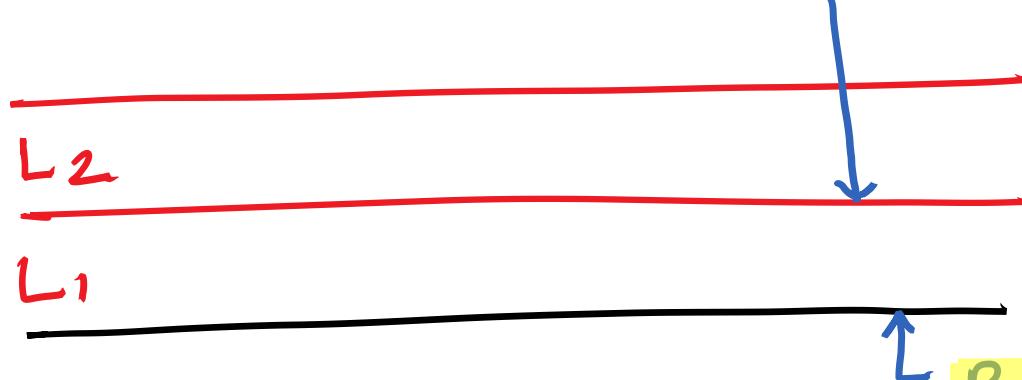
Two possible different situations depending on which Layer becomes unstable First.

Parameters that can decide which Layer will become unstable first

(1) Relative value of their μ (μ_1, μ_2)

(2) Their thickness \rightarrow vis-a-vis - $\overline{\Pi}_1, \overline{\Pi}_2$

(L-L) Deformable interface



* L_1, L_2 are both thin

$$(L_1 + L_2) < 100 \text{ nm}$$

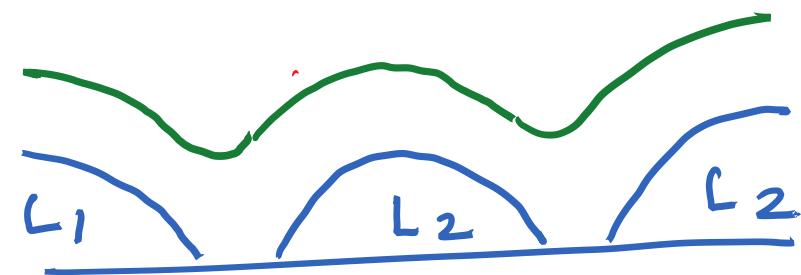
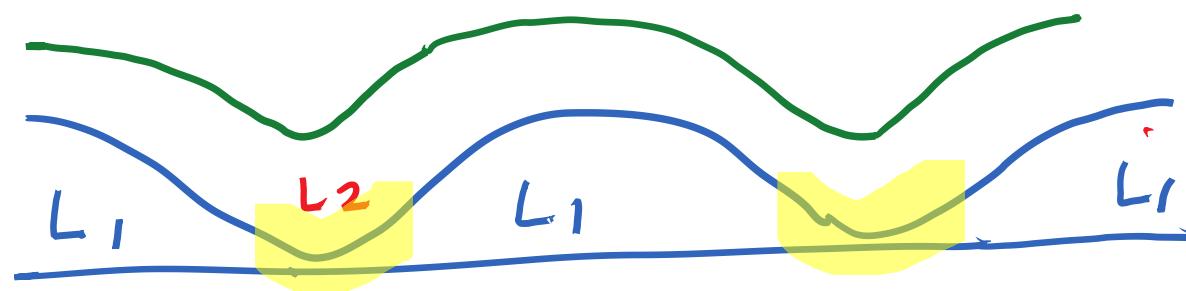
$T > T_{G1}, T_{G2} \rightarrow$

Stacked liquids.

Evolution Sequence:-

Rigid (S-L interface)

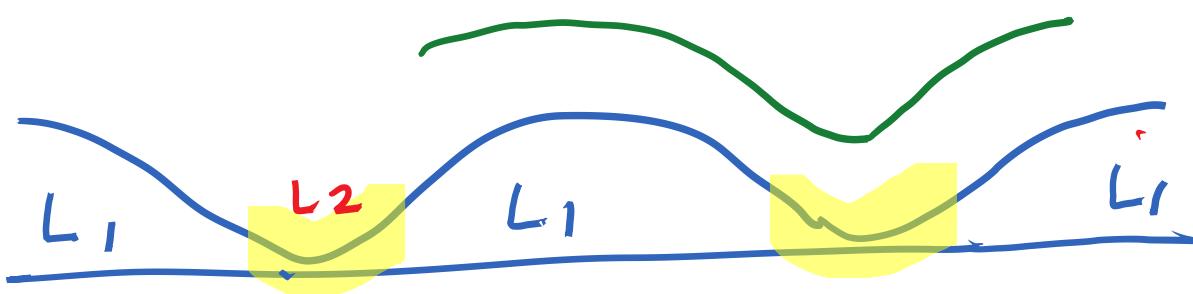
Case 1: Bottom layer becomes unstable first



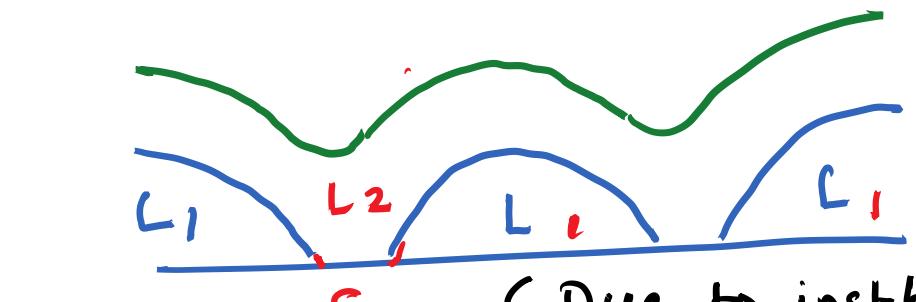
* Bottom layer ruptures first

* Top layer deforms in phase with
the bottom layer

Bottom layer becomes unstable first



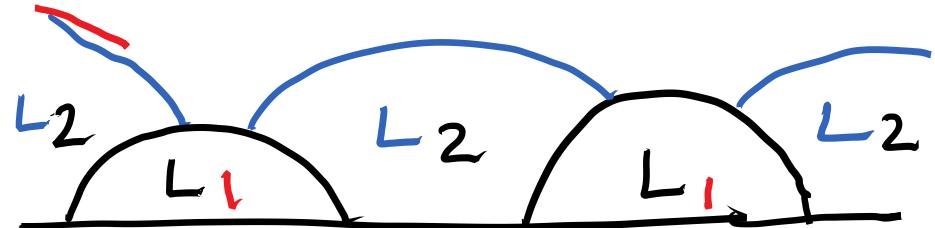
Can be the final configuration



* Bottom layer ruptures first

* Top layer deforms in phase with the bottom layer

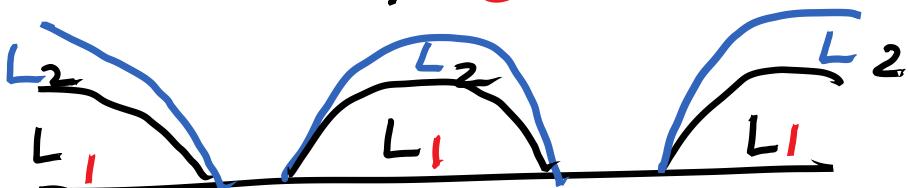
Case A: The intact top layer has ruptured over the L₂ droplets



L₂ is unstable over L₁

Case B:

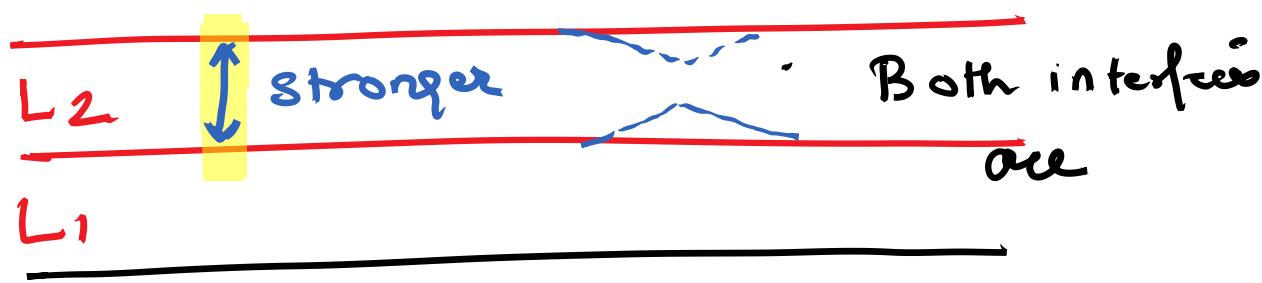
The intact L₂ unstable over the solid surface.



If any of the Disjoining Pressure → S-L₂ interaction { L₁-L₂ interaction } will lead to instability →

Core Shell droplets

Interaction is between two deformable interfaces -



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 $(L_1 + L_2) < 100 \text{ nm}$

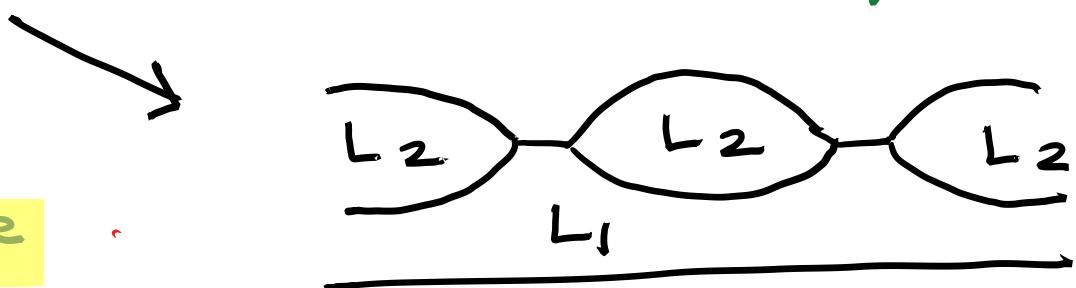
$T > T_{G1}, T_{G2} \rightarrow$
Stacked liquids.

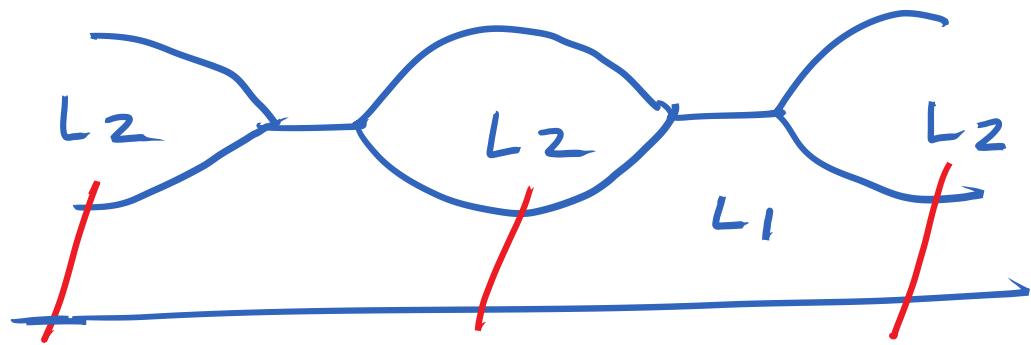
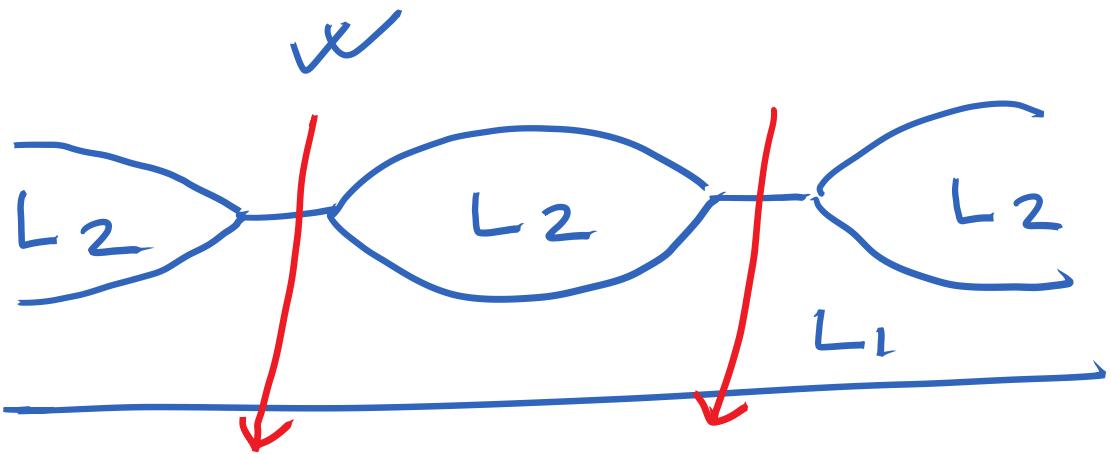
Evolution Sequence:-

Case 2: Top layer becomes unstable first:

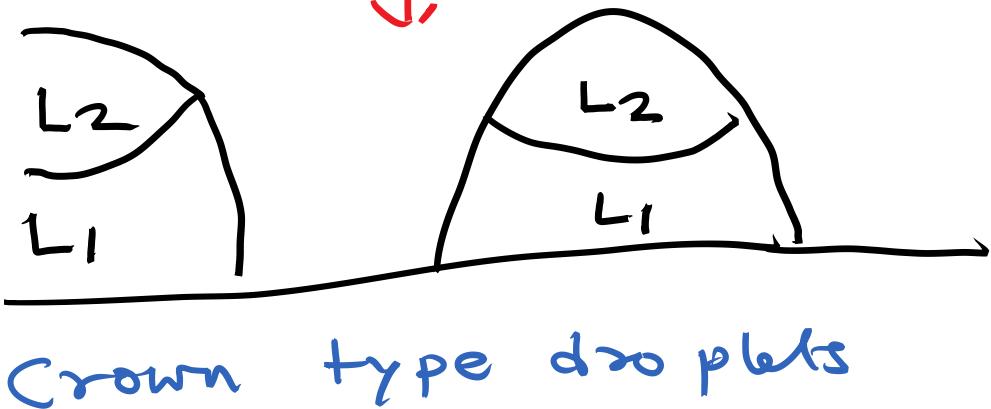
The instability is manifesting between two deformable Interfaces

- * Top layer ruptures first
- * Bottom layer deforms out of phase wrt the top layer

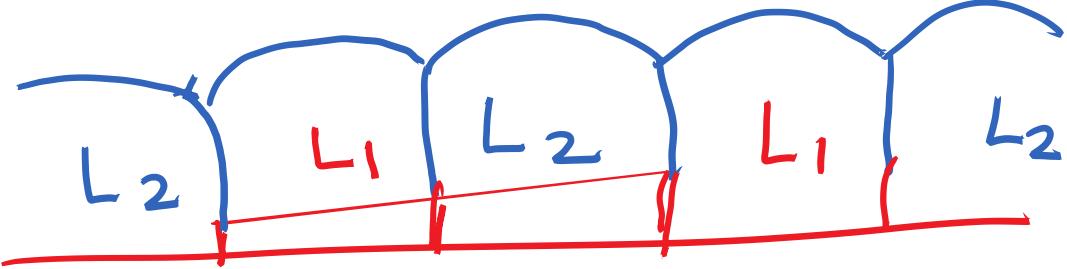




Either the



Crown type droplets



Laterally co-existing
Structures.