

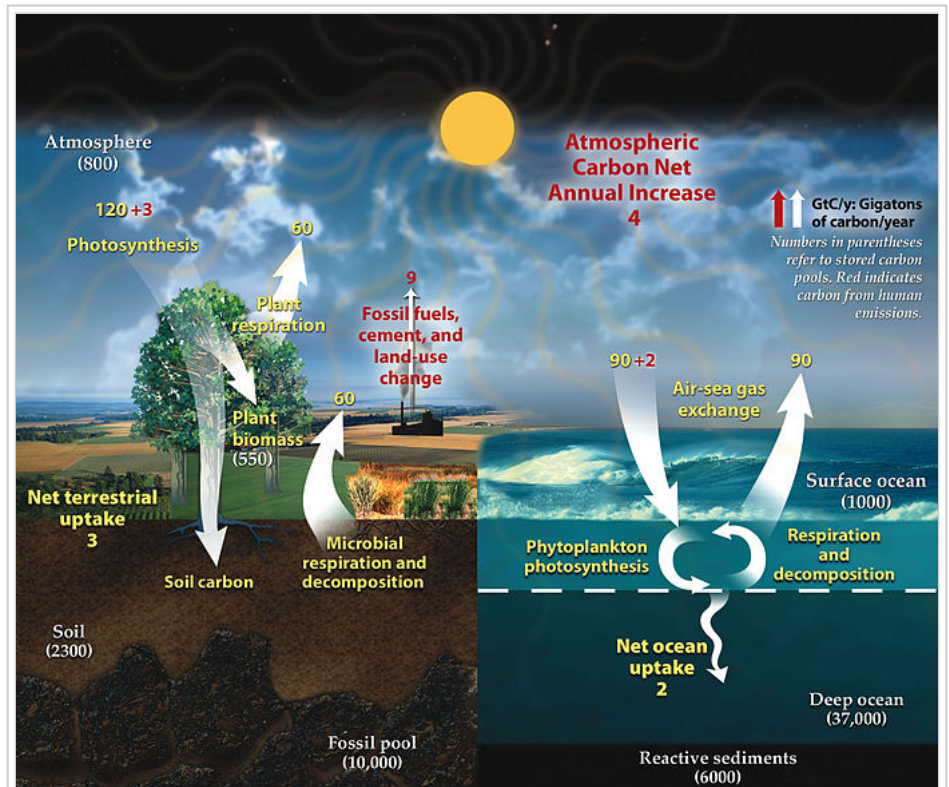
# Carbon cycle

From Wikipedia, the free encyclopedia

The **carbon cycle** is the biogeochemical cycle by which carbon is exchanged among the biosphere, pedosphere, geosphere, hydrosphere, and atmosphere of the Earth. Along with the nitrogen cycle and the water cycle, the carbon cycle comprises a sequence of events that are key to make the Earth capable of sustaining life; it describes the movement of carbon as it is recycled and reused throughout the biosphere, including carbon sinks.

The *global carbon budget* is the balance of the exchanges (incomes and losses) of carbon between the carbon reservoirs or between one specific loop (e.g., atmosphere <-> biosphere) of the carbon cycle. An examination of the carbon budget of a pool or reservoir can provide information about whether the pool or reservoir is functioning as a source or sink for carbon dioxide. The carbon cycle was initially discovered by Joseph

Priestley and Antoine Lavoisier, and popularized by Humphry Davy.<sup>[1]</sup>



This diagram of the fast carbon cycle shows the movement of carbon between land, atmosphere, and oceans in billions of tons per year. Yellow numbers are natural fluxes, red are human contributions, white indicate stored carbon. Note this diagram does not account for volcanic and tectonic activity, which also sequesters and releases carbon.

## Contents

- 1 Global climate
- 2 Main components
  - 2.1 Atmosphere
  - 2.2 Terrestrial biosphere
  - 2.3 Oceans
  - 2.4 Geological carbon cycle
  - 2.5 Human influence
- 3 See also
- 4 References
- 5 Further reading
- 6 External links

## Global climate

Carbon-based molecules are crucial for life on Earth, because it is the main component of biological compounds. Carbon is also a major component of many minerals. Carbon also exists in various forms in the atmosphere. Carbon dioxide (CO<sub>2</sub>) is partly responsible for the greenhouse effect and is one of the most important human-contributed greenhouse gases.<sup>[2]</sup>

In the past two centuries, human activities have seriously altered the global carbon cycle, most significantly in the atmosphere. Although carbon dioxide levels have changed naturally over the past several thousand years, human emissions of carbon dioxide into the atmosphere exceed natural fluctuations.<sup>[2]</sup> Changes in the amount of atmospheric CO<sub>2</sub> are considerably altering weather patterns and indirectly influencing oceanic chemistry. Current carbon dioxide levels in the atmosphere exceed measurements from the last 420,000 years and levels are rising faster than ever recorded,<sup>[3]</sup> making it of critical importance to better understand how the carbon cycle works and what its effects are on the global climate.<sup>[2]</sup>

## Main components

The global carbon cycle is now usually divided into the following major reservoirs of carbon interconnected by pathways of exchange:

- The atmosphere
- The terrestrial biosphere
- The oceans, including dissolved inorganic carbon and living and non-living marine biota
- The sediments, including fossil fuels, fresh water systems and non-living organic material.
- The Earth's interior, carbon from the Earth's mantle and crust. These carbon stores interact with the other components through geological processes

The carbon exchanges between reservoirs occur as the result of various chemical, physical, geological, and biological processes. The ocean contains the largest active pool of carbon near the surface of the Earth.<sup>[2]</sup> The natural flows of carbon between the atmosphere, ocean, terrestrial ecosystems, and sediments is fairly balanced, so that carbon levels would be roughly stable without human influence.<sup>[4][5]</sup>

### Atmosphere

Carbon in the Earth's atmosphere exists in two main forms: carbon dioxide and methane. Both of these gases absorb and retain heat in the atmosphere and are partially responsible for the greenhouse effect. Methane produces a larger greenhouse effect per volume as compared to carbon dioxide, but it exists in much lower concentrations and is more short-lived than carbon dioxide, making carbon dioxide the more important greenhouse gas of the two.<sup>[6]</sup>

Carbon dioxide leaves the atmosphere through photosynthesis, thus entering the terrestrial and oceanic biospheres. Carbon dioxide also dissolves directly from the atmosphere into bodies of water (oceans, lakes, etc.), as well as dissolving in precipitation as raindrops fall through the atmosphere. When dissolved in water,

Carbon pools in the major reservoirs on earth.<sup>[2]</sup>

| Pool                          | Quantity (gigatons) |
|-------------------------------|---------------------|
| Atmosphere                    | 720                 |
| Oceans (total)                | 38,400              |
| Total inorganic               | 37,400              |
| Total organic                 | 1,000               |
| Surface layer                 | 670                 |
| Deep layer                    | 36,730              |
| Lithosphere                   |                     |
| Sedimentary carbonates        | > 60,000,000        |
| Kerogens                      | 15,000,000          |
| Terrestrial biosphere (total) | 2,000               |
| Living biomass                | 600 - 1,000         |
| Dead biomass                  | 1,200               |
| Aquatic biosphere             | 1 - 2               |
| Fossil fuels (total)          | 4,130               |
| Coal                          | 3,510               |
| Oil                           | 230                 |
| Gas                           | 140                 |
| Other (peat)                  | 250                 |



carbon dioxide reacts with water molecules and forms carbonic acid, which contributes to ocean acidity. It can then be absorbed by rocks through weathering. It also can acidify other surfaces it touches or be washed into the ocean.<sup>[7]</sup>

Human activities over the past two centuries have significantly increased the amount of carbon in the atmosphere, mainly in the form of carbon dioxide, both by **modifying ecosystems' ability to extract** carbon dioxide from the atmosphere and by emitting it directly, e.g., by burning fossil fuels and manufacturing concrete.<sup>[2]</sup>

## Terrestrial biosphere

The **terrestrial biosphere** includes the organic carbon in **all land-living organisms**, both alive and dead, as well as carbon **stored in soils**. About 500 gigatons of carbon are stored above ground in plants and other living organisms,<sup>[4]</sup> while soil holds approximately 1,500 gigatons of carbon.<sup>[8]</sup> Most carbon in the terrestrial biosphere is organic carbon,<sup>[9]</sup> while about a third of soil carbon is stored in inorganic forms, such as calcium carbonate.<sup>[10]</sup> Organic carbon is a major component of all organisms living on earth. Autotrophs extract it from the air in the form of carbon dioxide, converting it into organic carbon, while heterotrophs receive carbon by consuming other organisms.

Because carbon uptake in the terrestrial biosphere is dependent on biotic factors, it follows a diurnal and seasonal cycle. In CO<sub>2</sub> measurements, this feature is apparent in the Keeling curve. It is strongest in the northern hemisphere, because this hemisphere has more land mass than the southern hemisphere and thus more room for ecosystems to absorb and emit carbon.

Carbon leaves the terrestrial biosphere in several ways and on different time scales. The combustion or respiration of organic carbon releases it rapidly into the atmosphere. It can also be exported into the oceans through rivers or remain sequestered in soils in the form of inert carbon. Carbon stored in soil can remain there for up to thousands of years before being washed into rivers by erosion or released into the atmosphere through soil respiration. Between **1989 and 2008 soil respiration increased by about 0.1% per year.**<sup>[11]</sup> In 2008, the global total of CO<sub>2</sub> released from the soil reached roughly 98 billion tonnes, about 10 times more carbon than humans are now putting into the atmosphere each year by burning fossil fuel. There are a few plausible explanations for this trend, but the most likely explanation is that increasing temperatures have increased rates of decomposition of soil organic matter, which has increased the flow of CO<sub>2</sub>. The length of carbon sequestering in soil is dependent on local climatic conditions and thus changes in the course of climate change. From pre-industrial era to 2010, the terrestrial biosphere represented a net source of atmospheric CO<sub>2</sub> prior to 1940, switching subsequently to a net sink.<sup>[12]</sup>

## Oceans



Epiphytes on electric wires. This kind of plant takes both CO<sub>2</sub> and water from the atmosphere for living and growing.



A portable soil respiration system measuring soil CO<sub>2</sub> flux

Oceans contain the greatest quantity of actively cycled carbon in this world and are second only to the lithosphere in the amount of carbon they store.<sup>[2]</sup> The oceans' surface layer holds large amounts of dissolved inorganic carbon that is exchanged rapidly with the atmosphere. The deep layer's concentration of dissolved inorganic carbon (DIC) is about 15% higher than that of the surface layer.<sup>[13]</sup> DIC is stored in the deep layer for much longer periods of time.<sup>[4]</sup> Thermohaline circulation exchanges carbon between these two layers.<sup>[2]</sup>

Carbon enters the ocean mainly through the dissolution of atmospheric carbon dioxide, which is converted into carbonate. It can also enter the oceans through rivers as dissolved organic carbon. It is converted by organisms into organic carbon through photosynthesis and can either be exchanged throughout the food chain or precipitated into the ocean's deeper, more carbon rich layers as dead soft tissue or in shells as calcium carbonate. It circulates in this layer for long periods of time before either being deposited as sediment or, eventually, returned to the surface waters through thermohaline circulation.<sup>[4]</sup>

Oceanic absorption of CO<sub>2</sub> is one of the most important forms of carbon sequestering limiting the human-caused rise of carbon dioxide in the atmosphere. However, this process is limited by a number of factors. Because the rate of CO<sub>2</sub> dissolution in the ocean is dependent on the weathering of rocks and this process takes place slower than current rates of human greenhouse gas emissions, ocean CO<sub>2</sub> uptake will decrease in the future.<sup>[2]</sup> CO<sub>2</sub> absorption also makes water more acidic, which affects ocean biosystems. The projected rate of increasing oceanic acidity could slow the biological precipitation of calcium carbonates, thus decreasing the ocean's capacity to absorb carbon dioxide.<sup>[14][15]</sup>

## Geological carbon cycle

The geologic component of the carbon cycle operates slowly in comparison to the other parts of the global carbon cycle. It is one of the most important determinants of the amount of carbon in the atmosphere, and thus of global temperatures.<sup>[16]</sup>

Most of the earth's carbon is stored inertly in the earth's lithosphere.<sup>[2]</sup> Much of the carbon stored in the earth's mantle was stored there when the earth formed.<sup>[17]</sup> Some of it was deposited in the form of organic carbon from the biosphere.<sup>[18]</sup> Of the carbon stored in the geosphere, about 80% is limestone and its derivatives, which form from the sedimentation of calcium carbonate stored in the shells of marine organisms. The remaining 20% is stored as kerogens formed through the sedimentation and burial of terrestrial organisms under high heat and pressure. Organic carbon stored in the geosphere can remain there for millions of years.<sup>[16]</sup>

Carbon can leave the geosphere in several ways. Carbon dioxide is released during the metamorphosis of carbonate rocks when they are subducted into the earth's mantle. This carbon dioxide can be released into the atmosphere and ocean through volcanoes and hotspots.<sup>[17]</sup> It can also be removed by humans through the direct extraction of kerogens in the form of fossil fuels. After extraction, fossil fuels are burned to release energy, thus emitting the carbon they store into the atmosphere.

## Human influence

Since the industrial revolution, human activity has modified the carbon cycle by changing its components' functions and directly adding carbon to the atmosphere.<sup>[2]</sup>

The largest human impact on the carbon cycle is through direct emissions from burning fossil fuels, which transfers carbon from the geosphere into the atmosphere. The rest of this increase is caused mostly by changes in land-use, particularly deforestation.

Another direct human impact on the carbon cycle is the chemical process of calcination of limestone for clinker production, which releases CO<sub>2</sub>.<sup>[23]</sup> Clinker is an industrial precursor of cement.

Humans also influence the carbon cycle indirectly by changing the terrestrial and oceanic biosphere. Over the past several centuries, direct and indirect human-caused land use and land cover change (LUCC) has led to the loss of biodiversity, which lowers ecosystems' resilience to environmental stresses and decreases their ability to remove carbon from the atmosphere. More directly, it often leads to the release of carbon from terrestrial ecosystems into the atmosphere. Deforestation for agricultural purposes removes forests, which hold large amounts of carbon, and replaces them, generally with agricultural or urban areas. Both of these replacement land cover types store comparatively small amounts of carbon, so that the net product of the process is that more carbon stays in the atmosphere.

Other human-caused changes to the environment change ecosystems' productivity and their ability to remove carbon from the atmosphere. Air pollution, for example, damages plants and soils, while many agricultural and land use practices lead to higher erosion rates, washing carbon out of soils and decreasing plant productivity.

Humans also affect the oceanic carbon cycle. Current trends in climate change lead to higher ocean temperatures, thus modifying ecosystems. Also, acid rain and polluted runoff from agriculture and industry change the ocean's chemical composition. Such changes can have dramatic effects on highly sensitive ecosystems such as coral reefs, thus limiting the ocean's ability to absorb carbon from the atmosphere on a regional scale and reducing oceanic biodiversity globally.

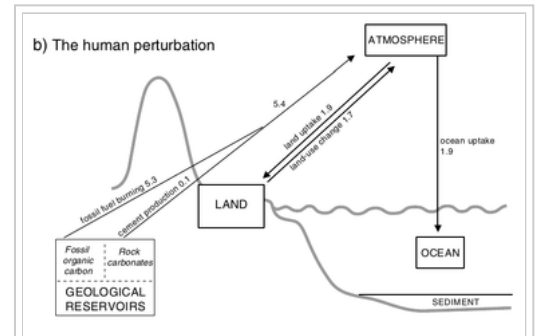
Arctic methane emissions indirectly caused by anthropogenic global warming also affect the carbon cycle, and contribute to further warming in what is known as climate change feedback.

On 12 November 2015, NASA scientists reported that human-made carbon dioxide (CO<sub>2</sub>) continues to increase above levels not seen in hundreds of thousands of years: currently, about half of the carbon dioxide released from the burning of fossil fuels remains in the atmosphere and is not absorbed by vegetation and the oceans.<sup>[19][20][21][22]</sup>

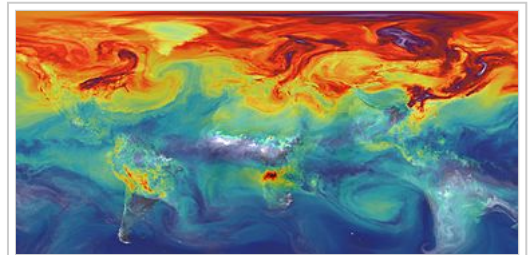
## See also

- Biochar
- Calvin cycle
- Carbon cycle re-balancing
- Carbon dioxide in Earth's atmosphere
- Carbon footprint
- Deficit irrigation
- Nitrogen cycle
- Ocean acidification
- Permafrost carbon cycle
- Snowball Earth and the "Slow carbon cycle"
- Soil plant atmosphere continuum
- Soil regeneration

## References



Human activity since the industrial era has changed the balance in the natural carbon cycle. Units are in gigatons.<sup>[4]</sup>



CO<sub>2</sub> in Earth's atmosphere if *half* of global-warming emissions are *not* absorbed.<sup>[19][20][21][22]</sup> (NASA computer simulation).

1. Holmes, Richard. "The Age Of Wonder", Pantheon Books, 2008. ISBN 978-0-375-42222-5.
2. Falkowski, P.; Scholes, R. J.; Boyle, E.; Canadell, J.; Canfield, D.; Elser, J.; Gruber, N.; Hibbard, K.; Höglberg, P.; Linder, S.; MacKenzie, F. T.; Moore b, 3.; Pedersen, T.; Rosenthal, Y.; Seitzinger, S.; Smetacek, V.; Steffen, W. (2000). "The Global Carbon Cycle: A Test of Our Knowledge of Earth as a System". *Science*. **290** (5490): 291–296. Bibcode:2000Sci...290..291F. doi:10.1126/science.290.5490.291. PMID 11030643.
3. Crowley, T. J. (2000). "Causes of Climate Change Over the Past 1000 Years". *Science*. **289** (5477): 270–277. Bibcode:2000Sci...289..270C. doi:10.1126/science.289.5477.270. ISSN 0036-8075. PMID 10894770.
4. Prentice, I.C. (2001). "The carbon cycle and atmospheric carbon dioxide". Climate change 2001: the scientific basis: contribution of Working Group I to the Third Assessment Report of the Intergovernmental Panel on Climate Change / Houghton, J.T. [edit.] Retrieved 31 May 2012.
5. "An Introduction to the Global Carbon Cycle" (PDF). University of New Hampshire. 2009. Retrieved 6 February 2016.
6. Forster, P.; Ramawamy, V.; Artaxo, P.; Berntsen, T.; Betts, R.; Fahey, D.W.; Haywood, J.; Lean, J.; Lowe, D.C.; Myhre, G.; Nganga, J.; Prinn, R.; Raga, G.; Schulz, M.; Van Dorland, R. (2007). "Changes in atmospheric constituents and in radiative forcing". *Climate Change 2007: the Physical Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*.
7. Planet, The Habitable. "Carbon Cycling and Earth's Climate". *Many Planets, One Earth*. **4**. Retrieved 2012-06-24.
8. Charles W. Rice Carbon in Soil: Why and How? ([http://www.agiweb.org/geotimes/jan02/feature\\_carbon.html](http://www.agiweb.org/geotimes/jan02/feature_carbon.html) Storing) Geotimes (January 2002). American Geological Institute
9. Balal Yousaf, Guijian Liu, Ruwei Wang, Qumber Abbas, Muhammad Imtiaz, Ruijia Liu: Investigating the biochar effects on C-mineralization and sequestration of carbon in soil compared with conventional amendments using stable isotope ( $\delta^{13}\text{C}$ ) approach. GCB Bioenergy 2016; DOI: 10.1111/gcbb.12401
10. Lal, Rattan (2008). "Sequestration of atmospheric CO<sub>2</sub> in global carbon pools". *Energy and Environmental Science*. **1**: 86–100. doi:10.1039/b809492f.
11. Bond-Lamberty, B. & Thomson, A.[1] (<http://www.nature.com/nature/journal/v464/n7288/full/nature08930.html>) Nature 464, 579-582 (2010)
12. Junling Huang and Michael B. McElroy (2012). "The Contemporary and Historical Budget of Atmospheric CO<sub>2</sub>" (PDF). *Canadian Journal of Physics*. **90** (8): 707–716. Bibcode:2012CaJPh..90..707H. doi:10.1139/p2012-033.
13. Sarmiento, J.L.; Gruber, N. (2006). *Ocean Biogeochemical Dynamics*. Princeton University Press, Princeton, New Jersey, USA.
14. Kleypas, J. A.; Buddemeier, R. W.; Archer, D.; Gattuso, J. P.; Langdon, C.; Opdyke, B. N. (1999). "Geochemical Consequences of Increased Atmospheric Carbon Dioxide on Coral Reefs". *Science*. **284** (5411): 118–120. Bibcode:1999Sci...284..118K. doi:10.1126/science.284.5411.118. PMID 10102806.
15. Langdon, C.; Takahashi, T.; Sweeney, C.; Chipman, D.; Goddard, J.; Marubini, F.; Aceves, H.; Barnett, H.; Atkinson, M. J. (2000). "Effect of calcium carbonate saturation state on the calcification rate of an experimental coral reef". *Global Biogeochemical Cycles*. **14** (2): 639. Bibcode:2000GBioC..14..639L. doi:10.1029/1999GB001195.
16. NASA. "The Slow Carbon Cycle". Retrieved 2012-06-24
17. The Carbon Cycle and Earth's Climate (<http://www.columbia.edu/~vjd1/carbon.htm>) Information sheet for Columbia University Summer Session 2012 Earth and Environmental Sciences Introduction to Earth Sciences I
18. A New Look at the Long-term Carbon Cycle ([http://www.ic.ucsc.edu/~mdmccar/ocea213/readings/02\\_C\\_cycle\\_Long\\_T/Berner\\_1999\\_GSAToday\\_A\\_new\\_look\\_Long\\_Term\\_C\\_cycle.pdf](http://www.ic.ucsc.edu/~mdmccar/ocea213/readings/02_C_cycle_Long_T/Berner_1999_GSAToday_A_new_look_Long_Term_C_cycle.pdf)) Vol. 9, No. 11 November 1999 GSA TODAY A Publication of the Geological Society of America
19. Buis, Alan; Ramsayer, Kate; Rasmussen, Carol (12 November 2015). "A Breathing Planet, Off Balance". NASA. Retrieved 13 November 2015.
20. Staff (12 November 2015). "Audio (66:01) - NASA News Conference - Carbon & Climate Telecon". NASA. Retrieved 12 November 2015.
21. St. Fleur, Nicholas (10 November 2015). "Atmospheric Greenhouse Gas Levels Hit Record, Report Says". *New York Times*. Retrieved 11 November 2015.
22. Ritter, Karl (9 November 2015). "UK: In 1st, global temps average could be 1 degree C higher". *AP News*. Retrieved 11 November 2015.
23. IPCC (2007) AR4. Climate Change 2007: Working Group III: Mitigation of Climate Change, section 7.4.5.1. [https://www.ipcc.ch/publications\\_and\\_data/ar4/wg3/en/4-5.html](https://www.ipcc.ch/publications_and_data/ar4/wg3/en/4-5.html)

## Further reading

- The Carbon Cycle (<http://earthobservatory.nasa.gov/Features/CarbonCycle/?src=eo-features>), updated primer by NASA Earth Observatory, 2011



- Appenzeller, Tim (2004). "The case of the missing carbon". *National Geographic Magazine*. – article about the missing carbon sink
- Bolin, Bert; Degens, E. T.; Kempe, S.; Ketner, P. (1979). *The global carbon cycle*. Chichester ; New York: Published on behalf of the Scientific Committee on Problems of the Environment (SCOPE) of the International Council of Scientific Unions (ICSU) by Wiley. ISBN 0-471-99710-2. Retrieved 2008-07-08.
- Houghton, R. A. (2005). "The contemporary carbon cycle". In William H Schlesinger (editor). *Biogeochemistry*. Amsterdam: Elsevier Science. pp. 473–513. ISBN 0-08-044642-6.
- Janzen, H. H. (2004). "Carbon cycling in earth systems—a soil science perspective". *Agriculture, Ecosystems & Environment*. **104** (3): 399–417. doi:10.1016/j.agee.2004.01.040.
- Millero, Frank J. (2005). *Chemical Oceanography* (3 ed.). CRC Press. ISBN 0-8493-2280-4.
- Sundquist, Eric; Broecker, Wallace S., eds. (1985). *The Carbon Cycle and Atmospheric CO<sub>2</sub>: Natural variations Archean to Present*. Geophysical Monographs Series. American Geophysical Union.

## External links

- Carbon Cycle Science Program (<https://web.archive.org/web/20040602163557/http://www.carboncyclescience.gov:80/>) – an interagency partnership.
- NOAA's Carbon Cycle Greenhouse Gases Group (<http://www.esrl.noaa.gov/gmd/ccgg/index.html>)
- Global Carbon Project – initiative of the Earth System Science Partnership (<http://www.globalcarbonproject.org/>)
- UNEP – The present carbon cycle – Climate Change (<http://www.grida.no/climate/vital/13.htm>) carbon levels and flows
- NASA's Orbiting Carbon Observatory (<http://oco.jpl.nasa.gov/>)
- CarboSchools (<http://www.carboschools.org/>), a European website with many resources to study carbon cycle in secondary schools.
- Carbon and Climate (<http://carboncycle.aos.wisc.edu/>), an educational website with a carbon cycle applet for modeling your own projection.



Wikimedia Commons has media related to **carbon cycle**.

Retrieved from "https://en.wikipedia.org/w/index.php?title=Carbon\_cycle&oldid=758769337"

Categories: Geochemistry | Chemical oceanography | Photosynthesis | Soil biology | Soil chemistry | Carbon | Numerical climate and weather models | Biogeography | Biogeochemical cycle

- This page was last modified on 7 January 2017, at 13:49.
- Text is available under the Creative Commons Attribution-ShareAlike License; additional terms may apply. By using this site, you agree to the Terms of Use and Privacy Policy. Wikipedia® is a registered trademark of the Wikimedia Foundation, Inc., a non-profit organization.