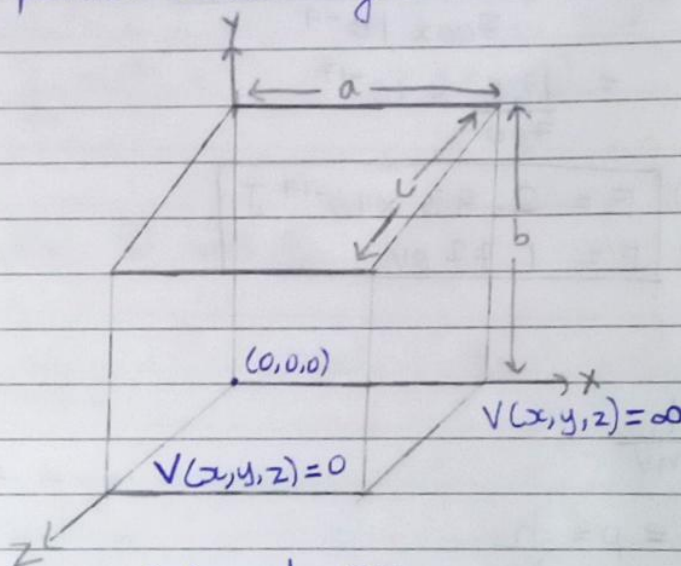


BPHY101L DA2-A⇒ PARTICLE IN A 3D BOX:

In quantum mechanics, the particle in a box model describes a particle free to move in a small space surrounded by impenetrable barriers.

The quantum particle in the 1D box can be expanded to consider a particle within higher dimensions such as a 3D box.



Schematic Diagram

→ Energy State:

Consider a particle which can move freely within rectangular box of dimensions $a \times b \times c$ with impenetrable walls.

- The potential can be written as:

$$V = \begin{cases} 0 & \text{Inside} \\ \infty & \text{At surfaces and outside} \end{cases}$$

- There is no force (i.e., no potential) acting on the particles inside the box $\Rightarrow V(\vec{r}) = 0$

$$- \vec{r} = a\hat{x} + b\hat{y} + c\hat{z}$$

- When the potential energy is infinite, then the wavefunction equals zero. When the potential energy is zero, then the wavefunction obeys the Time-Independent Schrödinger Equation:

$$-\frac{\hbar^2}{2m} \nabla^2 \psi(r) + V(r)\psi(r) = E\psi(r)$$

$$E\psi(x,y,z) = -\frac{\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2} \right] \psi(x,y,z) + V\psi(x,y,z) \quad - (1)$$

Using separation of variables to solve this partial differential equation

$$\psi(x,y,z) = X(x)Y(y)Z(z)$$

Solving,

$$\psi(x,y,z) = \sqrt{\frac{8}{abc}} \sin\left(\frac{n_x \pi x}{a}\right) \sin\left(\frac{n_y \pi y}{b}\right) \sin\left(\frac{n_z \pi z}{c}\right)$$

$$E_x + E_y + E_z = E = \frac{\hbar^2 \pi^2}{2m} \left[\frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} + \frac{n_z^2}{c^2} \right]$$



Energy State in a 3D Box