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# **BOARD OF INTERMEDIATE EDUCATION**

### JUNIOR INTER MODEL PAPER – 2020

#### **CHEMISTRY**

Max. Marks: 60 **TIME: 3 Hours** (English Medium)

#### **SECTION - A**

- i) Very short answer type questions.
- ii) Answer all the questions.
- iii) Each question carries two marks.

 $10 \times 2 = 20$ 

- 360 cm<sup>3</sup> of CH<sub>4</sub> gas diffused through a porus membrane in 15 minutes. Under similar conditions, 120 1. cm<sup>3</sup> of another gas diffused in 10 minutes. Find the molar mass of the gas.
- Calculate the oxidation number of Oxygen element in the given species. 2.
  - a) CaO<sub>2</sub>
- b) <u>O</u>F<sub>2</sub>
- When mixed with water, real cement will set immediatly, How to overcome this problem? 3.
- 4. What is 'sink'? Give two examples.
- 5. Name one use each of
  - a) Quartz
- b) Silica gel
- Write the biological importance of Ca+2, Mg+2 ions. 6.
- kibha. net 7. Define 'equilibrium constant'? Will it change with temperature?
- 8. Graphite function as lubricant. Why?
- What is 'COD'? 9.
- Write the functional isomers of the organic compound C<sub>3</sub>H<sub>6</sub>O. 10.

#### **SECTION - B**

- i) Short answer type questions.
- ii) Answer any six questions.
- iii) Each question carries 4 marks.

 $6 \times 4 = 24$ 

- a) Boyle's law b) Charles' law from Kinetic gas equation.
- Balance the following redox reaction by ion-electron method in basic medium.

$$MnO_4^-(aq.) + I^-(aq.) \to MnO_2(s) + I_2(s)$$

- 13. Explain hybridization of phosphorus in the formation of  $PCl_5$ .
- **14.** Define a) Enthalpy
- b) Entropy
- 15. Explain Lewis acid-base theory with suitable examples.
- **16.** Explain the following with one example each.
  - a) Electron deficient hydrides
- b) Electron rich hydrides
- c) Electron pricise hydrides
- d) Ionic hydrides

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- 17. a) What is 'inert pair effect'?
  - b) Give the formula, structure, common name of borazine.
- **18.** Explain the formation of coordinate covalent bond with example.

#### **SECTION - C**

- i) Long answer type questions.
- ii) Answer any two questions.
- iii) Each question carries 8 marks.

- a) Explain the difference between emission spectra and absorption spectra.
  - b) Discuss the importance of Bohr's model atom to explain various series of line spectra in Hydrogen
- 20. Define Electron gain enthalpy and Electronegativity. How these properties varies in groups and periods.
- 21. Give two methods of preparation of ethylene. How does it react with
  - a)  $Br_2/CCl_4$
- b) O<sub>3</sub>
- NNN. ek c) HBr
  - d) Cold, dilute alkaline KMnO<sub>4</sub>

-ANS Sankara Rao

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