

Energy Storage: Batteries

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Battery capacity

3 unities:

A,

Current: I

Ah or mAh

Charge capacity: Q

$$Q = I \cdot t$$

$$[A \cdot s]$$

$$[A \cdot h]$$

Wh

Energy capacity: E

$$E = P \cdot t$$

P – power

$$E = V \cdot I \cdot t [V \cdot A \cdot h]$$

V – voltage

$$E = V \cdot Q [V \cdot A \cdot h]$$

Units:

$$[V] = \left[\frac{J}{C}\right] = \left[\frac{W}{A}\right] \Rightarrow [W] = [V \cdot A]$$

$$[C] = [A \cdot s]$$

How much electric charge can be stored in battery, for example:

Battery has 10 Ah, means → battery can deliver 1 A for 10h,

or 2A for 5h

or 10A for 1h

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Specific capacity , energy density, power density

Ah/kg or mAh/g



Specific capacity: Q
 $Q = I * t / m$
 [A * h/kg]
 [mA * h/g]

Wh/kg



Energy density: E
 $E = P * t / m$
 $E = V * I * t / m$ [V * A * h/kg]
 $E = V * Q / m$ [Wh/kg]

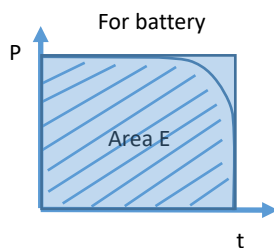
W/kg



Power density: P
 P / m
 $V * I / m$ [V * A /kg]
 [W/kg]

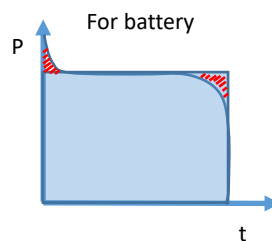
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Discharge curve

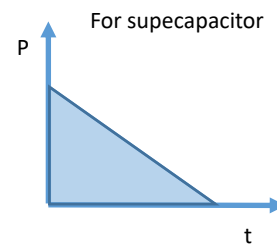


$$E = P * t$$

$$E = V * I * t$$



$$E = P * t$$

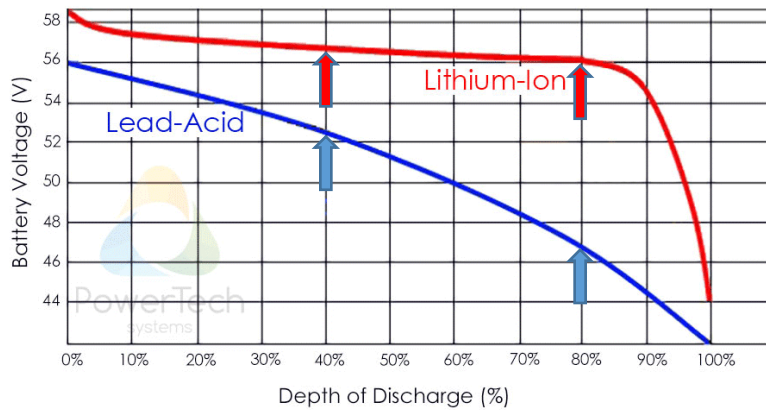


$$E = \frac{1}{2} P * t$$

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Examples

Discharge curve : Lithium-Ion vs Lead Acid



Potential difference between 40 and 80 % of DoD (Depth of Discharge):

- around 6 V for Pb-acid
- around 0.5 V for LiB



- good estimation of SoC (State of Charge) for Pb-acid
- for LiB very difficult due to flat discharge curve (the flattest one for Li iron phosphate technology)

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Problem 1

A 12 V battery has a charge capacity of 30 Ah:

- How long can this battery deliver an average current of 5 A?
- What is the energy capacity of this battery?

a)

$$\begin{aligned}
 Q &= I \cdot t \\
 \Rightarrow t &= Q / I \\
 \Rightarrow t &= 30 \text{ Ah} / 5 \text{ A} \\
 \Rightarrow t &= 6 \text{ h}
 \end{aligned}$$

b)

$$\begin{aligned}
 E &= V \cdot I \cdot t \\
 \Rightarrow E &= V \cdot Q \\
 \Rightarrow E &= 30 \text{ Ah} \cdot 12 \text{ V} & [W = V \cdot A] \\
 \Rightarrow E &= 360 \text{ Wh}
 \end{aligned}$$

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Problem 2

6 V battery has energy capacity of 300 Wh.

- What is the charge capacity of this battery?
- What is current that this battery can deliver if it's used continuously for 25 hours?

a)

$$\begin{aligned}
 E &= Q \cdot V \\
 \Rightarrow Q &= E / V \\
 \Rightarrow Q &= 300 \text{ [V} \cdot \text{A} \cdot \text{h]} / 6 \text{V} \\
 \Rightarrow Q &= 50 \text{ Ah}
 \end{aligned}$$

b)

$$\begin{aligned}
 Q &= I \cdot t \\
 \Rightarrow I &= Q / t \\
 \Rightarrow I &= 50 \text{ [Ah]} / 25 \text{h} \\
 \Rightarrow I &= 2 \text{ A}
 \end{aligned}$$

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Problem 3

A 1.2 V Ni-MeH battery has a charge capacity of 300 mAh. The battery is connected to a device with a load resistance of 200 Ohms.

- How much current is flowing in the circuit?
- Estimate how long this battery can sustain this current?
- What is the energy capacity of this battery?

a)

$$\begin{aligned}
 &\text{Ohm law} \\
 V &= I \cdot R \\
 \Rightarrow I &= V / R \\
 \Rightarrow I &= 1.2 \text{ [V]} / 200 \text{ [Ohm]} \\
 \Rightarrow I &= 0.006 \text{ [A]} = 6 \text{ mA}
 \end{aligned}$$

b)

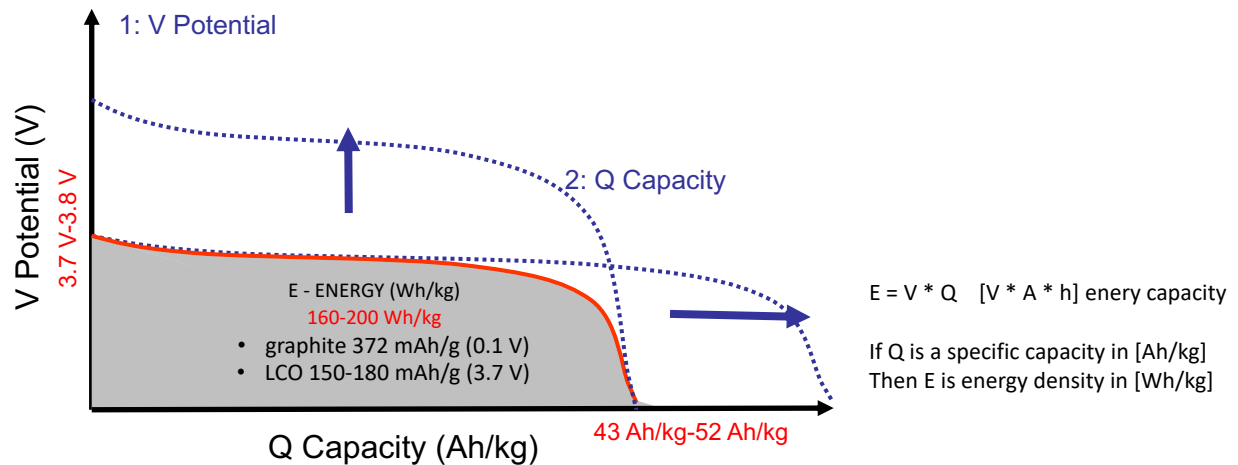
$$\begin{aligned}
 Q &= I \cdot t \\
 \Rightarrow t &= Q / I \\
 \Rightarrow t &= 300 \text{ [mAh]} / 6 \text{ [mA]} \\
 \Rightarrow t &= 50 \text{ h}
 \end{aligned}$$

c)

$$\begin{aligned}
 E &= Q \cdot V \\
 E &= 0.3 \text{ [Ah]} \cdot 1.2 \text{ [V]} \quad [W = V \cdot A] \\
 E &= 0.36 \text{ Wh}
 \end{aligned}$$

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Discharge curve for LiB – Energy density



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Energy density of LiB = $V * Q$

1: **V Potential** $V_{\text{cathode}} (3.8 \text{ [V]}) - V_{\text{anode}} (0.1 \text{ [V]}) = 3.7 \text{ V}$

2: **Q Capacity**

$Q_{\text{anode}} = 372 \text{ [mAh/g]}, 1/Q_a = 1000/372 = 2.7 \text{ [g/Ah]}$

$Q_{\text{cathode}} = 170 \text{ [mAh/g]}, 1/Q_c = 1000/170 = 5.9 \text{ [g/Ah]}$

$1/Q_a + 1/Q_c = 2.7 + 5.9 = 8.6 \text{ [g/Ah]}$

$Q_{\text{total}} = 1/8.6 * 1000 = 116 \text{ [mAh/g]}$

But it should be divided by 2, which is the weight of Al (for cathode) and Cu (for anode) current collectors, so

$Q_{\text{total}} = 116/2 = 58 \text{ [mAh/g]}$

3: **E Energy density**

$E = V * Q$

$E = 3.7 \text{ [V]} * 58 \text{ [mAh/g]} = 214.6 \text{ [Wh/kg]}$

Theoretical capacity of electrode materials:

$$Q_{\text{th}} = \frac{n F * 1000}{M * 3600} \text{ (mAh/g ou Ah/kg)}$$

$$E_{\text{bat}} = \frac{V_{\text{cathode}} - V_{\text{anode}}}{1/Q_{\text{cathode}} + 1/Q_{\text{anode}}} \text{ (Wh/kg)}$$

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