

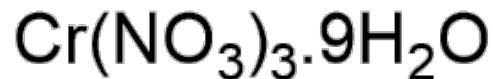
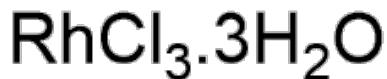
# EA-MEJ-357: Preparation of RhCl<sub>3</sub> and Cr(NO<sub>3</sub>)<sub>3</sub> stock solutions

Date: 2025-10-09  
Tags: Stocksolution MEJ EA RhCl<sub>3</sub>\*3H<sub>2</sub>O  
Cr(NO<sub>3</sub>)<sub>3</sub>\*9H<sub>2</sub>O Solution  
Category: Prep work  
Status: Done  
Created by: Ebrahim Abedini

## Objective

Preparation of stock solutions of RhCl<sub>3</sub> and Cr(NO<sub>3</sub>)<sub>3</sub> salts.

## Reaction scheme/sample structure



9.874 mM

28.809 mM

ChemDraw File (linked): [EA-357.cdx](#)

## Literature/reference experiments

Literature	<a href="https://doi.org/10.1039/C9EE00310">https://doi.org/10.1039/C9EE00310</a>
Reproduction	/
Similar experiments	Prep work - NB-275: Preparation of RhCl <sub>3</sub> and Cr(NO <sub>3</sub> ) <sub>3</sub> solutions

## Reagents

Name	CAS Number / Experiment Number	Inventory number	Amount [mmol]	Equivale nts	Mass <sub>theo</sub> [mg]	Mass <sub>exp</sub> [mg]	Molar mass [g/mol]	Density (g/ml)	Volume [ml]	Concentrat ion [mM]
RhCl <sub>3</sub> .3H <sub>2</sub> O, BLD Pharmatech GmbH	13569-65-8	C138335	/	/	12.8	13	263.31	/	/	9.874
Cr(NO <sub>3</sub> ) <sub>3</sub> .9H <sub>2</sub> O	7789-02-8	/	/	/	57.62	57.64	400.15	/	/	28.809
milli-Q water	/	/	/	/	/	/	/	/	for solution preparation (up to final volumes of 5 ml)	/

## Procedure/observations

Date	Time	Step	Observations	Pictures
08.10.2025	16:30	Weighing RhCl <sub>3</sub> .3H <sub>2</sub> O inside the glovebox, transfer it to a 10 ml vial. Named: <b>EA-357-RhCl<sub>3</sub></b>	Dark red crystals <b>EA-357-RhCl<sub>3</sub></b>	20251008_1630-RhCl3 and Cr(NO <sub>3</sub> ) <sub>3</sub> salts before addition of water.jpg
	16:30	Weighing Cr(NO <sub>3</sub> ) <sub>3</sub> .9H <sub>2</sub> O in a 10 ml vial. Named: <b>EA-357-Cr(NO<sub>3</sub>)<sub>3</sub></b>	Dark blue crystals <b>EA-357-Cr(NO<sub>3</sub>)<sub>3</sub></b>	20251008_1630-RhCl3 and Cr(NO <sub>3</sub> ) <sub>3</sub> salts before addition of water.jpg
	16:34	2ml of milli-Q water was added to the <b>EA-357-RhCl<sub>3</sub></b> vial of RhCl <sub>3</sub> *3H <sub>2</sub> O using Eppendorf pipette 100-1000 µl.	Red solution	20251008_1634-RhCl3 and Cr(NO <sub>3</sub> ) <sub>3</sub> salts after addition of water.jpg
	16:34	2ml of milli-Q water was added to the <b>EA-357-Cr(NO<sub>3</sub>)<sub>3</sub></b> vial of Cr(NO <sub>3</sub> ) <sub>3</sub> .9H <sub>2</sub> O using Eppendorf pipette 100-1000 µl.	Blue solution	20251008_1634-RhCl3 and Cr(NO <sub>3</sub> ) <sub>3</sub> salts after addition of water.jpg
	16:43	Solution of the <b>EA-357-RhCl<sub>3</sub></b> vial was transferred into a 15ml falcon tube using an Eppendorf pipette. 1ml of milli-Q water was added to the empty vial and swirled for quantitative transfer of the solution.	Red solution	20251008_1643-solutions in the falcon tubes.jpg
	16:43	Solution of the <b>EA-357-Cr(NO<sub>3</sub>)<sub>3</sub></b> vial was transferred into a 15ml falcon tube using an Eppendorf pipette. 1ml of milli-Q water was added to the empty vial and swirled for quantitative transfer of the solution.	Blue solution	20251008_1643-solutions in the falcon tubes.jpg
	18:50	The tube was filled with water till 5 ml and covered with Al foil. Named: <b>EA-357-9.874 mM RhCl<sub>3</sub>, 08.10.2025</b>	Red solution <b>EA-357-9.874 mM RhCl<sub>3</sub>, 08.10.2025</b>	20251008_1850-EA-357-Rh & Cr stock solutions.jpg
	18:50	The tube was filled with water till 5 ml and covered with Al foil. Named: <b>EA-357-28.809 mM Cr(NO<sub>3</sub>)<sub>3</sub>, 08.10.2025</b>	Blue solution <b>EA-357-28.809 mM Cr(NO<sub>3</sub>)<sub>3</sub>, 08.10.2025</b>	20251008_1850-EA-357-Rh & Cr stock solutions.jpg

## Product characterization

Sample	Concentration mM	Mass [mg]	Purity	Volume, mL	Amount [µmol]	Yield [%]	Description	Image	Storage location
<b>EA-357-9.874 mM RhCl<sub>3</sub>, 08.10.2025</b>	9.874	/	/	5	/	/	Red solution	20251008_1850-EA-357-Rh & Cr stock solutions.jpg	CEEC II, E004, EA fume hood
<b>EA-357-28.809 mM Cr(NO<sub>3</sub>)<sub>3</sub>, 08.10.2025</b>	28.809	/	/	5	/	/	Blue solution	20251008_1850-EA-357-Rh & Cr stock solutions.jpg	CEEC II, E004, EA fume hood

# Results

Stock solution of  $\text{RhCl}_3$  and  $\text{Cr}(\text{NO}_3)_3$  salts were prepared with volume of 5ml and concentration of **9.874 mM** and **28.809 mM** respectively.

## Attached files

EA-357.cdx

sha256: ebfe1c00690b5ddf3130395d846d4152d4839a311c2b5dcc280090cda42adae

EA-357.png

sha256: e167c1a6cf057e97721fcaa9eaa4a6bc621a7aa4077a49d11857839081db91f3

$\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$   
9.874 mM

$\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$   
28.809 mM

20251008\_1643-solutions in the falcon tubes.jpg

sha256: f6b0c80212eb5cca38e80ea9d4b9ab2a5202d0afcead5d3f4b428fe7b3ffa468



20251008\_1850-EA-357-Rh & Cr stock solutions.jpg

sha256: 0d60834a62549fcf6d4652fa059c20d58f01931b1d6f3e4dd5e858138cf86c5



20251008\_1634-RhCl3 and Cr(NO3)3 salts after addition of water.jpg

sha256: 39134ec282cafd6e5368a67a5d5cafaf9025ce99d20f35cfda678d1b363a984e



20251008\_1630-RhCl3 and Cr(NO3)3 salts before addition of water.jpg  
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Unique eLabID: 20251009-79229c2bbb2fbc35d238bec1d72a030cd7a3e20c  
Link: <https://elab.water-splitting.org/experiments.php?mode=view&id=3155>