

EA-MEJ-357: Preparation of RhCl₃ and Cr(NO₃)₃ stock solutions

Date: 2025-10-09

Tags: Stocksolution MEJ EA RhCl₃*3H₂O Cr(NO₃)₃*9H₂O Solution

Category: Prep work

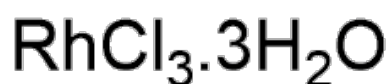
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Created by: Ebrahim Abedini

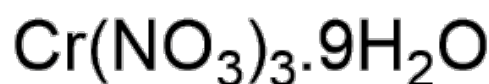
Objective

Preparation of stock solutions of RhCl₃ and Cr(NO₃)₃ salts.

Reaction scheme/sample structure



9.874 mM



28.809 mM

ChemDraw File (linked): [EA-357.cdx](#)

Literature/reference experiments

Literature	https://doi.org/10.1039/C9EE00310J
Reproduction	/
Similar experiments	Prep work - NB-275: Preparation of RhCl₃ and Cr(NO₃)₃ solutions I

Reagents

Name	CAS Number / Experiment Number	Inventory number	Amount [mmol]	Equivalents	Mass _{theo} [mg]	Mass _{exp} [mg]	Molar mass [g/mol]	Density (g/ml)	Volume [ml]	Concentration [mM]
RhCl ₃ ·3H ₂ O, BLD Pharmatech GmbH	13569-65-8	C138335	/	/	12.8	13	263.31	/	/	9.874
Cr(NO ₃) ₃ ·9H ₂ O	7789-02-8	/	/	/	57.62	57.64	400.15	/	/	28.809
milli-Q water	/	/	/	/	/	/	/	/	for solution preparation (up to final volumes of 5 ml)	/

Procedure/observations

Date	Time	Step	Observations	Pictures
08.10.2025	16:30	Weighing $\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$ inside the glovebox, transfer it to a 10 ml vial. Named: EA-357-RhCl₃	Dark red crystals EA-357-RhCl₃	20251008_1630-RhCl3 and Cr(NO3)3 salts before addition of water.jpg
	16:30	Weighing $\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ in a 10 ml vial. Named: EA-357-Cr(NO₃)₃	Dark blue crystals EA-357-Cr(NO₃)₃	20251008_1630-RhCl3 and Cr(NO3)3 salts before addition of water.jpg
	16:34	2ml of milli-Q water was added to the EA-357-RhCl₃ vial of $\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$ using Eppendorf pipette 100-1000 μL .	Red solution	20251008_1634-RhCl3 and Cr(NO3)3 salts after addition of water.jpg
	16:34	2ml of milli-Q water was added to the EA-357-Cr(NO₃)₃ vial of $\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ using Eppendorf pipette 100-1000 μL .	Blue solution	20251008_1634-RhCl3 and Cr(NO3)3 salts after addition of water.jpg
	16:43	Solution of the EA-357-RhCl₃ vial was transferred into a 15ml falcon tube using an Eppendorf pipette. 1ml of milli-Q water was added to the empty vial and swirled for quantitative transfer of the solution.	Red solution	20251008_1643-solutions in the falcon tubes.jpg
	16:43	Solution of the EA-357Cr(NO₃)₃ vial was transferred into a 15ml falcon tube using an Eppendorf pipette. 1ml of milli-Q water was added to the empty vial and swirled for quantitative transfer of the solution.	Blue solution	20251008_1643-solutions in the falcon tubes.jpg
	18:50	The tube was filled with water till 5 ml and covered with Al foil. Named: EA-357-9.874 mM RhCl₃ 08.10.2025	Red solution EA-357-9.874 mM RhCl₃ 08.10.2025	20251008_1850-EA-357-Rh & Cr stock solutions.jpg
	18:50	The tube was filled with water till 5 ml and covered with Al foil. Named: EA-357-28.809 mM Cr(NO₃)₃ 08.10.2025	Blue solution EA-357-28.809 mM Cr(NO₃)₃ 08.10.2025	20251008_1850-EA-357-Rh & Cr stock solutions.jpg

Product characterization

Sample	Concentration mM	Mass [mg]	Purity	Volume, mL	Amount [μmol]	Yield [%]	Description	Image	Storage location
EA-357-9.874 mM RhCl₃ 08.10.2025	9.874	/	/	5	/	/	Red solution	20251008_1850-EA-357-Rh & Cr stock solutions.jpg	CEEC II, E004, EA fume hood
EA-357-28.809 mM Cr(NO₃)₃ 08.10.2025	28.809	/	/	5	/	/	Blue solution	20251008_1850-EA-357-Rh & Cr stock solutions.jpg	CEEC II, E004, EA fume hood

Results

Stock solution of RhCl_3 and $\text{Cr}(\text{NO}_3)_3$ salts were prepared with volume of 5ml and concentration of **9.874 mM** and **28.809 mM** respectively.

Attached files

EA-357.cdx

sha256: ebfe1c00690b5ddf3130395d846d4152d4839a311c2b5dccb280090cda42adae

EA-357.png

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$\text{RhCl}_3 \cdot 3\text{H}_2\text{O}$	$\text{Cr}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$
9.874 mM	28.809 mM

20251008_1643-solutions in the falcon tubes.jpg

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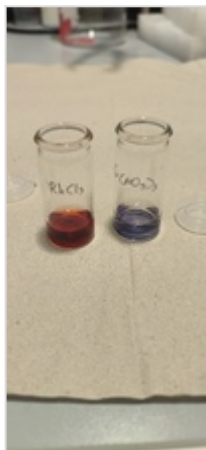
20251008_1850-EA-357-Rh & Cr stock solutions.jpg

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20251008_1634- RhCl_3 and $\text{Cr}(\text{NO}_3)_3$ salts after addition of water.jpg

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20251008_1630-RhCl₃ and Cr(NO₃)₃ salts before addition of water.jpg

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Link: <https://elab.water-splitting.org/experiments.php?mode=view&id=3155>