# AEI-053: Synthesis of [RuCl2(CO)(PNP)] with 2,6-Bis(P,P-di(tertbutyl)phosphinomethyl)pyridin

**Date:** 2022-12-15 **Tags:** Dichloro PNP [RuCl2(CO)(PNP)]

AEI P(tBu)N(py)P(tBu)

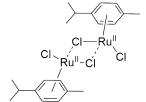
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Created by: Alexander Eith

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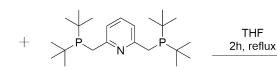
## Reaction scheme/sample structure

similar to [Experiment] AEI-050: Synthesis of [RuCl2(CO)(PNP)] with 2,6-Bis(P,P-di(tertbutyl)phosphinomethyl)pyridine



[RuCl<sub>2</sub>(p-cymene)]<sub>2</sub>

Molecular Weight: 612.38 g/mol



2,6-bis((di-*tert*-butylphosphaneyl)methyl)pyridine Chemical Formula: C<sub>23</sub>H<sub>43</sub>NP<sub>2</sub>

Molecular Weight: 395,55152

formic acid
acetic anhydride
2 NEt<sub>3</sub>

MeCN
15min.

N CI P CO CI

 $[RuCl_2(CO)(P^{tBu}N^{py}P^{tBu})]$ 

Chemical Formula: C<sub>24</sub>H<sub>43</sub>Cl<sub>2</sub>NOP<sub>2</sub>Ru Molecular Weight: 595,53152

## Reagents

Name	Amount [mmol]	Equivalents	Mass <sub>theo,</sub>	Purity [%]	Mass <sub>theo</sub> [mg]	Mass <sub>exp</sub> [mg]	Molar mass [g/mol]	Volume <sub>theo</sub> [ml]	Volume <sub>exp</sub> [ml]	Density [g/mL]
[RuCl2(p-cymene)]2	0.125	0.5	76.5	1	1	76.5	612.38	1	1	1
PNP, 2,6-Bis(P,P-di(tertbutyl)phosphinomethyl)pyridine ([Experiment] AEI-047: Synthesis of 2,6-Bis(P,P-di(tertbutyl)phosphinomethyl)pyridine)	0.25	1	98.89	96.5	102.5	102.63	395.55152	/	1	/
THF	1	/	/	1	/	/	1	15	1	/
n-heptane	1	1	1	1	1	/	1	10 + 3	/	1
formic acid	8.3	27	384.65	/	1	/	46.03	0.315	0.32	1.22
acetic anhydride	8.3	27	853.16	1	1	1	102.09	0.790	0.79	1.08
triethylamine	16.6	54	1679	1	1	1	101.19	2.30	2.3	0.73
toluene	1	1	1	1	1	1	1	15	1	1

## **Procedure/observations**

All steps (unless stated otherwise) were carried out under argon using either standard Schlenk technique or in an argon filled glovebox (MBRAUN). [Protocol] Schlenk Technique

Date	Time	Step	Observations
15.12	9:30	A 10mL and a 50 mL-Schlenk flask were prepared.	
	9:40 - 10:50	A small reflux condenser was prepared, by evacutating it for approx.1 h	

	10:30 - 10:45	PNP was weighed into the 10 mL-flask in a glovebox	white solid
	11:05	[RuCl2(p-cymene)]2 was weighed into the 50 mL flask under air and the flask was subsequently set under argon	red solid
	11:30	To the PNP THF (7.5 mL) and to the [RuCl2(p-cymene)]2 THF (7.5 mL) were added and the mixtures were stirred for 5 min	PNP: turbid solution, Ru: some solid did not dissolve, orange-red solution
	11:35	The PNP mixture was added to the 50 mL flask containing the [RuCl2(p-cymene)]2 mixture	no significant change was observed
	11:35 - 11:55	The mixture was stirred at 500 rpm for approx. 20 min. Thereby the reaction flask was equipped with a small reflux condenser	see: before heating.jpg
	11:55 - 13:55	The mixture was heated to reflux for 2 h, at 400 rpm At 12:15 the mixture started to reflux.	during the heating the colour got darker and more red.
	13:55 - 15:05	The mixture was cooled to rt	clear, dark yellow, orange solution, see: after heating.jpg After some time at rt solids formed, see: solid.jpg
	13:50 - 14:50	The setup for the generation of CO was build up according to [Protocol] Addition of CO by bubbling of CO through the solution	
	15:05	Toluene (15 mL) was added into the flask for the CO generation according to [Protocol] Addition of CO by bubbling of CO through the solution	
	15:10 - 15:15	formic acid and acetic anhydride were added into the flask for the CO generation according to [Protocol] Addition of CO by bubbling of CO through the solution	
	15:23 - 15:41	Triethylamin was added slowly over the next 15 min into the flask for the CO generation according to , both flask were stirred at 500 rpm	Bubbles developed in the clear colorless solution. The reaction mixture became less cloudy and finally formed a solution after the addition of 0.8 mL and approx. 8 min, see clear by CO.jpg
	15:45 - 7:00	The solution was stirred under CO atmosphere at 500 rpm and rt	after CO.jpg
16:12	7:00 - 11:25	The stirring was turned off and the reaction mixture was left standing at rt	orange solution, see: after stirring over night.jpg
	11:25 - 12:55	The volume of the solution was reduced to approx. 5 mL	see: reduced.jpg

	11:55 - 12:15	10 mL n-Heptane were added to the solution and the obtained suspension was stirred for approx. 5 min	see: after precipitation.jpg
	13:45 - 13:55	The cold solution was filtered according to [Protocol] Filtration with frit technique	
	13:55 - 14:05	The obtained solid was washed with n- Heptane (3 * 1 mL)	
	14:10 - 14:55	The solid was dried under reduced pressure	
	14:55	The solid and filtrate were stored under argon at rt	orange-yellow liquid, see filtrate with crystals greenish solid, see filter residue.jpg
19.12	14:20	A NMR-Sample of the solid was prepared according to [Protocol] Preparation of NMR Sample (AEI-050-1)	range brown solution, 13.8 mg
		In the filtrate crystals formed.	See: filtrate with crystals

# Analysis

Date	Time	Sample name	Analysis method	Solvent	Raw Data	Processed Data	Interpretation
19.12	15:24	AEI-053-1	1H (-30 - 20 ppm), 31P quant cryo	CD2Cl2	AEI-053-1.zip		Main species can be observed, as in [Experiment] AEI-050: Synthesis of [RuCl2(CO)(PNP)] with 2,6-Bis(P,P-di(tertbutyl)phosphinomethyl)pyridine. But more of the 73 ppm product. In 1H hydrides can be observed as well as very brod peak, suggesting a Ru(III) species.
1	1	SH-03013_A2_a-3	singel crystal XRD	I	SH_03013_A2_a-3.res, SH_03013_A2_a.docx	SH_03013_A2_are sinP1_1.png, SH_03013_A2_are sinP1_2.png, SH_03013_A2_are sinP1_3.png, SH_03013_A2_are sinP1.png	Nice crystal structure, dimer, trans species

## **Product characterization**

	m [mg]	purity	m <sub>pure</sub> [mg]	n [mmol]	Y [%]

### **Linked resources**

Protocol - Preparation of NMR Sample

Protocol - Filtration with frit technique

Protocol - Schlenk Technique

Protocol - Addition of CO by bubbling of CO through the solution using syringe

#### **Attached files**

SH 03013 A2 aresinP1.png

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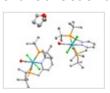
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SH 03013 A2 aresinP1 1.png

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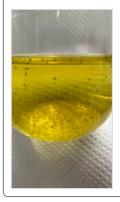
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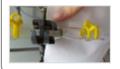
filtrate-with-crystals

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filter-residue.jpg

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after-precipitation.jpg

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reduced.jpg

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after-stirring-over-night.jpg

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after-CO.jpg

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clear-by-CO.jpg

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solid.jpg

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before-heating.jpg

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Unique eLabID: 20221215-82804a85eff1107de87b29a0a2642c45e0673de9 Link: https://elab.water-splitting.org/experiments.php?mode=view&id=342