

Week 11 - Monday - PHYS 107

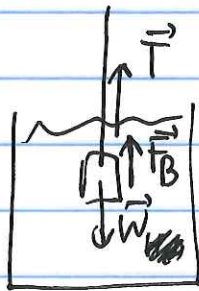
* Buoyancy:

$$\begin{array}{ccc} F_B = & W_{\text{displaced liquid}} \\ \uparrow & \uparrow \\ \text{buoyant force} & \text{weight of displaced} \\ \text{upward} & \text{liquid (number, would be downward)} \end{array}$$

$F_B > W_{\text{object}} \rightarrow$ float because net force up when submerged

$F_B < W_{\text{object}} \rightarrow$ sinks because net force down

Example: Archimedes measured the weight of the King's crown in air and in water



$$T = W_{\text{air}}$$

$$T = W - F_B \quad \alpha < W$$

↓
scale reads lower number

* Galileo thermometer (image)

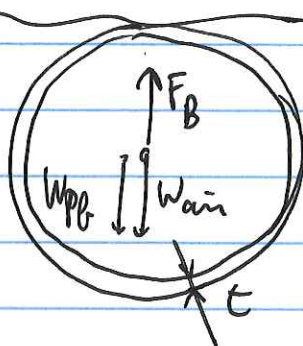
Floating bells have constant V and constant $m \rightarrow \rho$ is constant inside bell

What changes? ρ_{H_2O} !

As temperature increases $\rightarrow \rho_{H_2O}$ will increase $\rightarrow F_B = \rho_{H_2O} V_{\text{bell}}$
↑
increases
 bells start to rise one by one

* Example: lead balloon, $\rho = 11.3 \times 10^3 \text{ kg/m}^3$

How thin do the walls have to be for this air-filled balloon to float in water? $R = 10 \text{ cm}$



very small

$$0 = F_B - W_{Pb} - W_{air}$$

$$0 = \rho_{H_2O} \left(\frac{4}{3} \pi R^3 \right) g - \rho_{Pb} (4 \pi R^2) t g$$

$$\downarrow$$

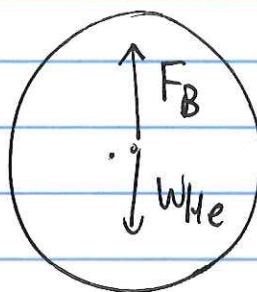
$$t = \frac{\rho_{H_2O} R}{3 \rho_{Pb}} = 3 \times 10^{-3} \text{ m} = 3 \text{ mm}$$

* Example: Can 42 He-filled balloons (5' diameter) lift a person (70kg)?

$$F_{\text{lift}} = F_B - W_{\text{He}}$$

$$F_B = W_{\text{air}} \downarrow$$

$$F_{\text{lift}} = W_{\text{air}} - W_{\text{He}}$$



F_{lift} = net force that is then used to lift person

$$= V \rho_{\text{air}} g - V \rho_{\text{He}} g = V (\rho_{\text{air}} - \rho_{\text{He}}) g$$

$$5' \text{ diameter} \rightarrow R = 0.76 \text{ m} \rightarrow V = \frac{4}{3} \pi R^3 = 1.85 \text{ m}^3$$

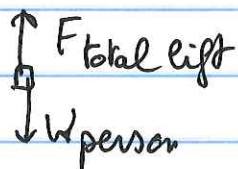
$$\text{For each balloon: } F_{\text{lift}} = (1.85 \text{ m}^3) \left(1.29 \frac{\text{kg}}{\text{m}^3} - 0.18 \frac{\text{kg}}{\text{m}^3} \right) g$$

$$= 20.2 \text{ N}$$

$$\text{Total lift} = (42)(20.2 \text{ N}) = 850 \text{ N} \rightarrow \text{yes, person is lifted}$$

$\frac{850 \text{ N}}{686 \text{ N}} > 1$

What is the acceleration?



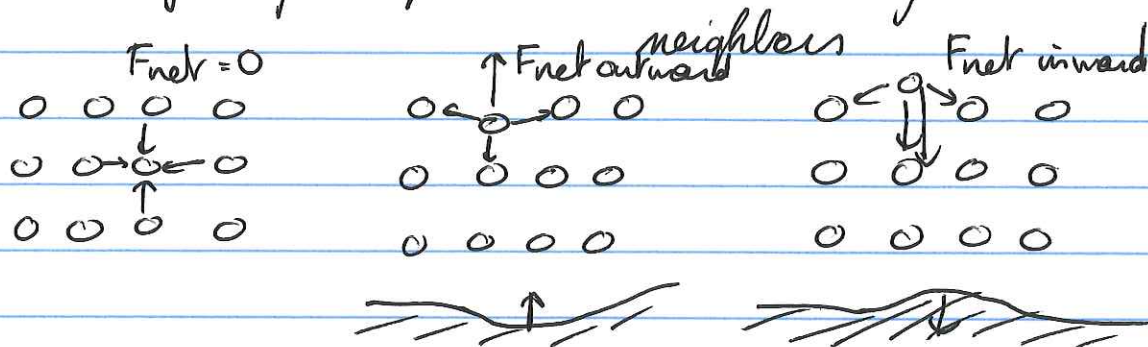
$$\rightarrow ma = F_{\text{total lift}} - W_{\text{person}}$$

$$a = \frac{F_{\text{total lift}} - W_{\text{person}}}{m} = +2.3 \text{ m/s}^2$$

Balloon flight video

* Surface tension : cohesive forces between molecules of the same type in a liquid

Consider surface of a liquid : molecules exert forces on



Surface of liquid "wants" to be as smooth as possible

For suspended liquid \rightarrow droplets become spherical

$$\text{Surface tension} = \gamma = \frac{F}{l} \quad \text{in units } \frac{N}{m}$$

where l is the circumference of the contact area

Water spider

Example : what is the angle at the contact with the water?

$$R = 1.5 \times 10^{-4} \text{ m} \quad \rightarrow \quad F = \gamma l = \gamma (2\pi R)$$

$$m = 2 \times 10^{-5} \text{ kg}$$

$$\gamma = 0.0728 \text{ N/m}$$

$$F_y = F \cos \theta \times 6 \quad (\text{legs})$$

$$F_{\text{net}} = F_y - W = 0 \rightarrow \cos \theta = \frac{mg}{12\pi\gamma R} = 0.47 \rightarrow \theta = 62^\circ$$

what happens when $\cos \theta > 1$? bug sinks

* Surfactants: substances in liquid that reduces γ ,
e.g. detergent in H_2O makes it "stick" ^{cling}
more to grease

In the lungs: pulmonary alveoli (mucus liquid)
tiny sacs with $r = 10^{-2} \text{ cm}$

inhalation: muscles expand chest cavity \rightarrow negative pressure
of $\sim -3 \text{ mmHg}$

surfactant of long lipoproteins in mucus
when extended: γ increases \rightarrow alveoli don't expand
too much

exhalation: surface tension when extended causes alveoli
to shrink again, γ decreases, alveoli don't
collapse

{ Drowning: water in lungs: surfactant is diluted $\rightarrow \gamma$ is
too large and can't inhale
{ Infants
Hyaline membrane disease

Emphysema: alveoli walls deteriorate \rightarrow larger bubbles/sacs
 \rightarrow reduced pressure \rightarrow can't exhale