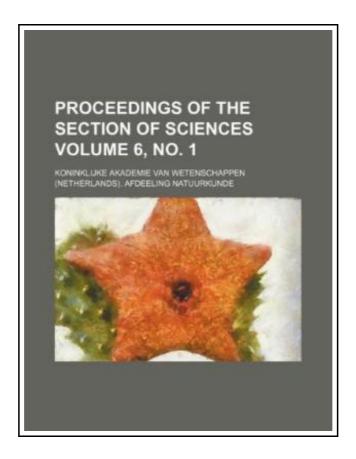
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1904 Excerpt: . and the critical curve aax p concur. This implies that at this temperature the curve /, p for the vapour coexisting with solid B is the prolongation of the curve p/ for the solution coexisting with solid B. The same occurs at a great many higher temperatures. That a continuation of the lines cp and ep is imaginary, clearly appears from this diagram, as the vapour-and the liquid-line, if both were prolonged, would change places, which is impossible.) Whereas from /, to t, saturated solutions are absolutely impossible, at tt the same phenomenon occurs as at /; here also the solubility curve dctq, the vapour-line de, q and the critical curve bat a q converge and the critical phenomenon is observed with a saturated solution. At higher temperatures a convergence of the three curves can no longer occur and in consequence all critical temperatures between /, and /s, just as between /, and tt, are critical temperatures of unsaturated solutions. If between p and q solution + vapour-fsolid B be impossible, it is conceivable, as suggested before, that we may succeed in getting supersaturated solutiors and observing their critical phenomena. In such a case the dotted critical curve if prolonged might be realized between p and q, so this dotted line is metastable. For a thorough knowledge of the phenomenon a p, r, /-diagram is most desirable and a r, /-diagram indispensable. Both space ) In the first communication, Zeilschr. f. Elcktr. 33, GG3,...

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