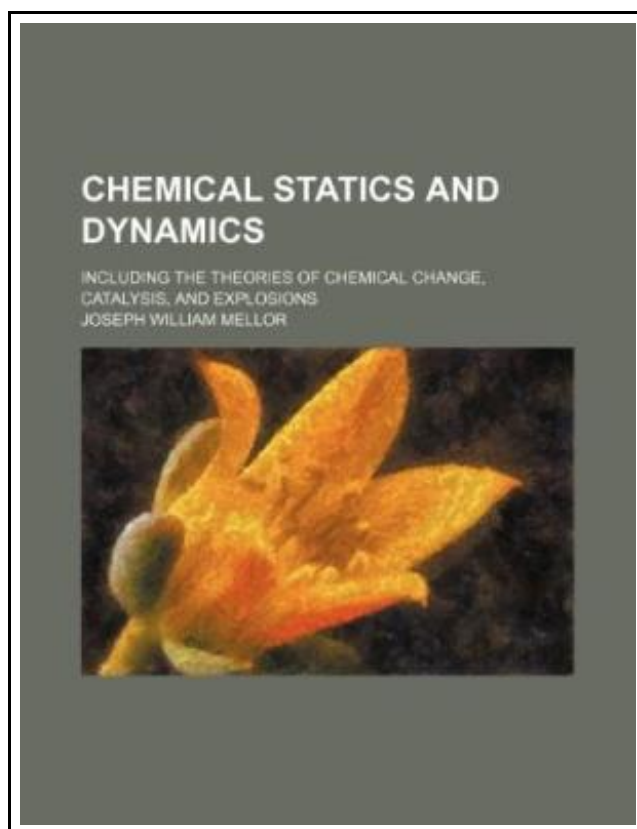


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1909 Excerpt: . attempt is made to measure the rate of combination of hydrogen and oxygen. Berthelot<sup>1</sup> found that the rate of combination of hydrogen and oxygen, at 250-300, was accelerated by the presence of barium hydrate, by alkalies, and by traces of manganese salts. Now, these substances may be regarded as decomposition products of glass, and are therefore always present on the surface of glass vessels, and Berthelot thinks that the discordant results obtained when the speed of the reaction is measured in different vessels might be due to the presence of varying amounts of the decomposition products of the glass. Reactions in liquid menstrea are not so sensitive to the nature of the walls of the vessel as gaseous reactions. But still the nature of the vessel does exercise a measurable influence on the course of some reactions. For example, Rayman and Sulc<sup>2</sup> find that metal vessels exert a specific influence on the inversion of cane sugar. Polished platinum vessels, however, 1 M. Berthelot, Compt. Rend., 125. 271, 1897. 1 B. Rayman and O. Sulc, Zeit. phys. Chem., 21. 481, 1896; 33. 47, 1900; F. Plzak and B. Husek, ib., 47. 733, 1904. do not influence the decomposition of hydrogen peroxide.<sup>1</sup> The rate of reduction of Fehling s solution by invert sugar is augmented by the use of vessels with a large internal capacity.<sup>2</sup> The cuprous oxide formed during the reaction also seems to accelerate the velocity of the reduction. 80. J. J. Thomson s Surface...



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