2019 Consumer Confidence Report

Water System Name: Foresthill Public Utility District Report Date: May 20, 2020

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 to December 31, 2019 and may include earlier monitoring data.

Este informe contiene información muy importante sobre su agua para beber. Favor de comunicarse Foresthill Public Utility District a (530) 367-2511 para asistirlo en español.

Type of water source(s) in use: Reservoir, spring and emergency backup supply wells

Name & general location of source(s): Sugar Pine Reservoir is located north of Foresthill, Mill Creek Spring is

located north east of Foresthill, and there are two well sites near Todd Valley

Drinking Water Source Assessment information: A Watershed Sanitary Survey was completed June 19, 2009 and is available at the district office. The source is most vulnerable to contamination through recreational activities.

Time and place of regularly scheduled board meetings for public participation:

The board meets at 2:00 PM on the

second Wednesday of the month at the Veteran's Memorial Hall, 24601 Harrison Street, Foresthill, CA 95631

For more information, contact: Henry N. White Phone: (530) 367-2511

TERMS USED IN THIS REPORT

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (U.S. EPA).

Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Primary Drinking Water Standards (PDWS): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

Secondary Drinking Water Standards (SDWS): MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the MCL levels.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Variances and Exemptions: Permissions from the State Water Resources Control Board (State Board) to exceed an MCL or not comply with a treatment technique under certain conditions.

Level 1 Assessment: A Level 1 assessment is a study of the water system to identify potential problems and determine (if possible) why total coliform bacteria have been found in our water system.

Level 2 Assessment: A Level 2 assessment is a very detailed study of the water system to identify potential problems and determine (if possible) why an *E. coli* MCL violation has occurred and/or why total coliform bacteria have been found in our water system on multiple occasions.

ND: not detectable at testing limit

ppm: parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (μg/L)

ppt: parts per trillion or nanograms per liter (ng/L)

ppq: parts per quadrillion or picogram per liter (pg/L)

pCi/L: picocuries per liter (a measure of radiation)

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

• *Microbial contaminants*, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

- *Inorganic contaminants*, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- Pesticides and herbicides, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the U.S. EPA and the State Board prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. The U.S. Food and Drug Administration regulations and California law also establish limits for contaminants in bottled water that provide the same protection for public health.

Tables 1, 2, 3, 4, 5, and 6 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old. Any violation of an AL, MCL, MRDL, or TT is asterisked. Additional information regarding the violation is provided later in this report.

the violation is provided		•								
TABLE 1 – No microbiological contan				WING THE D inpling period. 7					ACTERIA	
				OWING THE					COPPER	
Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of Samples Collected	90 th Percent Leve Detect	tile No. Sites Exceeding		PHG	No. of Schools Requesting Lead Sampling		Typical Source of Contaminant	
Lead (ppb)	June 19, 2017	22	1.71	0	15	0.2		3	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits	
Copper (ppm)	June 19, 2017	22	0.034	4 0	1.3	0.3	Not applicable		Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives	
	TABLE	3 – SAMPI	LING R	ESULTS FOR	SODIUM	AND I	IARDI	NESS		
Chemical or Constituent (and reporting units)	Sample Date	Leve Detec	-	Range of Detections	MCL		PHG (MCLG)		Typical Source of Contaminant	
Sodium (ppm)	June 2014	1 2.8		N/A	None	N	•		ent in the water and is naturally occurring	
Hardness (ppm)	March 2011	22.0)	N/A	None	N	the wa		polyvalent cations present in r, generally magnesium and and are usually naturally g	
TABLE 4 – DET	TECTION	OF CONT.	AMINA	NTS WITH A	PRIMAR	Y DRIN	IKING	WATE	R STANDARD	
Chemical or Constituent (and reporting units)	Sample Date	Leve Detec	-	Range of Detections	MCL [MRDL]	(M(HG CLG) Typica RDLG]		d Source of Contaminant	
Total Trihalomethanes (ppb)	August 5, 2019	, 30	30		80	N			ct of drinking water ion	
Haloacetic Acids (ppb)	August 5, 2019	, 25		24 - 25	60	N	N/A Byprodu disinfect		ct of drinking water ion	
Control of DBP	Quarterly	, 0.79	9	0.75 - 0.79	TT	N	J/A	Various	natural and man-made	

Precursors (TOC, ppm)

2019

TABLE 5 – DETECTION OF CONTAMINANTS WITH A <u>SECONDARY</u> DRINKING WATER STANDARD						
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	SMCL	PHG (MCLG)	Typical Source of Contaminant
Color (units)	January 2007	3.0	N/A	15.0	N/A	Naturally occurring organ materials
Odor (units)	June 2010	3	N/A	3	N/A	Naturally occurring organic materials
Chloride (ppm)	June 2014	2.7	N/A	500	N/A	Runoff from natural deposits
Specific Conductivity (µS/cm)	February 23, 2012	48	N/A	1,600	N/A	Substances that form ions when in water
Total dissolved solids (ppm)	May 2016	37	N/A	1,000	N/A	Runoff from natural deposits

TABLE 6 - DETECTION OF UNREGULATED CONTAMINANTS

No unregulated contaminants were detected in the sampling period

Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Foresthill Public Utility District is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (1-800-426-4791) or at http://www.epa.gov/lead.

For Water Systems Providing Groundwater as a Source of Drinking Water

No ground water was used during the reporting period

For Systems Providing Surface Water as a Source of Drinking Water

TABLE 7 - SAMPLING RESULTS SHOWING TREATMENT OF SURFACE WATER SOURCES					
Treatment Technique ^(a) (Type of approved filtration technology used)	In-line Filtration				
	Turbidity of the filtered water must:				
Turbidity Performance Standards (b)	1 – Be less than or equal to 0.1 NTU in 95% of measurements in a month.				
(that must be met through the water treatment process)	2 – Not exceed 1.0 NTU for more than eight consecutive hours.				
	3 – Not exceed 5.0 NTU at any time.				
Lowest monthly percentage of samples that met Turbidity Performance Standard No. 1.	100%				
Highest single turbidity measurement during the year	0.090 May 28, 2019				
Number of violations of any surface water treatment requirements	None				

⁽a) A required process intended to reduce the level of a contaminant in drinking water.

⁽b) Turbidity (measured in NTU) is a measurement of the cloudiness of water and is a good indicator of water quality and filtration performance. Turbidity results which meet performance standards are considered to be in compliance with filtration requirements.