

The Integral, Riemann Summing and Atmospheric Precipitable Water

Calc 1 and Atmospheric Physics

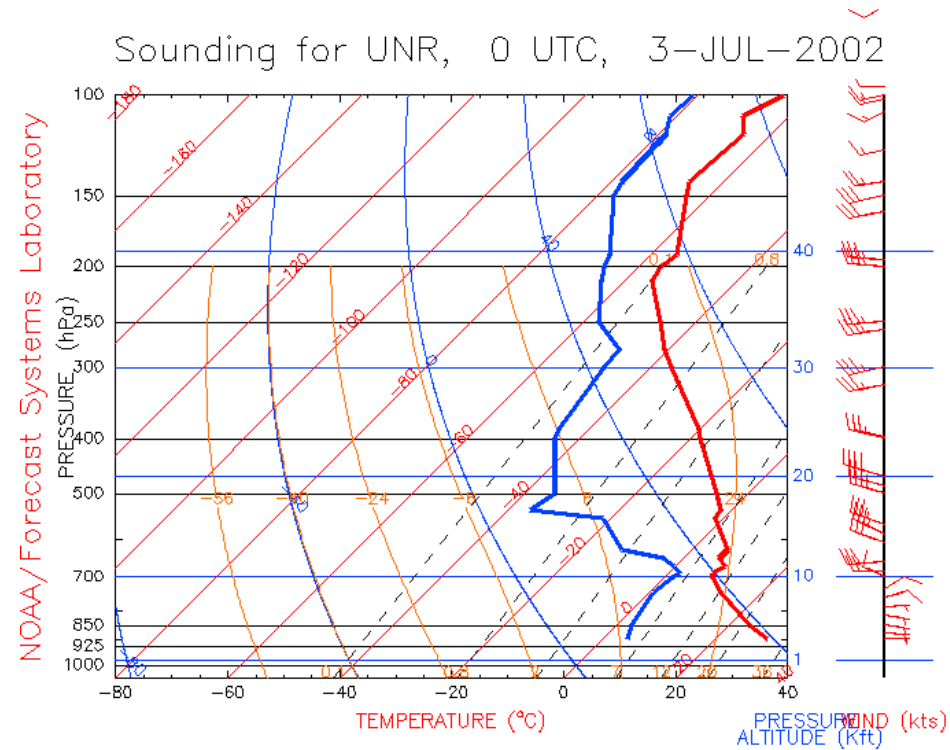
Dr. Bill Capehart
Institute of Atmospheric Sciences
SD School of Mines

Water in the Atmosphere

- Water plays an important role in the atmosphere.
- Obviously it's correlation with rain, snow and clouds is an important factor.
- But it also plays an important role in the buoyancy of air and other processes that indirectly lead to rainfall and storm development.
- It is also an important greenhouse gas
- We also use the presence of water to view atmospheric motion with satellites.

Water in the Atmosphere

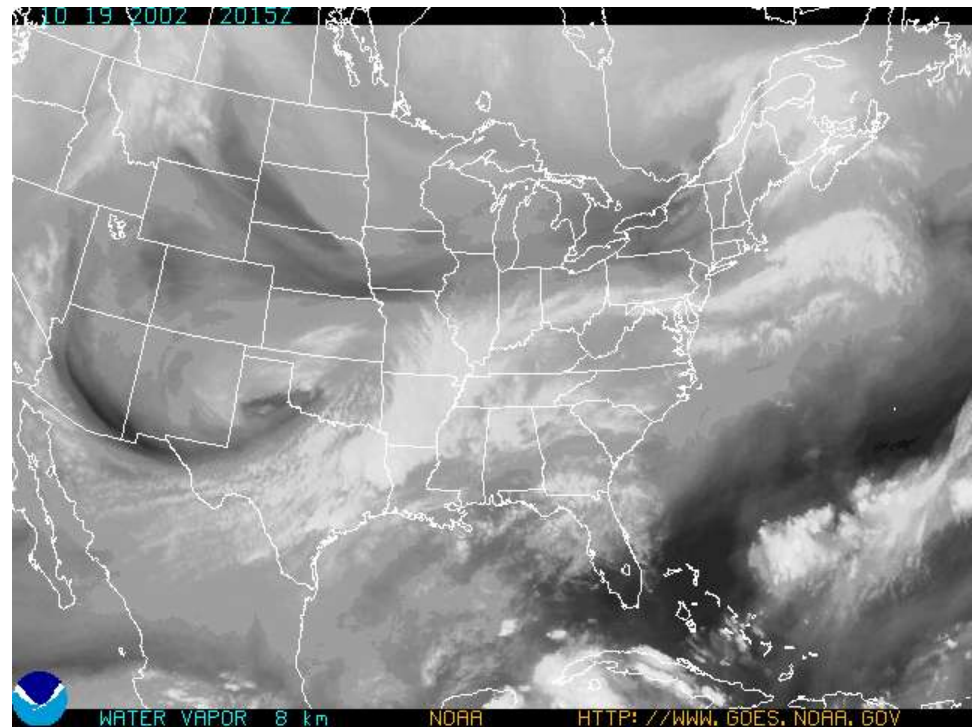
- Atmospheric Water can be measured by **weather balloons**, satellites that can see the emission of heat from the atmosphere (where water vapor is a key greenhouse gas) and surface instruments that look up into the atmosphere.



A vertical graph of temperature and dewpoint taken from a Rapid City Weather Service weather balloon <http://raob.fsl.noaa.gov/>
You'll see these charts in Synoptic and Thermodynamics

Water in the Atmosphere

- Atmospheric Water can be measured by weather balloons, **satellites that can see the emission of heat from the atmosphere (where water vapor is a key greenhouse gas)**, and surface instruments that look up into the atmosphere.
- You'll learn about the significance of the bright and dark swirls in Synoptic Meteorology



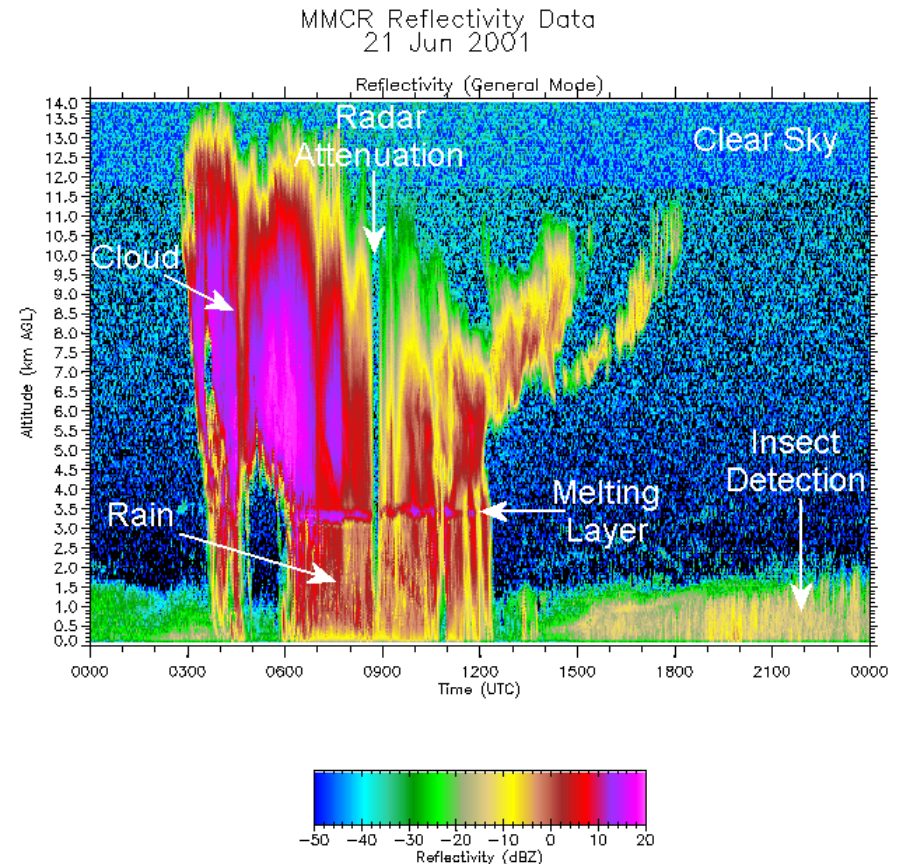
A image form the NOAA GOES weather satellite – available at

<http://www.goes.noaa.gov/>

You'll see these images pretty much a lot of the time!

Water in the Atmosphere

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Data from an upward-looking cloud radar from the Dept of Energy's atmospheric observation in Kansas and Oklahoma

<http://www.nsdsl.arm.gov/Visualization/mmcr/frame.htm>

Precipitable Water

- One important way to describe water in the atmosphere is through “Precipitable Water”
- Put simply, the total amount of water (in inches or cm or mm) in a column in the atmosphere.
- This includes not only ice and water in rain or clouds, but also the water vapor in the atmosphere.



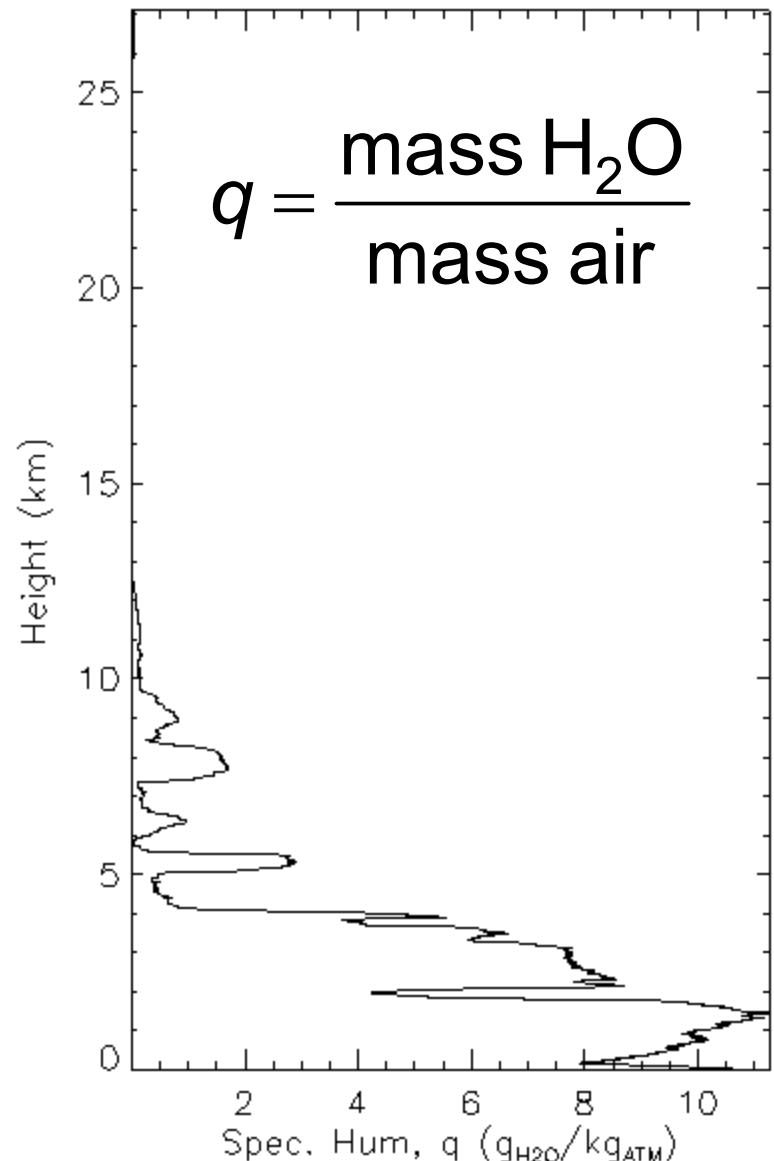
Precipitable Water

- In this example, we are going to use a very detailed sounding from the DOE's Kansas and Oklahoma network to show how basic calculus, the integral and Riemann summing can help you estimate the total amount of water in a column.



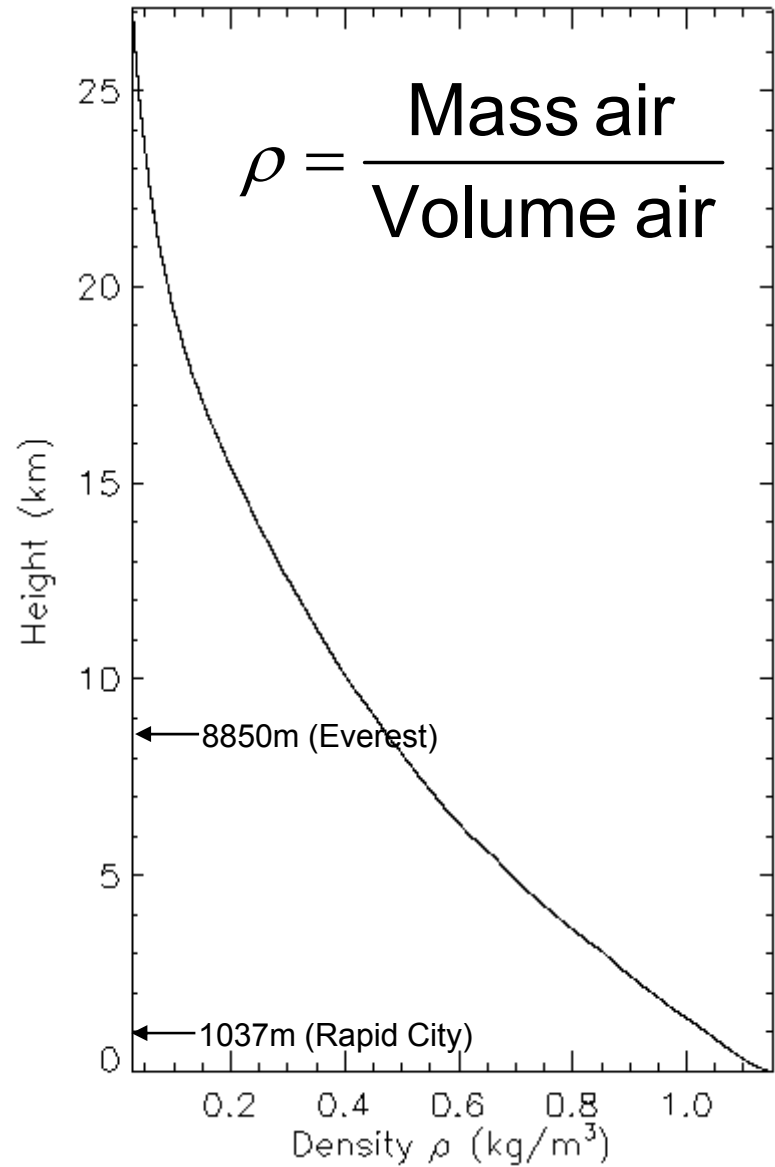
Humidity Profile

- You're probably accustomed to looking at humidity in terms of relative humidity or, perhaps, dew point.
- Here, we use a parameter called "absolute humidity" (q) which is defined as the ratio of total mass of water vapor to the total mass of air for a given layer.
- This will give us a more direct way to estimate the total amount of water vapor a column
- Notice that there is much less water in a column than total air (even in a cloud). As such we use a ratio of grams of water per kilogram of atmosphere



Density Profile

- It also helps if you have a profile of atmospheric density (ρ).
- Notice that the profile of density decreases exponentially with height.
- (Which is why people who climb Everest and other high mountains, need oxygen tanks)
- For more about how this graph is calculated read the natural log and exponential decay module



Precipitable Water and Integration

- Precipitable Water is effectively the *integrated* columnar water. Not the emphasis of “integrated.”

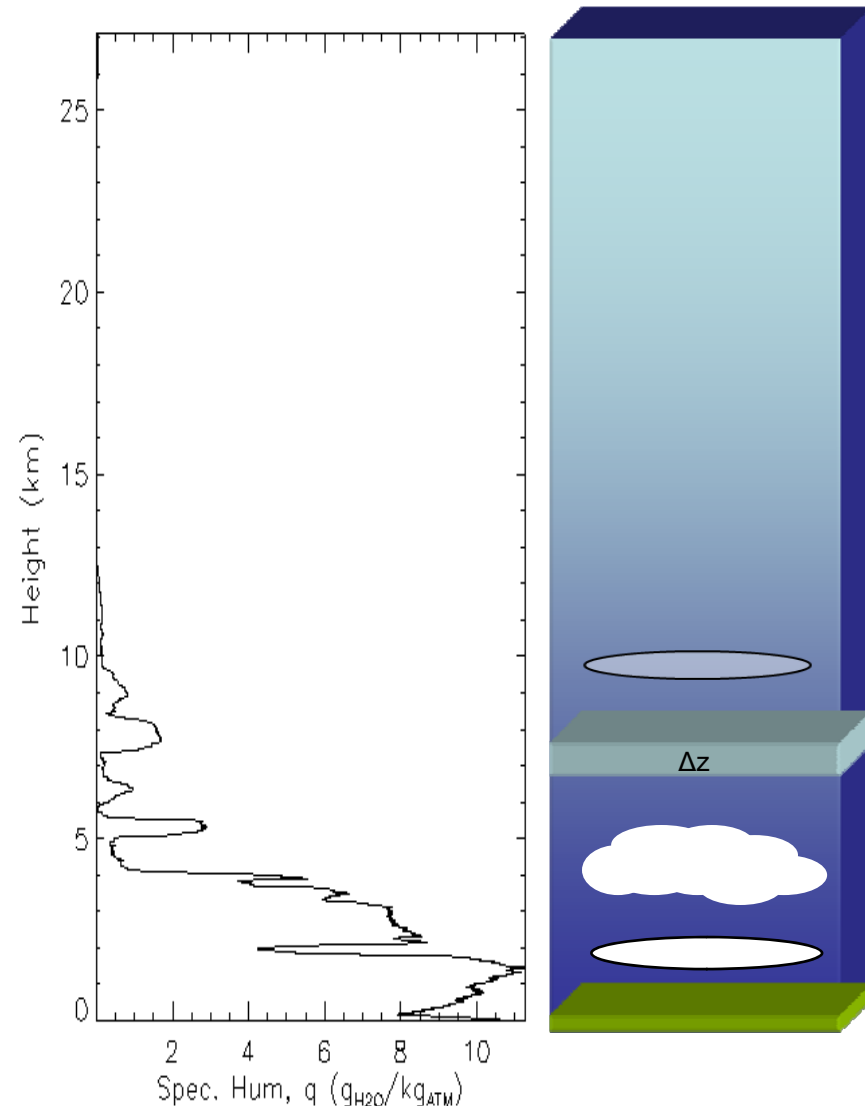
- Indeed, the definition is

$$W = \int_0^{z_{top}} \rho_a q dz$$

- By looking at the formula we can see why the formula is structured as it is. Let’s look at the units...

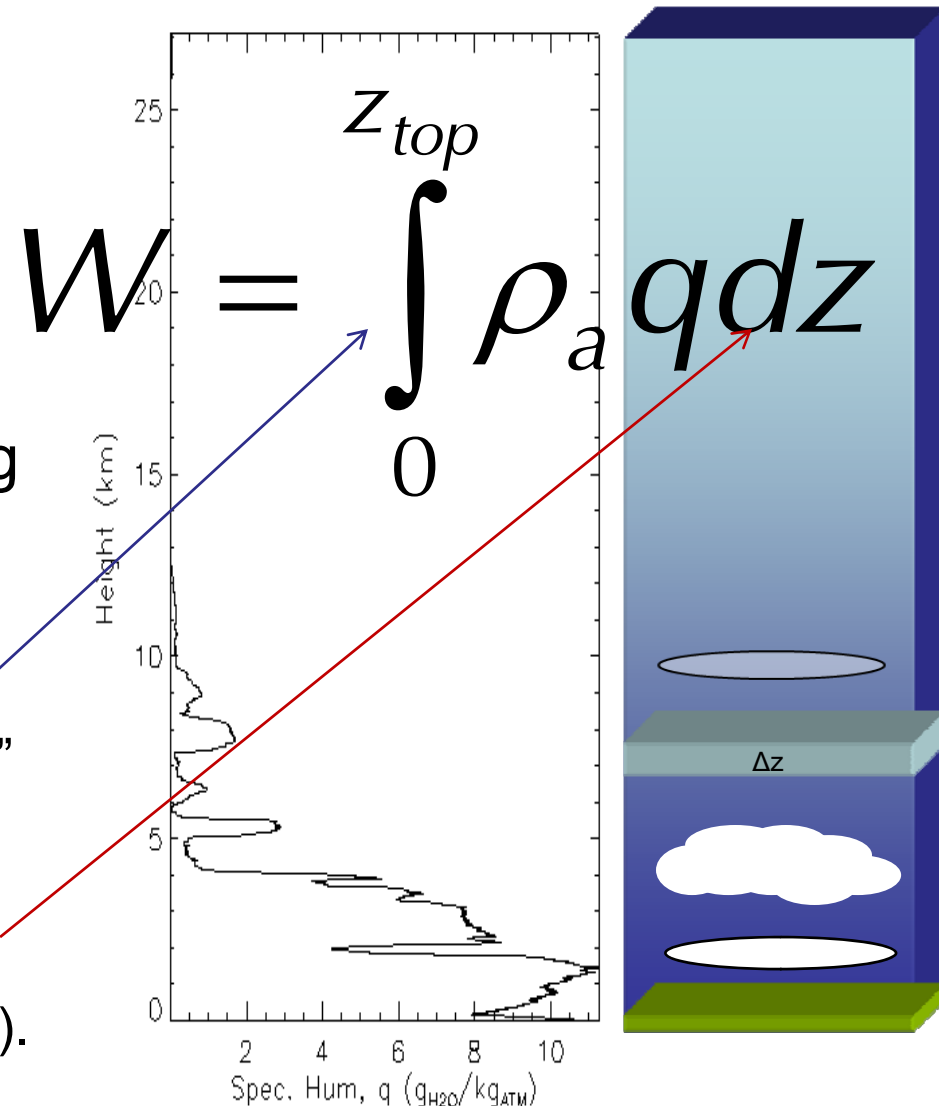
$$W \left(\frac{\text{kg}_{\text{water}}}{\text{m}^2_{\text{space}}} \right) = \int_0^{z_{top}} \rho_a \left(\frac{\text{kg}_{\text{air}}}{\text{m}^3_{\text{air}}} \right) q \left(\frac{\text{kg}_{\text{water}}}{\text{kg}_{\text{air}}} \right) dz (\text{m}_{\text{air}})$$

- Unit integrity is very important in these types of calculations.
 - Notice how we’ve not used km or g – only “MKS” units



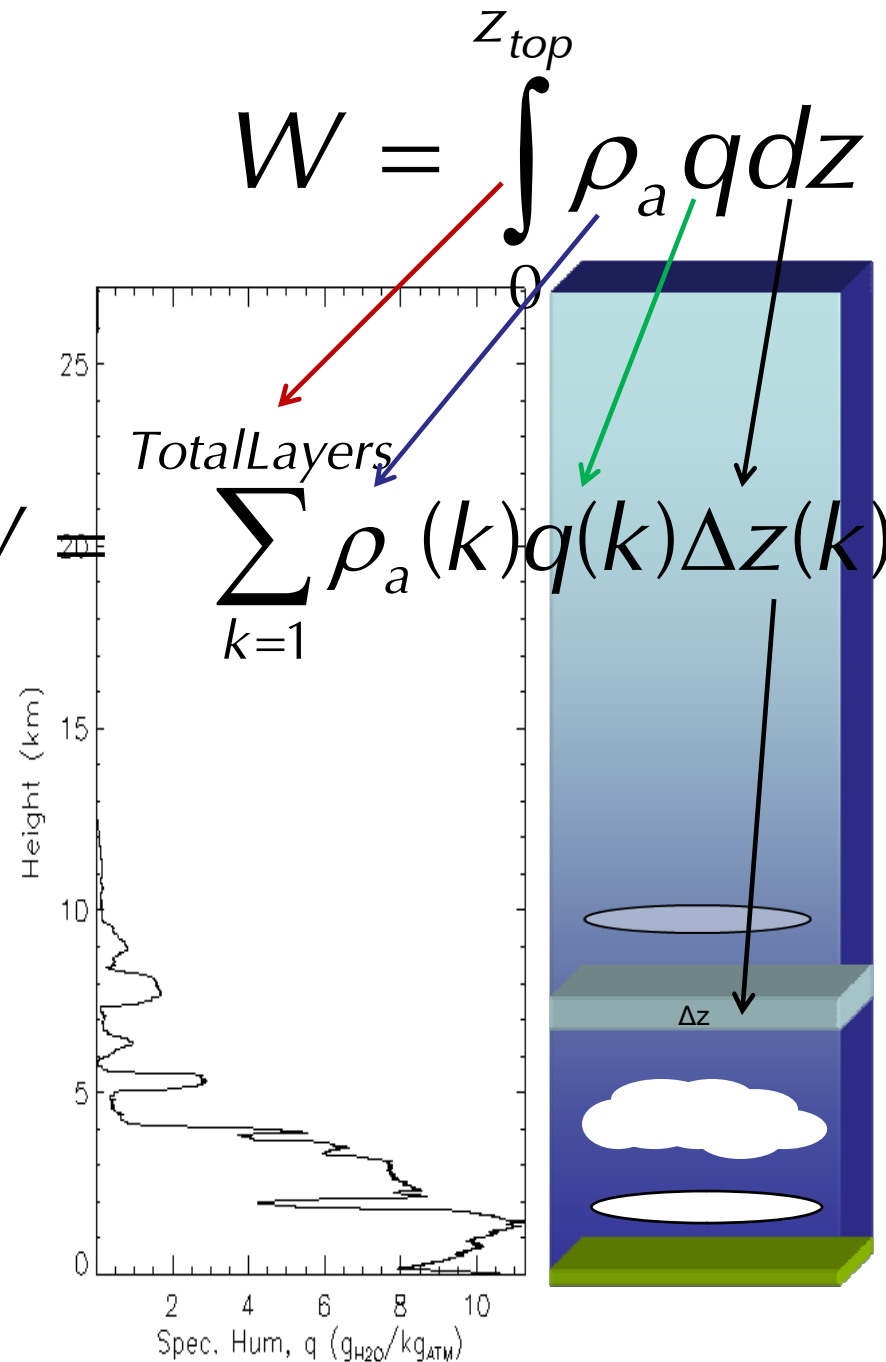
Intermission

- Let's Pause and second look at that Integral
- In Physical Equations, an Integral often implies that we will be adding things up along a path (ds, dx, etc) or over time (dt)
 - Here we are going from the surface (z=0) to a fictional “top” of the atmosphere (z=z_{top})
 - And we are doing this in tiny (infinitesimal) increments of Δz (hence the dz and not the delta).



Intermission

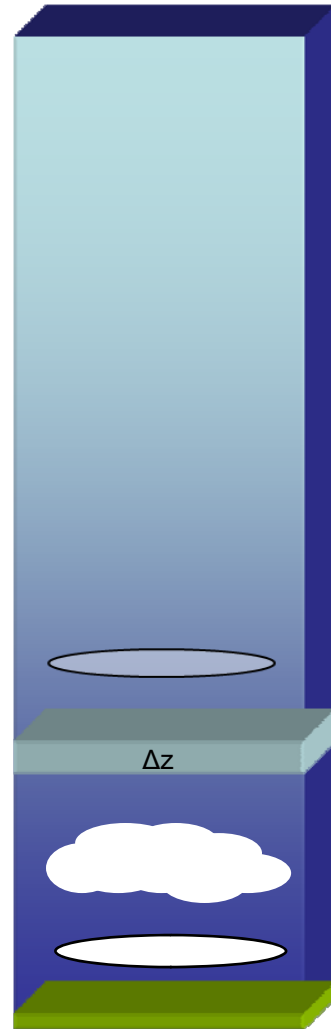
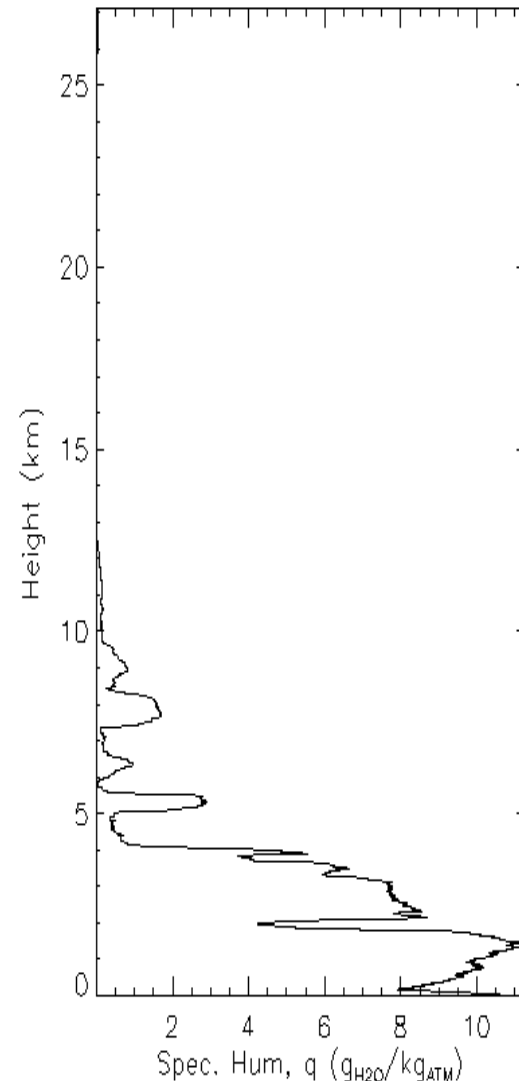
- Often when we work with integrals in the real world, especially in this example, we do what you do in Calc 1. Break it down into little boxes that we *can* measure.
- If you need to physically interpret an equation (and you should do this every time you see one), you can often use the Riemann Sum translation in your head!
 - (here we use “k” as a counter for the height layers)



Precipitable Water and Riemann Sums

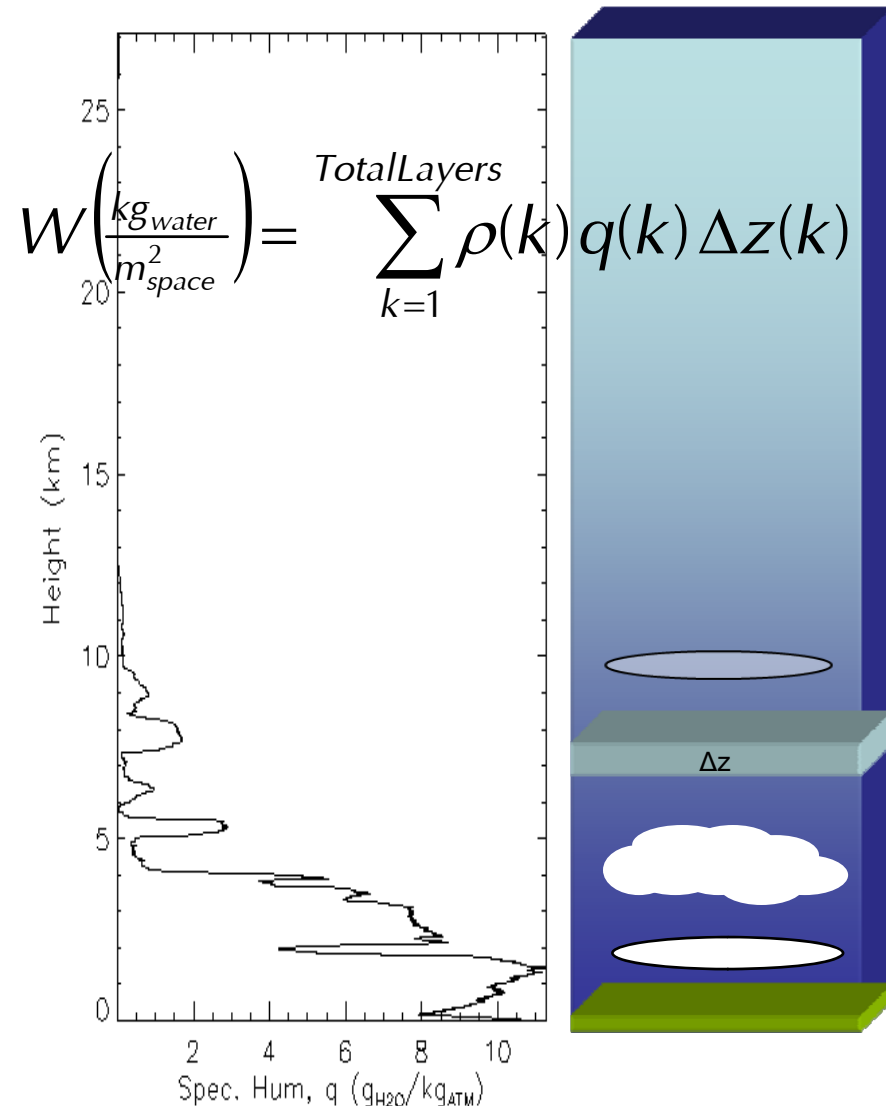
- When we solve this using real data, we actually don't use anything as elegant as integrals. Looking at the graph of specific humidity to the right, we can see the complexity in the plot. Getting a simple function of $q = f(z)$ would simply not provide an acceptable result. Therefore, we rely in the precursor to integration: The Riemann Sum and use the integral form to present the idealized principal.

$$W\left(\frac{kg_{water}}{m^2_{space}}\right) = \sum_{k=1}^{TotalLayers} \rho(k) q(k) \Delta z(k)$$



Precipitable Water and Riemann Sums

- To solve this problem, we could use this equation in a spreadsheet, if we don't have that many Δz increments.
- For a large number of Δz 's (this case to the side has over 3000 layers – well in excess of conventional weather balloon observations!)
- For this case, which represents a rather wet atmosphere, the result is about $34.46 \text{ kg}_{\text{water}}$ sitting over a square meter of real estate
 - Prove that units estimate
- However, we often prefer to express this quantity in terms of water depth (not mass).



One Last Unit Conversion

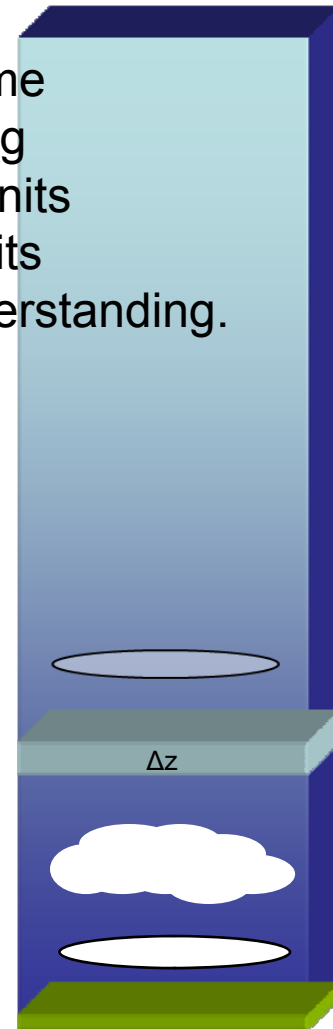
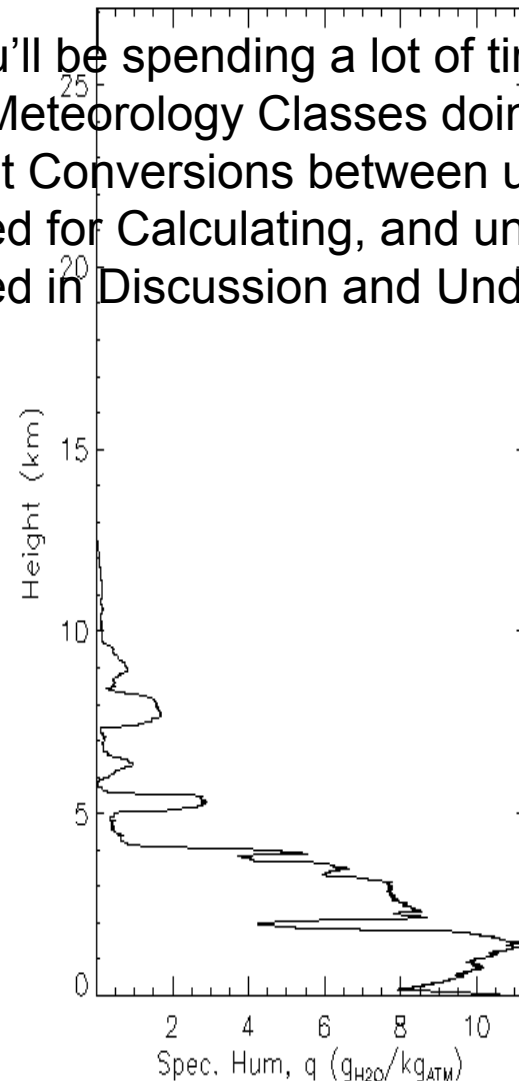
- To convert our 34.46 $\text{kg}_{\text{water}}/\text{m}^2_{\text{air}}$ of column moisture into a depth we need one last conversion.

$$W(m_{\text{water}}) = W\left(\frac{\text{kg}_{\text{water}}}{\text{m}^2_{\text{space}}}\right) \left[\rho_w = \left(\frac{m^3_{\text{space}}}{1000 \text{ kg}_{\text{water}}} \right) \right]$$

$$W(\text{cm}_{\text{water}}) = W(m_{\text{water}}) \left(\frac{100 \text{ cm}}{\text{m}} \right)$$

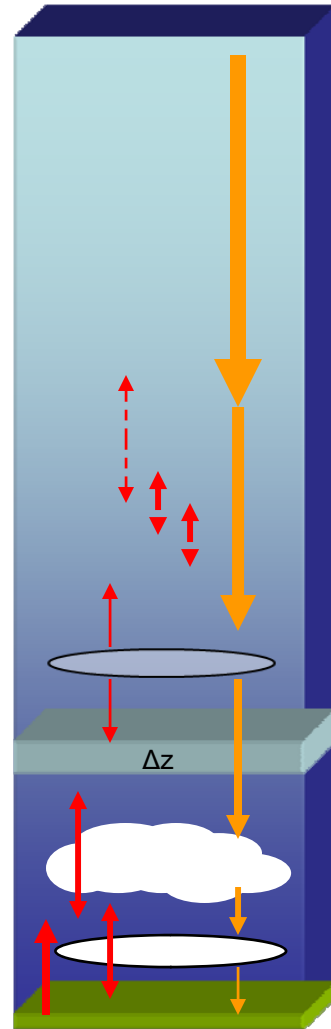
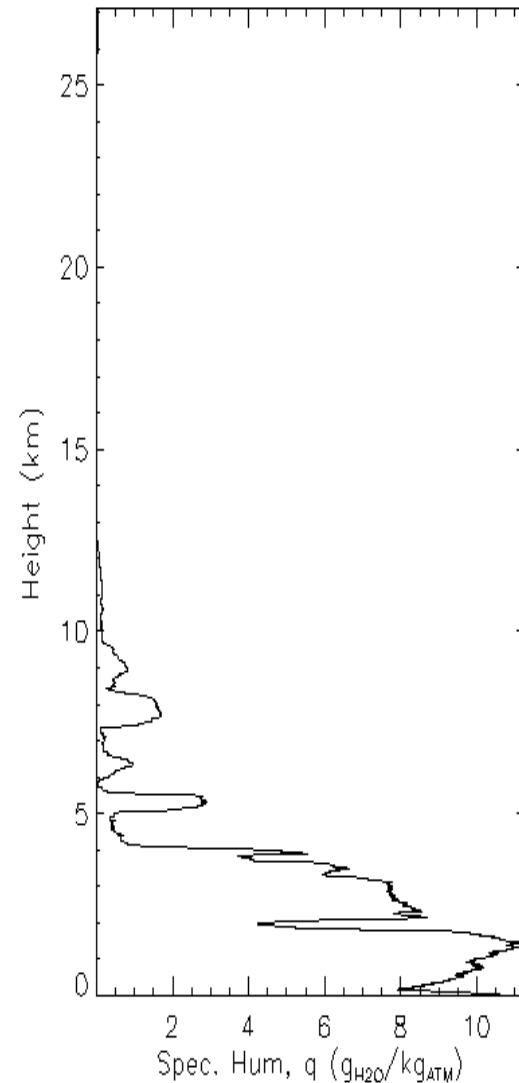
- Thus our depth of water in the column is 3.446 cm or (conveniently!) 34.46 mm

You'll be spending a lot of time in Meteorology Classes doing Unit Conversions between units used for Calculating, and units used in Discussion and Understanding.



Other Uses of the Integral and Summing

- We also use this application of the integral to add up the contributing components of atmospheric and solar radiation coming from, and traveling through, the atmosphere.
- You'll also see it when we do budgets where we tally up the properties within a volume or region



Bored? Oh! we can fix that!

- When you are at a point where you've covered integration by substitution consider this:
- Instead of Z (height) we often use p (pressure) as a vertical coordinate (this is often the case in meteorology. (Pressure decreases with height so you'll see a minus sign). This relationship is called the hydrostatic approximation. You will get it in thermo, dynamics, synoptics and atmospheric physics.

$$\frac{dp}{dz} = -\rho_a g$$

- Instead of integrating from $z=0$ to a fictional $z="z_{top}"$, do it from $p=p_{sfc}$ to $p=0$.

- Hint: this is what you'll get:
$$W = -\frac{1}{g} \int_{p_{surface}}^{p_{top}=0} q dp$$

Got Questions

- Talk to
 - Dr Helsdon and Dr Detwiler: Atmospheric Physics and Thermodynamics