TABLE B.7 Properties of Superheated Steam^a

P(bar)		Sat'd	Sat'd	Temperature (°C)→							
$(T_{\text{sat.}}^{\circ}\text{C})$		Water	Steam	50	75	100	150	200	250	300	350
0.0	\hat{H}	_	-	2595	2642	2689	2784	2880	2978	3077	317
(—)	Û	_	_	2446	2481	2517	2589	2662	2736	2812	2890
	Ŷ				-	_		-	-	_	
0.1	\hat{H}	191.8	2584.8	2593	2640	2688	2783	2880	2977	3077	3177
(45.8)	Û	191.8	2438.0	2444	2480	2516	2588	2661	2736	2812	289(
	Ŷ	0.00101	14.7	14.8	16.0	17.2	19.5	21.8	24.2	26.5	28.7
0.5	\hat{H}	340.6	2646.0	209.3	313.9	2683	2780	2878	2979	3076	3177
(81.3)	Û	340.6	2484.0	209.2	313.9	2512	2586	2660	2735	2811	2889
	Û	0.00103	3.24	0.00101	0.00103	3.41	3.89	4.35	4.83	5.29	5.75
1.0	\hat{H}	417.5	2675.4	209.3	314.0	2676	2776	2875	2975	3074	3176
(99.6)	Û	417.5	2506.1	209.2	313.9	2507	2583	2658	2734	2811	2889
	Ŷ	0.00104	1.69	0.00101	0.00103	1.69	1.94	2.17	2.40	2.64	2.87
5.0	\hat{H}	640.1	2747.5	209.7	314.3	419.4	632.2	2855	2961	3065	3168
(151.8)	Û	639.6	2560.2	209.2	313.8	418.8	631.6	2643	2724	2803	2883
	Ŷ	0.00109	0.375	0.00101	0.00103	0.00104	0.00109	0.425	0.474	0.522	0.571
10	\hat{H}	762.6	2776.2	210.1	314.7	419.7	632.5	2827	2943	3052	3159
(179.9)	Û	761.5	2582	209.1	313.7	418.7	631.4	2621	2710	2794	2876
	Ŷ	0.00113	0.194	0.00101	0.00103	0.00104	0.00109	0.206	0.233	0.258	0.282
20	\hat{H}	908.6	2797.2	211.0	315.5	420.5	633.1	852.6	2902	3025	3139
(212.4)	Û	906.2	2598.2	209.0	313.5	418.4	603.9	850.2	2679	2774	2862
	Û	0.00118	0.09950	0.00101	0.00102	0.00104	0.00109	0.00116	0.111	0.125	0.139
40	Ĥ	1087.4	2800.3	212.7	317.1	422.0	634.3	853.4	1085.8	2962	3095
(250.3)	Û	1082.4	2601.3	208.6	313.0	417.8	630.0	848.8	1080.8	2727	2829
	Ŷ	0.00125	0.04975	0.00101	0.00102	0.00104	0.00109	0.00115	0.00125	0.0588	0.0665
60	\hat{H}	1213.7	2785.0	214.4	318.7	423.5	635.6	854.2	1085.8	2885	3046
(275.6)	Û	1205.8	2590.4	208.3	312.6	417.3	629.1	847.3	1078.3	2668	2792
	V	0.00132	0.0325	0.00101	0.00103	0.00104	0.00109	0.00115	0.00125	0.0361	0.0422
80	\hat{H}	1317.1	2759.9	216.1	320.3	425.0	636.8	855.1	1085.8	2787	2990
(295.0)	Û	1306.0	2571.7	208.1	312.3	416.7	628.2	845.9	1075.8	2593	2750
	Ŷ	0.00139	0.0235	0.00101	0.00102	0.00104	0.00109	0.00115	0.00124	0.0243	0.0299
100	\hat{H}	1408.0	2727.7	217.8	322.9	426.5	638.1	855.9	1085.8	1343.4	7
(311.0)	Û	1393.5	2547.3	207.8	311.7	416.1	627.3	844.4	1073.4	1343.4	2926 2702
	Ŷ	0.00145	0.0181	0.00101	0.00102	0.00104	0.00109	0.00115	0.00124	0.00140	0.0224
150	\hat{H}	1611.0	2615.0	222.1	326.0	430.3	641.3	858.1			
(342.1)	Û	1586.1	2459.9	207.0	310.7	414.7	625.0	841.0	1086.2 1067.7	1338.2	2695
	Ŷ	0.00166	0.0103	0.00101	0.00102	0.00104	0.00108	0.00114	0.00123	1317.6 0.00138	2523 0.0115
200	\hat{H}	1826.5	2418.4	226.4	330.0	434.0	644.5	860.4			
(365.7)	Û	1785.7	2300.8	206.3	309.7	413.2	622.9	837.7	1086.7 1062.2	1334.3	1647.1
22121	Ŷ	0.00204	0.005875	0.00100	0.00102	0.00103	0.00108	0.00114	0.00122	1307.1 0.00136	1613.7 0.00167
$221.2(P_c)$	\hat{H}	2108	2108	228.2	331.7	435.7	645.8	861.4			
$374.15)(T_c)$	Û Ŷ	2037.8	2037.8	206.0	309.2	412.8	622.0	836.3	1087.0 1060.0	1332.8	1635.5
250		0.00317	0.00317	0.00100	0.00102	0.00103	0.00108	0.00114	0.00122	1302.9 0.00135	1600.3 0.00163
250	\hat{H}			230.7	334.0	437.8	647.7	862.8			
(-)	Ŷ		H.T	205.7	308.7	412.1	620.8	834.4	1087.5 1057.0	1331.1	1625.0
200				0.00100	0.00101	0.00103	0.00108	0.00113	0.00122	1297.5 0.00135	1585.0 0.00160
300	\hat{H} \hat{U}	HE AL	1 1 8	235.0	338.1	441.6	650.9	865.2			
	Ŷ	_		205.0	307.7	410.8	618.7	831.3	1088.4 1052.1	1328.7	1609.9
500	Ĥ			0.0009990	0.00101	0.00103	0.00107	0.00113	0.00121	1288.7 0.00133	1563.3 0.00155
()	Û			251.9	354.2	456.8	664.1	875.4			
()	Ŷ			202.4 0.0009911	304.0	405.8	611.0	819.7	1093.6	1323.7	1576.3
1000	Ĥ				0.00100	0.00102	0.00106	0.00111	1034.3 0.00119	1259.3	1504.1 0.00144
(—)	Û	_		293.9 196.5	394.3	495.1	698.0	903.5		0.00129	
	V	_		0.0009737	295.7 0.0009852	395.1	594.4	795.3	1113.0	1328.7	1550.5
				0.0007131	0.0009852	0.001000	0.00104	0.00108	999.0	1207.1	1419.0

^aAdapted from R. W. Haywood, *Thermodynamic Tables in SI (Metric) Units*, Cambridge University Press, London, 1968. Water is a liquid in the enclosed region between 50°C and 350°C. $\hat{H} = \text{specific enthalpy (kJ/kg)}$, $\hat{V} = \text{specific internal energy}$ (kJ/kg), $\hat{V} = \text{specific internal energy}$.