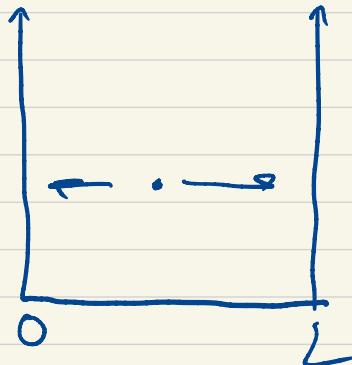


## Lecture 6 - Particle-in-a-box

Apply QM model to very simple system.

$$V(x) = \begin{cases} 0 & , 0 < x < L \\ \infty & , x \leq 0, x \geq L \end{cases}$$



Schrödinger eq

$$\left\{ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} + V(x) \right\} \psi(x) = E \psi(x)$$

Boundary conditions  $\psi(0) = \psi(L) = 0$

$\infty$  potential  $\rightarrow$  impenetrable

$$-\frac{d^2}{dx^2} \psi(x) = \frac{2mE}{\hbar^2} \psi(x)$$

$$\psi(x) = c_1 \sin kx + c_2 \cos kx$$

$$\begin{aligned}\psi'(x) &= c_1 k \cos kx - c_2 k \sin kx \\ \psi''(x) &= -c_1 k^2 \sin kx - c_2 k^2 \cos kx \\ &= -k^2 \cdot \psi(x)\end{aligned}$$

$$k^2 = \frac{2mE}{\hbar^2} , E = \frac{\hbar^2 k^2}{2m}$$

Boundary conditions

$$\psi(0) = c_1 \sin(0) + c_2 \cos(0) = 0$$

$$\psi(x) = c_1 \sin kx$$

$$\psi(L) = c_1 \sin kL = 0 \rightarrow kL = n\pi$$

$$n = 1, 2, 3, \dots$$

$$k = \frac{n\pi}{L}$$

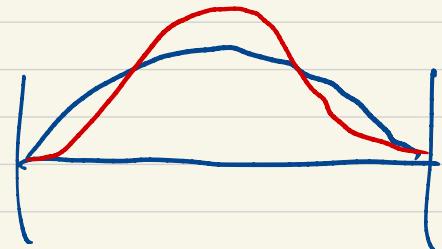
$$\psi_n(x) = c_1 \sin \left( \frac{n\pi x}{L} \right), E_n = \frac{n^2 \pi^2 \hbar^2}{2m L^2}$$

standing waves

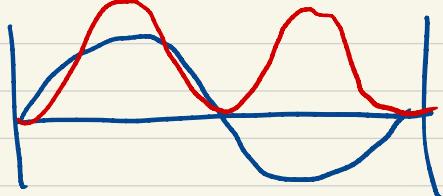
$$\lambda = \frac{2\pi}{k} = \frac{2L}{n}, \lambda = 2L, L, \frac{2L}{3}, \dots$$

$\sim \uparrow$   $\lambda \downarrow$  nodes  $\uparrow$   $E \uparrow$

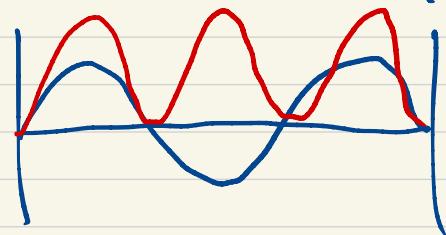
$n=1 E, 2L$



$n=2 4E, L$



$n=3 9E, \frac{2}{3}L$



normalize

$$\langle \psi_n | \psi_n \rangle = c_i^2 \int_0^L \sin^2 \frac{n\pi x}{L} dx$$

$$= c_i^2 \frac{L}{2} = 1$$

$$\Rightarrow c_i = \pm \sqrt{\frac{2}{L}}$$

orthonormal?

$$\langle \psi_m | \psi_n \rangle = \frac{2}{L} \int_0^L \sin \frac{m\pi x}{L} \sin \frac{n\pi x}{L} dx$$

○  $n \neq m$

# Wilson Ho paper

## position + momentum

$$\hat{p} \times \psi(x) \neq \psi(x)$$

$$p = \sqrt{2mE} = \pm n \frac{\pi \hbar}{L}$$

$$\psi(x) = \frac{e^{+i n \pi \hbar / L} + e^{-i n \pi \hbar / L}}{2}$$

superposition

$\hat{x} \psi(x) \neq \psi(x)$  } neither precise  
position or  
momentum  
known

$$\Delta x \sim L \Rightarrow \Delta p_x \sim \frac{\hbar}{2L}$$

By uncertainty principle,  
confined particle must have  
momentum and thus a  
kinetic energy!

## Zero point energy

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} \neq 0 !!$$

example

	<u>car</u>	<u>e<sup>-</sup></u>
m	1000 kg	$9.1 \times 10^{-31}$ kg
L	10 m	$10^{-9}$ m
$E_1$	$5 \times 10^{-76}$ J $\approx 10^{-52}$ kJ/mol	$6 \times 10^{-20}$ J 0.38 eV <u><math>1 \approx 40</math> kJ/mol</u>

Illustrates correspondence principle  
QM behaviour dissolves into classical behavior @ large sizes.

$$k_B T \approx 0.026 \text{ eV} @ 298 \text{ K}$$

$$N_{\text{car}} \approx 3 \times 10^{27} \quad N_{e^-} \approx 4$$

To center box, apply  $u = x - \frac{L}{2}$

$$-\frac{L}{2} < u < \frac{L}{2}$$

$$x = u + \frac{L}{2}$$

$$\sin \frac{n\pi x}{L} \rightarrow \sin \left\{ \left( \frac{n\pi}{L} \right) \left( u + \frac{L}{2} \right) \right\}$$

$$\rightarrow \sin \frac{n\pi u}{L} \cos \left( \frac{n\pi}{2} \right) + \cos \frac{n\pi u}{L} \sin \left( \frac{n\pi}{2} \right)$$

$$n=1 \quad \psi_1 = \cos \frac{\pi u}{L}$$

$$n=2 \quad \psi_2 = -\sin \frac{2\pi u}{L}$$

$$n=3 \quad \psi_3 = -\cos \frac{3\pi u}{L}$$

...

- - - - - - - - - - - - - - -

$$- \text{or} - \quad \psi = c_1 e^{iku/L} + c_2 e^{-iku/L}$$

$$n=1 \quad k=\pi \quad c_1 = -c_2$$

$$n=2 \quad k=2\pi \quad c_1 = c_2$$

...

## Multiple dimensions

Suppose box is cubic.

$$V(x, y, z) = \begin{cases} 0, & |x|, |y|, |z| < \frac{L}{2} \\ \infty, & |x| \text{ or } |y| \text{ or } |z| \geq \frac{L}{2} \end{cases}$$

$$\left\{ -\frac{\hbar^2}{2m} \nabla^2 + V(x, y, z) \right\} \psi(x, y, z) = E \psi(x, y, z)$$

$\psi$  vanishes outside box

Separate variables  $\psi(x, y, z) = X(x) Y(y) Z(z)$

Substitute & divide by  $X Y Z$

$$-\frac{\hbar^2}{2m} \left\{ \underbrace{\frac{1}{X} \frac{d^2}{dx^2} X}_{E_x} + \underbrace{\frac{1}{Y} \frac{d^2}{dy^2} Y}_{E_y} + \underbrace{\frac{1}{Z} \frac{d^2}{dz^2} Z}_{E_z} \right\} = E$$

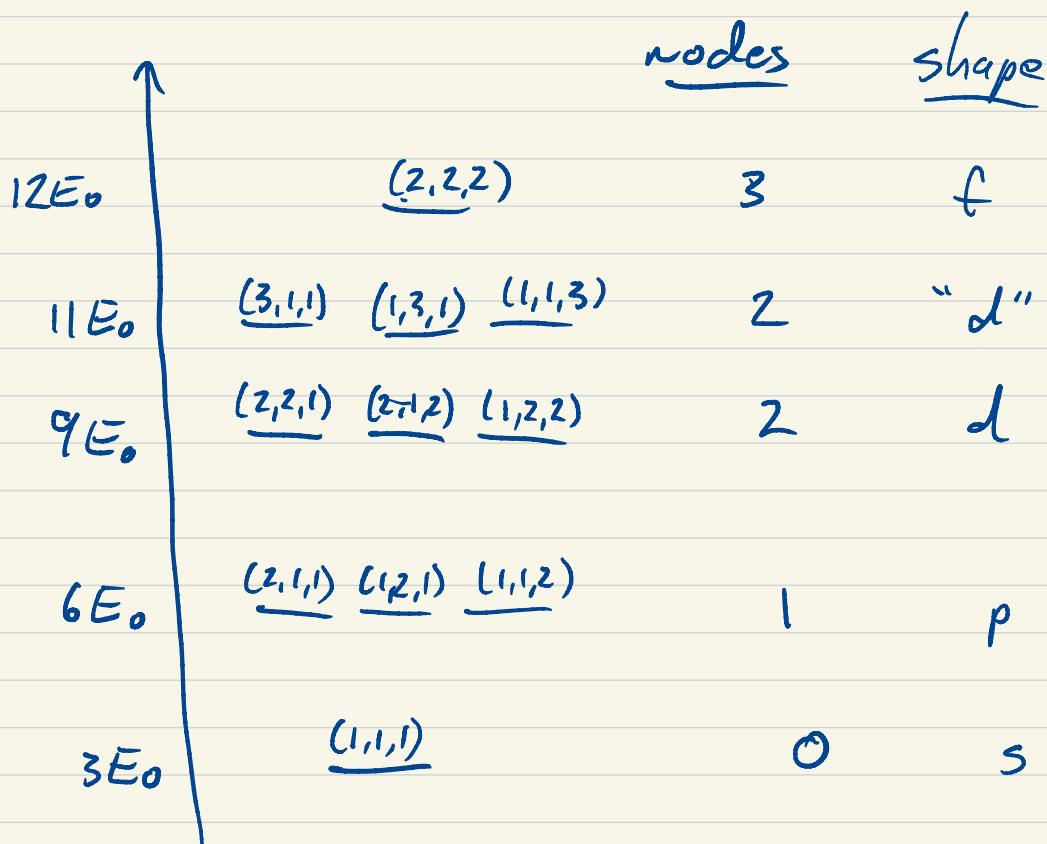
$$-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} X(x) = E_x X(x) \quad + y + z$$

identical to 1D problem...

$$X_e(x) = \sqrt{\frac{2}{L}} \sin \frac{e\pi x}{L} \quad E_e = \frac{e^2 \pi^2 \hbar^2}{2m L^2}$$

$$\Psi_{emn}(x, y, z) = X_e(x) Y_m(y) Z_n(z)$$

$$E = \frac{(e^2 + m^2 + n^2) \hbar^2 \pi^2}{2m L^2}$$



See python notebook

each dimension  $\rightarrow$  new Q.N.  
 $\rightarrow$  nodes

symmetry of problem leads to  
degenerate solutions

"degeneracy"

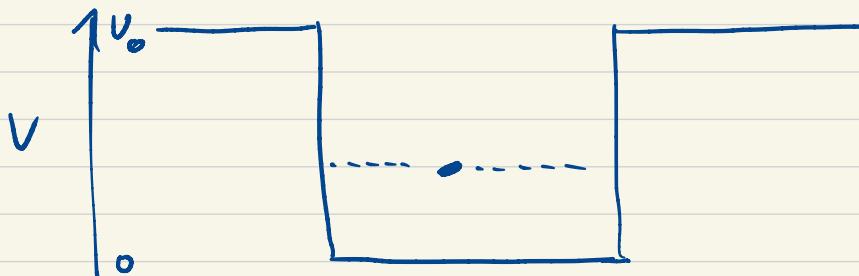
3d box states look like... an atom!

- - - - -  
independent e<sup>-</sup> approximation  
add e<sup>-</sup> from bottom-up  
e<sup>-</sup> have "spin":  $\uparrow$  &  $\downarrow$   
add 2/energy level

qualitatively useful, neglects  
e<sup>-</sup> - e<sup>-</sup> interactions.

Will improve later.

## finite depth well



Classically, 2 possibilities :

$E < V_0$ , bounce between walls

$E > V_0$ , fly right past well

QM? Boundary conditions change

If  $E < V_0$ , have  $\lim_{x \rightarrow \pm\infty} \psi(x) = 0$   
(bound state)

In well region ("classical" region)

$$-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \psi = E\psi$$

$$\psi \sim \sin kx \sim e^{ikx} - e^{-ikx}, \quad k = \sqrt{\frac{2me}{\hbar^2}} > 0$$

## Outside well

$$\left\{ -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \psi + V_0 \psi \right\} = E \psi$$

$$\frac{d^2}{dx^2} \psi = -\frac{2m(E-V_0)}{\hbar^2} \psi$$

$\qquad\qquad\qquad$

$> 0$

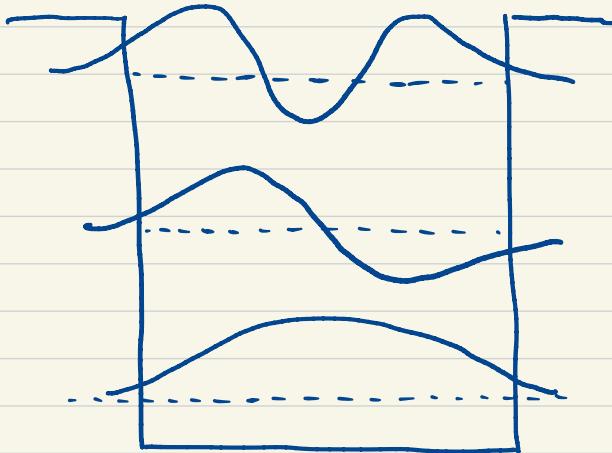
$$\frac{d^2}{dx^2} \psi = x^2 \psi \quad x = \sqrt{\frac{2m(V_0 - E)}{\hbar^2}}$$

$$\rightarrow 4 \sim e^{kx} + e^{-kx} \neq 0 !!$$

exponential decay in "forbidden" region

To solve exactly, would have to require wavefunction & derivatives to match @ boundaries.

would be nice to write this code !!

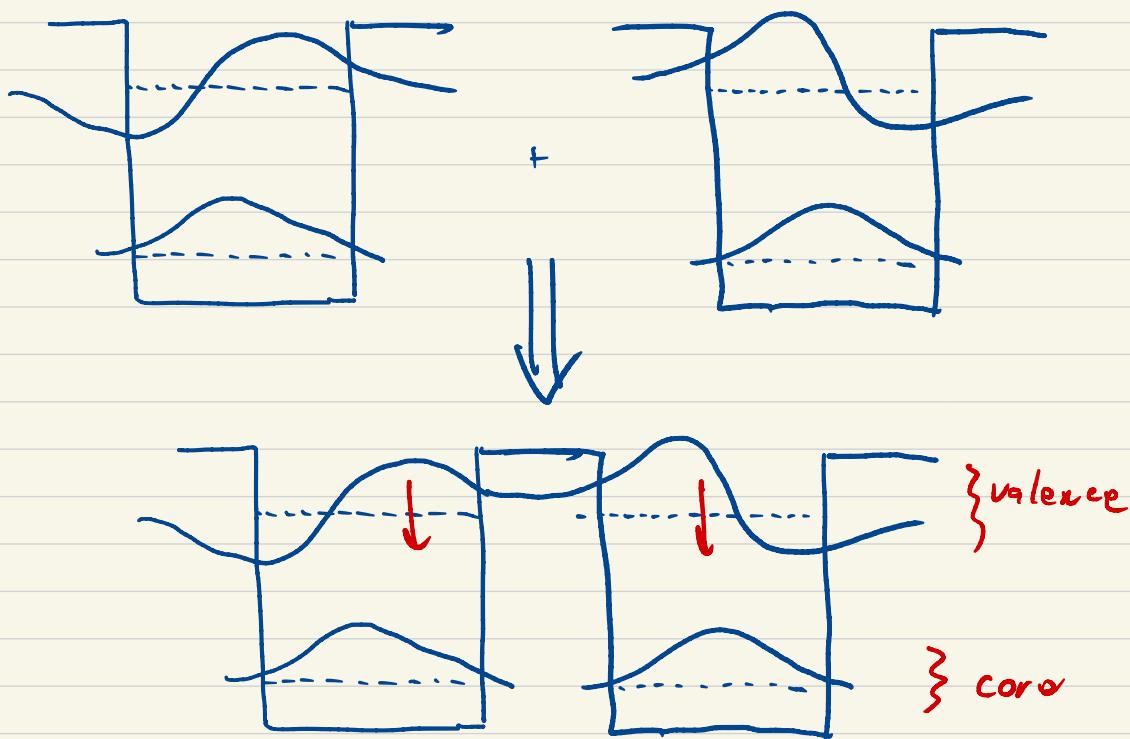


$V_0 - E$  small  
 $K$  small  
decay slow

$V_0 - E$  big  
 $K$  big  
decay fast

- probability to explore "forbidden" region  $\neq 0$
- probability  $\uparrow$  as  $V_0 - E \downarrow$
- energy of states  $\downarrow$  relative to infinite well  
explore more space  $\rightarrow KE \downarrow \star$
- retain nodal structure
- finite number of bound states.

Better representation of an atom ...



"overlap", or constructive interference, through barrier expands region explored by  $e^-$  and lowers energy

That's a chemical bond

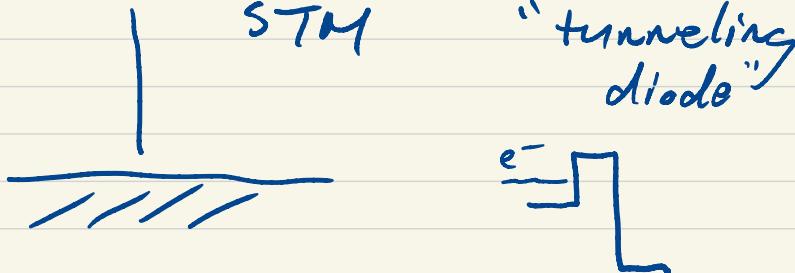
$$E_{A-B} < E_A + E_B$$

Only works for wavefunctions that sneak out of well ("valence") and that match each other energetically "homonuclear" bond.

contrast "heteronuclear"

Chemical bonding is one example of QM tunnelling *hugely important!*

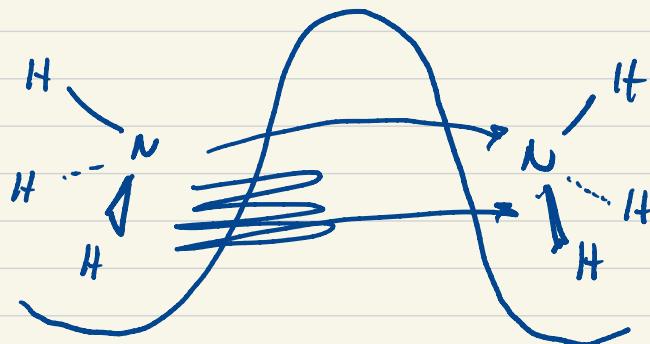
Other examples involving  $e^-$



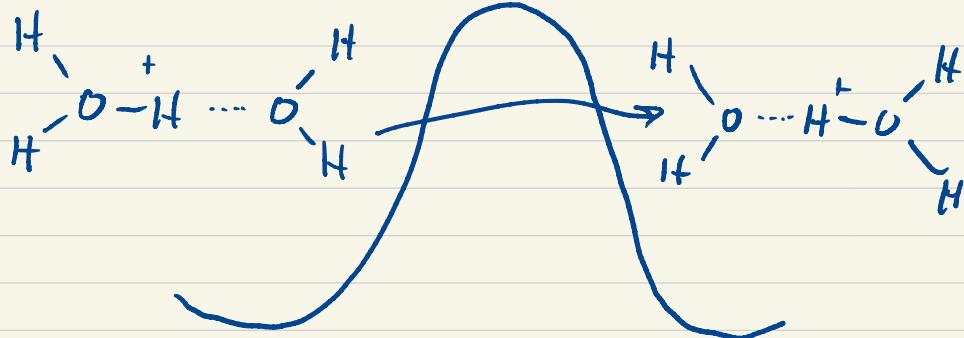
$K \propto m^{1/2} \rightarrow$  probability to tunnel ↓ as  $m \uparrow$

Only other particle for which tunneling is practically important in a proton

eg



or



will vibrate in one well, occasionally "appear" on other side !!