

Probability

Flipping a coin

two outcomes H

$\frac{1}{2}$

T

$\frac{1}{2}$

probability

postulate of equal a priori probabilities

probability gives relative likelihood
of possible outcomes. UNLESS

Flipping two coins

H H }
H T }
T H }
T T }

4 outcomes

distinguishability!!

1 state

Define $P(i)$ to be probability of
getting i heads

i is a random variable

$$\begin{aligned} P(0) &\propto 1 \\ P(1) &\propto 2 \\ P(2) &\propto 1 \end{aligned} - \left\{ \begin{array}{l} TH \\ HT \end{array} \right. \quad \begin{array}{l} \text{degenerate} \\ \text{outcomes} \end{array}$$
$$P(i>2) = 0$$

Normalize P s.t. $\sum \text{prob} = 1$

$$\tilde{P}(i) = \frac{P(i)}{\sum_i P(i)}$$

normalization factor

We are always free to normalize.

For discrete events

$$\tilde{P}(i) = \frac{\text{degeneracy of } i}{\text{total events/outcomes}}$$

Coin toss example of binomial trial

n tosses $\rightarrow 2^n$ possible outcomes

i heads out of n $\rightarrow {}^n C_i$ or $\binom{n}{i}$

$$\frac{n!}{i!(n-i)!} \quad (\# \text{ of ways to arrange } n \text{ coins})$$

$+ n-i$ tails)

$$\tilde{p}(i) = \frac{{}^n C_i}{2^n}$$

What happens to $\tilde{p}(i)$ as $n \rightarrow \infty$?

Example of discrete probabilities

50 students in class, 10 defense is 5 random students, what are your odds of being in that 1st group?

How many groups of 5? $\binom{50}{5} = 2118760$

How many groups contain you?

$$\binom{49}{4} = 211876$$

$P(\text{you}) = 1/10$ (obvious, $1/10$ chance each week!!)

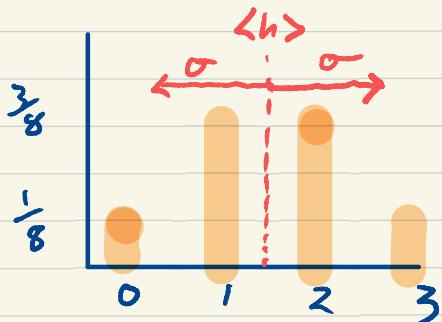
Probability all sophomores?

Back to coins

$$n=3 \quad 2^3=8$$

$\# h$

$$\binom{3}{0} = \binom{3}{3} = 1 \quad \binom{3}{1} = \binom{3}{2} = 3$$



most probable? $\tilde{P}(1) = \tilde{P}(2) = \frac{3}{8}$

expected value?

$$\begin{aligned} \langle h \rangle &\equiv \sum_h h \tilde{P}(h) = 0 \cdot \frac{1}{8} + 1 \cdot \frac{3}{8} + 2 \cdot \frac{3}{8} + 3 \cdot \frac{1}{8} \\ &= \frac{3}{2} \quad \text{not a possible value!} \end{aligned}$$

Can find expected value of any func of h

$$\langle f(h) \rangle = \sum_h f(h) \tilde{P}(h) = \sum_h f(h) P(h) / \sum_h P(h)$$

$$\text{mean} = \langle x \rangle$$

$$\text{mean squared} = \langle x^2 \rangle$$

$$\text{rms} = \langle x^2 \rangle^{1/2}$$

$$\text{variance } \sigma^2 = \langle x^2 \rangle - \langle x \rangle^2$$

$$\text{std deviation } \sigma = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

Gives "width of distribution"

3 Coins example

$$\begin{aligned}\langle h^2 \rangle &= 0^2 \cdot \frac{1}{8} + 1^2 \cdot \frac{3}{8} + 2^2 \cdot \frac{3}{8} + 3^2 \cdot \frac{1}{8} \\ &= \frac{3}{8} + \frac{12}{8} + \frac{9}{8} = 3\end{aligned}$$

$$\langle h^2 \rangle^{1/2} = \sqrt{3} \neq \langle h \rangle !!$$

$$\begin{aligned}\sigma^2 &= \langle i^2 \rangle - \langle i \rangle^2 \\ &= 3 - \left(\frac{3}{2}\right)^2 = \frac{3}{4}\end{aligned}$$

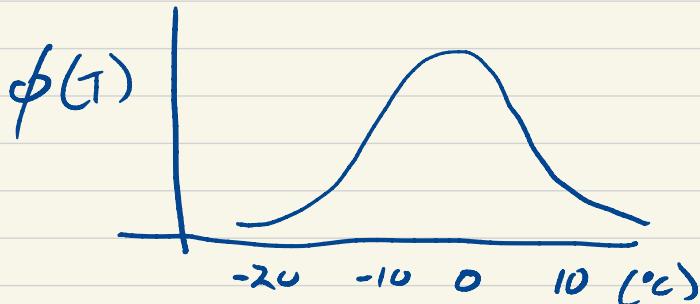
$$\sigma = \sqrt{3}/2$$

Interesting to think about how
 $\langle h \rangle$ & σ_h grow with n ...

Continuous probability densities

High T in S.B. in January

Could collect data, any real # possible



$\phi(T)$ has units $^{\circ}\text{C}$ probability density

Probability of $T = 0.000^{\circ}\text{C} \rightarrow 0!!$

Evaluate ranges

$$P(-10 < T < 0) \propto \int_{-10}^0 \phi(T) dT$$

unitless

$$P(T > 10) \propto \int_{10}^{\infty} \phi(T) dT$$

Can normalize

$$\tilde{\phi}(T) = \frac{\phi(T)}{\int_{-273}^{\infty} \phi(T) dT}$$

Example

$$\phi(T) = e^{-(T-20)^2/50}$$

see python notebook

Boltzmann probability central
probability dist of physical chem.
Discovered by Ludwig Boltzmann in 1800's.

Determines probability of an object to have an energy E , parametric in temperature T .

$$p(E) \propto e^{-E(q,p)/k_B T} \quad \text{always a ftn of some internal variables}$$

$$k_B = 1.3807 \times 10^{-23} \text{ J/K} \quad \text{Boltzmann's const.}$$

energy? fundamental physical quantity comes in many forms
kinetic, potential, light

ftn of internal DOFs $E(q,p)$

Conserved in a closed system

energy units

$$\text{SI: J: } \frac{\text{kg m}^2}{\text{s}^2} \rightsquigarrow \text{kJ/mol}$$

$$\text{eV} = 1.6 \times 10^{-19} \text{ J} = 96.5 \text{ kJ/mol}$$

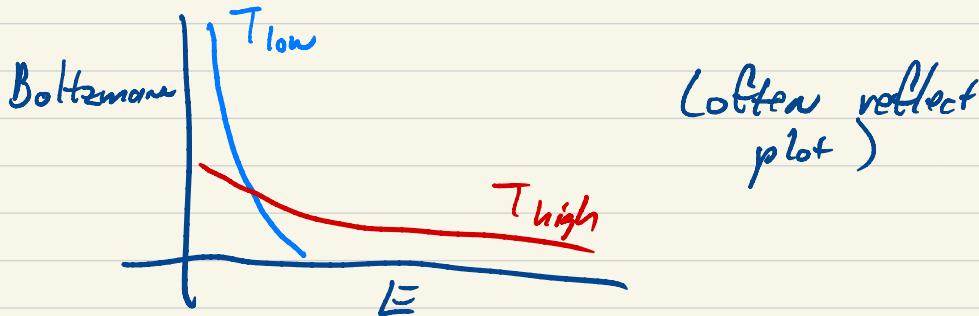
temperature units

absolute T, Kelvin

"Temperature" is a characteristic of a system that follows Boltzmann statistics. Characteristic of all large systems @ equilibrium.

Sets an energy scale

$\frac{1}{T}$	$\frac{k_B T}{0}$
0	0
298 K	$0.026 \text{ eV} = 2.5 \text{ kJ/mol}$
1000 K	$0.086 \text{ eV} = 8.3 \text{ kJ/mol}$



Ex: gases in atmosphere

gravitational potential energy:

$$E(h) = mgh$$

Probability density to be at height h at temperature T ?

$$\phi(h) \propto e^{-mgh/kT}$$

Normalize:

$$\int_0^{\infty} \phi(h) dh = k_B T / mg$$

$$\tilde{\phi}(h) = \frac{mg}{kT} e^{-mgh/kT}$$

Table 2: car vs gas molecule

	<u>car</u>	<u>molecule</u>
m	$\sim 1000 \text{ kg}$	$\sim 10^{-26} \text{ kg}$
mgh	9800 J $6.1 \times 10^{-22} \text{ eV}$	$9.8 \times 10^{-26} \text{ J}$ $6.1 \times 10^{-7} \text{ eV}$

$$kT @ 298K \quad 0.026 \text{ eV}$$

$$mgh/kT \quad 2.4 \times 10^{24} \quad 2.3 \times 10^{-5}$$

$$\tilde{\phi}(1m)/\tilde{\phi}(0m) \quad 0 \quad 0.99998$$

$$\langle h \rangle \quad 0 \quad 42 \text{ km !!}$$

Size matters !!

Basis of barometric law that pressure decays \sim exponentially w/ altitude. (right only if isothermal !!)

$$\text{Pres}(h.) = \frac{N(h>h_0) \cdot m \cdot g}{\text{Area}} \rightarrow$$

$$\frac{\text{Pressure}}{\text{Press}(0)} = e^{-mgh/kT}$$

Does atmosphere go forever? No...
gas molecules can exceed escape velocity and leak out. No He or H₂ in atmosphere !!

equipartition - Boltzmann applies to kinetic + potential energies each alone

Ex : kinetic energy

$$-\infty < v_z < \infty$$

$$\int_0^{v_z} dv_z$$

$$E_{\text{kin}}(v_z) = \frac{1}{2} m v_z^2$$

$$\phi(v_z) \propto e^{-\frac{1}{2} m v_z^2 / kT}$$

normalize

$$\int_{-\infty}^{\infty} \phi dv = \left(\frac{2\pi kT}{m} \right)^{1/2}$$

$$\tilde{\phi}(v_z) = \left(\frac{m}{2\pi kT} \right)^{1/2} e^{-mv_z^2/2kT}$$

Bell shaped curve \rightarrow "Gaussian"

$$G(x) = \frac{1}{\sigma\sqrt{2\pi}} e^{-(x-\mu)^2/2\sigma^2}$$

by inspection

$$\langle v_z \rangle = \mu = 0$$

$$\langle v_z^2 \rangle = \sigma^2 = k_B T/m$$

See Figure 2

car molecule

m

1000 kg

10^{-26} kg

$$\left(\frac{kT}{m}\right)^{1/2}$$

2×10^{12} m/s

640 m/s (!)

@298K

rms velocity

Expectation values

"average" value of a random variable

e.g. 2 coins

$$\langle i \rangle = \bar{i} = \frac{1}{4} \cdot 0 + \frac{1}{2} \cdot 1 + \frac{1}{4} \cdot 2 = 1$$

Generally $\langle i \rangle = \sum_i i \tilde{P}(i)$

$$= \sum_i i p(i) / \sum_i p(i)$$

Can evaluate for functions of i :

$$\langle i^2 \rangle = \sum_i i^2 \tilde{P}(i) = \frac{3}{2}$$

$$\langle i \rangle^2 \neq \langle i^2 \rangle !!$$

Generally $\langle f(i) \rangle = \sum_i \tilde{P}(i) \cdot f(i)$

- or -

$$\int \tilde{\phi}(x) \cdot f(x) dx$$

mean = $\langle x \rangle$

mean squared = $\langle x^2 \rangle$

rms = $\langle x^2 \rangle^{1/2}$

most probable?

variance $\sigma^2 = \langle x^2 \rangle - \langle x \rangle^2$

std deviation $\sigma = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$

Gives "width of distribution"

2 Coins example

$$\begin{aligned}\sigma^2 &= \langle i^2 \rangle - \langle i \rangle^2 \\ &= \frac{3}{2} - 1 = \frac{1}{2}\end{aligned}$$

T distribution example

$$\phi(T) = e^{-(T-20)^2/50}$$

see python notebook