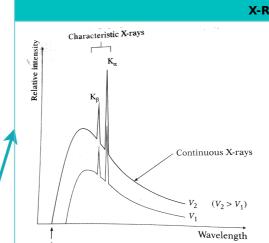
### Production of X-Rays

X-Rays are produced when cathode rays strike a target

Energy is released (photon) when accelerated electrons hit the target (tungsten).

Approximately 5% energy is released as form of heat, dissipated by cooling fins.



X-Rays

Bragg's Law

X-Rays

# X-Rays Spectra

Continuous(Bremsstrahlung) Spectrum $qV=hf_{max}=rac{hc}{\lambda_{min}}$ 

**Characteristic Line Spectrum** 

$$\Delta E = hf$$

## Bragg's Law

$$m\lambda = 2d\sin heta_m$$

$$\frac{m\lambda}{2d} \le 1 \quad m \le \frac{2d}{\lambda}$$

#### **Crystal Atomic Spacing**

Diffraction of x-ray occurs in the crystal atomic spacing.

Used to calculate wavelength of x-rays.

#### **Properties of X-Rays**

Neutral, not deflected by electric or magnetic fields

Can penetrate through solid matter

Can cause certain compounds to fluorocence, i.e. Zinc Sulphide

Can ionize atoms.

Can cause photoelectric emmision.

#### **Application of X-Rays**

Detection of bullet in patient's leg

Observation and photography of the healing of a broken bone.

Detection of contraband in baggage.

Discrimination between authentic gems and artificial gems.

Detection of pearl in oyster.