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Functioning?

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Plant Preference for Ammonium versus Nitrate: A Neglected **Determinant of Ecosystem Functioning?**

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ABSTRACT: Although nitrogen (N) availability is a major determinant of ecosystem properties, little is known about the ecological importance of plants' preference for ammonium versus nitrate (β) for ecosystem functioning and the structure of communities. We modeled this preference for two contrasting ecosystems and showed that β significantly affects ecosystem properties such as biomass, productivity, and N losses. A particular intermediate value of β maximizes the primary productivity and minimizes mineral N losses. In addition, contrasting β values between two plant types allow their coexistence, and the ability of one type to control nitrification modifies the patterns of coexistence with the other. We also show that species replacement dynamics do not lead to the minimization of the total mineral N pool nor the maximization of plant productivity, and consequently do not respect Tilman's R* rule. Our results strongly suggest in the two contrasted ecosystems that β has important consequences for ecosystem functioning and plant community structure.

Keywords: ammonium, biological invasion, coexistence, nitrate, nitrification, resource partitioning.

Introduction

Nitrogen (N) is generally considered as the main factor limiting plant growth in many temperate terrestrial ecosystems (Vitousek and Howarth 1991). Most studies on ecosystem function have considered N as a single resource despite the fact that plants are able to assimilate different forms of N, such as ammonium (NH₄), nitrate (NO₃),

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and organic N (Begon et al. 1998; Marschner 2008). The implications of plant organic N nutrition (Hodge et al. 2001; Weigelt et al. 2005) and symbiotic N fixation (Vitousek et al. 2002; Menge et al. 2008) for ecosystem function and plant communities have received considerable attention (Schimel and Bennett 2004). In contrast, few studies have examined the consequences of plant preference for NH₄ versus NO₃ on ecosystems' functioning, that is, at ecosystem scale, though this preference is well documented from the physiological point of view (Williams and Miller 2001; Marschner 2008) and the processes controlling the availability of NH₄ and NO₃ in soils are widely studied (Frank and Groffman 2009). More precisely, most studies focusing on the ecological consequences of plant preference for different N forms do not distinguish explicitly NH₄ from NO₃ but rather consider "inorganic N" versus "organic N" (Harrison et al. 2007; Kahmen et al. 2008).

From an energetic point of view, NH₄ uptake and assimilation are less costly than NO₃ uptake and assimilation (Salsac et al. 1987). This could constitute an advantage for plants being very competitive for NH₄ absorption. However, some studies show that NH₄⁺, when it is the only N source for plants, can cause severe toxicity symptoms (de Graaf et al. 1998; Britto et al. 2001). This NH₄ toxicity could counterbalance the energetic advantage in taking up NH₄ rather than NO₃. Moreover, since plants may also exhibit large demands in K+, Ca2+, and other cations, the absorption of NO₃ can lead to a more even charge balance for plants than the absorption of NH₄⁺. NO₃⁻ is also easier to store in plant tissues than NH₄⁺. Furthermore, due to its electronic charge, NH₄ is often adsorbed to organo-

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mineral complexes (Brady and Weil 2001; Marschner 2008) and thus can be retained in the soil. Conversely, the negative charge of NO₃⁻ makes this form of N much more mobile and thus more prone to loss and thereby less available for plants (Brady and Weil 2001; Marschner 2008). On the other hand, this higher mobility leads to a more rapid diffusion to roots and thus to an easier access for plants to the NO₃⁻ that is not lost than to the NH₄⁺ that is not lost. These physiological costs and physical limitations constrain plant strategy in opposite directions and might thus impose a trade-off between the absorption of the two N forms as suggested by Maire et al. (2009).

Some plants are able to inhibit (Lata et al. 1999, 2000, 2004; Subbarao et al. 2007a, 2007b) or stimulate (Lata et al. 2000; Hawkes et al. 2005) nitrification, that is, the microbially mediated conversion of NH₄ into NO₃, thereby altering the relative amount of NH₄ and NO₃ available in the soil for their own mineral N nutrition, as well as for the mineral N nutrition of their competitors. This should have large ecological consequences, and a recent model assessing the impact of nitrification inhibition by plants on ecosystem functioning (Boudsocq et al. 2009) has reported that such an inhibition of nitrification may increase primary productivity and ecosystem fertility in a sustainable way. Indeed, inhibiting nitrification enhances the conservation of mineral N in the soil, as NO₃ is more prone than NH₄ to being lost from the ecosystem as explained above. This could be of particular importance in highly constrained ecosystems with poor and/or welldrained soil. It has also been hypothesized that nitrification-inhibiting plants may also have developed a greater capacity for the absorption of NH₄ than for NO₃, and that the propensity of a species to take up either NH₄ or NO₃ could influence the recycling efficiency and thereby the productivity of the whole ecosystem (Barot et al. 2007; Boudsocq et al. 2009).

Furthermore, recent work supports the hypothesis that N preference may influence plant community structure through changes in competition for N between species (Aanderud and Bledsoe 2009). Indeed, though some species may possess an important plasticity in their absorption of different N forms (Houlton et al. 2007), others show a preference for a particular form, and the relative abundance of different N forms in the soil-due to microbial N transformations in the soil but also to the different forms of atmospheric N inputs (Holland et al. 1999; Galloway et al. 2004)—could affect the coexistence between plant species (Tilman 1985). Indeed, the concepts of resource partitioning and niche complementarity support the idea that species that differ in their resource use are more likely to coexist (Pacala and Tilman 1994; Holt et al. 2008). Ashton et al. (2010) published a study suggesting that these concepts should apply concerning NH₄ and NO₃ uptake

and assimilation. Similarly, controlling nitrification rates could play a key role in the dynamics of plant communities as stimulation and inhibition of this process have been shown to strongly affect plant invasions (Lata et al. 2004; Hawkes et al. 2005; Rossiter-Rachor et al. 2009). Taken together, plant influence on nitrification and preference for NH₄⁺ versus NO₃⁻ should interact to determine the availability of mineral N and plant competitiveness.

Using a modified version of the mathematical model from Boudsocq et al. (2009) the aim of our study is to assess the importance of plants' relative ability to take up NH₄ and NO₃ for ecosystem functioning and plant community structure. More precisely, our objectives are to address these questions: (i) How does the preference of plants for NH₄⁺ versus NO₃⁻ influence primary productivity, plant biomass, and soil mineral N stocks? (ii) Can plant productivity and biomass be maximized for a particular value of plant preference for NH₄ versus NO₃, and if that is the case, what are the ecological mechanisms leading to this maximization? (iii) Can plant types that differ only in their preference for NH₄ versus NO₃ coexist? (iv) Finally, how is the ability of a plant to become invasive or make a system resistant to invasion affected by its preference for NH₄⁺ versus NO₃⁻, and its ability to control nitrification?

The model is parameterized for two contrasting natural ecosystems whose N cycles are well described in the literature: the Pawnee site (Colorado; Woodmansee et al. 1978), a temperate short-grass prairie, and the Lamto site (Ivory Coast), a tropical humid savanna (Abbadie et al. 2006; Boudsocq et al. 2009).

Model and Simulation Approach

Our model (fig. 1) describes the dynamics of N between different compartments: dead organic matter (D), NH₄⁺ (N_A), NO₃⁻ (N_N) and one or two plant types (P_1 and P_2). The model is described by the following set of differential equations:

$$\frac{dP_{i}}{dt} = \beta_{i}uN_{A}P_{i} + (1 - \beta_{i})uN_{N}P_{i} - (d + l_{P} - f)P_{i}, \quad (1)$$

$$\frac{dD}{dt} = R_D + d\sum_i P_i - (m + l_D)D, \qquad (2)$$

$$\frac{dN_{A}}{dt} = R_{A} + mD - \sum_{i} (\beta_{i}P_{i})uN_{A}$$
$$- \left(l_{N_{A}} + ne^{-\sum i_{i}P_{i}}\right)N_{A}, \tag{3}$$

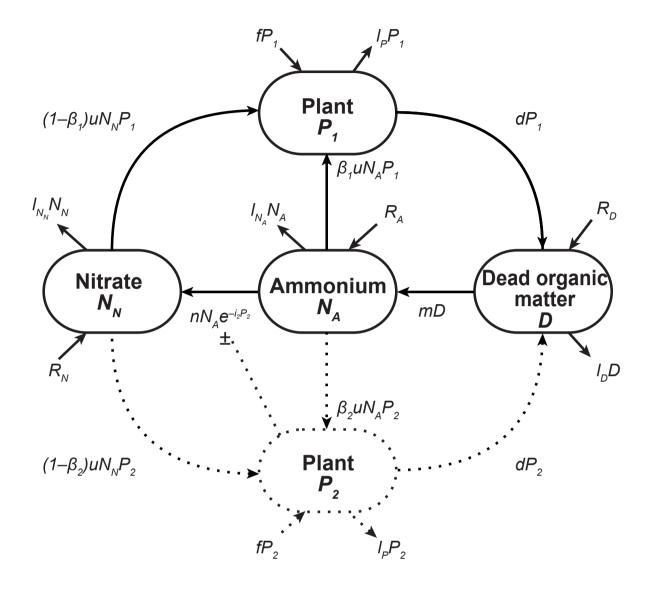


Figure 1: General model of N cycling in a terrestrial ecosystem. The labels associated with each arrow indicate the formula used for the corresponding flux. The dotted boxes and arrows are taken into account for testing coexistence and invasion situations only. Definitions of parameters can be found in table 1.

$$\frac{dN_{\rm N}}{dt} = R_{\rm N} + nN_{\rm A}e^{-\Sigma i_i P_i} - \left[\sum_i (1 - \beta_i)uP_i + l_{N_{\rm N}}\right]N_{\rm N}.$$
 (4)

The primary productivity (Φ) equation is

$$\frac{d\phi}{dt} = \sum_{i} (\beta_i u P_i) N_A + \sum_{i} [(1 - \beta_i) u P_i] N_N + f \sum_{i} (P_i). \quad (5)$$

Model parameters and definitions are given in table 1.

Plants build up their biomass by absorbing N from the two mineral N pools, that is, NH₄⁺ and NO₃⁻, with the rates of $\beta_i u$ and $(1 - \beta_i) u$, respectively. Term β_i represents the preference of the plant i for NH₄⁺, and $(1 - \beta_i)$ is the preference of the plant i for NO₃⁻ (β_i ranges between 0 and 1),

while u represents the N uptake capacity of plants. Note that to stay focused on plant preference for NH_4^+ versus NO_3^- , the model does not allow plants to take up organic N, which is assumed to be negligible in the studied ecosystems (Schimel and Bennett 2004). Plants' uptakes of N are modeled using "donor-recipient" functions, that is, the fluxes are proportional to the sizes of the mineral N stocks and the plant compartments. Note that another formalism, such as Monod equations for the absorption of N, would have been more realistic (donor-recipient functions do not saturate), but in this article, Monod equations do not affect qualitatively the results and make the model much more complex. In turn, plant mortality leads to a flux of N from

Parameter	Dimension	Definition	Value
f	yr^{-1}	Symbiotic N fixation rate	.01
d	yr^{-1}	Plant recycling rate	.258
$l_{\scriptscriptstyle P}$	$ m yr^{-1}$	Plant loss rate	0
R_D	$kg N ha^{-1} yr^{-1}$	Annual organic N input rate	0
m	yr^{-1}	DOM recycling rate	.00654
$l_{\scriptscriptstyle D}$	yr^{-1}	DOM loss rate	.00138
$R_{\rm A}$	kg N ha ⁻¹ yr ⁻¹	Annual NH ₄ input	3
и	ha kg N^{-1} yr ⁻¹	N uptake rate	.01336
n	yr^{-1}	Nitrification rate	.05
$l_{\rm NA}$	${ m yr}^{-1}$	NH ₄ loss rate	.05
$R_{\rm N}$	kg N ha ⁻¹ yr ⁻¹	Annual NO ₃ input	3
$l_{_{ m NN}}$	yr^{-1}	NO ₃ loss rate	.15
β	Dimensionless	Affinity for NH ₄ ⁺ versus NO ₃ ⁻	NA
i	ha kg N^{-1}	Nitrification inhibition/stimulation rate	$\pm .02$

Table 1: Parameters of the Pawnee site

Note: DOM = dissolved organic matter.

the P to the D compartment at a rate d. This dead organic matter (D; including litter and dead root biomass) is mineralized, leading to a flux of N to the N_A compartment, with a rate m. NH_4^+ is then absorbed by plants, or transformed into NO_3^- (i.e., nitrified) at a rate n. Plant type P_3 is assumed to be able to control the rate of nitrification, either inhibiting (Lata et al. 1999, 2000, 2004; Subbarao et al. 2007a, 2007b) or stimulating (Lata et al. 2000; Hawkes et al. 2005) by a coefficient i, so that the flux of nitrification is modeled as $nN_{A}e^{-\Sigma i_{i}P_{i}}$ (i > 0 when plants inhibit nitrification and i < 0 when plants stimulate nitrification). Finally, NO₃ is absorbed by plants.

In natural ecosystems, losses occur from these four compartments. Fire can cause significant losses from the P and D compartments (Bowman et al. 2009). Erosion and leaching can also lead to losses from the D compartment (Marschner and Rengel 2007). NH₄ can be lost through volatilization, making it unavailable for plants. Finally, NH₄ and NO₃ can be lost through leaching, and NO₃ can also be lost through denitrification (Marschner and Rengel 2007). These losses are included in our model with the rates of l_P , l_D , l_{NA} , and l_{NN} for losses from the P, D, N_A , and N_N compartments, respectively.

There are two main sources of N inputs to the ecosystem: atmospheric N deposition, and the biotic fixation of atmospheric N₂ by symbiotic and nonsymbiotic microorganisms (Galloway et al. 2004). Mineral atmospheric N deposition provides constant N inputs to the N_A and N_N compartments (R_A and R_N , respectively). Organic N deposition and N nonsymbiotic fixation are modeled by a constant N input to the D compartment of R_D . Finally, the symbiotic N fixation of N₂ provides an N input to the P compartment proportional to a constant rate f and to the size of the P compartment. Stocks are expressed in

kilograms of N per hectare and fluxes in kilograms of N per hectare per year.

In order to study the effect of β_i (preference for NH₄) on ecosystem properties, we determine the mathematical equations corresponding to the compartments' steady states by setting the system of differential equations to 0 and define the model equilibrium conditions (appendix A, available online). First considering only one plant type (P_1) , we then determine by numerical simulations the value of β_1 leading to the highest biomass and plant productivity for a given set of parameters (noted β_{opt1}) and analyze the sensitivity of β_{opt1} to ecosystem parameters including nitrification rates, NO₃⁻ loss rates, and NH₄⁺ and NO₃ deposition rates (app. B, available online). To do so, we use Mathematica, version 8.0, to find the value of β for which each equilibrium compartment derivative with respect to β is null (command: Solve [D (<compartment, $\beta > 0$ = 0]). Then, in order to study the ecological importance of plant NH₄ versus NO₃ uptake and its impact on plant invasion, we add a second plant compartment that differs from the first one by its preference for NH₄ and NO₃ (fig. 1) and by its ability to control nitrification. We thus distinguish two plant compartments $(P_1 \text{ and } P_2)$ having two distinct preferences for NH₄⁺ versus NO₃⁻ (β_1 and β_2 , respectively). We then perform simulations with one plant type being the "resident" species (i.e., its starting biomass is 100 kg N ha⁻¹, which is close to the plant biomass observed in the Pawnee and Lamto sites; Woodmansee et al. 1978; Abbadie et al. 2006) and the other one being the "invader" type (i.e., its starting biomass is 0.01 kg N ha⁻¹, 10,000-fold lower than the resident). The numerical simulations of the differential equations (eqq. [1]– [4]) have a time step of 1 year and are long enough to reach steady states (10,000 steps of time, as determined by preliminary simulations). Stable coexistence between plant types is checked using mutual invasibility (Chesson 2000), and the role of type (i.e., invader or resident) is switched for each tested pair of type. We first simulate the invasion of two plant types differing only by their preference for NH₄ (β) and then consider the case where P_2 is able to inhibit or to stimulate nitrification. In these latter cases, P_1 and P_2 plants only differ by their preference for NH₄ and NO₃ and their ability to control nitrification. Except for i, the parameters used in the main text of this study come from the Woodmansee et al. article (1978)

Results and Discussion

Because our results are qualitatively similar for both ecosystems, we show only the results for the Pawnee site in the main text. Results for the Lamto site are given in the appendixes B-D, available online.

Ecosystem Properties

Plant biomass and primary productivity are maximized for a particular value of plant NH₄ preference ($\beta_{opt} \approx 0.65$; fig. 2). Overall, variations in β lead to 10-fold variations in plant biomass and productivity (fig. 2A). This particular preference (β_{opt}) also leads to a minimization of the total mineral N losses (fig. 2B). We prove analytically in appendix A that this pattern always holds: whatever the value of β_{opt} when P* is maximized, plant productivity is maximized and mineral N losses are minimized. Moreover, using derivatives of equilibrium values with β , we show a positive relationship between the variation of P^* and the variation of D^* and between the variation of P^* and the variation of plant productivity as functions of β (preference for NH₄⁺). Conversely, we show a negative relationship between the variations of plant biomass/productivity and the variation of total mineral N losses as functions of β .

Concerning the variations of mineral N compartments as functions of β (see fig. 2B), the higher the preference for NH₄⁺, the lower the size of the NH₄⁺ pool. Symmetrically, the higher the preference for NO₃, the lower the size of the NO₂ pool. However, in the Pawnee site, there are two exceptions to this general pattern: (i) when β (preference for NH₄) ranges between 0 and 0.15, the higher the preference for NH₄⁺, the higher the availability of NH₄⁺, and (ii) when β ranges between 0 and 0.40, the higher the preference for NO₃, the higher the availability of NO₃. We give in appendix A an analytical study of the variations of the model compartments as functions of β . We identify in this appendix the general conditions that lead to such a pattern.

We show analytically that in Pawnee, the increase of NH_4^+ pool with β can be due to three complementary

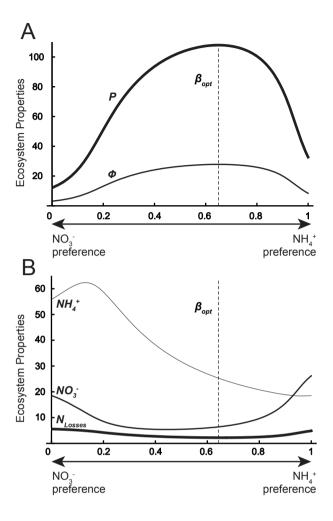


Figure 2: Variations in the size of ecosystem compartments (P[A]; NH_4^+ and $NO_3^-[B]$) and fluxes (primary productivity ϕ [A] and N_{losses} [B]) as a function of plant preference for NH₄ versus NO₃ (β) for the Pawnee site. The sizes of the ecosystem compartments are expressed in kilograms of nitrogen per hectare, and the fluxes are expressed in kilograms of nitrogen per hectare per year. As shown in the figure, $\beta_{\rm opt}$ corresponds to the maximal plant biomass and primary productivity and to the minimum of N losses.

conditions (see eq. [A15.2] in app. A): (i) the low preference of plant for NH_4^+ (since β ranges between 0 and 0.15, see fig. 2B), (ii) the low plant biomass associated with low values of β , and (iii) the high recycling efficiency of dead plant biomass into NH₄. These factors enable a significant amount of plant biomass that is transformed into NH₄ and that is not absorbed in high quantities by plants. So, NH₄ accumulates in the ecosystem as long as plants have their biomass increasing with β while being specialized in NO₃⁻ absorption (0 < β < 0.15). Concerning NO_3^- , when plants are specialized in NO_3^- absorption (β close to 0), NH₄ accumulates in the ecosystem, leading to high nitrification flux, and subsequently to a high NO₃ availability. Thus, despite the fact that plants prefer NO₃, the availability of this N form remains relatively high because of the high nitrification flux.

As a result of these trends, the total availability of mineral N (NH₄ plus NO₃) is larger for extreme values of β (either close to 0 or 1): the losses of mineral N are high while the productivity and plant biomass are low (fig. 2A, 2B). Term β_{opt} has an intermediate value that keeps both forms of mineral N to low levels, therefore preventing important N losses from the ecosystem and leading to a maximum plant biomass. Our results support conclusions from models considering only one limiting nutrient and reporting that the fewer nutrients are lost, the higher the plant biomass and productivity, as stated by Tilman's R* rule (Tilman 1985).

In our model, β_{opt} corresponds to a "generalist" strategy, so that being specialized in one of the two N mineral forms is not optimal when considering primary productivity. Since β affects in opposite ways the availability of both mineral N sources, being "specialized" in one of these N forms (β < 0.3 or β > 0.9; see fig. 2B) deprives the plant of the resource that is not used. This resource then becomes more vulnerable to losses since it is not taken up by plants. Consequently, the total losses of mineral N are high, resulting in a low plant biomass and productivity.

We show in appendix B that β_{opt} (optimal preference for NH₄) varies depending on N inputs, N losses, and nitrification rates. More precisely, we show that in Pawnee, the higher the rates of nitrification and NO₃ deposition, the higher the plant productivity for plants exhibiting a high preference for NO₃. Similarly, the higher the rates of NO₃ loss (for at least a twofold increase or more) and NH₄ deposition, the higher the plant productivity for plants exhibiting a high preference for NH₄⁺.

Coexistence and Invasions

Figure 3 displays the outcomes of a mutual invasion between two plants types P_1 and P_2 whose preferences for NH_4^+ versus NO_3^- are equal to β_1 and β_2 , respectively. When both plant types differ only by their preference for NH₄. (fig. 3A), while being not specialized in NH₄ absorption $(\beta \le 0.93)$, the one with the highest preference for NH₄ invades and excludes the other (see zones b and c): for example, if we consider an established population P_1 at $\beta_1 = 0.3$ being invaded by a type P_2 with $\beta_2 = 0.4$, this invader succeeds, excludes the resident population, and becomes the new resident population. Similarly, the plant type with the highest preference for NH₄ cannot be invaded by the other, that is, if a resident plant population P_1 exhibits a preference for NH₄ of $\beta_1 = 0.4$, an invader P_2 with a preference $\beta_2 = 0.3$ fails to invade the resident.

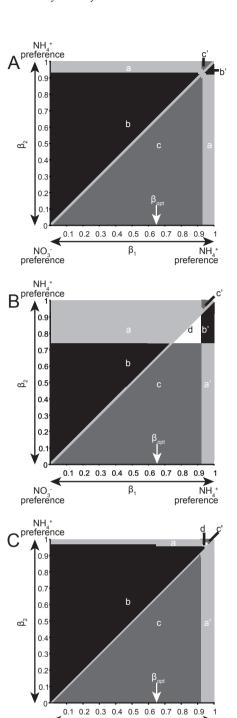


Figure 3: Patterns of invasion and coexistence between two plant types $(P_1 \text{ and } P_2)$ for the Pawnee site $(A, P_1, \text{ and } P_2 \text{ differ only in }$ their preference for NH₄⁺ vs. NO₃⁻; B, P₂ stimulates nitrification; C, P_2 inhibits nitrification). Terms β_1 and β_2 are the NH₄ preferences for P_1 and P_2 , respectively. Zones: light gray (a) = coexistence; black (b) = P_2 wins the competition; dark gray (c) = P_1 wins the competition; white (d) = no invasion and maintaining of residents.

 β_1

ŃΗ

preference

NO.

preference

If both types are specialized in NH₄⁺ absorption ($\beta \ge 0.93$ in fig. 3A), then the plant type with the lowest preference for NH₄⁺ invades the other, and cannot be invaded by the other (see zones b' and c' in fig. 3A). Coexistence occurs when a plant type specialized in NH₄⁺ absorption ($\beta \ge 0.93$) invades, or is invaded, by another one that is not specialized in NH₄⁺ ($\beta \le 0.93$ on fig. 3A; see zone a). When two plant types have exactly the same preference ($\beta_1 = \beta_2$), they obviously coexist. Note that the coexistence regions are entirely occupied by stable equilibria (numerically determined).

These results are in agreement with theoretical models showing that coexistence is possible for organisms that are dissimilar enough (Abrams 1996). Considering studies claiming that n competitors can not coexist in a stable equilibrium on fewer than n resources (Armstrong and McGehee 1980; Tilman 1985) our results suggest that N should not be considered as a single resource. Here, two plant types coexist on two mineral N forms. Some empirical studies have documented N partitioning between N mineral and organic forms (Miller and Bowman 2002; Weigelt et al. 2005; Fang et al. 2007) and also suggested that this partitioning could lead to coexistence.

When considering two plant types differing only by their preference for NH_4^+ (β), the plant having the closest preference to the value that enables coexistence ($\beta \approx 0.93$) always succeeds in invading the other. This suggests that from the population dynamics point of view, it can be expected that, in a community driven by competitive exclusion with local extinctions and repeated colonizations from a regional pool, plants' preferences will converge toward a value that enables coexistence ($\beta \approx 0.93$; see fig. 3A). This rationale should also be true on the evolutionary scale since figure 3A can be assimilated to a pairwiseinvasibility plot (see articles on the adaptive dynamics methodology: Metz et al. 1996; Geritz et al. 1997). Indeed, when considering a rare mutant in a large population, evolution should push plant preference toward the same value ($\beta \approx 0.93$). It is important to note that the value of β (preference for NH₄⁺) enabling coexistence is significantly different from β_{ont} , suggesting that the most productive plant is not necessarily the "best invader" and that evolution or species replacement does not optimize primary productivity (Boudsocq et al. 2011). Appendix D confirms that coexistence and complementarity between the resource use of the two plant types, lead to suboptimal values of primary productivity. Here, Tilman's R* rule is thus not respected. On one hand, a plant produces more biomass when relying on both NH₄ and NO₃ rather than on only one N form ($\beta_{\rm opt} \approx 0.65$). On the other hand, plants should benefit from relying more on NH₄⁺ than NO_3^- (fig. 3; $\beta > 0.93$) since they (i) should resist to invasion and (ii) should succeed in invading plants differing only by their preference. This result does not support other studies stating that plants exhibiting the highest biomasses and productivities are those that are the most competitive on the most abundant form of N and vice versa (McKane et al. 2002; Houlton et al. 2007). This result also shows that coexistence based on N partition does not necessarily increase primary productivity or plant biomass. Although this result would require further analyses for a full interpretation, this confirms that mechanisms of coexistence influence the impact of biodiversity on ecosystem functioning (Mouquet et al. 2002).

Nitrification Control by Plants and Its Effects on Invasion, Coexistence, and Exclusion

Though mineral N is available in two different forms, these two forms are not independent: mineralization first releases NH₄⁺ that is then converted in NO₃⁻ during nitrification, so the abundance of NO₃⁻ depends on both the abundance of NH₄ and the nitrification rate. This asymmetry between NH₄ and NO₃ should favor plants having a strong preference or being very competitive for the absorption of NH₄, since these plants shunt the N cycle and take up mineral N before it is nitrified into NO₃. However, atmospheric deposition of NO₃ constitutes an input into the system that cannot be shunted. Moreover, the competition between plants and microorganisms can also be stronger on NH₄ than NO₃ (Hodge et al. 2000), leading to a standing production of NO₃ by nitrifiers. This suggests that nitrification and its control by plants may play a central role in the outcome of competition for NH₄ and NO₃ between two species (Lata et al. 2004). To explore this, we consider two cases, the first one where one of the two plant types (P_2) was able to stimulate nitrification (fig. 3B), and the second one where one of the two plant types (P_2) was able to inhibit nitrification (fig. 3C). In both cases, we found new patterns of invasiveness and coexistence compared to the case where both types only differ by their preferences for NH_4^+ (fig. 3A).

Nitrification Stimulation. If P_2 is able to stimulate nitrification (fig. 3B), coexistence becomes possible over a greater range of combinations of plant preferences for NH₄⁺ (see zone a, larger in B than in A). The fact that coexistence is possible between two plant types having relatively low and similar preferences for NH₄⁺ when nitrification is stimulated (see fig. 3B, zone a), is due to the fact that stimulation of nitrification balances the availability of both N mineral forms. Moreover, for a relatively large range of combinations of plants' preference, stimulating nitrification prevents invasion and exclusion between two plant types (see fig. 3B, zone d). When the type stimulating nitrification is the resident, then the availability

of NH₄ is strongly decreased while NO₃ is greatly increased. An invader with a higher preference for NH₄⁺ than the resident can not succeed in its invasion, because its preference does not enable it to take advantage of the new balance between both N mineral forms. Conversely, if the resident does not affect nitrification while the invader stimulates it, the invasion fails because the resident keeps the advantage given its higher preference for NH₄. Here, the invader has too low a density to significantly affect the availability of N forms. This zone (d) can be considered as a "founder control" zone (Grime, 1998).

Hawkes and collaborators showed that several exotic annual grasses (Avena barbata and Bromus hordeaceous) were able to increase gross nitrification rates in the soil by a factor of 2 (Hawkes et al. 2005), and they suggested that such a control over nitrification gives a strong advantage in invading resident plants. This seems only partially in accordance with the results from our model, particularly when the considered plant has a higher preference for NO₃⁻ than the resident does (0.75 $< \beta_2 < \beta_1 < 0.93$; see fig. 3B, zone b'). However, it can be supposed that a more local modeling approach would give stronger results since nitrification stimulation by a plant should influence only the vicinity of its own roots and then drastically modify the local environment of plants sharing its own rhizosphere. Taken together, our results confirm that the control of nitrification is crucial for the outcome of competition for NH₄ and NO₃ between plant types.

Nitrification Inhibition. The ability of P_2 (fig. 3C) to inhibit nitrification (i) allows this type to exclude the other even when its preference for NH₄⁺ is very high (even for $0.93 < \beta_2 < 0.97$; see the size of the zones a and c' in fig. 3C; cf. zones a and c' in fig. 3A), and (ii) prevents exclusion from an invasive type with lower preference for NH₄⁺ when P_1 or P_2 have a high preference for NH₄ (see zone d in fig. 3C).

Also, when considering the zone d in figure 3C, if the nitrification inhibiting plant is the resident, the availability of NH₄ is increased while NO₃ is decreased, so that a plant type that exhibits a lower preference for NH₄⁺ than the resident cannot invade. Conversely, in the zone d in figure 3, if the resident type does not affect nitrification, then an invader with the ability to inhibit nitrification, while having a higher preference for NH₄⁺, cannot invade the resident. This is because the invader does not exploit NO₃⁻ even though the availability of this N form is relatively high.

Factors Influencing NO₃⁻ and NH₄⁺ Uptake by Plants

Some species are known to prefer NH₄ to NO₃ (Falkengren-Grerup and Lakkenborg-Kristensen 1994; Tavernier 2003; Zhao et al. 2009). However, it seems that sufficient empirical results are lacking for the derivation of general rules on plant preference as a function of the ecosystem they belong to, the physico-chemical conditions (e.g., atmospheric deposition, soil environmental conditions), their functional or taxonomic group, or their life stage. Are there some plant species that are able to use NH₄ and NO₃ indifferently (Hewins and Hyatt 2009)? Or, conversely, are there plants having a specific uptake capacity for each N form? It is likely that each plant species has a preference that depends on its genotype (rooting architecture, relationships with rhizospheric microorganisms, root uptake kinetics, transport and storage in plant cells, and ability to reduce NO₃⁻) and its environment (such as the temperature, pH, humidity, relative abundances of N forms). In their article, Houlton et al. (2007) showed that some tropical plant species can switch their dominant N source (NH₄⁺, NO₃⁻ or dissolved organic N) in response to changes in precipitation, by taking up the most available form of N. Similarly, Ashton et al. (2010) showed that some alpine plants can switch their main N source in function of the preference of their competitors for N. This strongly suggests that some plant species might be plastic in their ability to take up the two forms of N, but probably within a limited range determined by phylogenetic and energetic constraints. This plasticity should be taken into account in future models but we can consider for the moment that the β (preference for NH₄⁺) of our model describes the realized preference in given conditions. Future models should also include a detailed analysis of the effect of the uptake rate (u). Our model only allows for a constant intensity of uptake but plasticity in the total investment into N uptake could strengthen or attenuate the effects of β on ecosystems properties as well as on the patterns of invasion and coexistence.

Finally, our model does not take into account N immobilization by soil microorganisms (Harrison et al. 2007). The consequences of this competition for mineral N between microbes and plants have already been investigated in our previous model (Boudsocq et al. 2009), and we found that at equilibrium state, immobilization of N by microorganisms does not affect the availability of both N forms for plants. This general result still holds in this version of the model. However, we can also suppose that the microorganisms competitiveness and preference for NH₄ and NO₃ can have a significant impact on the availability of these N forms for plants and thus on the preference for NH₄ versus NO₃ of plants. Better knowledge and understanding of the competitive interactions between plants and microbes for mineral N uptake, and their preferences for mineral N forms, could significantly improve our assessment of the optimal preference of plants for NH₄ versus NO₃ in a given ecosystem and allow speculation on its potential evolution in a global change context.

Perspectives

Our results point out the large and overlooked potential consequences of plant preference for NH₄⁺ versus NO₃⁻ on terrestrial ecosystem functioning and plant community structure. To address this crucial point, we urgently need to experimentally assess the preference for NH₄⁺ versus NO₃⁻ of a large number of plant species and try to relate it with plant life stages and phylogeny or ecosystem properties. It would also enable assessing the plasticity of plants in their preference and their productivity, following what has been done on some species (Maire et al. 2009).

Determining the link between plants' preference for either form of N and their own productivity would be of particular interest for the current and future management of agrosystems. For example, our results suggest that plant production and mineral N losses depend on plant preference for NH₄ versus NO₃, on nitrification rates (and thus on plant ability to inhibit or stimulate nitrification) and on external inputs of the two mineral N forms (app. B). Our model could help increasing crop yield and decreasing N losses by choosing fertilization practices and selecting the relevant varieties of plants for their preference for NO₃ or NH₄ and their capacity to control nitrification (Subbarao et al. 2006; Maire et al. 2009). This is of paramount importance because modern agriculture urgently requires more sustainability and reducing N losses (Tilman et al. 2002), as mineral fertilization leads to the degradation of many terrestrial and aquatic ecosystems, and as the industrial production of mineral N is based on nonrenewable energy resources.

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